INSTITUTE OF ADVANCED STUDIES IN EDUCATION (DEEMED TO BE UNIVERSITY)

OF

GANDHI VIDYA MANDIR, SARDARSHAHR (CHURU) RAJASTHAN – 331403 Phone – 01564 – 220025, 223642, 223054 Web: www.iaseuniversity.org.in



SYLLABUS SCHEME OF EXAMINATION AND COURSE OF STUDY

FACULTY OF EDUCATION

B.Sc.B.Ed.(FOUR YEAR INTEGRATED COURSE)

Session: 2022-26



B.Sc. B.Ed.(FOUR YEAR COURSE)

The course of study shall extend over a period of four years as an integrated course in Science, Education, General Hindi,General English andEnvironmental Education leading to the composite degree of B.Sc.B.Ed.

A. Eligibility

Candidates who have passed Senior Secondary 10+2 examination or any other examination recognized board with at least 50% marks in the aggregate are eligible for admission to the course.

Or

The Institute will regulate admission tHRough selection on the basis of marks in the qualifying examination and/or in the entrance examination or any other selection process as per the policy of the state or NCTE and the University and in accordance with the state quota as decided.

Candidate will be eligible for admission to Physical Science (PCM) Group if she/he has qualified qualifying examination with Physics, Chemistry and Mathematics or Statistics and for admission to Biological Science (CBZ) Group if she/he has qualified qualifying examination with Physics, Chemistry and Biology or Biotechnology.

B. Reservation - Reservation of seats for SC/ST/OBC and Handicapped will be as per existing Rajasthan Govt. /Central Govt. /University rules.

C. Admission Procedure for BSCBED- Admission shall be made on merit on the basis of marks obtained in the qualifying Examination and/or in the entrance examination or any other selection process as per policy of the State Government and the University.

D. Duration and Working Days - **Duration** TheB.Sc.B.Ed.Programme shall be of duration of four Academic Years, which can be completed in a Maximum of five Years from the date of the admission to the programme.

Working Days

- There shall be at least two hundrade fifty (250) working days each year exclusive of the period of examination and admission.
- Institution shall work for a minimum of thirty-six hours(36) a week, during which physical presence in the institution of all the teachers and student teachers is necessary to ensure their availability for advice, guidance, dialogue and consultation as and when needed.
- The minimum attendance of student-teachers shall have to be 80% for all course work and practicum, and 90% for school internship.

E. Objectives of B.Sc.B.Ed.

- The objectives of this programme is to prepare teachers from upper primary to middle level (Classes VI-VIII) & Secondary Level (Classes IX-X) pre-service teacher education program are to enable the prospective teacher to –
- Understand nature of education and pedagogic processes tHRough enriched experiences. Contribute to fillup the gap between theory and practice by dovetailing both appropriately.
- Interactive processes wherein group reflection, critical thinking and meaning making will be encouraged.
- Understand various educational issues in the context of diverse socio-cultural & Multilingual Indian Society.
- Enable them to face the challenging of social, political and technological issues.
- Understand the nature, purpose, influencing factors and problems of secondary education in contemporary issues.
- Describe teaching learning process in the classroom and various factors that influence it.
- Understand various level learners, their needs, and interest and peculiar problems and motivate them for learning.
- Conduct Pedagogical content analysis in subject areas and use it for facilitating learning in the classroom.
- Develop and select tests, evaluate and keep records of student's progress-cognitive as well as non-cognitive.
- Adopt and develop enrichment learning& instructional material in subject areas.
- To develop problem solving ability through action research.
- Foster skills and attitude for involving the Community as an educational partner and use society resources in education.
- Become self-regulated learners, develop professional commitment and work as responsible professionals.
- Become aware about human values and gender, school and society.
- To prepare the students to further their education in graduate school and/or directly pursue productive professional careers in the private,state,federal,or educational sectors.
- Enable the students to demonstrate possession and application of higher-level learning skills in critical thinking and problem solving as applied to science issues.
- To enable the students to effectively apply the scientific method.
- Enable to work both on cross-disciplinary teams and function independently as specialists in a science or technical field.
- Demonstrate broad knowledge and technical skills in at least one area of science, and a basic understanding of science disciplines.
- Develop,apply,integrate and generate scientific knowledge in professional contexts to analyse challenges and to develop effective solutions
- Demonstrate understanding of the importance of science to the human endeavour.
- Work and learn both independently and collaboratively to encompass diverse abilities and perspectives, exercise personal, professional and social responsibility as a global citizen.

Programme Outcomes:

- Competence to teach effectively two school subjects at the Elementary & secondary levels.
- Ability to translate objectives of secondary education in terms of specific Programmes and activities in relation to the curriculum.
- Ability to understand children's needs, motives, growth pattern and the process of learning to stimulate learning and creative thinking to faster growth and development.
- Ability to use-
 - 1. Individualized instruction.
 - 2. Dynamic methods in large classes.
- Ability to examine pupil's progress and effectiveness of their own teaching tHRough the use of proper evaluation techniques.
- Equipment for diagnosing pupil progress and effectiveness of their own teachings tHRough the use of proper evaluation techniques.
- Readiness to spot talented and gifted children and capacity to meet their needs.
- Ability to organize various school programmes, activities for pupils.
- Developing guidance point of view in educational, personal and vocational matters.
- Ability to access the all-round development of pupils and to maintain a cumulative record.
- Developing certain practical skill such as:
 - 1. Black board work
 - 2. Preparing improvised apparatus
 - 3. Preparing teaching aids and ICT.
- Interest and competence in the development of the teaching profession and education. Readiness to participate in activities of professional organizations.

Programme Specific outcomes:

Students will be able to:

- Understand basic concepts and ideas of educational theory.
- Build understanding and perspective on the nature of the learner, diversity and learning.
- Comprehend the role of the systems of governance and structural functional provisions that support school education.
- Develop understanding about teaching, pedagogy, school management and community involvement.
- Critically examine key universal constructs in developmental psychology and educational psychology.
- Develop a knowledge setup of content, contexts and circumstances of society.
- Build skills and abilities of communication, self-expression and ICT.
- Develop the potential for perspective building located in the Indian socio-cultural context.
- Analyse the structure of knowledge as reflected in disciplinary streams and subjects
- Develop an understanding of the concept of assessment and its practices.

The syllabus of this course comprises of the following –

F. Course Design: The syllabus of this course comprises of the following -

(a) Perspectives in Education – Course

- Course Childhood,Growing Up & Learning.
- Course Educations in Contemporary India.
- Course Teaching and Learning.
- Course Gender, School and Society
- Course Knowledge and Curriculum
- Course Creating an Inclusive School

(b) Curriculum and Pedagogic Studies -

- Course Yoga for Holistic Health
- Course Pedagogy of School Subject (I&II)
- Course Assessment for Learning

(c) Engagement with the Field– the Self, the Child, Community and School -Task and assessment that run tHRough all the courses as indicated in the year wise distribution of the syllabus.

• School Internship

(d) **Specilization courses** will be offered in areas such as given below or an additional pedagogy course (In another course at the secondary level or the same course at the high secondary level):-

- Health & Physical Education
- Educational aspects of The Geeta
- Environmental Studies
- Chetna Vikas Moolya Shiksha
- (e) Core Courses
 - Physics
 - Chemistry
 - Botany
 - Zoology
 - Mathematics

(f) Compulsory Course

- GeneralEnglish
- GeneralHindi
- ICT in Education
- Environmental Studies

G. COURSES OF STUDY AND SCHEME OF EVALUATION FIRST YEAR

		EVALUATION			
Course Code	Title of the Course	External	Internal	Practical	Total
BSCBED110	Genral English(compulsory)*	70	30	-	100
BSCBED-151	Childhood, Growing up & Learning	70	30		100
BSCBED-152	Education in Contemporary India	70	30	-	100
BSCBED-153	Yoga for Holistic Health	35	15	-	50
BSCBED-154 Chetna Vikas Mulya Shiksha (Specialization) 35 15 -					50
BSCBED-155	Content: (PCB & PCM)			I	I
BSCBED-155 a I	1. Physics I	40	10	50	150
BSCBED-155 a II	2. Physics II	40	10		150
BSCBED-155 b I	3. Chemistry I	40	10	50	150
BSCBED-155b II	4. Chemistry II	4. Chemistry II 40 10 50			
BSCBED-155 c I	CBED-155 c I 5. Zoology I 40 10 50				150
BSCBED-155 c II	6. Zoology II	40	10		
BSCBED-155 d I	7. Botany I	40	10	50	150
BSCBED-155 d II	8. Botany II	40	10		
BSCBED-155 e I	9. Mathematics I	60	15		150
BSCBED-155 e II	10. Mathematics II	60	15		
ССА					25
Prayer, Yoga, Meditation & Festival etc					25
Total					800

*Marks of compulsory subject shall notbe added inthe total marks.

SECOND YEAR

Course Code	Title of the course		EVALUA	ATION	
		External	Internal	Practical	Total
BSCBED-210	Gen. Hindi(Compuls	ory)* 70	30	_	100
					100
BSCBED-251	Knowledge & Curric	ulum 70	30	-	100
BSCBED-252	Teaching & Learning	g 70	30	-	100
BSCBED-253	Health & Physical Ed	lucation 35	15		50
BSCBED-254 Content: (BCZ& PCM)					
BSCBED-254 a I	1. Physics I	40	10		
BSCBED-254 a II	2. Physics II	40	10	50	200
BSCBED-254 a III	3. Physics III	40	10		
BSCBED-254 b I	4. Chemistry I	40	10		
BSCBED-254 b II	5. Chemistry II	40	10	50	200
BSCBED-254 b III	6. Chemistry III	40	10	50	
BSCBED 250 c I	7. Zoology I	40	10		
BSCBED 250 c II	8. Zoology II	40	10	50	200
BSCBED-254 c III	9. Zoology III	40	10	50	
BSCBED-254 d I	10. Botany I	40	10		
BSCBED-254 d II	11. Botany II	40	10		200
BSCBED-254 d III	12. Botany III	40	10	50	
BSCBED-254 e I	13. Mathematics	I 60	7		
BSCBED-254 e II	14. Mathematics		7	1	200
BSCBED-254 e III	15. Mathematics	III 60	6	1	
ССА					25
Prayer, Yoga, Meditation & Festival etc					25
Total					900

*Marks of compulsory subject shall notbe added inthe total marks.

THIRD YEAR

Course code	Title of the course EVALUATION				
		External	Internal	Practical	Total
BSCBED-350	ICT in Education*	70	30		100
BSCBED-351	Gender. School & Society	35	15		50
BSCBED-352	Creating an Inclusive School	35	15		50
BSCBED-353	Educational aspects of Geeta	35	15		50
BSCBED-354	Pedagogy of School Subjec	t-I			
BSCBED-354-I	1. Mathematics	35	15		50
BSCBED-354-II	2. General Science	35	15		50
BSCBED-354-III	3. Chemistry	35	15		50
BSCBED-354-IV	4. Biology	35	15		50
BSCBED-354-V	5. Physics	35	15		50
BSCBED-355	BSCBED-355 Content: (BCZ& PCM)				
BSCBED-355 a I	1. Physics I	40	10	50	150
BSCBED-355 a II	2. Physics II	40	10	- 50 15	
BSCBED-355 b I	3. Chemistry I	40	10	- 50	150
BSCBED-355b II	4. Chemistry II	40	10	- 50	150
BSCBED-355 c I	5. Zoology I	40	10	50	150
BSCBED-355 c II	6. Zoology II	40	10		50 50 50 50
BSCBED-355 d I	7. Botany I	40	10	50	150
BSCBED-355 d II	8. Botany II	40	10		150
BSCBED-355 e I	9. Mathematics I	60	15		150
BSCBED-355e II	10. Mathematics II	60	15		150
	CCA				25
	Prayer, Yoga, Meditation	n & Festival	etc		25
Internship (4 Weeks)					50
Total					750

*Marks of compulsory subject shall notbe added inthe total marks.

Internship (4 weeks) included in total marks.

FOURTH YEAR

Course code	Title of the course	EVALUATION			
		External	Internal	Practical	Total
BSCBED-450	Environmental Studies*	70	30	-	100
BSCBED-451	Assessment for Learning	70	30	-	100
BSCBED-452	Agriculture	35	15	-	50
BSCBED-453	Pedagogy of school subject-II				
BSCBED-453 I	1. Mathematics	35	15	-	50
BSCBED-453 II	2. General Science	35	15	-	50
BSCBED-453 III	3. Chemistry	35	15	-	50
BSCBED-453 IV	4. Biology	35	15	-	50
BSCBED-453 V	5. Physics	35	15	-	50
BSCBED-454 Content: (BCZ& PCM)					
BSCBED-454 a I	1. Physics I	40	10	50	100
BSCBED- 454 b I	2. Chemistry I	40	10	50	100
BSCBED- 454 c I	3. Zoology I	40	10	50	100
BSCBED- 454 d I	4. Botany I	40	10	50	100
BSCBED- 454 e I	5. Mathematics I	80	20	-	100
	CCA				
Prayer, Yoga, Meditation & Festival etc					25
BSCBED- 455 Internship (16 weeks)				200	
Total				750	

^{*}Marks of compulsory subject shall notbe added inthe total marks. Internship (16weeks) included in total marks.

SCHOOL INTERNSHIP

As the title suggests in this component of the programme, the student- teachers are actually placed in a school for duration of four and sixteen weeks, in two time slots. Initially, they will be attached to particular School for four weeks as 'school attachment'.

This shorter period is to provide them adequate exposure to have a 'feel' of dealing with teaching- Learning.

A time gap after this school attachment will provide opportunity to student-teachers to share experiences, reflect, clarify several things with teacher educators and internalize them.

After about four weeks, they will go for 'school placement' of sixteen weeks. During this period, their role in the school is something like an 'apprentice' and its specific contours need to be worked out by course faculty.

They will be engaged in the school functioning in all its aspects.

MAIN OBJECTIVE

- Student teacher will be enabled to reflect on their practice, and learn to adapt and modify their visualisation/implementation towards betterment of student learning; involve in various school activities and processes in order to gain a 'feel' of the multiple roles of a teacher.
- Develop understanding of the 'school culture'; and learn to reflect upon, consolidate and share their school experiences; and to recognize one's own development as a teacher.
- To provide multiple components of field learning getting to know the school, observing children, observing teaching and learning in real classroom contexts & practicing teaching.
- To developing capacities to think with educational theories and applying concept in concrete teaching learning situations,managing classroom learning,evaluating learners and providing feedback,learning to work with colleagues,reflecting on one's own professional practice are drawn upon to provide appropriate learning experiences for the student teacher that is critical to the education of teachers.
- To undertake responsibility for planning and implementation of learning situations for specific units of studyin the context of their school.

INSTRUCTION

Student teacher with supervision of their teacher educator must actively engaged in all the school activities at least their internship,(from school management,PTM (MNC) to classroommanagement)

Phase-I internship of third year	= 50 marks
Phase-II internship of fourth year	= 200 marks
Total Marks of Internship	= 250 marks

Phase-I: Internship (4 weeks)

	Assessment is based on the following activities –	
Planning	Content Analysis and mode of transaction (Assignment in teaching subject)	5
	Creating and maintaining teaching learning material for the school (which can become valuable resource for the regular teachers of the school). a) TLM in any teaching subject	5+5=10
	Make lesson plan using 10- different methods in which 5 must involve,student could develop their own method (fusion based) with the help on teacher educator.	5
Planning & Execution	Identify a problem of action research and draft proposal on it.	5
Execution	*Innovative Micro Teaching (5 Skills) (Teaching Subjects at secondary to senior secondary level)	5
Execution	One-week, regular observation of regular teacher (at the beginning of practice teaching.)	5
	Delivery of Four lessons based on model of teaching. (After each lesson of practice teaching student teacher need to discuss with subject teacher on their pedagogy and new practices it must be seconded.)	5
Assessment & Evaluation	Draft a report based on: - continuously and comprehensively evaluating students' learning for feedback into curriculum and pedagogic practice.	5
Regularity and involvement in different school activities	Observation of day-to-day * school activities and report of an in- depth study of four activities.	5
	TOTAL MARKS	50

* Note: Any of the above activity may be replaced as per the need of the course

SUGGESTED SCHOOL ACTIVITIES: -

Select any one activity from each group given below: -

A Group: -

- Organization of cultural activities,
- Organization of literary activities
- Organization of games/sports.
- Framing of time table.
- Water resource management tHRough traditional methods.
- Prepare a report after interview of effective/good teachers.

B Group-

- Attending and organizing morning assembly
- Maintenance of classroom discipline
- Review of School Records
- Guidance and Counseling
- Gardening

CGroup -

- Organizing science fair, exhibition, science club, nature study
- Maintenance of School library
- Maintenance of School laboratories.
- Health and hygiene.
- Study on role of community for school improvement
- School mapping

D Group-

- Sensitization for environmental problems.
- Cleaning campaigning in school.
- School climate/Environment (any one aspect)
- Voluntary services.
- Mass awareness of social evils and taboos.
- Any other activity/s decided by the institute.

Internship (16 weeks)

TEACHING PRACTICE AND PRACTICAL WORK

Planning &	Preparation, administration & analysis of diagnostic test (s)	10
Execution	followed by remedial teaching any selected lesson.	
Execution	Execution of action research project	10
	Observation and preparation of report	10
	Working with community project of social welfare. (submission	10
	of report)	
	Exhibition of TLM in school prepared by student teacher	10
	Regular classroom teaching delivery of 70 lessons	5+5=10
Assessment &	Two Criticism Lesson in teaching subject	10+10=20
Evaluation	5 Lessons to be observe by teacher educator.	5+5=10
	Final Lesson (External evaluation)	50
Regularity &	Student teachers function in liaison with the regular teachers in	05
involvement in	the school in all day-to-day functioning along with teaching-	
different schoo	learning by mentor teachers	
activities	Participating in various 'out of classroom activities' in school	5
	Organizing events	
	Participation in any two co-curricular activity and preparation of	10
	report	
	Study (and preparation) of school calendar, time table, assessment	5
	schedule,library and laboratory.	

Portfolio, including detailing of teaching-learning plans, resources	10
used, assessment tools, student observations and records.	
observations of work done by the students during the internship	05
programme.(Seeking reactions of students,headmasters/	
principals/ cooperating teachers and supervisors)	
Preparation and maintenances of feedback diary	10
A journal by student teacher in which he/she records one's	10
experiences, observations, and reflections.	
TOTAL MARKS	200

Examination & Evaluation:

S. No.	Course	Distribution of Marks
1	Perspectives in Education	650
2	Curriculum and Pedagogic Studies (50+50)	100
3	Specialization	200
4	CCA & Prayer Yoga	200
5	Engagement with the Field – Internship	250
6	Core Courses	1800
7	Compulsory course	
	GRAND TOTAL	3200

Marks Distribution of yearwise			
Year Total Marks			
I Year		800	
II Year		900	
III Year		750	
IV Year		750	
	Grand Total	3200	

Marks Distribution	Year of Study			
Areas of Marks	I Year	II Year	III Year	IV Year
Content Subjects	450	600	450	300
Perspective in Edu.	250	200	100	100
Specialization	50	50	50	50
Pedagogy Subject			50	50
CCA	25	25	25	25
Prayer & Yoga	25	25	25	25
Internship			50	200
Total	800	900	750	750
	Grand Total			3200

Successful candidates will be awarded division on the basis of the aggregate marks of all the Core Courses, Pedagogy Courses and perspective Courses

	As per the following:	
I.	First Division	60% or more
II.	Second Division	50% or more (but less than
		60%)
III.	Third Division	40% or more in theory (but
		less than 50%)

Note-

The minimum pass marks in each year examination shall be 40% for each theory paper & practicum and 50% for internship in teaching separately.

Candidates can apply for Re-evaluation in any of the theory courses as per rules stipulated by the University for BSCBEDdegree. Changes in Statutes/ Ordinances/ Rules/ Regulations/ Syllabi and books may from time to time be made by amendment or remaking and a candidate shall, except in so far as the University determines otherwise, comply with any change that applies to years she/he has not completed at the time of change.

Notes

There shall be a University examination at the end of each year as per details of the scheme of examination.

A candidate will be permitted to appear in the TERM END EXAMINATION only if he/she has pursued a regular course of study and attended at least 80% of the classes for all the course work and practicum and 90% for school internship.

A candidate shall be admitted to the next higher class only if s/he passes his/her Part I/ Part II / Part III Examination as per rules mentioned herein after.

In order to qualify for B.Sc. B.Ed. degree a candidate should obtain a minimum of 40% marks in theory and practical and 50% for internship in teaching separately, wherever applicable in each subject in each year of the course and 40% marks in Pre-Internship in III Year and also in Internship in Teaching in the Fourth Year.

In Part I,there will be two core courses; compulsory course is General Hindi/ General English,ICT in Education and Environmental Education. In order to pass,a candidate must secure atleast 40% marks in each Compulsay course. However,the marks obtained in these papers will not be taken into account for awarding the division. In case a candidate fails in the compulsory course,she/he has to clear the same in consequent year.

The minimum pass marks in the supplementary examination shall be the same as prescribed for the main examination for internship may pass % shall be 50% separately for internship. i.e. (250)

Rules & Regulation: -

• Final examination is dividing in to two parts external and internal in all the courses, except the courses having practical exams. Internal assignment will be included two tests (7.5 marks each) and any two assignments in course having 100 marks and one assignment of 5 marks in the course having 50 marks along with two tests. In all the core course internal marks will be given by two tests.

- The candidate who has passed any year of B.Sc.B.Ed. Programme after taking supplementary examination will be awarded minimum pass marks in the concerned subject irrespective of marks actually obtained in the supplementary examination.
- A candidate who fails in 50% courses or less than 50% courses in first year of the programme will be eligible to take admission in second year and give the supplementary examination in the course(s) in which s/he fails along with subsequent examination and get two more chances for clearing this course. In second year, if a candidate is failed, she/he has only one more chance to qualify the exam. Non-appearance or absence from the examination of this paper will be counted as a chance.
- If a student is failed more than 50% courses in first year, he/she has to appear as exstudent in next subsequent year.
- A candidate who fails in one or two courses inany year of the programme will be eligible to take the supplementary examination in the course(s) in which s/he fails and get two more chances for clearing this course either along with the supplementary examination in next subsequent year. Non-appearance or absence from the examination of this paper will be counted as a chance.
- In case the candidate is not able to pass supplementary examination in two subsequent years,s/he can appear only as an ex-student in all courses again at the main examination of the next subsequent year. she/he will not be required to appear in practical(s) if she/he has already cleared the same and have to pay extra one third fee as ex-student. A candidate shall be deemed to be an ex-student if s/he completed a regular course of study at the Institute and fulfilled the required attendance and appeared in University examination but failed or did not take the examination.
- A candidate who fails in the practical/theory/field work of a course at the main examination shall be required to appear only in the corresponding practical/ theory of the supplementary examination.
- A candidate who appears for the supplementary examination may take provisional admission to the next higher class at his/her own risk. Such a candidate will,however,be allowed to appear in the University examination of the next higher-class course to his/her passing the supplementary examination,fulfilling the attendance requirement as a regular candidate and completion of courses of study as per scheme of examination. If a candidate getting supplementary does not take provisional admission to the next higher class by the notified last date of admission and passes the supplementary examination at a later stage,she/he will not be admitted to the next higher class. However,such a candidate may take admission to the next higher class in the next academic session.
- A candidate who fails in more than two courses (except General Hindi/General English/ Environmental studies and ICT in Education) in any year of the course shall be declared failed and will not be promoted to the next class. Such a candidate will be permitted to appear at the main examination of the subsequent year in all the courses only as an exstudent.
- Candidate who fails in more than two courses but passes in practical he/she will be required to appear again in all the courses (theory) except practical only as an ex-student.
- A candidate will be given a maximum of tHRee chances at the main examination and the corresponding supplementary examination in any year of the course. If he/she does not

pass the examination even thereafter,she/he will not be eligible for readmission to any year of the programme.

Pedagogical course – Learning to function as a teacher will be conducted in two phases. Phase I (Pre internship) of 50 marks and phase II (Internship in Teaching) of 200 marks will be conducted in B.Sc.B.Ed. Programme in Part III and IV respectively.

If a candidate fails in the Learning to function as a teacher (Pre-Intern - ship/Internship in Teaching) or is unable to complete Pre-Internship/Internship in teaching but passes in all other subjects he/she will be required to repeat the complete Pre-Internship/_Internship in Teaching' in the next academic session along with regular candidates. Division will be awarded to the successful candidates only after the Part IV examination and on the basis of cumulative total of marks obtained in all the four years of the course in all the subjects including Internship in Teaching but excluding the core subjects i.e.,General Hindi/General English,and Environmental studies.

FIRST YEAR

Course code	Title of the course	EVALUATION			
		External	Internal	Practical	Total
BSCBED110	Gen. English(compulsory)*	70	30	-	100
BSCBED-151	Childhood, Growing up & learning	70	30	-	100
BSCBED-152	Education in Contemporary India	70	30	-	100
BSCBED-153	Yoga for Holistic Health	35	15	-	50
BSCBED-154	Chetna Vikas Mulya Shiksha (Specialization)	35	15	-	50
BSCBED-155	Content: (BCZ& PCM)		1	1	
BSCBED-155 a I	1. Physics I	40	10	50	150
BSCBED-155 a II	2. Physics II	40	10		150
BSCBED-155 b I	3. Chemistry I	40	10	50	150
BSCBED-155b II	4. Chemistry II	40	10	50	150
BSCBED-155 c I	5. Zoology I	40	10	50	150
BSCBED-155 c II	6. Zoology II	40	10		
BSCBED-155 d I	7. Botany I	40	10	50	150
BSCBED-155 d II	8. Botany II	40	10		
BSCBED-155 e I	9. Mathematics I	60	15		150
BSCBED-155 e II	10. Mathematics II	60	15		
	ССА				25
Prayer, Yoga, Meditation & Festival etc					25
	Total				800

*Marks of compulsory subject shall not be added in the total marks.

B.Sc.B.Ed I YEAR					
COURSE CODE:	BSCBED-110	(COURSE 7	FYPE : CO	MPULOSRY
COURSE TITLE :	GENERAL ENGLISH				
MAX.MARKS:	100	MI	IN. PASS M	IARKS:	40
THEORY	70	MI	IN. PASS M	IARKS:	28
EXAMINATION					
CONTINUOUS	30	MI	IN. PASS M	IARKS:	12
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIVE YE	EAR			
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATI	ION		MONTI	HLY TEST
	03 HR			1	HR

Objectives :

- To Develop proficiency in English
- To Develop Listening abilities and skills.
- To Develop Reading abilities and skills.
- To Develop writing abilities and skills.
- To Develop basic skills in grammar, enriching their vocabulary.

Learning outcomes:

After completion of the course, student-teachers will be able to-

- Develop proficiency in English
- Understand the demands of audience, course, situation and purpose and the use of language for effective communication.
- Annalise language in context to gain an understanding of grammar, vocabulary, spelling, punctuation and speech.
- Learn authentic literary and non-literary texts
- Develop insight and appreciation.
- Sharpen writing abilities and skills.
- Make students read English poetry & prose with a view to enhancing their comprehension of the language and encourage them to develop reading habits. Develop basic skills in grammar, enriching their vocabulary and enabling them to write simple and correct English.

	• Subject verb agreement
	• Transforming an Affirmative sentence into Negative and Interrogative sentence
	Passive voice
	Direct /Indirect Speech
20)	• Auxiliaries
S (2	Conditional sentences
B	• PHRasal verbs
UNIT-1 NG HOURS (20)	• Joining sentences
	Transformation
5 ž	Simple sentence into compound and complex sentences
IH	Other varied transformations
U TEACHIN	• Degree (Positive/ comparative/superlative degree)
LE	Phonetic Transcription and word stress
	• common idioms and pHRases
	• Tenses
	• English sound & phonetic symbols
	• Preposition
•	

UNIT-2 TEACHING HOURS (20)	 Reading Comprehension and types of text Questions based on a passage from the prescribed text to test the candidate's comprehension and vocabular Questions based on an unseen passage to test the candidate's comprehension and vocabulary Synonyms and antonyms
UNIT-3 TEACHING HOURS (20)	 John Milton- On his blindness William Blake- To the evening star Alfred Lord Tennyson- break,break,break Charls Lamb- A bachelor's complaint against the behaviour of married people J.B Priestly-On getting off to sleep
UNIT-4 TEACHING HOURS(20)	 Summary writing Précise writing Letter/Application writing Report Writing
UNIT-5 TEACHING HOURS(20)	 Advertisement Notice Invitation E-mail
TEACHING AND LEARNING STRATEGI ES	 Lectures E-learning Videos Extension Lectures Content Review Self-Learning Group Discussions Field Visit Survey Documentaries Short Films Team Teaching * The teaching strategies are subject to change as per requirement of the students and their capabilities.

CONTINUO US &	Details of	Continuous and Comprehensive Assessme	nt (CCA) are as follows:
COMPREH ENSIVE	SR. NO.	CCA: COMPONENT	MAXIMUM MARKS
ASSESSME	1	Monthly Test	10X6 Test = 60
NT (CCA)	2	Presentation	10
	3	Group Discussion	10
	4	Debate	10
	5	Participation and Presentation in Seminar	
	_		
	6	Report Writing	10
	7	Viva Voce	10
	8	Attendance*	10
	9	Co-curricular Activity	10
	10	Team Teaching	10
		ATION (METHOD TO ASCERTAIN MA	· · · · · · · · · · · · · · · · · · ·
	CCA will	be reduced to 30 marks or 15 marks (as per c	ourse weightage).
	Formula:	Marks obtained/Total marksX30	
	For exam	ple: 60»160X30 =11.25	
		O-I: Provided that a candidate shall be granted	
	form of ex	kemption from CCA components, however, no	ot more than 3 in a
	respective	e course.	
	PROVIS	O-II: Provided further that this will be manda	tory for a candidate
	to appear	in the monthly test conducted in the respective	e course.
	*Attenda	nce in Lectures and Practical	
		Percentage Marks Allotted	
		75% to 80% 02	
		81% to 85% 04	
		86% to 90% 06	
		91% to 95% 08	
		Above 96% 10	
EXAMINA	Term-end	examinations are organized by the univers	ity in the prescribed format to
TION		e scholars to achieve success in contempora	
PATTERN	their goals	-	5 1
PERIODIC	1. A		
AL		owever, the Universitymay revise the syllab	us at any time during the
REVISION		nning year after giving a notice for a period	•
OF			
SYLLABUS			
SELECTED	• Abra	ams, M. H. (2005). A Glossary Of Literary Te	rms NewDelhi: Macmilan
READINGS		ams, M. H., & Harpham, G. G. (2018).	
		venth Ed.). New Delhi: Cengage LearningIndi	• •
		h, D. (2009). The Oxford Companion Engli	
		h, Ed.) New York, United States: OxfordUniv	-
		lick, C. (2015). The Oxford Dictionary Of	Literary rerms. United Kingdom:
		ord University Press.	
		, &Jonathan. (2010).English Literary:-A Ver	y ShortIntroduction. New Delhi:
		ord University Press.	
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Macmillan

B.Sc.B.Ed. 1 Year				
BSCBED-151	BSCBED-151 COURSE TYPE: CORE		CORE	
Childhood, Growing up and	Learn	ing		
100	MIN	. PASS MARKS:	40	
70	MIN	. PASS MARKS:	28	
30	MIN	. PASS MARKS:	12	
80 PERCENT IN RESPECT	IVE Y	EAR		
TERM END EXAMINATIO	TERM END EXAMINATION MONTHLY TEST			
03 HR 01 HR		۲. Element of the second se		
Objectives:				
• Understand the meaning, nature and characteristics of growth & development.				
	BSCBED-151 Childhood, Growing up and 1 100 70 30 80 PERCENT IN RESPECT TERM END EXAMINATIO 03 HR	BSCBED-151 Childhood, Growing up and Learni 100 MIN 70 MIN 30 MIN 30 MIN 80 PERCENT IN RESPECTIVE Y TERM END EXAMINATION 03 HR ning, nature and characteristics of grow	BSCBED-151 COURSE TYPE: Childhood, Growing up and Learning IOO 100 MIN. PASS MARKS: 70 MIN. PASS MARKS: 30 MIN. PASS MARKS: 80 PERCENT IN RESPECTIVE YEAR TERM END EXAMINATION MONTHLY 03 HR 01 HE	

- Understand the principles and theories of development.
- Acquire knowledge on different stages of development with its multi-dimensional aspects.
- Analyze the characteristics of development of children from different psycho-sociological and cultural background.
- Examine the impact of urbanization, virtualization, social and economic change on the development of adolescents.
- Critically analyze the impact of different agencies on child development.

• Understand the significance of gender, caste, social class and their influence on children.

Learning Outcomes: After completion of the course, student-teachers will be able to :-

- Understand children of different ages by interacting & observing them in diverse social, economic and cultural context rather than tHRough an exclusive focus on child development.
- Study of childhood, child development and adolescence.
- Understand learning as divergent process.
- Make aware about the importance of healthy liking and preventing diseases.
- Introduce psychological traits of learners.
- Become health aware & sensitize children about mental and physical health.
- Understand the role of the family and the school in the child's development.

	Childhood and Child Development
	• Education Psychology- its meaning, scope and implications for teacher in classroom
	situation. Various psychologists and their contributions in education.
	• Importance of psychology for teacher and learner.
54	• Childhood: Meaning, Concept and Characteristics.
S	• Concept of Growth and Development,
UR	• Dimensions and Principles of Development.
UNIT-1 ING HOURS (24)	• Factors affecting Development (especially in the context of family and school) and their
	relationship with learning.
5Ž	 Childhood and child Development implication in teaching and learning
U TEACHIN	• Role of Heredity and Environment.
V	Assignment:
EL	• Organize creative activities for children of diverse socio-cultural background with aim
_	to learn to communicate and relate with them.
	• Create child based new activity to learn to listen to children with attention and empathy.
	• Creating and applying appropriate tools to measure the growth and development of
	children in school.

	Adolescent Development & Personality Factors
	Adolescent: Meaning, Concept and Characteristics.
	• Adolescent Development implication for teachers, teaching and learning.
<u>-</u>	• Cognitive, Physical, Social, Emotional and Moral Development patterns and
(54	characteristics of Adolescent's Learner.
RS	• Personality: Meaning, concept, types of personality and affecting factors. Theories and
UNIT-2 TEACHING HOURS (24)	Assessment, Adjustment and its Mechanism, Maladjustment.
HC HC	• Individual Differences: Meaning, Types and Factors Affecting Individual differences.
N S	• Piaget, Kohlberg and Vygotsky: constructs and critical perspectives.
	Addressing the Talented, Creative, especially abled-Learners. Assignment:
C	 Identifying the adolescence's problem, a sound of discussion with parents will be held.
ΈA	Suggest the remedial majors to being about positive change.
F	• Seminar or workshop for student teacher to observe, interact with and study
	adolescents of different ages in and outside the school, in diverge social-economic,
	cultural, linguistic and regional contexts.
	• Organizing seminars for the psycho-development of children outside the school.
	Learning & Learning Difficulties
	• Learning: Meaning, Concept, Types and Nature of Learning, factors influencing
	Learning, theories of Learning, Learning implication for teachers.
7	 Learning process. Cognition and Emotions, Motivation and Learning
5	• Factors contributing to learning– Personal and Environmental.
R	• Domains of learning, Cognitive, Affective and Psychomotor.
<u>0</u>	• Understanding diverse learners: Backward, Mentally Retarded, Gifted, Creative,
H	disadvantaged-deprived, CWSN, Children with learning disabilities.
UNIT-3 TEACHING HOURS (24)	• Motivation: meaning, concept and its Implications for Learning and Achievement.
H	Assignment:
AC	• The pupil teacher will prepare at least two lessons based on computer assisted instruction and study its effectiveness.
LE	• Observe two learners, in natural setting to study and write a report on their domain of
-	learning, the report present class in the presence of teacher education.
	• To study children of diverse, economic and societal & cultural context for
	understanding learners' thinking and learning and prepare a report.
	Mental Health & Hygiene
	• Mental Health & Hygiene: Meaning, Concept and its affecting factors
	• Development of good Mental Health.
52)	• Personal and Environmental Hygiene.
S	• Physical & Mental Hygiene for teachers and learners.
CB	• Adjustment: Concept and ways of Adjustment and its Mechanism, Maladjustment. Role
Г-4 НО	of Teacher in the Adjustment.
UNIT-4 ING HOU	Assignment:
D Z	• Examine the personal, domestic and physical hygiene of school-student.
UNIT-4 TEACHING HOURS (22)	• Organize prayer meetings, yoga and meditation camps by the trainees for the healthy
EA	personality development of the students.
E	• Conduct a study of psychological variables such as stress, mental health, conflict,
	anxiety, depression, self-esteem among school students.Organizing seminar/symposia in the community with students for awareness about
	cleanliness and health.

	Development and Implications in Education
	• Self-concept, Social Skills of Learner.
	• Intelligence: Concept, Theories and its Measurement.
53	• Multiple-Intelligence: Meaning, definitions, concept and theories.
UNIT-5 TEACHING HOURS (22)	• Multi-Dimensional Intelligence, Critical perspective of the construct of Intelligence and its implication.
I-5 HO	• Creativity: Meaning, definitions, concept and theories.
	Assignment:
	• Collecting and analyzing comparative data on the development of students at different levels of the school.
AC	• Find out the IQ of the students in different subjects by the trainees.
LE	• Organize thematic activities to ascertain the thematic originality/creativity of the
	students.
	• Apply any two psychological tests on two students and on the basis of the conclusion,
	make a comprehensive profile at least ten students for each test.
TEACH	1. Lectures
ING	2. E-learning
AND	3. Videos
LEARN	4. Extension Lectures
ING	5. Content Review
STRAT	6. Self-Learning
EGIES	7. Group Discussions
	8. Field Visit
	9. Survey
	10. Documentaries
	11. Short Films
	12. Team Teaching
	* The teaching strategies are subject to change as per requirement of the students
	and their capabilities.

CONTI NUOUS	Details o	f Continuous and Comprehensive Assessment	(CCA) are as follows:		
& COMP	SR. NO.	CCA: COMPONENT	MAXIMUM MARKS		
REHEN	1	Monthly Test	10X6 Test = 60		
SIVE	2	Presentation	10/10/10/10/10/10/10/10/10/10/10/10/10/1		
ASSESS	3	Group Discussion	10		
MENT	4	Debate	10		
(CCA)	5	Participation and Presentation in Seminar	10		
~ ,	6	Report Writing	10		
	7	Viva Voce	10		
	8	Attendance*	10		
	9	Co-curricular Activity	10		
	10	Team Teaching	10		
		NATION (METHOD TO ASCERTAIN MAR			
		ll be reduced to 30 marks or 15 marks (as per cou	· · · · · · · · · · · · · · · · · · ·		
		: Marks obtained/Total marksX30	inse werginage).		
		mple: $60 \div 160 X 30 = 11.25$			
		SO-I: Provided that a candidate shall be granted	a relaxation in the		
		exemption from CCA components, however, not			
		ve course.			
	-	SO-II: Provided further that this will be mandate	ry for a candidate		
		r in the monthly test conducted in the respective	•		
		lance in Lectures and Practical			
		Percentage Marks Allotted			
	75% to 80% 02				
		81% to 85% 04			
		86% to 90% 06			
		91% to 95% 08			
		Above 96% 10			
EXAMINA TION		l examinations are organized by the university in ars to achieve success in contemporary competition	-		
PATTERN			2		
PERIODI CAL	1. Annua				
REVISIO		ver, the Universitymay revise the syllabus at	any time during the running		
N OF	year a	fter giving a notice for a period one month.			
SYLLABU S					
SELECTE	• Shriv	astav, D.N. and Verma, P. (2007). Child Developme	ent and Child Psychology. Vinod		
D READING	Pusta	k Mandir: Agra.			
KEADING S		k, M. (2002). Child Development and Family Rel	ationship. Research Publication:		
	Jaipu		nent Arva Book Denot: New		
	• Mangal, S.K. and Mangal,S. (2005). Child Development. Arya Book Depot: New Delhi.Sharma, R.K. and Saharma,				
	 H.S.(2006). Psychological Foundation of Child Development. Radha Prakashan Mandir: 				
	Agra.				
	Singh Desce	n, D.P. and Talang Prakash (2002). Psycho-Social ba	sis of learning and development.		
		arch Publication: Jaipur. vastva. D.N. Verma, P. (2010). Modern Experimenta	l Psychology and Teshing SHRi		
		d Pustak Mandir: Agra.	a rependicipy and resining, orner		
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		cation: Agra.			
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B.Sc.B.Ed. 1 Year						
COURSE CODE:	BSCBED-152		COURSE TYPE: CORE			
COURSE TITLE:	COURSE TITLE: Education in Contemporary India					
MAX.MARKS:	100 MIN. 1		MIN. PASS MARKS:	40		
THEORY	70		MIN. PASS MARKS:	28		
EXAMINATION						
CONTINUOUS	30		MIN. PASS MARKS:	12		
COMPREHENSIVE						
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RESPECTIVE	YEAR	R			
ELIGIBILITY				_		
EXAMINATION	TERM END EXAMINATION		MONTHLY TEST	ſ		
DURATION	03 HR		01 HR			
 Objectives: To learn the concepts of social Change and social transformation in relation to education. Acquire knowledge of the concept of Educational Administration and Management. Understand the role of center, state and local agencies in managing education. Develop an understanding of the main issues related to Indian educational system. To develop understanding about the social realities of Indian society and its impact on education. To identify the contemporary issues in education and its educational implications. To know the different values ensHRined in the constitution of India and its impact on education. Learning Outcomes: After completion of the course, students will be able to: Contextualize contemporary India with development of education. Understand the Classroom as a social context. Appreciate diverse perspectives of social, cultural, economic and political issues. Critically analyses human and child rights. Equips the teacher with proactive perspective and sense of agency. Engage with concepts which are drawn from a diverse set of disciplines. Learn about policy debates overtime the implementation of policies and actual shaping of school education. 						
 Diversity in contemporary Indian Society & Education Indigenous Meaning, Concept & Characteristics. Education: Meaning, Concept and Nature. Social and Cultural Diversity: Meaning, Concept and their impact on Education. Social, Cultural, Economic and Political and Technological perspective of Society and Education. The role of Educational Institution for creating new social orders. Classroom as a social context. Assignment: Prepare a report on role of Educational Institution for creating new social order in your area. Prepare a report on Parents and teacher experiences about their and others' culture and diversity. 						

	Educational Management
UNIT-2 TEACHING HOURS (24)	 Concept and functions of Education Management, Education Management in Rajasthan, School as a Unit of Decentralized planning, Educational Management Information System (EMIS), Institutional Planning, School Mapping, Block Resource Centre (BRC), School Management Committee (SMC), District Information System for Education (DISE), Samagra Shiksha Abhiyan (SMSA). National integration and National security. Assignment: Prepare a report after studying School Management Committee (SMC) in nearby school. Make a presentation on Education Management Information system.
	Constitutional Provision as the guideline to Education
UNIT-3 TEACHING HOURS (24)	 Constitutional provisions related to Education. Constitutional provisions on Human and Child Right, Values & Education. Role of NCPCR (National Commission on Protection of Child Right). Constitution direction for Issues & Problems in Education. Assignment: Conduct an awareness programme on Child Rights with students, parents and community. Organise a seminar on Constitutional provisions related to Education.
ଲ	Emerging Indian Concerns and their Educational Implications
UNIT-4 TEACHING HOURS (22)	 Meaning, Concept and Impact on Education of Liberalization, Globalization, Govermintization, Privatization. Stratification of Education: Concept and Process. Nationalist critique of Colonial Education and Experiments with a Alternatives Education for Marginalized group like Women, Dalits and Tribal people on personal family and Community Hygiene. Assignment: Oragnize a group disscussion on Education for Marginalized group. Oragnize a debate on Govermintization v/s Privatization of education.
	Organization of Educational Setup
6	Organization of Educational Setup at Primary and Secondary
UNIT-5 TEACHINGHOURS (22)	 Functions of RIE, SIERT, SBER, CTE, DIET. Educational Initiative: Balika Shiksha Foundation, Kasturba Gandhi Balika Vidalaya, Rajasthan Text Book Board, Bharat Scouts and Guides. RastHRiya Military School, Sainik School, Model School, E-Mitra, E-Governance, Rajshiksha, Edu sat, Gyandarshan, Gyanvani. Right to Education, SSA, Policies for UEE, Nayee Taleem. Assignment: Examine Policy & Constitutional provision on equality and Right to Education.
L	• Train students in any five Handicrafts on the basis of the Nayee Taleem (such as
	paper Meshi, Mithi Kutti, Handloom etc) and other related to cottage industries
TEACHING AND	and organize an exhibition on handicraft material. 1. Lectures
LEARNING	2. E-learning
STRATEGIES	3. Videos
	4. Extension Lectures
	5. Content Review

		Y		
		F-Learning		
		 Group Discussions Field Visit 		
	9. Sur	•		
		cumentaries		
		rt Films		
		m Teaching		
	* The teaching strategies are subject to change as per requirement of the students and their capabilities.			
CONTINUOUS &				
CONTINUOUS & COMPREHENSIVE	Details of	Continuous and Comprehensive Assessment	(CCA) are as ionows:	
ASSESSMENT				
	SR. NO.	CCA: COMPONENT	MAXIMUM	
(CCA)			MARKS	
	1	Monthly Test	10X6 Test = 60	
	2	Presentation	10	
	3	Group Discussion	10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	
	EXPLAN	ATION (METHOD TO ASCERTAIN MAR	KS FOR CCA):	
	CCA will	be reduced to 30 marks or 15 marks (as per cou	rse weightage).	
	Formula:	Formula: Marks obtained/Total marksX30		
	For exam	For example: 60÷160X30 =11.25		
	PROVISO-I: Provided that a candidate shall be granted a relaxation in the			
	form of exemption from CCA components, however, not more than 3 in a			
	respective course.			
	PROVISO-II: Provided further that this will be mandatory for a candidate to			
	* *	appear in the monthly test conducted in the respective course.		
	*Attendance in Lectures and Practical			
	Percentage Marks Allotted			
	75% to 80% 02			
	81% to 85% 04			
	86% to 90% 06			
	91% to 95% 08			
		Above 96% 10		
EXAMINATION	Term-end e	examinations are organized by the university in	the prescribed format to	
PATTERN	enable the scholars to achieve success in contemporary competitions and to			
	achieve the	*		
PERIODICAL	1. An			
REVISION OF	2. Ho	wever, the Universitymay revise the syllabus	at any time during the	
SYLLABUS	run	ning year after giving a notice for a period o	ne month.	
SELECTED		न.के. (2009). शिक्षा के दार्शनिक व सामाजिक आध		
READINGS	हाऊसः र			
	• रूहेला,	एस.पी. (2009). शिक्षा के दार्शनिक व समाज	शास्त्रीय आधार. अग्रवाल	
		न्सः आगरा.		
		रयूप्रसाद. (2009). शिक्षा के दार्शनिक, ऐतिहासिक	व समाजशास्त्रीय आधार.	

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पब्लिकेशन्स : आगरा.
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आर. लाल बुक डिपो.
• त्यागी, ओंकार सिंह (2007). उदीयमान भारतीय समाज और शिक्षा. जयपुर : अरिहंत
प्रकाशन.
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• Sexena, N.R. (2001). Principles of Education. International Publishing House:
Meerut (UP)

	B.Sc.B.Ed. 1 st Year			
COURSE CODE:	BSCBED-153		COURSE TYPE:	
		CORE		
COURSE TITLE:	Yoga for Holistic Health			
MAX.MARKS:	50	MIN. PASS MARKS:	20	
THEORY	35	MIN. PASS MARKS:	14	
EXAMINATION				
CONTINUOUS	15	MIN. PASS MARKS:	6	
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE Y	EAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION	MONTHLY T	EST	
DURATION	03 HR	01 HR		

Objectives:

- To develop the understanding of the Fundamentals of holistic health.
- To develop the understanding of Metaphysical Bases of Yoga.
- To develop the understanding of Philosophy of Yoga Education.
- To know Different Yoga Systems and Characteristics of Yoga Practitioner.
- To develop Awareness of COVID -19.
- To develop the understanding of selected instruments of yoga and their effective uses.

Learning Outcomes:

After completion of the course, students will be able to:

- Understanding about the concept, scope & need of Holistic Health.
- Apply their Knowledge about yoga and fundamentals of holistic health in daily life.
- To compare the Education of holistic health between Indian & Western Context.
- Conceptualize Metaphysical Bases of Yoga.
- Integrate yoga & meditation in their daily life.
- Examine Different Yoga Systems and Characteristics of Yoga Practitioner.
- Able to perfectly select instrument of yoga and their effective uses.

UNIT-1 TEACHING HOURS (12)	 Fundamentals of holistic health Concept of Holistic Health. Need & scope of education for Holistic Health. The Indian context and Western context of Education for Holistic Health. Dimensions of development of Holistic Health. Assignment:
TE HO	 Comparative study of the concept of holistic health in Indian context & Western context. Oraganize an Institutional programme for development of Holistic Health tHRough yoga.
UNIT-2 TEACHING HOURS (12)	 Metaphysical Bases of Yoga Concept of Purush and Prakriti as Basic Component of Cosmic Reality. Concept of Antahkaran Man, Budhdhi, Chitt, Ahankar. Assignment: Oraganize workshop on Metaphysical Bases of Yoga. Oraganize awareness programme on Manviya Prakriti in school.

	Philosophy of Yoga Education
	• The meaning and definition of yoga.
rh 🙃	• Needs, importance and scope of yoga education.
(17 17 17	• Yoga as a way of healthy and integrated living.
E H S	• Yoga as a way of socio-moral upliftment of man.
UNIT-3 TEACHING HOURS (12)	• Yoga as a way of spiritual Enlightment, Atmanubhuti, Pratyakshanubhuti.
LE HO	Assignment:
	• Practice any five yogasana in school with involvement of parents for socio-moral and
	physical health upliftment of student.
	• Orientation programme of school level students for integration of yoga in their daily life.
	Different Yoga Systems and Characteristics of Yoga Practitioner
	• Ashtang yoga of Pantajali (Yam,Niyam,Asan,Pranayama-Pratyahar-Dharna-Dhyana,
UNIT-4 FEACHING HOURS (11)	Samadhi).
LT-L	 Gyan-Bhakti-Karma yoga of Bhagvadgita. Integral yang of Aurhinda and madem asheel of yoga
	 Integral yoga of Aurbindo and modern school of yoga. Characteristics of a super mustification.
IO IE∀	• Characteristics of a yoga practitioner.
	Assignment:Discuss characteristics of a yogi purush with student and make routine for student.
	 Oraganize a training camp for school student on Ashtang yoga of Pantajali
	Instrument of Yoga Different Assures and Preneurom to promote a sound Physical and montal health
	 Different Asanas and Pranayam to promote a sound Physical and mental health. Dhene and its therementies are less
1)	 Dhyan and its therapeutic value. Selected many of Dhyang
	Selected ways of Dhyana.Awarness of COVID - 19: Mechanism during Quarantine and Home Isolation.
UNIT-5 FEACHING HOURS (11)	• Awarness of COVID - 19. Mechanism during Quarantine and Home Isolation. Assignment:
EA U	• Practice dhyana for five minutes to students before teaching and learning anlyse its
ΕĦ	impact on their performance.
	• Conduct a Mechanism strategy of Yoga for Quarantine and Isolate people and prepare a
	Report of Yoga Aasana special for Isolate people.
TEACHI	1. Lectures
NG AND	2. E-learning
LEARNI	3. Videos
NG STRATE	4. Extension Lectures
GIES	5. Content Review
GILD	6. Self-Learning
	7. Group Discussions
	8. Field Visit
	9. Survey 10. Documentaries
	11. Short Films
	12. Team Teaching
	* The teaching strategies are subject to change as per requirement of the students and
	their capabilities.

CONTI NUOUS	Details of	Continuous and Compre	hensive Assessment (CCA) are as follows:	
&COMP REHEN	SR. NO.	CCA: COMPONENT MAXIMUM MARKS			
SIVE	1	Monthly Test10X6 Test = 60			
ASSESS	2	Presentation		10	
MENT	3	Group Discussion 10			
(CCA)	CA)3104Debate105Participation and Presentation in Seminar10				
	6	Report Writing 10			
	7	Viva Voce		10	
	8	Attendance*		10	
	9	Co-curricular Activity		10	
	10	Team Teaching		10	
	EXPLAN	ATION (METHOD TO	ASCERTAIN MAR	KS FOR CCA):	
		be reduced to 30 marks of			
	Formula:	Marks obtained/Total mar	·ksX30		
		ple: $60 \div 160 \times 30 = 11.25$			
		O-I: Provided that a candi			
	of exempt	ion from CCA componen	ts, however, not more	than 3 in a respective	
	course.				
		O-II: Provided further that		•	
	+ +	the monthly test conducte	A	rse.	
	*Attenda	nce in Lectures and Prac	ctical		
		Doroontago	Marks Allotted		
		Percentage 75% to 80%	02		
		81% to 85%	02 04		
		86% to 90%	06		
		91% to 95%	08		
		Above 96%	10		
EXAMI	Term-end e	examinations are organize	ed by the university in	the prescribed format to	o enable
NATION		s to achieve success in con			
PATTE					
RN		•			
PERIOD ICAL	1. Ani		• 41 11 1	· · · · ·	
REVISI		wever, the Universityma		-	unning
ON OF	yea	r after giving a notice fo	r a period one month	•	
SYLLAB					
US					
SELECT		ो (2016). योगप्रतिभा. आगराः राध			
ED	• किशनलाल	(2016).योगासन एवं ध्यान क्रिया	एं. दिल्लीः राजा पॉकेट बुक्स.		
READIN GS	• रंजन, राज	कुमार (2016).योग शिक्षा. आगराः	राखीप्रकाशन.		
Go	• दुबे एवं शग	र्ना (2016). योगशिक्षा. आगराः राध	ा प्रकाशन.		
	~	Shankar (2002). Classical		hes to Yoga.New Delhi:	Pratibha
	Prakasha	. ,			
	• Goel, A.	(2007). Yoga education:	Philosophy and pract	ice. New Delhi: Deep at	nd Deep
	Publicati				
	• Kumar, I	K. (2012). Yoga Education	n. New Delhi: Shipra I	Publication.	

• Paraddi, Kasuma Mallapa and Ganesh, Shankar (2006). Ashtanga Yoga in relation to Holistic Health. New Delhi: Satyam Publication.
• Singhal. J.C. (2009).Yoga Perecived, Practised. Saga of India.New Delhi: AbhishekPrakashan.
• Swami, A. P. & Mukerjee (2008).Yoga lessons for developing spiritual consciousness. New Delhi: Cosmo Publication.
• Yadav, S. & Kumar, S. (2014). To Study the Effectiveness of Yoga Education on
Holistic Development of Teacher Trainees. International Journal of Innovative Research & Development. 3(1).

B.Sc.B.Ed. I Year					
COURSE CODE: BSCBED-154 COURSE TYPE		PE: CORE			
COURSE T	ITLE:	Chetna Vikas Moolya Shi	etna Vikas Moolya Shiksha		
MAX.MAR	KS:	50	MIN. PA	ASS MARKS:	20
THEORY		35	MIN. PA	ASS MARKS:	14
EXAMINAT	ΓΙΟΝ				
CONTINUC	DUS	15	MIN. PA	ASS MARKS:	6
COMPREH	ENSIVE				
ASSESSME	NT (CCA)				
ATTENDAM	NCE	80 PERCENT IN RESPE	CTIVE Y	EAR	
ELIGIBILI	ТҮ				
EXAMINAT	ΓΙΟΝ	TERM END EXAMINA	TION	MONTHLY	TEST
DURATION	N	03 HR		01 HR	1
Objectives:					
To de	evelop human	moral values, peace and har	mony of p	oupil teachers.	
• To de	evlop Human	relationship, Spirituality and	Social de	velopment of pupil-	teachers.
• Deve	lop an ability	in the pupil teachers to distin	nguish bet	ween good and bad.	
	•	ity in human beings.			
	•				
Learning ou	tcome:After	completion of the course, stu	ident-teac	hers will be able -	
• Life a	and values per	rtaining to Individual, Family	y and Soci	ety.	
• Harm	ony in Enviro	onment, Nature and Existenc	e.	•	
	•	n Mindset, Humane Condu		Jniversal Order in	Nature and
Exist	•	·····, ·····			
• Evalu	ating and und	derstanding the difference be	tween An	imal and Human cor	isciousness.
	-	able happiness and prosper			
	in beings?			· ··· ··· ··· ···	
	•	evelopment of a holistic per	spective a	among students tow	ards life and
		as towards happiness and p			
*		ty and the rest of existence.			
		ing of Human Being and H	uman Va	lues and Humane C	Conduct
5 C		nding the harmony in self and			
E Z E		nding Human being and Hun	•		
UNIT-1 EACHIN OURS (J		Character and Morality.	iun ooui.		
	 Relationships and Justice (Nyaya) within Family & Society. 				
UNIT-1 TEACHIN HOURS (1	Relationships and Justice (Nyaya) within Family & Society. Palationship of Human being with Nature (Environment)				
 Relationship of Human being with Nature (Environment). Human Behavior, Humane Instincts and Personality. 					
Significance of Value Education					
5 3			ad to Uum	van Valua	
• Problems faced by the Humanity related to Human Value.					
• Social and Family disintegration.					
UNIT-2 TEACHING HOURS 12)	• Stress and conflict in Individuals.				
H	• Significar	nce of Value Educations for s	solving the	e human problems.	
	The Journa of the	ing the Velue			
UNIT-3 TEACHING HOURS (12)		ing the Values	ation C	Cuidance D	
UNIT-3 ACHING URS (12)		alues: Trust, Respect, Affe	ction, Car	e, Guidance, Rever	ence, Glory,
L C C	Gratitude		0		
EA OU		al Values: Complementari		-	ommitment,
ĒĤ		y, Self-Restraint, Obedience,	-	•	C
		Values Self (Jeevan) Valu		iness, Peace, Satis	faction, and
	Continuo	us Happiness Material Value	s.		

	Understanding Exis	stence and Co-existence and t	he Interrelationship	s in		
UNIT-4 TEACHING HOURS (11)	<u>Nature</u>	Existence & Co-existence /Nature.				
VIT CH JRS	• Understanding the	e Order, Co-existence and I	nterrelationships, M	utual		
LEA IOC	•	clicity (Avartansheelata) in Nature.				
	 Problems faced by a Ecological and Env 	ironmental imbalances.				
	Harmony and Value					
11)		nships: Meaning, Need, Importance	and Purpose.			
S	• Human relation: typ					
UR		se of fulfillment in relationships an ues in family-relations, Importance				
T-5 HO		Goal (Living with resolution and a				
IN D	participation in soci		i prosperioj	,		
		ng up of Families and Relationships		-		
ACI		of Marriage (vivaah sambandh) an king up of marriages.	nd a study of reasons	s and		
UNIT-5 TEACHING HOURS (11)		of concept of a family and a family	y-based village govern	nance		
_	order.					
TUTORI ALS	One tutorial class on	ce a week (12)				
TEACHI	1. Lectures					
NG AND	2. E-learning					
LEARNI	3. Videos					
NG	4. Extension Lec					
STRATE GIES	 Content Review Self-Learning 					
GILD	7. Group Discussions					
	8. Field Visit					
	9. Survey 10. Documentaries					
	11. Short Films					
	12. Team Teaching					
	* The teaching strategies are subject to change as per requirement of the					
CONTIN	students and their capabilities.Details of Continuous and Comprehensive Assessment (CCA) are as follows:					
UOUS &		s and Comptenensive Assessmen	(CCA) are as follow	0.		
COMPR	SR. CCA: CO	MPONENT	MAXIMUM			
EHENSI	NO.		MARKS			
VE ASSESS	1 Monthly T	est	10X6 Test =			
MENT	2 Presentatio	n and a state of the state of t	<u> </u>			
(CCA)	2Presentation103Group Discussion10					
	4 Debate		10			
		on and Presentation in Seminar	10			
	6 Report Wr	iting	10			
	7 Viva Voce 8 Attendance	o*	<u> </u>			
		e* lar Activity	10			
			10			
	10 Team Teac	ening	10	10Team Teaching10		

	EXPLANATION (METHOD TO ASCERTAIN MARKS FOR		
	CCA):		
	CCA will be reduced to 30 marks or 15 marks (as per course		
	weightage).		
	Formula: Marks obtained/Total marksX30		
	For example: 60÷160X30 =11.25		
	PROVISO-I: Provided that a candidate shall be granted a relaxation		
	in the form of exemption from CCA components, however, not more		
	than 3 in a respective course.		
	PROVISO-II: Provided further that this will be mandatory for a		
	candidate to appear in the monthly test conducted in the respective		
	course.		
	*Attendance in Lectures and Practical		
	Percentage Marks Allotted		
	75% to 80% 02		
	81% to 85% 04		
	86% to 90% 06		
	91% to 95% 08		
	Above 96% 10		
EXAMIN	Term-end examinations are organized by the university in the prescribed format to		
ATION	enable the scholars to achieve success in contemporary competitions and to achieve		
PATTER	their goals.		
N	then goals.		
PERIODI	1. ANNUAL		
CAL	2. However, the Universitymay revise the syllabus at any time during the		
REVISIO	running year after giving a notice for a period one month.		
N OF	running year arter giving a notice for a period one month.		
SYLLAB			
US			
SELECT	•Gaur, R.R. & Sangal, R. Bagaria, G.P. (2009). A Foundation Course in Human		
ED	Values and Professional Ethics. Excel Books: New Delhi.		
READIN	•Nagraj, A. (1998). Jeevan Vidya Ek Parichay. Divya Path Sansthan:		
GS	Amarkantak.		
	•Dhar, P.L. and R.R. Gaur (1990). Science and Humanism. Common Wealth Publisher.		
	• Tripathi, A.N (2003). Human Values. New Age International Publishers.		

B.Sc. B.Ed. I Year					
COURSE CODE:	BSCBED-155 a I			COURS	E TYPE : CORE
COURSE TITLE :	Paper-I: Relativity, Mec	hanics, Os	cillatio	ns and V	Vaves
MAX.MARKS:	50	MIN.PAS	S MAF	RKS	20
THEORY	40	MIN.PAS	S MAF	RKS	16
EXAMINATION:					
CONTINUOUS AND	10	MIN.PAS	S MAF	RKS	4
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPE	CTIVE YE	AR		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATION MONTHLY TEST				
DURATION	03 HR 01 HR			1 HR	
OBJECTIVE:					
• The aim and objective of this course on Relativity, Mechanics, Oscillations and Waves to give					
the students fundamental ideas on special theory of relativity, conservation laws, dynamics of rigid					
hodies escillations and waves. This source will get as a strong heateround if he/she shapes to					

the students fundamental ideas on special theory of relativity, onservation laws, dynamics of rigid bodies, oscillations and waves. This course will act as a strong background if he/she chooses to pursue higher studies in physics.

Learning Outcome: After completion of the course, student-teachers will be able to-

- Understand Newtonian mechanics, special theory of relativity.
- Apply Newton's laws to explain natural physical phenomena.
- Discuss on the simple harmonic motion and its equation.
- Differentiate between damped oscillator and driven oscillator.
- Identify the coupled oscillator and some electrically coupled oscillators.
- Explain acoustics and waves in media.

	• Relativity: Reference systems, inertial and non-inertial frames, law of
	•
	motion, Galilean transformation, Galilean invariance and conservation
	laws, propagation of light, Michelson-Morley experiment. Postulates
5 2	of the special theory of relativity, Lorentz transformations , length
UNIT-1 TEACHING HOURS(12)	contraction, time dilation, addition of relativistic velocities, Doppler
	effect, variation of mass with velocity, mass-energy equivalence,
O O O O O O O O O O O O O O O O O O O	particle with a zero-rest mass, <i>simple application to a decay process</i> .
D F H	Four dimensional momentum vector, Covariance of equation of physics.
	• Mechanics:Motion under a central force, Conservation of angular
4	momentum, Kepler's laws, Gravitational law and field, Potential due
UNIT-2 TEACHIGN HOURS(14)	to a spherical body, Gauss and Poisson equations for gravitational
S S S S S S S S S S S S S S S S S S S	potential, gravitational self-energy. Rigid body motion, Rotational
5	
0 -7	motion, Degree of Freedom, Moment of inertia and their products,
	theorem of parallel and perpendicular axes, principal moments and
	axes, Euler's theorem, equation of motion for rotation, Molecular
	rotations; Di and tri-atomic molecules, Precessional motion, top,
CE	gyroscope. System of particles, Centre of mass, angular momentum,
A	equation of motion, single stage and multistage rocket, Conservation
E	
	theorems for energy, momentum and angular momentum, Elastic
	and inelastic collisions.

UNIT-3 TEACHING HOURS(12)	• Oscillations: Potential well and periodic oscillations, cases of Harmonic oscillations, different equations and its solutions, kinetic and potential energy, Simple Harmonic Oscillations in spring and mass system, simple and compound pendulum, Torsional pendulum, Bifilar oscillations, Helmholtz resonator, LC circuits, Vibration of a magnet, Oscillation of two masses connected by a spring, superposition of two simple harmonic motions of same frequency along the same line, Interference, Superposition of two mutually perpendicular simple harmonic vibrations of same frequency, Lissajou's figures, cases of different frequency.				
UNIT-4 TEACHIN G HOURS (10)	• Coupled oscillations:Damped harmonic oscillators, Power dissipation, Quality factor, Driven harmonic oscillator & Resonance, transient and steady state, power absorption, resonance in system with many degrees of freedom, Two coupled oscillators, Normal modes, N-				
S-LING S-LING TEACHING AND LEARNING STRATEGIES	 Maily degrees of freedom, Two coupled oscillators, Frequency response Waves in media: Speed of transverse waves on a uniform string, speed of longitudinal waves in a fluid, energy density and energy transmission in waves, waves over liquid surfaces, gravity waves and ripples, group velocity and phase velocity, superposition of waves, linear homogeneous equations and the superposition principle, nonlinear superposition and consequences. Acoustics: Noise and Music, the human ear and its responses, limits of human audibility, intensity and loudness, Bel and decibel. Production and detection of ultrasonic and infrasonic waves and applications 1. Lecture method 2. Problem Solving method 3. Graphical method 4. Seminar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion 				
	 9. Self-Learning/e-Learning 10. Workshops/Experiments. * The teaching and learning strategies may be change as per requirement of the students and their capabilities. 				
CONTINUOUS COMPREHENSIVE	Details o follows:	f Continuous and Comprehensive A	ssessment (CCA) are as		
ASSESSMENT	S. No.	CCA- Components	Max. Marks Allocation		
(CCA)	1.	Monthly test	20*3 Test=60		
	2.	Quizzes and Assignments	10		
	3.	Viva-voce	10		
	4.	Seminar/Symposia	10		
	5. Report writing 10				
	6.	1			
-	7.	Review of literature	10		
	7. 8.	Review of literature Creativity/Innovation	10 10		
	7.	Review of literature	10		

	Total 160 marks equivalent reduced to CCA original marks 30.		
EXAMINATION	Term-end examinations are organized by the university in the prescribed		
PATTERN	format to enable the scholars to achieve success in contemporary		
	competitions and to achieve their goals.		
PERIODICAL	1. ANNUAL		
REVISE OF	2. HOWEVER, THE UNIVERSITY may revise the syllabus at any		
SYLLABUS	time during the running year after giving a notice for a period one		
	month.		
SELECTED	• Resnick, R. (1971). Introduction to special relativity.(Wiley India Pvt.		
READINGS	Ltd., 2005) Charles Kittel, Berkeley Physics Course vol1, Mechanics		
	(Mc Graw-Hill,1965)		
	• Stephani, H. (2004). Relativity: An introduction to special and general relativity. Cambridge university press.		
	• Feynman, R. P., Leighton, R. B., & Sands, M. (1965). The Feynman lectures on physics; vol. 1. American Journal of Physics, 33(9), 750-752.		
	• Feynman, R. (2018). Feynman lectures on gravitation. CRC Press.		
	• Bajaj, N. K. (1988). The physics of waves and oscillations. Tata		
	McGraw-Hill Education.		
	• Gambir, R.S. (2006) Mechanics (CBS Publishers, New Delhi.		
	• Ghosh, R. K. (1975). The mathematics of waves and vibrations, (MC		
	Milan, 1975)		
	• Frank, S. Crawford J. R., Berkely (1968). Physics course: vol.3, waves		
	(Mc Graw-Hill book company, 1968)		

B.Sc. B.Ed. I Year					
COURSE CODE:	BSCBED-155 a II		COU	RSE TYPE : CORE	
COURSE TITLE :	Paper-II: Mathematical	Backgro	und, Properti	es of Matter and	
	Electromagnetic Waves	C	· -		
MAX.MARKS:	50	MIN.PA	SS MARKS	20	
THEORY	40	MIN.PA	SS MARKS	16	
EXAMINATION:					
CONTINUOUS	10	MIN.PA	SS MARKS	4	
&COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPE	CTIVE Y	EAR		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINA	TION	MON	THLY TEST	
DURATION	03 HR			01 HR	
Objective:					
• This course will enab	ble the students to develop	the kno	wledge and u	inderstanding of the	
	to solve problems in a nun				
Classical mechanics, E	Electromagnetic theory, Stati	istical Phy	vsics, Thermal	Physics etc.	
Learning outcomes: Aft	er completion of the course,	student-te	eachers will be	e able to:-	
• Know the mathematica	al background, properties of	matter an	d electromagn	netic waves.	
• Get familiar with conc	epts of scalars and vectors.		-		
• Apply the principles of	of Kinematics of moving fl	uids and	Electromagne	tic induction in real	
situations.	C		U		
• Learn the Electromagn	etic wave and Electromagne	etic field a	and Energy de	nsity.	
	• Scalars and Vectors: I			*	
a C	of scalar field and its ge				
	of a vector field, line, su				
UNIT-1 TEACHING HOURS(15)	field, Gauss divergenc				
	theorem. Functions of tw	wo and tH	IRee variables	s, Partial derivatives,	
H H	geometrical interpretation	on of tota	l differential	of a function of two	
	and tHRee variables, higher order derivatives and their applications.				
	• Elasticity: Hook's Law, Small deformations, Young's modulus,				
	Bulk modulus and Mo	odulus of	rigidity for	an isotropic solid,	
UNIT-2 IEACHIN(HOURS(8)	Poisson's ratio, relatio			-	
O O O O			0	· · · · · ·	
H H	bending of beams and cantilever, Torsion of a cylinder, Bending				
	moments and Shearing forces.				
3) <u>(</u>	• Kinematics: Kinematic		0 1		
	Euler's equation, Berno	-	-		
UNIT-3 TEACHING HOURS(13)	and turbulent flow,				
	Reynold's number, Stokes law and applications, surface tension				
H H	and surface energy, molecular interpretation of surface tension,				
	 Pressure on a curved liquid surface, Wetting. Electromagnetic induction, Faraday's law (its integral and 				
5 <u>C</u>	•			· •	
UNIT-4 TEACHING HOURS(12)		Lenz's law, Mutual and Self-inductance,			
UNIT-4 EACHIN OURS(1	Transformers, Energy		-		
self-inductance by Rayleigh's method, Maxwell's displacen current, Maxwell's equations , Electromagnetic field and Energy			_		
HCH	density.	iauons, i	Electromagnet	ic field and Energy	

STRATEGIES	 Electromagnetic Waves: Plane electromagnetic wave in vacuum, Wave equation for E and B of linearly, Circularly and elliptically polarized electromagnetic waves, Poynting vector, Reflection and Refraction at a plane boundary of dielectrics, Polarization by Reflection and total internal Reflection, Faraday effect, Wave in conducting medium, Reflection and Refraction by the ionosphere. 1. Lecture method 2. Problem Solving method 3. Graphical method 4. Seminar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion 8. Videos/Animation 9. Self-Learning/e-Learning 10. Workshops/Experiments. 		
		aching and learning strategies may b	
CONTINUOUS		nent of the students and their capabi of Continuous and Comprehensive A	
COMPREHENSIVE	as follow	-	assessment (CCA) at
ASSESSMENT	S. No.	CCA- Components	Max. Marks
(CCA)			Allocation
	1.	Monthly test	20*3 Test=60
	2.	Quizzes and Assignments	10
	3.	Viva-voce	10
	4.	Seminar/Symposia	10
	5. 6.	Report writing Workshop	10 10
	6. 7.	Review of literature	10
	8.	Creativity/Innovation	10
	9.	Experimental Skill	10
	10.	Co-curricular activity	10
	11.	Attendance	10
		marks equivalent reduced to CCA ori	
EXAMINATION		d examinations are organized by	2
PATTERN	-	ed format to enable the scholars t	
DEDIODICAT	-	orary competitions and to achieve their	goals.
PERIODICAL REVISE OF	1. ANN 2. How	UAL ever, the Unviersitymay revise the	syllabus at any time
SYLLABUS		ig the running year after giving a n	•
	mont		since for a period one
SELECTED		erman, G. E. (1970). Electricity and ma	agnetism. Elma.
READINGS	• Bleaney, B. I., Bleaney, B. I., & Bleaney, B. (2013). Electricity		
	 and Magnetism, Volume 2(Vol. 2). Oxford University Press. Resnick, R., Walker, J., & Halliday, D. (1988). Fundamentals of physics (Vol. 1). Hoboken: John Wiley. Blatt, F. J., (1989). Principles of physics. Boston, London: Allyn 		
		Bacon. iths, D. J. (2005). Introduction to ele	ectrodynamics. Prentice

Hall of India.Sarwate, V. V., (1993). Electromagnetic fields and waves. Bohem
 press. Ghosh, S. N., (2002). Electromagnetic theory and wave propagation. CRC Press.
 Kakani and Hemrajani (2008). Electromagnetism theory and Problems, New Delhi: CBS Publishers.

PHYSICS PRACTICALS-I Duration: 4 HR MAX.MARKS: 50 Min. Marks: 20 Total number of experiments to be performed by the students during the session should be 16, selecting any eight from each section. In examination two experiments are to be performed taking at least one from each section. The distribution of marks in the practical examination will be as follows: (i) Two experiments 30 Marks For each experiment, distribution of marks will be as follows: 2 Figure: 2 Formula/Theory: Observation: 7 Calculation and Result: 3 Precautions: 1 (ii) Viva voce 10 (iii) Records 10 Total 50 Marks LIST OF EXPERIMENTS Section A Study of conservation of momentum in two dimensional oscillations. • Study of a compound pendulum. •

- Study of damping of a bar pendulum under various mechanics.
- Study of oscillations of a mass under different combinations of springs.
- Study of bending of a cantilever or a beam
- Study of torsion of a wire (static and dynamic methods.)
- Study of flow of liquids tHRough capillaries.
- Determination of surface tension of a liquid by different methods.
- Study of viscosity of a fluid by different methods.
- To find M.I. of an irregular body by inertia table.
- To study the random decay and determine the decay constant by statistical method.
- Determine Y by Searle's apparatus.

Section B

- Study of magnetic field due to a current
- Conversion of galvanometer into an ammeter
- Conversion of galvanometer into voltmeter.
- To determine the internal resistance of primary cell using potentiometer.
- Measurement of low resistance by Carey-Foster Bridge.
- Measurement of inductance using impedance at different frequencies.
- Measurement of capacitance using impedance at different frequencies.
- I-V characteristics of a P-N junction diode
- Determine unknown resistor by post office box.
- To determine Poisson's ratio of rubber tube.
- To determine the frequency of A.C mains.

	B.Sc. B.Ed. I Year				
COURSE CODE:	BSCBED-155 b I			COURSE TYPE :	
				CORE	
COURSE TITLE :	Paper I-Inorganic Chemistry				
MAX.MARKS:	50	MIN.	PASS N	MARKS	20
THEORY	40	MIN.	PASS N	MARKS	16
EXAMINATION:					
CONTINUOUS	10	MIN.	PASS I	MARKS	4
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIVE SESS	SION			
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATION		Ν	IONTHLY	TEST
DURATION	03 HR 01 HR		R		
OBJECTIVE :					

• The aim and objective of this course is to teach the fundamental concepts of chemistry of Atomic Structure and Bonding, main group elements, Ionic solids, Acids and Bases, transition metals and their magnetic behavior and Environmental chemistry to the students.

Course Learning outcomes : After completion of the course, student-teachers will be able to:-

- Understanding of principles of Atomic structure and Chemical Bonding.
- Get familiar with chemistry of main group elements.
- Apply the principles of Acids and Bases in real life situation.
- Determine the crystal structure of NaCl, KCl and CaCl (Laue's method and powder method).
- Define the characteristics of different type of bond.
- Define the Basic properties of main group elements&Transition Elements.

A	Atomic	Structure	and Bonding	
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	Atomic Structure and Donung				
	• Atomic Structure: Fundamental particles atomic models and their				
	limitations.Idea of de-Broglie matter waves, Heisenberg uncertainty				
	principle, atomic orbitals, Schrodinger wave equation, significance of φ				
	and φ^2 , quantum numbers, abfbau principale, Pauli's exclusion principle.				
(15)	Hund's rule, (n+1) rule, Electronic configuration of elements. Atomic				
S	mass. Molecular mass. Equivalent mass. Mole concept. Symbols. Ions.				
5	Radicals. Type of formulas empirical formula, molecular formula, shapes				
UNIT-1 TEACHIGN HOURS	of s, p, d, orbitals.				
	• Structure and Bonding: Ionic bond, Covalent Bond, coordinate bond,				
55	general properties of ionec and covalent bond Valence bond theory and				
IH	itslimitations, directional characteristics of covalent bond, various types of				
C	hybridization and shapes of simple inorganic molecules and ions, valence				
EA	shall electron pair repulsion (VSEPR) theory to $NH_3, H_3O^+, SF_4, CIF_3, ICl_2^-$				
E	and H_2O,MO theory,homonuclear and heteronuclear (CO and NO)				
	diatomic molecules, multicenter bonding in electron deficient				
	molecules, bond strength and bond energy, polarisation, fajion's rule				
	,concept of resonance				
	· •				

UNIT-2 TEACHIGN HOURS (15)	 <u>Chemistry of main group elements</u> <u>S-Block Elements</u>: Comparative study, diagonal relationship, salient features of hydrides, solvation and complexation tendencies including their function in bio-systems, an introduction to alkyls and aryls. <u>P-Block Elements</u>: Comparative study (including diagonal relationship) ofgroups 13-17 elements, compounds like hydrides, oxides, oxy acids and halides of group 13-16,hydrides of boron-diborane and higher boranes,borazine,borohydrides,fullerenes,carbides,fluorocarbons,silicates (structural principle), tetrasulphur tetra nitride, basic properties of halogens,interhalogens and polyhalides. <u>Chemistry of Noble Gasses</u>: Chemical properties of the noble gases, chemistry of xenon,structure and bonding in xenon compounds.
	Ionic compounds: Bonding and Structure
UNIT-3 TEACHING HOURS (14)	 Ionic solids: solid state –classification of solidsIonic structures,radius ratio effect and coordination number,limitation of radius ratio rule,lattice defects,semiconductors,lattice energyand Born-Haber cycle,solvation energy and solubility of ionic solids,polarizing power and polarisability of ions,Fajan's rule,Metallic bond-free electron,valence bond and band theories. Structure of Ionic solids: Definition of space lattice, unit cell, Laws of crystallography (i) Law of constancy of interfacial angles(ii) Law of rationality of indices (iii) Law of symmetry,Symmetry elements in crystals. X-ray diffraction by crystals,Derivation of Bragg equation,Determination of crystal structure of NaCl,KCl and CsCl (Laue's method and powder method). Weak Interactions- Hydrogen bonding,van der Waals forces.
	Acids and Bases
UNIT-4 TEACHING HOURS(15)	 Theories of Acids and Bases: Arrhenius, Bronsted- Lowry, the Lux-Flood, solvent system and Lewis concept of acids and bases. Concept of Hard and Soft Acids and Bases (HSAB): Classification ofacids and bases as hard and soft,Pearson's HSAB concept,acid-base strength and hardness and softness,Symbiosis,theoretical basis of hardness and softness,electro negativity and hardness and softness. Solvent Systems: Physical properties of a solvent, types of solvents and their general characteristics reactions in non-aqueous solvents with reference to liquid NH3 and liquid SO2

	Chamistry	of Transition Floments		
		of Transition Elements		
		y of Elements of First Transition S		
16	properties	of d block elements, properties of the e	lements of the first	
	transition series ,their binary compounds (hydrides, carbides and oxides)			
RS I	and complexes with respect to relative stability of their oxidation states,			
00	coordination number and geometry.			
NG HOI	• Chemistry of Elements of Second and Third Transition Series: General			
	characteristics , comparative treatment of Zr/Hf , Nb/Ta, Mo/W in respect			
	of ionic radii, oxidation states , magnetic behaviour,			
H H	• Elimentary M.O. approach for metallic bond and bond order. Conducters,			
	-	semiconductors and super conducters		
UNIT-5 TEACHING HOURS (16)		ental chemistry :- air , water, and soil	pollution affects of	
		of ozone layer, green house effect and glob		
			ai walling , sualigy	
		of environmental pollution		
TEACHING AND		ure method		
LEARNING		lem Solving method		
STRATEGIES	-	bhical method		
		inar/Symposia		
		ew of literature		
	-	ort writing		
		ip Discussion		
		eos/Animation		
		Learning/e-Learning		
	10. Workshops/Experiments.			
	* The teach	ing and learning strategies may be change	e as per requirement	
	of the students and their capabilities.			
CONTINUOUS &	Details o	f Continuous and Comprehensive Assessn	nent (CCA) are as	
COMPREHENSIVE		follows:		
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)		L L	Allocation	
	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		arks equivalent reduced to CCA original man		
EXAMINATION		xaminations are organized by the university		
PATTERN	format to	enable the scholars to achieve succes	ss in contemporary	
	competition	s and to achieve their goals.		
PERIODICAL	1. ANN	NUAL		
REVISE OF	2. How	vever, the Universitymay revise the syllabu	s at any time	
	during the running year after giving a notice for a period one			
SYLLABUS	duri	ng the running year after giving a notice f	or a period one	
SYLLABUS	duri mon	e e e	or a period one	
SYLLABUS		e e e	or a period one	
SYLLABUS		e e e	or a period one	

SELECTED	•Lee, J.D. Concise Inorganic Chemistry, ELBS.				
READINGS	•Douglas,B.E. and Mc Daniel,D.H.,Concepts & Models of Inorganic Chemistry				
	• Day, M.C. and Selbin, J. Theoretical Inorganic Chemistry, ACS Publications.				
	• SHRiver and Atkins Inorganic Chemistry, W. H. Freeman and Company				
	• James Huheey, Inorganic chemistry: Principles of Structure and Reactivity,				
	Pearson Education India.				
	•B.N. Figgis, J. E Huheey, P.W. Atkins Inorganic Chemistry, Pearson				
	Education.				
	• Duward SHRiver, Inorganic Chemistry, W. H. Freeman.				
	• Gary Wulfsberg, Inorganic Chemistry, University Science Books.				
	• A. R. West, Solid State Chemistry and its Applications, Wiley.				

	B.Sc. B.Ed. I Ye	ar			
COURSE CODE:	BSCBED- 155 b II	~~	COURSE TY	PE : CORE	
COURSE TITLE :	Paper II: Physical Chemistry	7			
MAX.MARKS:	50			20	
THEORY	40	MIN.PAS	SS MARKS	16	
EXAMINATION:					
CONTINUOUS	10	MIN.PAS	SS MARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80PERCENT IN RESPECTIV	E SESSIC	DN		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATIO	N	MONTHLY		
DURATION	03 HR		01 HF	R	
OBJECTIVES:					
	e of this course is to teach				
	viorof gases and liquid and co	lloidal Sta	ates, Solutions,	and Chemical	
Equilibrium to the stude					
C	nes -After completion of the cou				
	lear chemistry, behavior of gases	-		ates.	
	olutions, Dilute Solutions and C	-	-		
	s of concept of Equilibrium in re				
-	d crystal, solid and liquid, Class	ification, s	structure of nemal	tic	
And cholestric phases.					
	<u>Nuclear Chemistry</u>			C 1	
• IS	• Radioactive decay–decay law, disintegration constant, half-life and average				
UNIT-1 TEACHING HOURS (15)	life alpha and beta disintegration reactions ,group displacement law,				
	nuclear reactions fission, fusion		•	* *	
D EA U	radioactivity, nuclear power, ca				
E H	types of radiations, nuclear ch	emistry fo	or peace, Nuclear	r chemistry in	
	Medicine and diagnostic technic	ques.			
Ē	Sehaviour of Gases				
•	Gaseous States: gas laws, ide	eal gas ec	uation, Dalton's	law of partial	
	pressure Postulates of kinetic	-	-	-	
(25	behavior, and Vander Waals equ	•			
S S	· · · · ·			ty of states, the	
	Critical Phenomena: PV isotherms of real gases, continuity of states, the				
UNIT-2 IIGN HO	 isotherms of vander Waals equation, critical temperature and its importance. Relationship between critical constants and vander Waals constants, the law of corresponding statesreduced equation of state. Molecular velocities: Root mean square, average and most probable 				
ZZ					
•					
 Signification of the pressure Postulates of kinetic theory of gases, deviation from identification behavior, and Vander Waals equation of state. Critical Phenomena: PV isotherms of real gases, continuity of states, the isotherms of vander Waals equation, critical temperature and importance. Relationship between critical constants and vander Waals constants, the law of corresponding states reduced equation of state. Molecular velocities: Root mean square, average and most probable velocities, Qualitative discussion of the Maxwell's distribution molecular velocities.collision number, mean free path and collision. 					
E molecular velocities, collision number, mean free path and collision			and collision		
	diameter, Liquification of gase	es (based o	on Joule – Thomso	on effect).	

	Liquid and Colloidal States
UNIT-3 TEACHING HOURS(22)	 Liquid State: Kelvin equation, surface tension and surface energy,wetting and contact angle ,interfacial tension and capillary action. Intermolecular forces,structure of liquids (a qualitativedescription).Structural differences between solids,liquids and gases,Liquid crystals: Difference between liquid crystal,solid and liquid, vapoure pressure,surface tension and viscosity coefficient and its application Classification,structure of nematic and cholestric phases,Thermography and seven segment cells. Colloidal States: Definition of colloids,classification of colloids, Solids in liquids (sols): properties kinetic, optical and electrical, stability of colloids,protective action,Hardy-Schulze law,gold number. Liquids in liquids (gels): classification,preparation and properties,inhibition,general application of colloids,colloidal electrolytes.
	Solutions ,Dilute Solutions and Colligative Properties
UNIT-4 TEACHING HOURS (25)	• Ideal and no-ideal solutions, methods of expressing concentrations of solutions, activity and coefficient. Dilute solutions, colligative properties, Raoult's law relative lowering of vapour pressure, molecular weight determination. Osmosis, law of osmotic pressure and its measurement, determination of molecular weight from osmotic pressure. Elevation of boiling point and depression of freezing point. Thermodynamic derivation of relation between molecular weight and elevation in boiling point and depression in freezing point. Experimental methods for determining various colligative properties.
	Chemical Equilibrium
UNIT-5 TEACHING HOURS (33)	 Equilibrium constant and free energy. Thermodynamic derivation of law of mass action. Le Chatelier's principle andits application to physical and chemical system, factors affecting chemical eqilibria. Reaction isotherm and reaction isochore, Clapeyron equation and Clausius-Clapeyron equation, applications. Equilibrium: statement and meaning of the terms-phase, component and degree of freedom, derivation of Gibbs phase rule, phase equilibria of one component system-water, CO₂ and S systems. Phase equilibria of two component system – solid – liquid equillibria, Solid solutions-compound formation with congruent melting point (Mg-Zn) and incongruent melting point, (NaCl-H₂O), (FeCl₃-H₂O) system. Freezing mixtures, acetone-dry ice. Liquid-liquid mixtures-Ideal liquid mixtures, Raoult's and Henry's law. Non-ideal system-azeotropes- HCl-H₂O and ethanol-water systems. Partially miscible liquids.

TEACHING AND	1 L	ecture method		
LEARNING	2. Problem Solving method			
STRATEGIES	3. Graphical method			
STRATEOILS	4. Seminar/Symposia			
	5. Review of literature			
	 Keview of interature Report writing 			
		roup Discussion		
		ideos/Animation		
		elf-Learning/e-Learning orkshops/Experiments.		
			ahanga ag nan	
		ching and learning strategies may be tent of the students and their capabilit		
CONTINUOUS &		f Continuous and Comprehensive A		
COMPREHENSIVE	follows:			
ASSESSMENT	S. No.	CCA- Components	Max. Marks Allocation	
(CCA)	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		marks equivalent reduced to CCA origin		
EXAMINATION		examinations are organized by the un		
PATTERN	format to enable the scholars to achieve success in contemporary			
		ons and to achieve their goals.	r y	
PERIODICAL	1. ANN			
REVISE OF	2. How	ever, the Universitymay revise the syl	labus at any time during	
SYLLABUS		unning year after giving a notice for a	•	
SELECTED	 Bajpai, D. N. (1998). Advanced physical chemistry. S. Chand. 			
READINGS	 CHRistmann, K. (2013). Introduction to surface physical 			
	chemistry(Vol. 1). Springer Science & Business Media.			
	• Donnan, F. G. (1916). A System of Physical Chemistry.			
		e,98(2446).	<u>j</u>	
		,T.Drobny,G., & Reid,P. J. (2008). Phy	sical chemistry for the life	
	-	ces. Prentice Hall.	, <u></u>	
		,D.,& Glasstone,S. (1960). Elements	of physical chemistry.	
	Macn		1	
	• Linga	felter. E. C. (1960). Elements	of Physical Chemistry	
	0	stone,Samuel).	<u> </u>	
	•	e, M. A. (2018). Physical properties of ma	aterials. CRC press.	
L		, (reprived of me	Press.	

CHEMISTRY PRACTICALS- I	
Duration:5 HR MAX.MARKS: 50 Min. Marks: 20	
Inorganic Chemistry	
Ex.1 Separation and identification of 3 cations and 3 anions in the mixture	15
Organic Chemistry	
Ex.2 Laboratory Techniques	03
Ex.3 Qualitative Analysis Detection of elements and detection of functional g	roup 10
Physical Chemistry	1
Ex.4 Perform one of the experiments mentioned in the syllabus.	12
Ex.5 Vive-Voce	05
Ex.6 Practical-Record	05
Total	50
Marks	
LIST OF EXPERIMENTS	
A. Inorganic Chemistry	
Semi micro analysis: Detection of the presence of tHRee cations and the	Ree anions (including
interfering) in a given mixture qualitatively.	ζ υ
B.Organic Chemistry	
Laboratory techniques	
Calibration of Thermometer	
Naphthalene (80-82°C), Acetanilide (113.5-114°C), Urea (132.5-133°C), Distill	ed Water (100°C)
Distillation:	
Simple distillation of ethanol-water mixture using water condenser	
 Distillation of nitrobenzene and aniline using air condenser 	
 Crystallization 	
 Concept of induction of crystallization 	
	nal)
• Phthalic acid from hot water (using fluted filter paper and stem less fun	ner)
Acetanilide from boiling water	
Naphthalene from ethanol	
Benzoic acid from water	
• Decolourisation and crystallization using charcoal	
• Decolourisation of brown sugar (sucrose) with animal charcoal using g	•
• Crystallization and decolorisation of impure naphthalene (100g of naph	thalene mixed with 0.3g
of Congo Red using 1g decolorizing carbon) using ethanol	
Sublimation (simple and Vacuum):	
• Camphor, Naphthalene, phthalic acid and Succinic acid.	
Determination of melting point:	
Naphthalene,Benzoic acid,Urea,Succinicacid,Cinnamicacid,Salid	cylicacid, Acetanilide, m-
Dinitrobenzene, p-chlorobenzene, Aspirin.	
Determination of boiling points: Ethanol, Cyclohexane, Toluene, Anilineand	Nitrobenzene.
C. Physical Chemistry	
Chemical Kinetics:	
To determine the specific reaction rate of the hydrolysis of methyl acetate/ethy	yl acetate catalyzed by
hydrogen ions at room temperature.	
To study the effect of acid strength on the hydrolysis of an ester.	
To compare the strength of HCI and H_2SO_4 by studying the kinetics of hydrol	ysis of ethyl acetate.
To study kinetically the reaction rate of decomposition of iodide by H_2O_2	
Colloids:	
To prepare arsenious sulphide sol and compare the precipitating power of i	monobi- and trivalent
anions.	
Viscosity: To determine the percentage composition of a given mixture (non	interacting systems) by
	interacting systems, by
	r

viscosity method.

To determine the percentage composition of a given binary mixture by viscosity method.(acetone & ethyl

Surface Tension:

To determine the surface tension of amyl alcohol in water at different concentrations and calculate the excess viscosity of these solutions.

To determine the percentage composition of a given binary mixture by surface tension method (acetone & ethyl methyl ketone).

	B.Sc.B.Ed. I Yes	ar		
COURSE CODE:	BSCBED-155 c I	COU	JRSE TY	PE : CORE
COURSE TITLE :	Non-Chordata			
MAX.MARKS:	50	MIN.PASS M.	ARKS	20
THEORY	40	MIN.PASS M	ARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN. PASS M	ARKS:	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPEC	TIVE YEAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMINAT	ION	MON	THLY TEST
DURATION	03 HR			01 HR
 evolutionary trends in external morphology and internal structure; identification and classification with examples, to enable them to understand various modes of adaptations in animals. To learn about the get familiar with various kind of nonchordata Amoeba to Asterias. To learn about the apply the General principles of taxonomy and classification in real life situation and further studies. Learning Outcomes: This course will enable the students to: Understand invertebrates, the organizational hierarchies and complexities, the evolutionary trends in external morphology and internal structure; identification and classification with examples, to enable them to understand various modes of adaptations in animals. Get familiar with various kind of nonchordata Amoeba to Asterias. Apply the General principles of taxonomy and classification in real life situation and further studies. 				
UNIT-1 TEACHINGHOURS (15)	 General principles of t five kingdom scheme, symmetry,coelom,segn classification of Protozo General Structural organ Habit and habitat, str reproduction of Paramee Locomotion in Protozoa Nutrition in Protozoa Reproduction in Protozoa Reproduction in Protozoa Protozoan'sparasites: o structure, life cycle pathogenic protozoa 	Basis of classifinentation, embr oa up to order. nization of Amo ructure, nutritio cium ans- pseudopodia oa. distribustion, ha and diseases	cation of yogeny. eba, Eugl on, osmo al, ciliary bit a s causeo	non-chordata: Outline ena. oregulation and and flagellar. nd habitat, l by selected

	• Oi	utline classification of Porifera an	d Coelenterata up to order.	
-2 ING (15)		abit and habitat, morphology, inte Sycon	rnal structure, reproducation	
UNIT-2 LACHIN DURS (J	• Ca	anal system and skeleton in Spong	ges	
UNIT-2 TEACHING HOURS (15)		abit, habitat, morphology, internal producation of Obelia	structure, nutrition and	
	• Pc	lymorphism in coelenterates, cor	al reefs	
-3 ING (15)	• O	utline classification of Platyhelmi to order.		
UNIT-3 TEACHING HOURS (15		 Habit and habitat, morphology, internal structure, reproduction and life – cycle of Fasciola, and Ascaris 		
E H	• Pa	rasitic adaptations in Helminthes		
S S S S S S S S S S S S S S S S S S S		ine classification of Annelida and	ArtHRopoda up to order.	
UNIT-4 CHING HOURS (15)	excr	it and habitat, structure, nutrit etion, nervous system and r <i>emon</i>	ion, respiration, circulation, eproduction of Hirudinaria	
UNIT-4 HING F (15)	• Peri	patus: structure and affinities		
CHC	• Mou	th parts and feeding habits of Inse	ects.	
'EA		val form of crustacean.		
E	• Dros	• Drosophila-structure and life history.		
		tline classification of Mollusca	a and Echinodermata up to	
2 3	order			
UNIT-5 TEACHING HOURS (15)	• Habit and habitat, structure, nutrition, respiration, blood vascular system, excretion, nervous system and reproduction of Pila			
	-	orsion in Gastropoda.	ind reproduction of Tha	
L IOI		abit and habitat, structure, nutritic	on, respiration, blood vascular	
	system, excretion, nervous system and reproduction of Asterias.			
	• La	rval forms of Echinoderms.		
TEACHING AND		ecture method		
LEARNING		oblem Solving method		
STRATEGIES		aphical method minar/Symposia		
		eview of literature		
		eport writing		
		oup Discussion		
		deos/Animation		
		lf-Learning/e-Learning		
	 10. Workshops/Experiments. * The Teaching and Learning Strategies may be change as per 			
	requirement of the students and their capabilities.			
CONTINUOUS &				
COMPREHENSIVE	follows:	-		
ASSESSMENT	S. No.	CCA- Components	Max. Marks Allocation	
(CCA)	1. Monthly test 20*3 Test=60			
	2.	Quizzes and Assignments	10	
	3. Viva-voce 10			

	4.	Seminar/Symposia	10
	5.	Report writing	10
	6.	Workshop	10
	7.	Review of literature	10
	8.	Creativity/Innovation	10
	9.	Experimental Skill	10
	10.	Co-curricular activity	10
	11.	Attendance	10
		marks equivalent reduced to CCA	_
EXAMINATION PATTERN	Term-end format to	examinations are organized by th enable the scholars to achiev ons and to achieve their goals.	e university in the prescribed
PERIODICAL REVISE OF SYLLABUS	1. ANNU 2. HOW	JAL EVER, THE UNIVERSITYmay luring the running year after giv	
SELECTED READINGS	 Sauna Brool labor Brool labor Brool labor Curti gener Hegn zoolo Hyma McG Jorda revise Jorda Chan Light zoolo McEa press Parke Macn Scott cladis Sedgy Allen Verm Willn evolu 	er, T. J., & Haswell, W. A. (19 nillan International Higher Educat Ram, N. R., & Scott-Ram, N stics, taxonomy and evolution. Can wick, A. (1905). A Student's Tex & Unwin. a, P. S. (2001). Invertibrate Zoolo ner, P. (1990). Invertebrate relat tion. Cambridge University Press. ogy, I. by EL Jordan and PS Verr	 f invertebrate zoology: for ino. f invertebrate zoology: for ino. rs, K. R. (1938). Textbook of G. (1968). Invertebrate lan. rates: mollusca I (Vol. 6). Invertebrate Zoology sixth and Company, Ltd. 857pp. 9). Invertebrate zoology. S. field text in invertebrate larvae. CRC 967). Textbook of zoology. ion. N. R. (1990). Transformed mbridge University Press. t-book of Zoology (Vol. 2). gy. S. Chand Publishing. ionships: patterns in animal

	B.Sc. B.Ed. I Yea	r			
COURSE CODE:	BSCBED-155 c II		COURSE	TYPE : CORE	
COURSE TITLE :	Animal Cell Biology and Ge	netics			
MAX.MARKS:	50	MIN.	PASS MARKS:	20	
THEORY	40	MIN.	PASS MARKS:	16	
EXAMINATION:					
CONTINUOUS	10	MIN.	PASS MARKS:	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIV	VE YE	AR		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATIO	DN	MONTHI	LY TEST	
DURATION	03 HR		01 H	łR	
environment. • To learn about th and modern cond Learning Out Come : A • Comprehend the	 To learn about the appreciate the role and significance of cell in human welfare and environment. To learn about the Comprehend the modern concepts and applied aspects of Cell Biology and modern concepts of Genetics and to create awareness regarding inheritance. Learning Out Come : After completion of the course, student-teachers will be able: Comprehend the modern concepts and applied aspects of Cell Biology and modern concepts of Genetics and to create awareness regarding inheritance. 				
UNIT-1 TEACHING HOURS (15)	 Introduction of cell: Discov (bacterial) and eukaryotic cell viruses and viroids. Cell membrane: Ultra struct unit membrane concept, fluid membrane. Modifications (sp Transport across cell membra diffusion), facilitated (mediat and secondary) and Endocyto Mitochondria: Morphology functions, origin, electron tra molecules. 	lls (plan ture, ch lity, gly pecializ ane: Pa ted) dif psis and , ultra s	nt and animal cells emical composition ycocalyx and func- tations) of plasma ssive transport (os fusion; active trans l Exocytosis.	s), cell theory, on, models, tions of cell membrane. smosis, asport (primary l composition,	
UNIT-2 TEACHIG N HOURS (15)	• Ultrastructure, types, chemic of: Endoplasmic reticuium,Golgi-co Cilia, flagella and microtubules.				

UNIT-3 TEACHING HOURS (15)	 Nucleus: occurrence, number, shape, size and structure (nuclear envelopes, nuclear matrix and nucleolus) CHRomosomes: Introduction, discovery morphology, structure (cHRomatids, primary and secondary constrictions, nucleolar organizer and telomeres) types. Chemical composition and functions. CHRomosomal organizations: EucHRomatin, heterocHRomatin, nucleosome concept. Cell reproduction: Cell cycle, phases, process and significances of mitosis and meiosis. CHRomosomal mutations- Variation in cHRomosome number (aneuploidy and euploidy) Structural changes in cHRomosomes (delection, duplication, inversion and translocation). 		
RS	• Mendelian principles of inheritance- monohybrid and dihybrid cross, back cross and test cross.		
UNIT-4 TEACHING HOURS (15)	• Deviation of Mendelism– incomplete dominance, codominance and lethal genes, modification of 3: 1 and 9: 3: 3: 1 with examples and problems.		
TEACH	 Gene interactions: Epistasis, complementary, supplementary, duplicate genes with cumulative effects and collaborator genes. Multiple alleles: Characters, examples, pseudoalleles, inheritance of A, B, AB, O and Rh blood groups (antibody reactions) 		
UNIT-5 TEACHING HOURS (15)	 Sex-determination: Genetic (sex cHRomosome, genic balance and haplo-diploidy mechanisms), hormonal and environmental control of sex determinations with examples. Sex-linked inheritance: white eye color in Drosophila, colour blindness and hemophilia in man. Linkage: Definition difference between linkage and independent assortment, cHRomosomal theory of linkage, kinds, linkage groups and significances. Crossing over: Definition, mechanism, theories, kinds, frequency, factors affecting crossing over and significances. 		
TEACHING AND LEARNING	1. Lecture method 2. Problem Solving method		
STRATEGIES	 Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be change as per requirement of the students and their capabilities. 		
CONTINUOUS COMPREHENSIVE	Details of Continuous and Comprehensive Assessment (CCA) are as follows:		
ASSESSMENT (CCA)	S. No. CCA- Components Max. Marks Allocation		
	1.Monthly test20*3 Test=602.Quizzes and Assignments10		
	2. Quizzes and Assignments 10		

	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.		10	
		Workshop		
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
	Total 160 n	narks equivalent reduced to CCA original	inal marks 30.	
EXAMINATION	Term-end e	examinations are organized by the univ	versity in the prescribed	
PATTERN		enable the scholars to achieve such		
		ns and to achieve their goals.		
	-	-		
PERIODICAL	1. ANNUA	L		
REVISE OF	2. HOWEV	VER, THE UNIVERSITYmay revi	se the syllabus at any	
SYLLABUS	time du	ring the running year after giving	a notice for a period	
	one mon	th.		
SELECTED	• Alberts,	B., Johnson, A., Lewis, J., Raff, M.,	Roberts, K., & Walter,	
READINGS	P. (2015). <i>Molecular biology of the cell</i> . Garland Science. <i>New York</i> ,			
	1227-1242.			
	• Blomquist, G. J., & Bagnères, A. G. (Eds.). (2010). Insect			
	hydrocarbons: biology, biochemistry, and chemical ecology.			
	-	Cambridge University Press.		
		• Conklin, K. F., Doerfler, W., Grafstron, R. H., Groudine, M.,		
		Hamilton, D. L., Jaenisch, R., & Langner, K. D. (2012). DNA		
	methylation: biochemistry and biological significance. Springer			
	Science & Business Media.			
		P. K. (2005). Cell and molect	ular biology Postogi	
	Publicat	· · · · ·	nur bibibgy. Rasiogi	
		6. (2007). Cell and Molecular Biolog	w John Wilow & Song	
	Incorpor		y. John whey & John	
	-	G. (2009). Cell and molecular b	viology: concepts and	
	-	ents. John Wiley & Sons.		
	-	R. L. (1967). Annelida. Rastogi Public	cations	
	-	H., Berk, A., Kaiser, C. A., Krie		
		er, A., & Matsudaira, P. (2008). M	-	
	Macmil		toreeniar een biology.	
		F. S., Steiner, L., & Unanue, E. (1989) Macmillan dictionary	
	of immu		j. muchallan alchonary	
	v	, G. (2010). Cellular and Biod	phomical Science IV	
	-	ional Pvt Ltd.	inemical science. IK	
			amistry and Maland	
		E. B. K., & Walker, J. (2005). Bioch	lennstry and Molecular	
	biology.	. Kuudes painos.		

PRACTICAL

Duration: 4 Hours Objectives:

To understand internal organization and skills- of staining and mounting of materials.

(Temporary and permanent), of dissection, display and labeling, of preparation of cultures of invertebrates by using common culture methods, laboratory observation of animal cell division.

Course Content

Study of museum specimens: with respect to levels and patterns of organization biosytematics. Biodiversity, adaptations, development stages, population dynamics, ecological implications etc.

Porifera: Sycon, Spongilla, Euplectella, Leucosolnia, Hylonema, Hypospongia, Euspongia:

Coeleterata: Hydra, Tubularia, Millepora, Physalia, Porpita, Vellela, Aurelia, Tubipora, Alcyonium, Metridium, Pennatula, Grantia, Fungia, Gorgonia.

Helminthes: Fasciola, Taenia solium, Planaria, Ascaris, Ancyclostoma

Annelida: Nereis Heteroneresis, ApHRodite, Chaetoptreus. Arenicola, Pheretima, Hirudinaria ArtHRopoda: Palaemon, Eupagurus, Scolopendra, Apis, Peripatus.

Mollusca: Chiton, Pila, Aplysia, Helix, Dentalium, Mytilus, Pinctada, Unio, Sepia, Loligo Octopus.

Echinodermata: Antedon, Holothuria, Cucumaria, Astropecten, Asterias, Echinus

Microscope: Simple and compound microscope, working mechanism and maintenance. •

Study of Permanent slides:

Paramecium, Paramecium Conjugation, paramecium binary fission, Euglena, Vorticella, Sycon-L.S., Sycon T.S., Hydra L.S., Hydra T.S, Cercaria larva, Metacercaria, Miracidium larva, Sporocyst larva, Redia larva, Ascaris male and female T.S., T.S. thorough, pharynx region, Gizzard and intestinal region of Earthworm, T.S. tHRough buccal cavity of Hirudinaria, Crustacean larva- Zoea, Metazoea, Nauplius, Mysis, T.S. of gill of Unio, T.S. of the shell & mantle of Unio. Glochidium larva of Unio.

Dissections and/or its demonstration tHRough Charts/Models/Video/CD/digital alternative etc and/or preparation of working models of the different systems of the following animals.

Earthworm: Alimentary canal Nervous system, Reproductive system

Leech: Alimentary canal

Cockoach: Mouthparts, Digestive system, nervous system

Prawn: Nervous system

Pila: Nervous system

Microsopic preparation or their observation of the following:-

- Paramecium, Eunglena, Sponge-spicules, gemmules, Obelia, Hydra, parapodium of Nereis, statocyst of Prawn, mouth parts of Cockroach, radula of Pila, gill of Unio.
- Study of bacterial and eukaroytic cell
- Slides of sub cellular components(Cell organelles)
- ErytHRocyte plasma membrane permeability.
- Study of Karyotype and Idiogram of man.
- Study of Barr Bodies in human buccal epithelial cells. •
- Drosophila culture and life cycle.
- Sexual Dimorphism in Drosophila, Identification of wild or mutant varieties. •
- Study of salivary gland Chromosomes of Drosophila •
- Problems on pedigree analysis. •
- Meiotic studies of testes of cockroach.

MAX.MARKS: 50

MAX.MARKS: 50 S. No.	Exercise	Time: 4 HR. Marks
1.	Dissection* (Exposition, labeled diagram)	6
2.	Temporary mounting –one (Staining, identification, sketch	3
3.	Museum specimens – five (identification and classification	10
4.	Permanent slides – two or four (Identification with reasons)	4
5.	Preparation of cHRomosome slide (root tip/grasshopper testis)	5
6.	Drosophila spotting / pedigree	5
7.	Practical record and slides	5
8.	Viva	4
9.	Project report and assignment	8

Guidelines/ Instructions for Practical Examination

Note:

- Use of animals for dissection is subject to the conditions that these are not banned under the Wildlife Protection Act or any other legislation.
- Students are required to submit the following during examination.
- One assignment on the instrument/ technique about its principle, working, precautions and applications, and /or reagents / solutions preparation.
- Report on study of animals from their natural habitat from their local surroundings.Vermicompost etc.

B.Sc.B.Ed. I YearCOURSE CODE:BSCBED-155 d ICOURSE TYPECOURSE TITLE :Diversity of Microbes and Lower Plants& PlantsMAX.MARKS:50MIN.PASS MARKSTHEORY40MIN.PASS MARKSEXAMINATION:40CONTINUOUS10CONTINUOUS10MIN.PASS MARKS40COMPREHENSIVE40ASSESSMENT (CCA)80 PERCENT IN RESPECTIVE YEARELIGIBILITY40EXAMINATIONMONTHLDURATION03 HR01 H					
COURSE TITLE :Diversity of Microbes and Lower Plants& PlantsMAX.MARKS:50MIN.PASS MARKSTHEORY40MIN.PASS MARKSEXAMINATION:CONTINUOUS10MIN.PASS MARKSCOMPREHENSIVEASSESSMENT (CCA)ATTENDANCE80 PERCENT IN RESPECTIVE YEARELIGIBILITYEXAMINATIONTERM END EXAMINATIONMONTHL	t Pathology 20 16				
(Thallophyta)MAX.MARKS:50MIN.PASS MARKSTHEORY40MIN.PASS MARKSEXAMINATION:MIN.PASS MARKSCONTINUOUS10MIN.PASS MARKSCOMPREHENSIVE80PERCENT IN RESPECTIVE YEARASSESSMENT (CCA)80PERCENT IN RESPECTIVE YEARELIGIBILITYEXAMINATIONMONTHL	20 16				
THEORY 40 MIN.PASS MARKS EXAMINATION: MIN.PASS MARKS CONTINUOUS 10 MIN.PASS MARKS COMPREHENSIVE MIN.PASS MARKS ASSESSMENT (CCA) MIN.PASS MARKS ATTENDANCE 80 PERCENT IN RESPECTIVE YEAR ELIGIBILITY MONTHL	16				
EXAMINATION:Image: Construction of the second s	-				
CONTINUOUS10MIN.PASS MARKSCOMPREHENSIVE10MIN.PASS MARKSASSESSMENT (CCA)80 PERCENT IN RESPECTIVE YEARATTENDANCE80 PERCENT IN RESPECTIVE YEARELIGIBILITYEXAMINATIONTERM END EXAMINATIONMONTHL	4				
COMPREHENSIVE ASSESSMENT (CCA) Image: Comparison of the second	4				
ASSESSMENT (CCA)ATTENDANCE ELIGIBILITY80 PERCENT IN RESPECTIVE YEAREXAMINATIONTERM END EXAMINATIONMONTHL					
ATTENDANCE ELIGIBILITY80 PERCENT IN RESPECTIVE YEAREXAMINATIONTERM END EXAMINATIONMONTHL					
ELIGIBILITYEXAMINATIONTERM END EXAMINATIONMONTHL					
EXAMINATION TERM END EXAMINATION MONTHL					
DURATION 03 HR 01 H					
	IK				
Objectives:					
• To learn about the diversity that exists in microorganisms.					
• To learn about the morphology, organization, structure, and reproduction in 1					
• To learn the role and significance of microbes in human welfare and environ	ment.				
• To learn about the symptoms of selected diseases caused by microbes.					
Learning Outcomes: This course will enable the students to:					
• Acquaint students with the diversity that exists in microorganisms.					
Understand the morphology, organization, structure, and reproduction in mic					
• Appreciate the role and significance of microbes in human welfare and envir	onment.				
• Study the symptoms of selected diseases caused by microbes.					
Viruses and Bacteria:					
Structure, Multiplication, transmission and disease sy	Structure, Multiplication, transmission and disease symptoms of				
\vec{z} \vec{z} viruses, Structure and economic importance of mycoplas	viruses, Structure and economic importance of mycoplasma, Bacteria				
Structure , Multiplication, transmission and disease synthesis viruses , Structure and economic importance of mycoplast structure, nutrition, reproduction and economic Gram's staining, Disease symptoms of bacteria in plants Canker, Viruses e.g., Mosaic Viruses disease in tobacco	structure, nutrition, reproduction and economic importance,				
Gram's staining , Disease symptoms of bacteria in plants	_				
E H Canker, Viruses e.g.:- Mosaic Viruses <i>disease</i> in tobacco.	8				
General Account and Economic importance of Cyanobacte	ria				
	/114.				
ن بن بن <u>Algae:</u> Communication (Tri	() 1025				
General characters, occurrence, classification (Fri Pigment constitution, fine structure of algal plastids, life-c and evolution of sex and thallus in algae . Cyanophyceae: Life Cycle of Nostoc, Anabena and Oscillat	General characters, occurrence, classification (Fritsch, 1935),				
Contractors, occurrence, classification (Find Pigment constitution, fine structure of algal plastids, life-c and evolution of sex and thallus in algae.	Pigment constitution, fine structure of algal plastids, life-cycles, Origin				
$5 \neq 3$ and evolution of sex and thallus in algae .	nd evolution of sex and thallus in algae.				
E H Cyanophyceae: Life Cycle of Nostoc, Anabena and Oscillat	toria				
Structure, reproduction and evolutionary significance	of following				
γ \ddot{X} \ddot{C} <u>genera:</u>	alaashaata				
Sectorgenera:Chlorophyceae: Chlamydomonas, Volvox, UlotHRix and ControlXanthophyceae: Vaucheria,Phaeophyceae: Ectocarpus, Sargassum	oieocnaete				
Xanthophyceae:Vaucheria,					
Phaeophyceae: Ectocarpus, Sargassum					
Rhodophyceae: Polysiphonia					

UNIT-4 TEACHING HOURS (15)	Fungi:General Characters, Occurrence, Classification (Alexopolus and Mims, 1979;G. C. Ainswort,1986), Types of Mycelia, Structure of Fungal Cell, Fungal Flagella, Nutrition and Economic Importance of Fungi. Host-parasite interaction, Control of plant diseases. Disease Symptoms of Fungi in Plants.			
UNIT-5 TEACHING HOURS (15)	Structure reproduction and life history of:Mastigomycotina – Synchytrium&PhytophthoraAscomycotina – Saccharomyces,Eurotium,PezizaBasidiomycotina – Puccinia, Agaricus, UstilagoDeuteromycotina – Alternaria, Cercospora,Colletotrichum, Generalaccount and economic importance of Lichens			
TEACHING AND LEARNING STRATEGIES	 Lecture method Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be 			
CONTINUOUS COMPREHENSIVE		change as per requirement of the students and their capabilities. Details of Continuous and Comprehensive Assessment (CCA) are as		
ASSESSMENT (CCA)	S. No.	CCA- Components	Max. Marks Allocation	
	1.	Monthly test	20*3 Test=60	
	2. 3.	Quizzes and Assignments Viva-voce	10 10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		rks equivalent reduced to CCA original		
EXAMINATION PATTERN	Term-end examinations are organized by the university in the prescribed format to enable the scholars to achieve success in contemporary competitions and to achieve their goals.			
PERIODICAL REVISE OF SYLLABUS	 ANNUAL HOWEVER, THE UNIVERSITYmay revise the syllabus at any time during the running year after giving a notice for a period one months. 			

SELECTED	• Dube, H. C. (1990). Fungi, general characteristics. An introduction to
READINGS	fungi, 2nd revised edn. Vikas, New Delhi, 11-146.
	• Hays, J. (1986). <i>Genetics of bacteria</i> : Edited by J. Scaife, D. Leach, and A. Galizzi. New York: Academic Press. (1985).
	• Mudd, J. B. (Ed.). (2012). <i>Responses of plants to air pollution</i> . Elsevier.
	• Odom, J. M., & Singleton, R. (1993). The sulfate-reducing bacteria:
	contemporary perspectives (pp. 189-210). New York: Springer-
	Verlag.
	• Sharma, O. P. (1986). <i>Textbook of algae</i> . Tata McGraw-Hill
	Education.
	• Sharma, O. P. (1992). <i>Textbook of Thallophyta</i> . McGraw Hill Pub.
	Co.
	• Sharma, P. D. (1991). <i>The Fungi</i> . Rastogi Publications.
	• Sharma. P.D. 1991. The Fungi. Rastogi & Co. Meerut.
	• Smith Jr, J. P. (2018). The Herbarium.
	• Smith, GM. (1971). Cryptogamic Botany. Algae & Fungi. Vol. 1
	New Delhi: Tata McGraw Hill Publishing Co.

B.Sc. B.Ed. I Year				
COURSE CODE:	BSCBED-155 d II	BSCBED-155 d II COURSE TYPE : COR		
COURSE TITLE :	Diversity of Cryptoga	Diversity of Cryptogams (Bryophytes & Pteridophytes)		
MAX.MARKS:	50	MIN.PAS	S MARKS	20
THEORY	40	MIN.PAS	S MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PAS	S MARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE ELIGIBILITY	80 PERCENT IN RES	PECTIVE	IEAK	
EXAMINATION	TERM END EXAMI	ΝΑΤΙΟΝ	MON	THLY TEST
DURATION	03 HR			01 HR
Objective:	US IIK			
 To learn about the morphology, organization, structure, and reproduction in cryptogams. To learn about the role and significance of cryptogams in human welfare and environment. To learn about the Geological time scale, fossils and fossilization processes, significance of fossils. Learning Outcome: After completion of the course, student-teachers will be able: Acquaint students with the diversity that exists in cryptogams. Understand the morphology, organization, structure, and reproduction in cryptogams. Appreciate the role and significance of cryptogams in human welfare and environment. Study the Geological time scale, fossils and fossilization processes, significance of fossils. 				
UNIT-1 TEACHING HOURS (15 s	General characters, distribution, origin of the land habit in lants, classification, Evolutionary trends in Thallus and porophyte development; Alternation of generations lassification; and Economic importance of Bryophytes.			
UNIT-2 IFACHIGN HOURS (15)	ollowing genera: Iepaticopsida – Riccia	paticopsida – Riccia Marchantia thocerotopsida – Anthoceros		
UNIT-3 TEACHING HOURS (15) L	yopsida – <i>Funaria</i> eneral characters, distribution, classification, stelar olution, and Origin of seed habit, and life cycles in eridophytes. ructure, reproduction and evolutionary significance of the lowing genera: ilotum; Lycopodium			

UNIT-4 TEACHING HOURS (15)	Structure, reproduction and evolutionary significance of the following genera: Selaginella; Equisetum; Pteris and Marsilea.			
UNIT-5 TEACHING HOURS (15)	Geological time scale, fossils and fossilization processes, significance of fossils.Study of the following form genera of fossils (<i>a</i>) <i>Rhynia</i> (<i>b</i>) <i>Calamtes</i> (<i>c</i>) <i>Glossopteris</i>			
TEACHING AND LEARNING STRATEGIES	 Lecture method Problem Solving method Graphical method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be 			
CONTINUOUS	change as per requirement of the students and their capabilities. Details of Continuous and Comprehensive Assessment (CCA) are as			
COMPREHENSIVE	follows:	-		
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)	1	Monthly test	Allocation	
	<u> </u>	Monthly test	20*3 Test=60	
	-	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	<u>6.</u> 7.	Workshop Review of literature	10	
	7.		10 10	
	<u>8.</u> 9.	Creativity/Innovation Experimental Skill	10	
	9.	Co-curricular activity	10	
	10.	Attendance	10	
		rks equivalent reduced to CCA origi	-	
EXAMINATION		aminations are organized by the univ		
PATTERN		nable the scholars to achieve suc	• •	
		and to achieve their goals.	cess in contemporary	
PERIODICAL	1. ANNUAL	-		
REVISE OF SYLLABUS	2. However	, the Universitymay revise the he running year after giving a no	•	

SELECTED READINGS	• Smith, GM. 1971. Cryptogamic Botany. Vol. 1 Algae & Fungi. Tata McGraw Hill Publishing Co, New Delhi.
	 Sharma, O.P. 1992. Text Book of Thallophytes. McGraw Hill Pub. Co. Smith, GM. 1971. Cryptogamic Botany. Vol-II Bryophytes and
	Pteridophytes. Tata McGraw Hill Pub. Co. New Delhi.Puri, P. 1980. Bryophyta. Atma Ram & Sons Delhi.

	BOTANY-PRACTICALS:-I	
Duration: 4 hours	MAX.MARKS: 50	Min. Pass Marks : 20
	Practical List	

- Study of the genera included in Theory syllabus of Algae and Fungi by making temporary micro preparations and observation of permanent slides.
- Observation of disease symptoms in host infected by Fungi, Bacteria, Viruses and Mycoplasma. Section cutting of diseased material and identification of the pathogens as per the theory syllabus.
- 3. Study of morphology, anatomy and reproductive structures of genera included in Bryophytes and Pteridophytes by making temporary micro preparations and observation of permanent slides.
- 4. Gram staining of bacteria. Study of crustose, foliose & fruticose Lichens. Maintenance & submission of a record of all the Laboratory activities.

B.Sc. B.Ed. I Year					
COURSE CODE:	BSCBED-155e I		COURS	E TYPE :CORE	
COURSE TITLE :	Calculus				
MAX.MARKS:	75	MIN.PASS MARKS 30		30	
THEORY	60	MIN.PASS MARKS		24	
EXAMINATION:					
CONTINUOUS	10	MIN.PASS N	IARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RE	SPECTIVE YI	EAR		
ELIGIBILITY					
EXAMINATION	TERM END EXAM		MO	NTHLY TEST	
DURATION Objectives:	03 HR			01 HR	
 curvature, asympto function of several calculus and their Learning Outcomes: This Learn conceptual calculus. Apply multivariab Inter-relationship at Know the mathem by revolution. Applications of n understanding the Realize important mathematics. 	 Apply multivariable calculus in optimization problems. Inter-relationship amongst the line integral, double and triple integral formulations. Know the mathematical background, properties of volume and surface of solids formed by revolution. Applications of multivariable calculus tools in physics, economics, optimization, and understanding the architecture of curves and surfaces in plane and space etc. Realize importance of Green, Gauss and Stokes' theorems in other branches of 				
UNIT-1 TEACHING HOURS (15)	Curvature, Asymptotes and Curve Tracing Curvature; Asymptotes of general algebraic curves, Asymptotes parallel to axes; Symmetry, Concavity and convexity, Points of inflection, Tangents at origin, Multiple points, Position and nature of double points; Tracing of Cartesian, polar and parametric curves.Limit and Continuity				
UNIT-2 TEACHIGN HOURS (20) D <	- δ definition of limit of afinite limits; Continuity ontinuous functions, In atterpretation of continuity, Differentiability: Differentiability of a real va- ifferentiability, Relation Differentiability and mor Darboux's theorem, Rolle's Cauchy's mean value theorem	Set $\overline{\delta}$ definition of limit of a real valued function, Limit at infinity ar ite limits; Continuity of a real valued function, Properties of inuous functions, Intermediate value theorem, Geometric pretation of continuity, Types of discontinuity; Uniform continuity			

5 C	Expansions of Maclaurin's	of Functions and Taylor's theorems for expansion	on of a function in an		
UNIT-3 TEACHING HOURS (12)		s, Taylor's theorem in finite form v			
UNIT-3 EACHIN OURS (1	and Roche-S	chlomilch forms of remainder; Maxi	ma and minima.		
	<u>Partial Differentiation</u> Functions of several variables, Level curves and surfaces, Limits and				
UNIT-4 TEACHING HOURS (20)		Partial differentiation, Tangent			
JRS		erivatives, The gradient, Maximal an			
101	the gradient, Tangent planes and normal lines. Extrema of functions two and more variables, Method of Lagrange multipliers, Constrained				
	optimization		lumphers, Constrained		
	Differentiati	L			
CH	U	partial derivatives, Total differentia			
'EA		hange of variables, Euler's theor	e		
F		aylor's theorem for functions of two velopes and evolutes.	vo variables and more		
		Triple Integrals			
33)	Reduction for	ormulae. Double integration over	e		
S S	-	regions, Double integrals in pola	-		
	-	a parallelepiped and solid region in cylindrical and	• •		
-1-5-T	integrals, Triple integration in cylindrical and spherical coordinates, Change of variables in double and triple integrals, Dirichlet integral.				
UNIT-5 ING HO	Green's, Stokes' and Gauss Divergence Theorem:				
		f vector field, Divergence, curl,	-		
UNIT-5 TEACHING HOURS (23)		ne integrals, Applications of line inte theorem for line integrals, Cons	-		
E		rem, Area as a line integral, Sur			
	1	Gauss divergence theorem.			
TEACHING AND LEARNING	 Lecture method Problem Solving method 				
STRATEGIES		nical method			
	4. Semir	nar/Symposia			
		w of literature			
	6. Repor	t writing Discussion			
		os/Animation			
	9. Self-L	Learning/e-Learning			
		shops/Experiments.			
		ing And Learning Strategies may of the students and their capabilit			
CONTINUOUS		on the students and then capability ontinuous and Comprehensive Ass			
COMPREHENSIVE	follows:	-			
ASSESSMENT	S. No.	CCA- Components	Max. Marks		
(CCA)	1.	Monthly test	Allocation 20*3 Test=60		
	2.	Quizzes and Assignments	10		
	2.Quilles and Hospinionis103.Viva-voce10				
	4.Seminar/Symposia10				

	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
	Total 160 mar	rks equivalent reduced to CCA origi	nal marks 30.	
EXAMINATION	Term-end exa	minations are organized by the univ	versity in the prescribed	
PATTERN	format to en	able the scholars to achieve suc	ccess in contemporary	
	competitions	and to achieve their goals.		
PERIODICAL	1. ANNU	UAL		
REVISE OF	2. However, the Universitymay revise the syllabus at any time			
SYLLABUS	during the running year after giving a notice for a period one			
	montl	hs.		
SELECTED	• Jerrold M	larsden, Anthony J. Tromba & A	lan Weinstein (2009).	
READINGS	Basic Mul	tivariable Calculus, Springer India l	Pvt. Limited.	
	• James Stewart (2012). Multivariable Calculus (7th edition).			
	Brooks/Cole. Cengage.			
	• Monty J. Strauss, Gerald L. Bradley & Karl J. Smith (2011).			
	• Monty J. Strauss, Geraid L. Bradley & Kall J. Sinut (2011). Calculus (3rd edition). Pearson Education. Dorling Kindersley			
	(India) Pv		I. Dorning Kindersiey	
	. ,			
	0	. Thomas Jr., Joel Hass, CHRistoph	her Heil & Maurice D.	
	Weir (201	8).		
	• Thomas' C	Calculus (14th edition). Pearson Edu	ication.	

B.Sc. B.Ed. I Year					
COURSE CODE:	BSCBED-155e II	CO	URSE TYPE	E : CORE	
COURSE TITLE :	Algebra and Geome	ora and Geometry			
MAX.MARKS:	75	MIN.PASS MARKS 30		30	
THEORY	60	MIN.PASS MA	RKS	24	
EXAMINATION:					
CONTINUOUS	15	MIN.PASS MA	RKS	6	
COMPREHENSIVE					
ASSESSMENT (CCA)			-		
ATTENDANCE	80 PERCENT IN RE	SPECTIVE YEA	R		
ELIGIBILITY					
EXAMINATION	TERM END EXAM			ILY TEST	
DURATION	03 H	K	01	HR	
Objective:		1	. 11		
	signed to provide a deeper a	C	0	1	
-	olynomial, relations and fu		•	•	
	ces, characteristics and equ				
sciences.	es in spaces and their utiliza	and importan		it branches of	
	s course will enable the stud	ents to:			
_	ortance of roots of real and c		als and learn	various	
methods of obtaining		omplex polynomia	and rearn	various	
	tions, equivalence relations	and partitions			
	_	_	numerical p	roblems	
 Employ De Moivre's theorem in a number of applications to solve numerical problems. Recognize consistent and inconsistent systems of linear equations by the row echelon form 					
e	of the augmented matrix, using rank.				
 Find eigenvalues and corresponding eigenvectors for a square matrix. 					
 Explain the properties of tHRee dimensional shapes. 					
	Theory of Equations and (*	s		
	Elementary theorems on the roots of an equations including Cardan's				
10	method, The remainder			U U	
	Factored form of a polynoi		•		
	Relations between the ro			-	
	equations, Imaginary roc				
AC		e		-	
LE	representation of complex n		•		
	Relations and Basic Numb				
Z (S	Relations, Equivalence re	· •			
1-2 11 5 (1 5 (1	Composition of functions, 1	osition of functions, Inverse of a function; Finite, countable an			
UNIT-2 TEACHIGN HOURS (15)	uncountable sets; The divisi	on algorithm, Div	isibility and	the Euclidean	
D O	algorithm, The fundamental	theorem of arith	metic, Modu	lar arithmetic	
ΗH	and basic properties of	basic properties of congruences; Principles of mathematical			
	induction and well ordering.				
	0				

	Dow Fob	lon Form of Matrices and Applicatic	ang l	
RS		elon Form of Matrices and Applicatio		
DC DC	•	of linear equations, Row reduction and		
UNIT-3 TEACHING HOURS (15)	-	nce, The rank of a matrix and applic		
UNIT-3 HING F (15)	linear tran	sformations, The matrix of a linear	transformation, Matrix	
	operations	, Determinants, The inverse of a matri	ix, Characterizations of	
CHC	invertible	matrices; Applications to Computer	Graphics; Eigenvalues	
EA.	and eigen	vectors, The characteristic equation an	d the Cayley-Hamilton	
	theorem.			
70		raight Lines and Spheres		
		vistance of a point from a plane, Angl	le between two planes.	
б		nes, Bisectors of	I,	
4 H		ween two planes; Straight lines: Equ	ations of straight lines,	
UNIT-4 HING F (23)	-	of a point from a		
	straight lin	ne, Distance between two straight line	es, Distance between a	
UNIT-4 TEACHING HOURS (23)	-	ne and a plane; Spheres: Different for		
E	-	Orthogonal intersection, Tangents and	-	
F		ne, Coaxial system of spheres, Pole, Pol	lar and Conjugacy.	
		rfaces, Curves and Conicoids		
UNIT-5 EACHIN : HOURS (22)	-	ves, Algebraic curves, curvature and		
		Ruled surfaces, Some standard surf		
D H B	-	urfaces, Cone, Cylinder, Central con	icolds, Tangent plane,	
		olar planes, and Polar lines.		
TEACHING AND LEARNING		cture method		
STRATEGIES		oblem Solving method		
SINALOILS	 Graphical method Seminar/Symposia 			
		view of literature		
		port writing		
		oup Discussion		
	8. Videos/Animation			
	9. Self-Learning/e-Learning			
	10. Workshops/Experiments.			
	* The TEACHING AND LEARNING STRATEGIES may be			
	U	per requirement of the students and	<u> </u>	
CONTINUOUS		Continuous and Comprehensive Ass	sessment (CCA) are as	
COMPREHENSIVE	follows:			
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)	1	Monthly test	Allocation 20*3 Test=60	
	1. 2.	Monthly test Quizzes and Assignments	20*3 Test=60 10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		marks equivalent reduced to CCA origi		
	Total 100 marks equivalent reduced to CON original marks 50.			

TEACHING AND	1. Lecture method
LEARNING	2. Graphical method
METHODS	3. Seminar/Symposia method
	4. Extension activity method
	5. Project and report writing
CONTINUES	1. Seminar/Symposia
ASSESSMENT	2. Project and report writing
METHODS	3. Viva-voce
	4. Monthly test
EXAMINATION	Term-end examinations are organized by the university in the prescribed
PATTERN	format to enable the scholars to achieve success in contemporary
	competitions and to achieve their goals.
PERIODICAL	1. ANNUAL
REVISE OF	2. HOWEVER, THE UNIVERSITY may revise the syllabus at any
SYLLABUS	time during the running year after giving a notice for a period
	one months.
SELECTED	• Titu Andreescu,& Dorin Andrica (2014). Complex Numbers from A
READINGS	toZ. (2nd edition). Birkhäuser.
	• Robert J. T. Bell (1994). An Elementary Treatise on Coordinate
	Geometry of THRee Dimensions. Macmillan India Ltd.
	• Chatterjee, D. (2009). Analytical Geometry: Two and THRee
	Dimensions. Narosa Publishing House.
	• Leonard Eugene Dickson (2009). First Course in the Theory of
	Equations. The Project Gutenberg EBook
	(http://www.gutenberg.org/ebooks/29785)
	• Edgar G. Goodaire & Michael M. Parmenter (2015). Discrete
	Mathematics with Graph Theory (3rd edition). Pearson Education Pvt. Ltd. India.
	• Bernard Kolman & David R. Hill (2003). Introductory Linear Algebra with Applications (7th edition). Pearson Education Pvt. Ltd.
	India.
	• David C. Lay, Steven R. Lay & Judi J. McDonald (2016). Linear Algebra and its Applications (5th edition). Pearson Education Pvt.
	Ltd. India.
	Liu. mula.

SECOND YEAR

Course Code	Title of the course		EVALUA	ATION	
		External	Internal	Practical	Total
BSCBED-210	Gen. Hindi(Compulsory)*	70	30	-	100
BSCBED-251	Knowledge & Curriculum	70	30	-	100
BSCBED-252	Teaching & Learning	70	30	-	100
BSCBED-253	Health & Physical Education	35	15		50
BSCBED-254	Content: (BCZ& PCM)	I			
BSCBED-254 a I	1. Physics I	40	10		
BSCBED-254 a II	2. Physics II	40	10	50	200
BSCBED-254 a III	3. Physics III	40	10	-	
BSCBED-254 b I	4. Chemistry I	40	10		
BSCBED-254 b II	5. Chemistry II	40	10	50	200
BSCBED-254 b III	6. Chemistry III	40	10	50	
BSCBED 250 c I	7. Zoology I	40	10		
BSCBED 250 c II	8. Zoology II	40	10	50	200
BSCBED-254 c III	9. Zoology III	40	10	50	
BSCBED-254 d I	10. Botany I	40	10		
BSCBED-254 d II	11. Botany II	40	10	50	200
BSCBED-254 d III	12. Botany III	40	10	50	
BSCBED-254 e I	13. Mathematics I	60	7		
BSCBED-254 e II	14. Mathematics II	60	7]	200
BSCBED-254 e III	15. Mathematics III	60	6		
	CCA				25
Prayer, Yoga, Meditation & Festival etc					25
Total					900

B.Sc.B.Ed. II YEAR					
COURSE CODE:	BSCBED-220		COURSE TY	PE : CORE	
COURSE TITLE :	GENERAL HINDI				
MAX.MARKS:	100	MIN. PASS MARKS: 40		40	
THEORY	70	MIN. P	ASS MARKS:	28	
EXAMINATION					
CONTINUOUS	30	MIN. P	ASS MARKS:	12	
COMPREHENSIVE					
ASSESSMENT					
(CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIVE	E YEAR			
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATIO	N	MONTHLY		
DURATION	03 HR		01 HF	2	
उद्देश्यः		0.0			
	ठ प्रमुख कवियों एवं रचनाकारों व	-			
	जे भाषा के साथ भारतीय सम्यत				
	हिन्दी साहित्य की रचनाओं	की सम	कालीन प्रमुख	रचनाओं से	
तुलनात्मक अध्य		0			
• हिन्दा साहित्य व करना।	के गद्य पद्य शैली द्वारा विभिन्न	। विकार	गत्मक प्रवृतिया	का जानकरा	
	े प्रमुख रचनाओं एवं उनके काल	में की ज	गवकारी कावा।		
	0				
• हिन्दी साहित्य के व्याकरण एवं भाषागत विकास की समझ विकसित करना।					
 हिन्दी साहित्य के विभिन्न पारिभाषिक शब्दों एवं प्रारूप से परिचय करना। जिन्दी भाषा परं साहित्य के परि सरकार अधिकर्ति परं दरियों कर विस्तय करना। 					
• हिन्दी भाषा एवं साहित्य के प्रति सकारात्मक अभिरूचि एवं वृतियों का विकास करना।					
 हिन्दी भाषा तत्वों एवं साहित्य के विविध रूपों का अध्ययन करवाना। िन्दी साहित्य परं विन्ती अपूर्ण की विविध विषयों का अध्ययन करवाना। 					
 हिन्दी साहित्य एवं हिन्दी भाषा की विविध विधाओं का ज्ञान करवाना। अधिगम सम्प्राप्तियाँ : 					
 विद्यार्थी हिन्दी भाषा तत्वों एवं साहित्य के विविध रूपों को जान सकेंगे। 					
 विद्यार्थी गहन्दा भाषा तत्वा एव साहत्य क विविध रुपा को जान सकगे। विद्यार्थी गद्य साहित्य एवं हिन्दी भाषा की विविध विधाओं की व्याख्या कर सकेंगे। 					
 विद्यार्थी पाठ्य प्रकरण के अन्तर्गत आयी किसी कहानी एवं नाटक का रूपान्तरण करवाने में रुचि ले सकेंगे। 					
 हिन्दी साहित्य के प्रमुख कवियों एवं रचनाकारों की विस्तृत जानकरी प्राप्त कर सकेंगे। 					
• विद्यार्थी हिन्दी साहित्य की भाषा के साथ भारतीय सम्यता एवं संस्कृति को जान					
संकेंगे।					
• आदिकाल की हिन्दी साहित्य की रचनाओं की समकालीन प्रमुख रचनाओं से					
तुलनात्मक अध्ययन प्राप्त करेंगे।					
• हिन्दी साहित्य के गद्य-पद्य शैली द्वारा विभिन्न विकासात्मक प्रवृतियों की जानकरी					
प्राप्त कर सकेंगे।					
	रु प्रमुख रचनाओं एवं उनके कालों की जानकारी प्राप्त कर सकेंगे।				
	हिन्दी साहित्य के व्याकरण एवं भाषागत विकास को समझ सकेंगे।				
 विद्यार्थी हिन्दी साहित्य के विभिन्न पारिभाषिक शब्दों एवं प्रारूप से परिचित हो सकेंगे। 					
_	म एवं साहित्य के प्रति सकारात्मक अभिरूचि एवं वृतियों का विकास हो				
सकेगा ।					

साहित्य खण्ड (पद्य भाग) साहित्य खण्ड (पद्य भाग) अयोध्या रिंह उपाध्याय 'हरिऔध' – कर्मवीर कर्मवीर अयेध्या रिंह उपाध्याय 'हरिऔध' – कर्मवीर भीवेतीशरण गुप्त – भूतोक का गौरव (भारत-भारती) स्यंकाल त्रिपाठी 'जिराला' – तोइती पत्थर जयाशंकर प्रसाद – अरुण यह मधुमय देशहमारा स्युभद्रा कुमारी चौहान – झाँसी की रानी हरियंशया बच्चन – पथ की पहचान स्यात्रिय खग्ड - यरकी पहचान स्वाती प्रसाद किख्यना हरियंशया बच्चन – पथ की पहचान स्वाती प्रसाद किख्यन - अरफरोश स्वाती प्रसाद तिमश्र - सरफरोश स्वाती प्रसाद (तत्सम, तद्भव, देशज, विदेशी) पारिशाधिक शब्दावली भारिशाधिक शब्दावली शब्द वीचार (तत्सम, तद्भव, देशज, विदेशी) पारिशाधिक शब्दावली शब्द गुद्धिकरण पारिशाधिक शब्दावली शब्द शुद्धिकरण शब्द शुद्धिकरण (ख) वाक्य के अंज (ख) वाक्य के अंज (ख) वाक्य के अंज (ख) वाक्य के प्रकार संक्षेपण पत्ल्वव (कोकोक्ति, मुहावरे पत्त प्राठप् रिंग साल प्राठप् राक्य एप्रच्या (किवव्य पर पाँच जिबव्य पूछे जायेंगे, जिनमें		<u>साहित्य खण्ड (गद्य भाग)</u>			
• नता नहा नागारक चाहए - रामधारा सिंह दिवेक साहित्य खण्ड (पद्य भाग) • अयोध्या सिंह उपाध्याय 'हरिऔध' - कर्मवीर • भीथेलीशरण गुप्त - कर्मवीर • भारत-भारती) • अयोध्या सिंह उपाध्याय 'हरिऔध' - कर्मवीर • ग्रेतका का गौरव (भारत-भारती) • युर्य्ववान्त त्रिपाठी 'निराला' - तोइती पत्थर • जयशंकर प्रसाद - अरुण यह मधुमय देशहमारा • उपशंकर प्रसाद - अरुण यह मधुमय देशहमारा • हरियंशराय बच्चन - • युर्ग युंका कुमारी चौहान - झाँसी की रानी • हरियंशराय बच्चन - पय की पहचान • वार्ग विचार • याकरण खण्ड • मवानी प्रसाद मिश्र - • मवानी प्रसाद मिश्र - • मवानी प्रसाद मिश्र - • याकरण खण्ड • • याकरण खण्ड • • याकरा खण्ड • • याकरा खण्ड • • याति प्रात • • याकरा ता प्रात • • याकरा का खावरा ता ता सां कि याना • • याकरा ता ता ति कि याना • • याकरा का शब्दकोण का आवत • • याकरा का खावरा ता ता कि का ता	(24)	• भारतनर्ष की रत्नति कैसे हो	_	भारतेन्द	
• नता नहा नागारक चाहए - रामधारा सिंह दिवेक साहित्य खण्ड (पद्य भाग) • अयोध्या सिंह उपाध्याय 'हरिऔध' - कर्मवीर • भीथेलीशरण गुप्त - कर्मवीर • भारत-भारती) • अयोध्या सिंह उपाध्याय 'हरिऔध' - कर्मवीर • ग्रेतका का गौरव (भारत-भारती) • युर्य्ववान्त त्रिपाठी 'निराला' - तोइती पत्थर • जयशंकर प्रसाद - अरुण यह मधुमय देशहमारा • उपशंकर प्रसाद - अरुण यह मधुमय देशहमारा • हरियंशराय बच्चन - • युर्ग युंका कुमारी चौहान - झाँसी की रानी • हरियंशराय बच्चन - पय की पहचान • वार्ग विचार • याकरण खण्ड • मवानी प्रसाद मिश्र - • मवानी प्रसाद मिश्र - • मवानी प्रसाद मिश्र - • याकरण खण्ड • • याकरण खण्ड • • याकरा खण्ड • • याकरा खण्ड • • याति प्रात • • याकरा ता प्रात • • याकरा का खावरा ता ता सां कि याना • • याकरा ता ता ति कि याना • • याकरा का शब्दकोण का आवत • • याकरा का खावरा ता ता कि का ता	SS		_	9	
• नता नहा नागारक चाहए - रामधारा सिंह दिवेक साहित्य खण्ड (पद्य भाग) • अयोध्या सिंह उपाध्याय 'हरिऔध' - कर्मवीर • भीथेलीशरण गुप्त - कर्मवीर • भारत-भारती) • अयोध्या सिंह उपाध्याय 'हरिऔध' - कर्मवीर • ग्रेतका का गौरव (भारत-भारती) • युर्य्ववान्त त्रिपाठी 'निराला' - तोइती पत्थर • जयशंकर प्रसाद - अरुण यह मधुमय देशहमारा • उपशंकर प्रसाद - अरुण यह मधुमय देशहमारा • हरियंशराय बच्चन - • युर्ग युंका कुमारी चौहान - झाँसी की रानी • हरियंशराय बच्चन - पय की पहचान • वार्ग विचार • याकरण खण्ड • मवानी प्रसाद मिश्र - • मवानी प्रसाद मिश्र - • मवानी प्रसाद मिश्र - • याकरण खण्ड • • याकरण खण्ड • • याकरा खण्ड • • याकरा खण्ड • • याति प्रात • • याकरा ता प्रात • • याकरा का खावरा ता ता सां कि याना • • याकरा ता ता ति कि याना • • याकरा का शब्दकोण का आवत • • याकरा का खावरा ता ता कि का ता			_		
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 भवानी प्रसाद मिश्र – सरफरोश व्याकरण खण्ड वर्ण विचार वर्ण विचार शब्द विचार (तत्सम, तद्भव, देशज, विदेशी) पारिभाषिक शब्दावली शब्दों को शब्दकोष क्रम में लिखना। युग्म शब्द, उपसर्ग, प्रत्यय, संधि, समास संज्ञा, सर्वनाम, क्रिया, विशेषण, लिंग, वचन काल रात्रा खण्ड – अ शुद्धिकरण (क) शब्द शुद्धिकरण वाक्य रचना (ख) वाक्य के अंग (ख) वाक्य के प्रकार संक्षेपण पल्लवन लोकोक्ति, मुहावरे पत्र (प्रारूप) 	NII See			<u> </u>	
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 भवानी प्रसाद मिश्र – सरफरोश व्याकरण खण्ड वर्ण विचार वर्ण विचार शब्द विचार (तत्सम, तद्भव, देशज, विदेशी) पारिभाषिक शब्दावली शब्दों को शब्दकोष क्रम में लिखना। युग्म शब्द, उपसर्ग, प्रत्यय, संधि, समास संज्ञा, सर्वनाम, क्रिया, विशेषण, लिंग, वचन काल रात्रा खण्ड – अ शुद्धिकरण (क) शब्द शुद्धिकरण वाक्य रचना (ख) वाक्य के अंग (ख) वाक्य के प्रकार संक्षेपण पल्लवन लोकोक्ति, मुहावरे पत्र (प्रारूप) 	'EA		-		
व्याकरण खण्ड • वर्ण विचार • शब्द विचार (तत्सम, तद्भव, देशज, विदेशी) • पारिभाषिक शब्दावली • शब्दों को शब्दकोष क्रम में लिखना। • युग्म शब्द, उपसर्ग, प्रत्यय, संधि, समास • संज्ञा, सर्वनाम, क्रिया, विशेषण, लिंग, वचन काल रचना खण्ड - अ • शुद्धिकरण (क) शब्द शुद्धिकरण (क) शब्द शुद्धिकरण • वाक्य रचना (क) वाक्य के अंग (ख) वाक्य के प्रकार • संक्षेपण • संक्षेपण • पल्लवन • लोकोकित, मुहावरे • पत्र (प्रारुप)	E	2	_		
• वर्ण विचार • शब्द विचार (तत्सम, तद्भव, देशज, विदेशी) • पारिभाषिक शब्दावली • शब्दों को शब्दकोष क्रम में लिखना। • राठम शब्द, उपसर्ग, प्रत्यय, संधि, समास • रांजा, सर्वनाम, क्रिया, विशेषण, लिंग, वचन काल रचना खण्ड – अ • शुद्धिकरण (क) शब्द शुद्धिकरण (क) शब्द शुद्धिकरण (ख) वाक्य शुद्धिकरण • वाक्य रचना (क) वाक्य के अंग (ख) वाक्य के प्रकार • संक्षेपण • पल्लवन • लोकोक्ति, मुहावरे • पत्र (प्रारूप)				संस्थरम्	
रचना खण्ड - अ शुद्धिकरण (क) शब्द शुद्धिकरण (ख) वाक्य शुद्धिकरण • वाक्य रचना (क) वाक्य के अंग (ख) वाक्य के अंग (ख) वाक्य के प्रकार • संक्षेपण • पल्लवन • लोकोक्ति, मुहावरे • पत्र (प्रारूप)	इकाई – 3 TEACHING HOURS (24	 शब्द विचार (तत्सम, तद्भव, देशज, विदेशी) पारिभाषिक शब्दावली शब्दों को शब्दकोष क्रम में लिखना। युग्म शब्द, उपसर्ग, प्रत्यय, संधि, समास 			
(क) शब्द शुद्धिकरण (ख) वाक्य शुद्धिकरण • वाक्य रचना (क) वाक्य के अंग (ख) वाक्य के प्रकार • संक्षेपण • पल्लवन • लोकोक्ति, मुहावरे • पत्र (प्रारूप)			.,		
श्वमा खण्ड – ब भग मा	इकाई –4 TEACHING HOURS (2	 शुद्धिकरण (क) शब्द शुद्धिकरण (ख) वाक्य शुद्धिकरण वाक्य रचना (क) वाक्य के अंग (ख) वाक्य के प्रकार संक्षेपण पल्लवन लोकोक्ति, मुहावरे पत्र (प्रारूप) 			
• निबन्ध (किसी भी विषय पर पाँच निबन्ध पूछे जायेंगे, जिनमें । पि १ मे दे एक करना होगा।)	IRS	रचना खण्ड – ब			
िम एक करना होगा।)		• निबन्ध (किसी भी विषय पर पाँच निबन्ध पूछे जायेंगे, जिनमें से			
10 × 2	, -5 H (एक करना होगा।)			
• भाषा सौंदर्य (भाव सौंदर्य, विचार सौंदर्य, नाद सौंदर्य, शिट सौंदर्य)	इकाई EACHING (22)	·	सौंदय	र्ग, नाद सौंदर्य, शिल्प	

Teaching and	• Lecture methods
LEARNING	 Presentation (PPT)
METHODS	
	Groups discus methods
	Seminal/Symposia methods
	Extinction activate methods
	Project and report writing
CONTINUOUS	Academic presentence in person
ASSESSMENT	Seminal/Symposia
METHODS	Extinction activate
	Vive voce
	• Monthly test
EXAMINATION	Term-end examinations are organized by the university in the prescribed
PATTERN	format to enable the scholars to achieve success in contemporary
	competitions and to achieve their goals.
PERIODICAL	1. Annual
REVISE OF	2. However the University may revise the syllabus at any time during
SYLLABUS	the running year after giving a notice for a period one months.
SELECTED	• शर्मा, राजकुमारी (२००६). हिन्दी शिक्षण. राधा प्रकाशन मन्दिर.
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	 शर्मा, अनुराधा (२०१२). भाषा विज्ञान तथा सिद्धान्त. विश्व भारती पब्लिकेशनः नई दिल्ली.
	 गुप्ता, प्रभा (२०१२). मातृभाषा व विविध योजनाएँ. साहित्य प्रकाशनः आगरा.
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	 नारंग, वैश्ना (2013). सम्प्रेषणपरक हिन्दी भाषा शिक्षा. ए.पी. भार्गव बुक हाऊसः आगरा.
	 शर्मा, प्रसाद (2007). हिन्दी शिक्षण. साहित्यागार. धामाणी मार्केट की गली. चौड़ा रास्ताः जयपुर.
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	 प्रसाद, भगवती (2002). प्रारंभिक स्तर पर हिन्दी शिक्षण. सुखपाल गुप्त आर्य बुक डिपोः नई दिल्ली.
	• त्यागी, एस.के. (२००८). हिन्दी भाषा शिक्षण. अग्रवाल पब्लिकेशन्सः आगरा.

	B.Sc.B.Ed. II Yea	r				
COURSE CODE:	BSCBED-251	COURSE TYPE: CORE				
COURSE TITLE:	Knowledge and Curriculur					
MAX.MARKS:	100			40		
THEORY	70			28		
EXAMINATION	/0					
CONTINUOUS	30	MIN. PASS MARKS: 12				
COMPREHENSIVE	50	MIN. PASS MARKS: 12				
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RESPEC	PECTIVE YEAR				
ELIGIBILITY						
EXAMINATION	TERM END EXAMINA	TION	MONTHL	Y TEST		
DURATION	03 HR		01 H			
Objectives:						
-	Meaning and Concept of know	ledge				
	various dimensions of the curri	0				
	cept and the process of knowle		ation and their lin	kage with		
• To understand cond society.	cept and the process of knowle	uge genera		kage with		
•	nction between Data, knowled	ge and Inf	ormation			
	cally analyze various curriculu	•				
• To understand entry	carry anaryze various curriculu		к.			
 To identify various To identify relation To understand the r 	se various/ curriculum textbook. s dimensions of the curriculum and their relationship. nship between the curriculum framework and syllabus. relationship between ideology and the curriculum.					
	 <u>nowledge Generation</u> Meaning Concept and nat 	ure of kno	wledge.			
	• Distinction between Da		-	on and skill		
	Teaching and Training, k		-			
	belief.	thewreage	and mitorination	i, reason and		
(1				Charles a		
Ő	• Process of Knowledge	generatio	n: Local windo	ow, Sharing,		
RS	practice and creation.					
100	• Various structures of soci	ety and kn	owledge and their	r linkage and		
HC	relationship.					
Z						
Ass	ignment					
UNIT-1 TEACHING HOURS (24)	• Conduct a group discussion on Distinction between Data			tween Data.		
EA		nformation and write a report.				
E	•					
	issue/innovation/ Discovery such as aeroplane or Robot.			•		
		i y sucii ds	acroptane of Kot			

<u> </u>	Curriculum and origin of knowledge
(24	• Knowledge as the bases of curriculum.
RS	• Curriculum: Meaning, concept, nature component and its basis.
00	• Type of curriculum: Explicit, hidden, absent or null curriculum.
LI H	• Curriculum, Syllabus, content and co-curriculum for knowledge
	generation.
H	Assignment
AC	• Identified the transaction of curriculum in a prescribed text book.
UNIT-2 TEACHING HOURS(24)	• Organize a workshop on types of curriculum.
	Features of Curriculum Framework
(24	• The salient features of NCF- 2005 and NCFTE-2010 and analysis of
RS	these documents with respect to various aspects of foundations,
8 00	concerns and the changes made with important considerations.
UNIT-3 TEACHING HOURS (24)	Assignment
	• Make a comparative Analysis of curriculum of school at any one
	level in the light of NCF 2005.
AC	• Organize an orientation program for school teachers on NCF-2005
LE	and NCFTE-2010.
	Text book and curriculum
2	• Text book – Criteria of selection, and critical analysis of Text
5	Book, Children's literature and teacher's handbooks.
RS	• Relationship among curriculum, syllabus and textbook.
40	• Selection of materials; Development of activities and tasks.
H	• Connecting learning to the world outside - Moving away from
UNIT-4 TEACHING HOURS (22)	rote-learning to constructivism.
	Assignment
AC	• Critical review or analysis of the text book at upper primary and senior secondary level.
E	 Organize a seminar on relationship among power, ideology and
	Curriculum.
×.	Modernity, Post Modernity and Knowledge in Curriculum
	Meaning and Concept of knowledge in modernity
	Meaning and Concept of knowledge in Post-modernity
2 H 2 C	• Knowledge process tHRough curriculum transaction.
UNIT-5 FEACHING HOURS (22)	Assignment
CHC	 Make a report on the journey of knowledge from modernity to post-modernity.
EA	• The students will be asked to collect the facts regarding current
E	knowledge and then prepare report.
TEACHING AND	1. Lectures
LEARNING	2. E-learning
STRATEGIES	3. Videos
	 Extension Lectures Content Review
	6. Self-Learning
	7. Group Discussions
	8. Field Visit
	9. Survey

	10 Dc	oumantarias					
	10. Documentaries 11. Short Films						
		12. Team Teaching * The teaching strategies are subject to change as per requirement					
		lents and their capabilities.	as per requirement				
CONTINUOUS		Continuous and Comprehensive Ass	ossmont (CCA) are as				
&COMPREHENSIV	follows:	Continuous and Comprehensive Ass	essment (CCA) at c as				
E ASSESSMENT	10110 H 5+						
(CCA)	SR. NO.	CCA: COMPONENT	MAXIMUM MARKS				
	1						
		Monthly Test Presentation	10X6 Test = 60 10				
	$\frac{2}{3}$	Group Discussion	10				
	4	Debate	10				
	5	Participation and Presentation in Semi					
	6	Report Writing	<u>10</u>				
	7	Viva Voce	10				
	8	Attendance*	10				
	9	Co-curricular Activity	10				
	10	Team Teaching	10				
		NATION (METHOD TO ASCERTA)					
	CCA):	ATION (METHOD TO ASCENTA					
		l be reduced to 30 marks or 15 marks (a	as per course				
	weightag						
	0 0	Marks obtained/Total marksX30					
		Formula: Marks obtained/ rotal marks x_{50} For example: $60 \div 160 X_{30} = 11.25$					
		PROVISO-I: Provided that a candidate shall be granted a relaxation					
	in the form of exemption from CCA components, however, not more						
	than 3 in a respective course.						
	PROVISO-II: Provided further that this will be mandatory for a						
	candidate	e to appear in the monthly test conducte	•				
	course.						
	*Attenda	ance in Lectures and Practical					
		PercentageMarks Al75% to 80%02	lotted				
		75% to 80% 02 81% to 85% 04					
		81% to 85% 04 86% to 90% 06					
		91% to 95% 08					
		Above 96% 10					
EXAMINATION	Term and	examinations are organized by the univ	versity in the prescribed				
PATTERN		enable the scholars to achieve suc	v 1				
		ons and to achieve their goals.	cess in contemporary				
PERIODICAL	1. ANNU						
REVISION OF			vllabus at anv time				
SYLLABUS	2. However the Unviersitymay revise the syllabus at any time during the running year after giving a notice for a period one						
	month.		u periou one				
SELECTED		est, W.H. (2012), Curriculum: Pros	pective Paradigm and				
READINGS		ty. M.C MILLAN publication.	and and and and				
		ram mohan (2009). Curriculum Instru	uction and Evaluation				
		val publication, Agra.	action and Drundulon.				
	00	AV. (2009). The curriculum: Theo	ry and Practice sage				
	- iteliy,	Try, (2007). The currentum. Theo.	i j una i luctice, sage				

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• Shulman L. S. (1986) Those Who Understand: Knowledge Growth
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• Tagore, R. (2003) Civilization and Progress in Crisis in Civilization
and other essays: rupa &Co. New Delhi.
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publication Agra.
• श्रीवास्तव, एच.एस.एवं चतुर्वेदी, एम. जी (२०१०). पाठ्यचर्या और
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• यादव, शियाराम, (२०११) पाठ्यक्रम विकास अग्रवाल प्रकाशन
आगरा.
• पूनम मदन (२०१६). पाठ्यचर्या के सैद्धान्तिक आधार, अग्रवाल
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B.Sc.B.Ed. II Year					
COURSE CODE:	BSCBED-252 COURSE TYPE: CORE			PE: CORE	
COURSE TITLE:	Teaching and Learning				
MAX.MARKS:	100	MIN.	PAS	S MARKS:	40
THEORY	70	MIN.	PAS	S MARKS:	28
EXAMINATION					
CONTINUOUS	30	MIN.	PAS	S MARKS:	12
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR				
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATIONMONTHLY TEST			TEST	
DURATION	03 HR 01 HR				
Objectives:					
• To develop Cognitive Psychology, Social Development and its effect on Teaching &					

- To develop Cognitive Psychology, Social Development and its effect on Teaching & Learning.
- To develop the emotional Development and Spiritual Development of the students.
- To develop effective Teaching
- To teach about Learning & its factors influencing Learning etc.
- To acquaint with the Diversity among learners and learning needs
- To make aware about the Role of ICT in Learning Enhancement.
- To relate with Teaching learning processes, Teaching learning strategies and methods in the context of NCF.
- To acquaint with Action Research. Right to Education Act 2009.

Learning Outcomes: After completion of the course, students will be able to:

- To develop scientific attitude for the process of teaching & learning.
- To develop understanding about the relationship of cognitive, social and emotional development with learning process.
- To provide an overall view on teaching & learning style and ideas to enhance these activities.
- To make aware about teaching skills, components and parameters of effective teaching.
- To relate various Psychological domains of Teaching & Learning.
- To familiarize with ICT in teaching & learning process.

	Psychological Domains of Teaching & Learning
	• Cognitive Psychology: - Meaning, Concept, Important, domains and
Ŧ	its relationship with Learning & Teaching.
(5)	• Social Development – Meaning, Importance, Social Process and its
RS	effect on Teaching & Learning, Theory of Social Construction.
UNIT-1 NG HOURS (24)	• Emotional Development - Meaning, Process, need to study and its
HC.	effect on Teaching and Learning process.
Z J	• Spiritual Development: - Meaning, Concept, important domains and
	its relationship with Learning & Teaching.
HO	Assignment:
TEAC	• Make a critical analysis of school situation in terms of its role in promoting learner's cognitive and non-cognitive learning outcome and report on entire activities.
	• Prepare learners' profile based on cognitive and non cognitive
	characteristics of two adopted students.

UNIT-2 TEACHING HOURS (24)	 Effective Teaching Effective Teaching : Meaning, Component and Parameters of Effective Teaching, Identification of Teaching Skills, Principles of Teaching, Classroom Instruction Strategies, Teacher as a Learner, Responsibilities of Teacher. Teaching for Culturally Diverse Students, Theory of Culturally Relevant Pedagogy. Values and personal relationship between Teachers and Learners, relationship among learners, self–esteem and freedom experienced by learner. Teaching Models & factor effecting teaching & learning. Assignment: Draft a survey based report on an effective Teacher's behaviour . Prepare an academic schedule of students after a thorough analysis of school dairy of students. Focus on the analysis of record of Teaching and Learning.
UNIT-3 TEACHING HOURS (24)	 Learning Learning – Meaning, and Characteristics, Learning Factors, Influencing Factors, Kind of Learning, tradition and changes in view of the learning process a shift from teaching to Learning. Principles of Learning, Quality of Learning. Discovering Learning - Meaning, Concept, Principles to assess quality of Learning. A Learning Cycle for Discovery, Classroom Instruction Strategies Theories Supporting the new view of the Learning process. Learning as construction of Knowledge (NCERT, 2005) learning in and outside school and its relationship with learner's motivation learning in diverse socio-cultural condition. Assignment: Conduct a Study of a case and prepare a report on factors influencing learning of two adopted students. Conduct a test of at least two students to determine which side of the brain is more powerful.
UNIT-4 TEACHING HOURS (22)	 Learning Style Diversity among learners and learning needs (with reference to special needs) background & Concept of Multilingual. Learning & thinking Style - Concept, Types and importance in Teaching – Learning process, factor effecting and relationship between Learning & Thinking Style. Role of ICT in Learning Enhancement. Assignment: Prepare an academic record of two students justifying your contribution in academic enhancement of them. Observe differential learning needs of the learners with regard to learning styles tHRough active involvement of students (playing, storytelling, puzzle games, reading etc) and draft a report for presentation.

	Teaching Style					
	• Teaching learning processes, Teaching learning strategies and					
(53	methods in the context of National Curriculum Framework					
S.	2005.					
0 0	• Action Research, Right to Education Act 2009 (Role and Responsibilities of Teachers)					
UNIT-5 NG HO	Responsibilities of Teachers)					
	• Teaching as Profession, Effective Classroom Management.					
	• ICT & Teaching.					
	Assignment:					
UNIT-5 TEACHING HOURS (22)	• Draft a report on Teachers' Teaching Style for one week based					
F	on the Classroom observation of two teachers.					
	• Make your own lesson Plan by studying Teacher's diaries.					
	1. Lectures					
TEACHING AND	2. E-learning					
LEARNING	3. Videos					
STRATEGIES	4. Extension Lectures					
	5. Content Review					
	6. Self-Learning					
	7. Group Discussions					
	8. Field Visit					
	9. Survey					
	10. Documentaries					
	11. Short Films					
	12. Team Teaching					
	* The teaching strategies are subject to change as per requirement					
	of the students and their capabilities.					
	L					

CONTINUOUS &		s of Continuous and Comprehensive Assessme	nt (CCA) are as
COMPREHENSIVE	follow	s:	
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM
(CCA)	NO.		MARKS
	1	Monthly Test	10X6 Test =
			60
	2	Presentation	10
	3	Group Discussion	10
	4	Debate	10
	5	Participation and Presentation in Seminar	10
	6	Report Writing	10
	7	Viva Voce	10
	8	Attendance*	10
	9	Co-curricular Activity	10
	10	Team Teaching	10
	EXP	LANATION (METHOD TO ASCERTAIN MA	RKS FOR
	CCA		
		will be reduced to 30 marks or 15 marks (as per c	ourse
		ntage).	
	U U	ula: Marks obtained/Total marksX30	
		xample: $60 \div 160 \times 30 = 11.25$	
		VISO-I: Provided that a candidate shall be grante	d a relaxation
		form of exemption from CCA components, how	
		3 in a respective course.	
		VISO-II: Provided further that this will be manda	tory for a
		date to appear in the monthly test conducted in the	
	cours		- Inspective
		endance in Lectures and Practical	
		Percentage Marks Allotted	
		75% to 80% 02	
		81% to 85% 04	
		86% to 90% 06	
		91% to 95% 08	
		Above 96% 10	
EXAMINATION	Term-	end examinations are organized by the university	in the prescribed
PATTERN		to enable the scholars to achieve success i	
		titions and to achieve their goals.	n contemporary
	compe	throns and to demote their gould.	
PERIODICAL	1	ANNUAL	
REVISION OF		HOWEVER, THE UNIVERSITY may revise t	he syllabus at
SYLLABUS		any time during the running year after giving	•
SILLADUS		period one month.	a notice for a
SHI HI I HII	<u></u>	*	प्रवोगित्याच के
SELECTED READING		बे.एस.पी. (२००५). बाल विकास व व	
SELECTED READING	मूल	बे.एस.पी. (२००५). बाल विकास व ब न तत्व.Concept Publishing Company Priva	
	मूल Gai	बे.एस.पी. (२००५). बाल विकास व व न तत्व.Concept Publishing Company Priva rden: New Delhi.	te Ltd. Mahan
	मूट Gai • भूष्	बे.एस.पी. (२००५). बाल विकास व व न तत्व.Concept Publishing Company Priva rden: New Delhi. 1ण,शैलेन्द्र (२००७–०८).शैक्षिक तकनीव	te Ltd. Mahan
	मूट Gai • भूष पवि	बे.एस.पी. (२००५). बाल विकास व व तत्व.Concept Publishing Company Priva rden: New Delhi. ाण,शैलेन्द्र (२००७–०८).शैक्षिक तकनीव देलकेशनः आगरा–७.	te Ltd. Mahan की. अग्रवाल
	मूल Gai • भूष पवि • शम	बे.एस.पी. (२००५). बाल विकास व व न तत्व.Concept Publishing Company Priva rden: New Delhi. ाण,शैलेन्द्र (२००७–०८).शैक्षिक तकनीव लकेशनः आगरा–७. र्ता, आर.ए. (२००८). शिक्षा के मनोवि	te Ltd. Mahan की. अग्रवाल
	मूल Gai • भूष पवि • शम	बे.एस.पी. (२००५). बाल विकास व व तत्व.Concept Publishing Company Priva rden: New Delhi. ाण,शैलेन्द्र (२००७–०८).शैक्षिक तकनीव देलकेशनः आगरा–७.	te Ltd. Mahan की. अग्रवाल

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	र्तन आर. लाल बुक डिपोः मेरठ.
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lea	rner and teaching learning process. R.lal Bookdepot: Meerut.
	•

	B.Sc.B.Ed. II Y	'ear		
COURSE CODE:	BSCBED-253		COURSE T	YPE: CORE
COURSE TITLE:	Health and Physical Edu	cation		
MAX.MARKS:	50	MIN.	PASS MARKS:	20
THEORY	35	MIN.	PASS MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN.	PASS MARKS:	6
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPE	CTIV	E YEAR	
ELIGIBILITY				
EXAMINATION	TERM END EXAMINA	ΓΙΟΝ		ILY TEST
DURATION Objectives:	03 HR		01	HR
 adolescents. To get acquainted y fitness, & their Fun To know the physic To learn about Sa prevention etc. To familiarize with To understand and institutions. 	eal fitness, strength, enduran afety and Security — dis various Games and Sports a learn various Yogic practio	Muscu ce and asters and thei ces, hea	lar, Digestive in self-defense activ in and outside s r impact on health lth services, poli	relation to Health ities. schools, ways of h cies and major of
• To develop positive to achieve it.	completion of the course , as e attitude towards health as th status, identify health pro-	individ blems a	ual and be collec	tively responsible or taking remedial

- To generate awareness about rules of safety in hazardous situation (illness, accident and injury) and equip them with first aid measures about common sickness and injuries.
- To learn and to form right habits about exercise, games and sports, sleep, rest and relaxation.
- To sensitize, motivate and help them to acquire the skills for physical fitness, learn correct postural habits and activities for its development.
- To understand various policies and programmes related to health, physical education and Yoga.
- To understand the process of assessment of health and physical fitness.

	• Concept of Health, Importance, Dimensions and Determinants of
	Health; Health needs of children and adolescents, including differently-abled children.
	• Physical Education- Meaning, concept and importance.
(12) NG	• Health and Physical Education and its relationship with other
	subject areas like Science, Social Science and Languages.
UNIT- TEACHI HOURS (Assignment:
	• Analysis various text book in light of health education, prepare report.
	• Prepare an awareness program on healthy life style of children or
	adolescence.

UNIT-2 TEACHING HOURS (12)	 The body system-Skeleton, Muscular, Respiratory, Circulatory and Digestive in relation to Health fitness, Bones, Muscles and Joints, their Functions. Food and Nutrition, Food habits, Timing of food, Nutrients and their functions, diversity of Indian food, seasonal foods and festivals, economics of food, preservation of food value during cooking, indigenous and modern ways to persevere food, shift in food practices food and waterborne and deficiency diseases and prevention. Assignment: Ask school student to prepare a first aid box and conduct a training program on first aid treatment of injury. Conduct an awareness program in community on preservation of food value during cooking.
UNIT-3 TEACHING HOURS (12)	 Safety and Security — disasters in and outside schools, ways of prevention, safety from snake and dog bites, animal attacks, prevention and treatment. Physical fitness, strength, endurance and flexibility, its components, sports skills, indigenous and self-defense activities. Assignment: Organize a Training program for girl students of school on self-defense activities. Organize a workshop on prevention of disasters in school.
UNIT-4 TEACHING HOURS (11)	 Games and Sports — athletics (general physical fitness exercises), games (lead-up games, relays and major games) rhythmic activities, gymnastics and their impact on health. Fundamental skills of games and sports; Sports for recreation and competition; Rules and regulations of sports; sports, ethics; sports awards and scholarships, sports-person ship. Assignment: Organization of inter school games and sports tournaments in your district/village/town. Orientation program on Fundamental Sports Skills : Basics of track and field (100 mts. 200 mts., long jump, shot-put, 4 × 50 mts. Relay) Gymnastics.
UNIT-5 TEACHING HOURS (11)	 Yogic practices—importance of yoga, yogasanas, kriyas and pranayams of (school, family and sports), health services, policies and major of institutions. Assignment: Learning and performing of basic yogic activities. Conduct yoga activities for development of physical fitness, i.e strength, speed, endurance, flexibility and body composition of students in school.
TUTORIALS	One tutorial class once a week (12)
TEACHING AND LEARNING STRATEGIES	 Lectures E-learning Videos Extension Lectures Content Review

	<u> </u>	alf Lagming		
		elf-Learning		
		roup Discussions ield Visit		
		urvey ocumentaries		
		hort Films		
*		eam Teaching	.	
		aching strategies are subject t	o change as per	requirement
		idents and their capabilities.		
	ollows:	f Continuous and Comprehen	isive Assessmen	at (CCA) are as
ASSESSMENT				
(CCA)	SR.	CCA: COMPONENT		MAXIMUM
	NO.			MARKS
	1	Monthly Test		10X6 Test =
				60
	2	Presentation		10
	3	Group Discussion		10
	4	Debate		10
	5	Participation and Presentation	on in Seminar	10
	6	Report Writing		10
	7	Viva Voce		10
	8	Attendance*		10
	9	Co-curricular Activity		10
	10	Team Teaching NATION (METHOD TO AS		10
	weighta Formula For exa PROVI in the fo than 3 i PROVI candida course.	a: Marks obtained/Total marks mple: 60÷160X30 =11.25 SO-I: Provided that a candidate orm of exemption from CCA co in a respective course. SO-II: Provided further that th te to appear in the monthly test lance in Lectures and Practic	(30 e shall be granted mponents, howe is will be mandat conducted in the	l a relaxation ver, not more cory for a
EXAMINATION T	erm_en	d examinations are organized by		n the prescribed
		o enable the scholars to ach		
		ions and to achieve their goals.		i contemporary
	. ANNI			
DEFECTION OF			nov rovice the	wllabug at any
SYLLABUS		EVER, THE UNIVERSITY	•	• •
		niring the running year ofte	r awina a notic	
	one m	luring the running year afte	i giving a nour	e for a period

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	B.Sc. B.Ed. II Y	ear		
COURSE CODE:	BSCBED-254 a I		COURSE TY	PE : CORE
COURSE TITLE :	Paper-I: Electricity and M			
MAX.MARKS:	50	MIN.PASS M	IARKS	20
THEORY	40	MIN.PASS M	IARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PASS M	IARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPEC	TIVE YEAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMINAT	ION		LY TEST
DURATION	03 HR		01	HR
Objective:				
• •	sm is one of the core courses	•		
-	tric and magnetic field, and	the electric and	magnetic fie	ld in matter.It
also includes analysis of		4		
_	er completion of the course, s			0-
	tual framework of electromag	-		
5	inding the difference between	-	ernating current	nt.
	pts of force on a moving char of Magnetic Fields of Matter		_	
• Apply their Knowledge	of Magnetic Fields of Matter			
	Electrostatics: Electrostat			-
	distributions of charges at	-		
$\widehat{\mathbf{e}}$	done on a charge in an ele	ctrostatic field,	conservative	nature of the
S(1	electrostatic field, Electric	potential, rela	tion between	electric field
UNIT-1 HING HOURS(13)	and electric potential, potential,	ential energy d	of a system, "	Forque on a
-10	dipole in a uniform field	and non-unif	orm electric	field and its
	energy, Potential and field	l due to dipole	, flux of the	electric field,
UNIT-I UNG HC	Gauss's law and its applic	ation for findir	ng E for sym	metric charge
	distributions, Poisson's	equation, Lap	lace equatio	n, boundary
TEAC	conditions and uniqueness t		1	, ,
E	Electric field around c		luced charge	s field and
	potential inside a conduct		e	
	method of images	or, neid near	the surface of	or conductor,
	Dielectrics: Electric field	in motton Di	lootnio modi	Im Enco and
		· · · · · ·		·
7.0	Bound Charges, dielectric	-		-
UNIT-2 TEACHIGN HOURS(10)	vector, Capacitor (parallel		•	,
IT. HIS RS	dielectric, displacement ve			
	Electric field due to a Diele	ectric Sphere pl	laced in a Uni	iform Electric
	Field, Gauss Law in dielec	tric medium, C	laussius-Moss	sotti Relation,
	boundary conditions satisfie	ed by E and D a	at the interface	e between two
	homogenous dielectrics.			
	- C			

	Current I	Electricity: Electric Currents (stead	dy and alternating).	
UNIT-3 TEACHING HOURS(12)		sity (J), non-steady currents and	•	
RS		s laws and analysis of multiloop circ	• •	
2 00		in LR and CR circuits, decay co	-	
UNIT-3 ING HO		-		
N US		its, AC circuits, complex numbers a		
	-	C circuit problems, complex imped		
G	LCRseries	and parallel resonance, Q factor,	power consumed by	
EA	an AC circ	uit, power factor, Y and delta network	s and transmission of	
E	electric pov	ver.		
<u> </u>	Magnetost	atics: Force on a moving charge: Lo	rentz force, equation	
UNIT-4 TEACHING HOURS(15)	-	ion of B, force on a straight conductor	-	
RS		agnetic field, Torque on a current lo		
40		ingular momentum and gyromagnet		
UNIT-4 ING HO		rticles in electric and magnetic fields.		
		-		
		ng field, CRO sensitivity. Transverse		
		energy determination of nuclear par		
LE ∕	•	Mutually perpendicular E and B fields	s, mass spectrograph,	
	velocity sel	ector, its resolution.		
	Magnetic	Fields in Matter:Biot-Savart lav	w and it's simple	
UNIT-5 TEACHING HOURS(10)	application	ns: straight wire and circular loop,	calculation of H in	
SS	simple geometrical situations, Ampere's Law and it's simple			
	application	ns: (1) Solenoid and (2) Toroid,	Ampere's law in	
HO HO	differential and integral form. Magnetic vector potential and			
UNIT-5 ING HO		equation, the divergence and curl of	-	
		lipole, magnetization current, magnet		
CE	-	intensity (H), magnetic permeabi		
EA	-	on of a bar magnet as a surface distr	•	
E	-	-	ibution of solenoidal	
		field of a magnetized object.		
TEACHING AND		ture method		
LEARNING STRATEGIES		blem Solving method phical method		
SIKAILGILS		ninar/Symposia		
		view of literature		
		ort writing		
	-	oup Discussion		
	8. Vid	eos/Animation		
		f-Learning/e-Learning		
		rkshops/Experiments.		
		CHING AND LEARNING STRAT	•	
CONTINUOUS	-	per requirement of the students and		
CONTINUOUS		Continuous and Comprehensive As	sessment (CCA) are	
COMPREHENSIVE	as follows: S. No.		Mor Morto	
ASSESSMENT (CCA)	5. INO.	CCA- Components	Max. Marks Allocation	
	1.	Monthly test	Allocation 20*3 Test=60	
	2.	Monthly test Quizzes and Assignments	20^{+3} Test=60	
	3.	Viva-voce	10	
	5.	1110 1000	10	

	4.	Seminar/Symposia	10
	5.	Report writing	10
	6.	Workshop	10
	7.	Review of literature	10
	8.	Creativity/Innovation	10
	9.	Experimental Skill	10
	10.	Co-curricular activity	10
	11.	Attendance	10
	Total 160 r	narks equivalent reduced to CCA orig	inal marks 30.
EXAMINATION		examinations are organized by th	
PATTERN	prescribed	format to enable the scholars to	achieve success in
	contempora	ary competitions and to achieve their g	goals.
PERIODICAL	1. ANNUA	AL	
REVISE OF	2. Howeve	er the Unviersitymay revise the sy	llabus at any time
SYLLABUS	during	the running year after giving a not	ice for a period one
	months	•	
SELECTED READINGS	 electrod Electric Hallida Mahaja magnet Portis, Wiley. Pugh, magnet Electric Purcell, Magnet Resnik, 2010. 	 h, D. J., & Ruppeiner, G. (198 dynamics.Prentice-Hall of India). bity and Magnetism (Addison-Wesley) by and Resnik (2010). Physics, Vol-2 (un, A. S., & Rangwala, A. A. (20) bism. Tata McGraw-Hill Education. A. M. (1978). Electromagnetic fields E. M. (1960). Principles of ism.Addison-Welsley). Panofsky and city and Magnetism (India Book House) bism. McGraw-Hill. c) H. Krane (2010). The Nature of Light c) R. Halliday D. & Krane, K. S. (199 	Reitz and Milford;). Wiley Eastern) 001). Electricity and : sources and media. of electricity and d Phillips; Classical e). purse: Electricity and ht" in Physics Vol II,

	B.Sc. B.Ed. II Yea	r		
COURSE CODE:	BSCBED-254 a II	-	COURSE T	YPE : CORE
COURSE TITLE :	Paper-II: Kinetic Theory and	d Thermo	dynamics	
MAX.MARKS:	50		SS MARKS	20
THEORY	40	MIN.PA	SS MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PASS MARKS 4		4
COMPREHENSIVE				
ASSESSMENT (CCA)	80 DEDCENT IN DESDECTION		•	
ATTENDANCE ELIGIBILITY	80 PERCENT IN RESPECTIV	E IEAK		
EXAMINATION	TERM END EXAMINATIO	N	MONTH	ILY TEST
DURATION	03 HR	11		HR
Objective:				
U	al with the relationship betwee	en the m	acroscopic pro	operties of the
	equilibrium. The students up		· ·	-
	's applications to various therm			
•	ive exposure to students about	•	•	-
-	in ideal gases, phase transitions		• •	-
*	er completion of the course, stud		U	
• Know the background	-			
• Get familiar with the la				
	•	.:	_	
	laws of thermodynamics in real	situation	s.	
• Learn the Classical the				
a	deal Gas: Kinetic model, dedu		•	*
	emperature, estimation of rms s	-		
5	stimate of the Avogadro numb			
Ţ [0] h	eat of monoatomic gas, ex			-
	Schaviour at low temperatures,		c expansion of	f an ideal gas,
	pplications to atmospheric physi			
	Real Gas: Van der Waals gas			
V V	Vaals forces, comparison with			
	onstants , gas and vapour, Joule er Waals gas, Joule coefficient ,			
Т	iquefaction of Gases: Boyle t			
		•		•
P RS	rinciple of regenerative cooling	and of ca	scade cooling,	liquefaction of
	ydrogen and helium, Refrigerati	on cycles	, meaning of ef	ficiency.
ΞĦ 1	ransport phenomena in Gase	s: Moleci	alar collisions	mean free path
				-
	nd collision cross sections, estin	mates of 1	molecular dian	heter and mean
D fi	ree path, Transport of ma	ass, mo	mentum and	energy and
ii ii	nterrelationship, dependence on	temperatu	re and pressure	е.
		racc	pressure	

			• • • •		
		s of Thermodynamics: The Zeroth la			
UNIT-3 TEACHING HOURS(15)	diagrams,	work done by and on the sys	stem, first law of		
KS S	thermodyn	namics, internal energy as a state func	tion, reversible and		
5	irreversit	ble processes, Carnot cycle and its efficient	ency, Carnot theorem		
HO HO		cond law of thermodynamics, Different v	•		
		-			
UNIT-3 ING HO	-	tical cycles used in internal combustio	-		
H		ngines), Entropy, principle of increa			
AC	•	namic scale of temperature, its identity			
E	scale, Imp	possibility of attaining the absolute zero to	emperature, third law		
	of thermo	dynamics.			
	Thermod	ynamic Relationships: Thermodynamic	variables: extensive		
		sive, Maxwell's general relationship			
ۍ ت		omson expansion and adiabatic cooling			
4 Ž Ž					
L H S		Waals gas, Clausius-Clapeyron equat	•		
UNIT-4 FEACHING HOURS(15	potentials	and equilibrium of thermodynamical sy			
HEH	thermodyn	namical variables, Cooling du	e to adiabatic		
	demagnet	tization, production and measurem	ent of very low		
	temperatu	res.			
	Classical	Theory of Radiation: Properties of The	ermal Radiation. Pure		
5 C		re dependence, Kirchhoff's law, Stef			
in N N N	-	-			
		_			
		eigh-iean's law Planck's quantum post	Pressure of Radiation, Wien's displacement law, Wien's distributions		
	-	law, Rayleigh-jean's law, Planck's quantum postulates, Planck's law,			
UNIT-5 TEACHING HOURS(10)	complete	fit with experiment, interpretation of			
UN TEA HOU	complete				
E E E E E E E E E E E E E E E E E E E	complete heats of g	fit with experiment, interpretation of			
	complete heats of g	fit with experiment, interpretation of ases at low temperature.			
TEACHING AND	complete heats of g 1. Le 2. Pro 3. Gr	fit with experiment, interpretation of ases at low temperature. Acture method oblem Solving method aphical method			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se	fit with experiment, interpretation of ases at low temperature. cture method oblem Solving method aphical method minar/Symposia			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re	fit with experiment, interpretation of ases at low temperature . Acture method oblem Solving method raphical method minar/Symposia eview of literature			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re	fit with experiment, interpretation of ases at low temperature . Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr	fit with experiment, interpretation of ases at low temperature . Incture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi	fit with experiment, interpretation of ases at low temperature . Incture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se	fit with experiment, interpretation of ases at low temperature . Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning			
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo	fit with experiment, interpretation of ases at low temperature . Incture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning orkshops/Experiments.	behavior of specific		
TEACHING AND LEARNING	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. We * The TE	fit with experiment, interpretation of ases at low temperature . Incture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation lf-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE	behavior of specific GIES may be		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE, change as	fit with experiment, interpretation of ases at low temperature . Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation lf-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE s per requirement of the students and the	behavior of specific GIES may be heir capabilities.		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of	fit with experiment, interpretation of ases at low temperature . Incture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation lf-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE	behavior of specific GIES may be heir capabilities.		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of follows:	fit with experiment, interpretation of (ases at low temperature. ecture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation lf-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE (s per requirement of the students and the Continuous and Comprehensive Asses	GIES may be heir capabilities. ssment (CCA) are as		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of	fit with experiment, interpretation of ases at low temperature . Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation lf-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE s per requirement of the students and the	behavior of specific GIES may be heir capabilities. ssment (CCA) are as Max. Marks		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of follows: S. No.	fit with experiment, interpretation of gases at low temperature. Interpretation of gases at low temperature. Interpretation of the students and the column of the students and the continuous and Comprehensive Assession (CCA- Components)	GIES may be heir capabilities. sment (CCA) are as Max. Marks Allocation		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of follows: S. No.	fit with experiment, interpretation of gases at low temperature. Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation lf-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE sper requirement of the students and the Continuous and Comprehensive Assess CCA- Components Monthly test	GIES may be heir capabilities. sment (CCA) are as Max. Marks Allocation 20*3 Test=60		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of follows: S. No. 1. 2.	fit with experiment, interpretation of gases at low temperature. Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE sper requirement of the students and the CCA- Components Monthly test Quizzes and Assignments	GIES may be heir capabilities. ssment (CCA) are as Max. Marks Allocation 20*3 Test=60 10		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE change as Details of follows: S. No. 1. 2. 3.	fit with experiment, interpretation of gases at low temperature. Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE sper requirement of the students and the CCA- Components Monthly test Quizzes and Assignments Viva-voce	GIES may be heir capabilities. sment (CCA) are as Max. Marks Allocation 20*3 Test=60 10 10		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wa * The TE change as Details of follows: S. No. 1. 2. 3. 4.	fit with experiment, interpretation of gases at low temperature. Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE sper requirement of the students and the CCA- Components Monthly test Quizzes and Assignments Viva-voce Seminar/Symposia	GIES may be heir capabilities. sment (CCA) are as Max. Marks Allocation 20*3 Test=60 10 10 10		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wo * The TE, change as Details of follows: S. No. 1. 2. 3. 4. 5.	fit with experiment, interpretation of gases at low temperature. Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE per requirement of the students and the CCA- Components Monthly test Quizzes and Assignments Viva-voce Seminar/Symposia Report writing	GIES may be heir capabilities. ssment (CCA) are as Max. Marks Allocation 20*3 Test=60 10 10 10 10 10		
TEACHING AND LEARNING STRATEGIES	complete heats of g 1. Le 2. Pro 3. Gr 4. Se 5. Re 6. Re 7. Gr 8. Vi 9. Se 10. Wa * The TE change as Details of follows: S. No. 1. 2. 3. 4.	fit with experiment, interpretation of gases at low temperature. Acture method oblem Solving method raphical method minar/Symposia eview of literature eport writing roup Discussion deos/Animation If-Learning/e-Learning orkshops/Experiments. ACHING AND LEARNING STRATE sper requirement of the students and the CCA- Components Monthly test Quizzes and Assignments Viva-voce Seminar/Symposia	GIES may be heir capabilities. sment (CCA) are as Max. Marks Allocation 20*3 Test=60 10 10 10		

	8.	Creativity/Innovation	10
	9.	Experimental Skill	10
	10.	Co-curricular activity	10
	11.	Attendance	10
	Total 160	marks equivalent reduced to CCA origination	al marks 30.
EXAMINATION	Term-end	examinations are organized by the unive	rsity in the prescribed
PATTERN	format to	enable the scholars to achieve succ	ess in contemporary
	competitio	ons and to achieve their goals.	
PERIODICAL	1. ANNU	JAL	
REVISE OF		ver, the Universitymay revise the sy	•
SYLLABUS		g the running year after giving a not	ice for a period one
	month	IS.	
SELECTED		one, S. (2007). Thermodynamics for chem	
READINGS	• Hoyst, R., & Poniewierski, A. (2012). Thermodynamics for chemists,		
	physicists and engineers. Springer Science & Business Media.		
	• K.S. Pitzer, L. Brewer (1961). Thermodynamcs (Mc Graw Hill Book		
	Company).		
	• Reif, F. (2009). Fundamentals of statistical and thermal physics.		
		and Press.	
		M., & Srivastava, B. N. (1935). Treatise	e on heat. The Indian
		(Publication) Pvt. Ltd. Allahabad.	
		l Glasston (2006). Thermodynamics for	Chemists (Affiliated
	East West Press Pvt. Ltd., New Delhi)		
	• Umezawa, H. (1995). Advanced field theory: Micro, macro, and		
	thermal physics. AIP.		
	• Zemansky, M. W., & Dittman, R. H. (1981). Heat and		
		odynamics (International Edition).	Mcgraw-Hill Book
	Compa	iny).	

B.Sc. B.Ed. II Year				
COURSE CODE:	BSCBED-254 a III COURSE TYPE			PE:
			CORE	12.
COURSE TITLE :	Paper-III: Optics and Lasers		con	
MAX.MARKS:	50	MIN PA	SS MARKS	20
THEORY	40		SS MARKS	16
EXAMINATION:	10	1,111,111,111		10
CONTINUOUS	10	MIN PA	SS MARKS	4
COMPREHENSIVE				·
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIV	E YEAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION	N	MONTHL	Y TEST
DURATION	03 HR		01 H	
Objective:				
 principles and application Learning Outcomes: After Gain knowledge on Classify the properti Apply the principle induction in real situ 	Physics. Further, the students is s of LASER in Industry, Science completion of the course, stud various theories of light. es of light like reflection, refracti es of Construction, Working plations. ce of a light, optics, Polarization is	and Tech ent-teach on, interfa rinciple	nology. ters will be abl e erence, diffraction and Application	e to - on etc. ns of laser
(Geometrical Optics: Fermat's	Principl	e,laws of refl	ection and
FERCIPE ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY ITENDIFY 			Lagrange's mbinations, inal points, m, Cardinal berrations , ated lenses, points of a	
CUNCE Interference of a light: <i>Young's Double slit Experiment</i> , The principle of superposition, Relation for Fringe Width, Incoherent and Coherent Sources, Division of wavefront and division of amplitude, Fresnel's Biprism, Interference in thin film, Wedge-shaped film, Newton's rings, application of interference in determination of wavelength and precision measurements. Haidinger fringes: fringes of equal inclination, Michelson interferometer, its application for precision determination of wavelength, wavelength difference and the width of spectral lines, Intensity distribution in multiple beam interference, Fabry-Perot interferometer.				

		fraction: Fresnel half-period zo		
	plates, Circ	ular aperture, Circular disc, Dif	fraction at a straight	
	edge, Const	ruction and working principle of	f Zone plate and its	
UNIT-3 TEACHING HOURS(15)	application a		-	
	* *	diffraction: Phasor diagram, Diffr	action at a single slit.	
		& N parallel slits and their intensi	6	
I II E II		grating, reflection grating and bla	-	
H H				
		ifferent mountings, diffraction by		
		terion of Resolution, resolving	power of Telescope,	
	-	Grating and Prism.		
		and Optical Rotation: Meaning		
9		ht, Types of polarized light, Product		
	circularly p	polarized light, Brewster law,	Malus law, double	
UNIT-4 FEACHING HOURS(10)	refraction,	Phase retardation plates, Analysis	of Polarized light as	
		ed, circularly polarized and Ellipt	-	
HC HC		plane of polarization, Specif	• •	
		determination, Polarimeter (Lauren		
	_	Holography:Einstein's A a	-	
		and Induced emissions, condition		
	1 I		0 1 1	
		of Ruby laser , He-Ne lase		
UNIT-5 IEACHING HOURS(10)		tics of Laser beam, directionality,		
[O]		Focusing of laser beams. THRee-l	÷	
H H		Basic concepts of Holography,		
	Hologram and reconstruction of the image. Applications of			
	Holography.			
TEACHING AND	1. Lecture method			
LEARNING	2. Problem Solving method			
STRATEGIES	3. Graphical method			
	4. Seminar/Symposia			
	5. Review of literature			
	6. Report writing			
		p Discussion		
		os/Animation		
		Learning/e-Learning		
		shops/Experiments.		
		CHING AND LEARNING STRAT	FFGIFS may be	
		er requirement of the students and		
CONTINUOUS		ontinuous and Comprehensive A	_	
&COMPREHENSIVE	as follows:	ontinuous and Comprehensive A	ssessment (CCA) are	
			Mar Maula	
ASSESSMENT (CCA)	S. No.	CCA- Components	Max. Marks	
			Allocation	
	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3. Viva-voce 10			
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	0.		-	
	7.	Review of literature	10	
		Review of literature		
	7.	I I I I I I I I I I I I I I I I I I I	10	

	10.	Co-curricular activity	10	
	11.	Attendance	10	
	Total 160 marks equivalent reduced to CCA original marks 30.			
EXAMINATION	Term-end e	xaminations are organized by the	he university in the	
PATTERN	prescribed f	format to enable the scholars to	achieve success in	
	contemporar	y competitions and to achieve their	goals.	
PERIODICAL	1. ANNUA	L	-	
REVISE OF	2. However	, the Universitymay revise the s	syllabus at any time	
SYLLABUS		he running year after giving a no	•	
	months.		-	
SELECTED	• Akhmano	v, S. A., & Nikitin, S. Y. (19	97). Physical optics.	
READING	Clarendon Press.			
	• Bhadra, S., & Ghatak, A. (Eds.). (2013). Guided Wave Optics and			
	Photonic Devices. CRC Press.			
	• Boyd, R. W. (2019). Nonlinear optics. Academic press.			
	•	. (Ed.). (2012). Laser spectroscopy	-	
	Business I		, spinger serence ee	
	• De Fornel	, F. (2001). Evanescent waves: from	n Newtonian optics to	
	atomic op	tics (Vol. 73). Springer Science & B	Business Media.	
	• Hann, R. A	A., & Bloor, D. (Eds.). (1989). Orga	anic materials for non-	
	linear opti	cs. London: Royal Society of Chem	istry.	
	• Jenkins, F. A., & White, H. E. (1937). Fundamentals of optics. Tata			
	McGraw-Hill Education.			
	• Longhurst, R. S., (1970). Geometrical and physical optics. Orient			
	Black Swa	· · · · · · · · · · · · · · · · · · ·	- 1	
	• Wood, R.	W. (1905). Physical optics. Macmil	lan.	

PHYSICS PRACTICALS-II					
Duration: 4 HR	MAX.MARKS: 50	Min. Marks: 20			
Total number of exp	periments to be performe	ed by the students during the session should be 16,			
selecting any eight	from each section. In e	examination two experiments are to be performed			
taking at least one fr	om each section.				
The distribution of r	narks in the practical exa	mination will be as follows:			
(i) Two experiments		30 Marks			
For each experiment	, distribution of marks w	ill be as follows:			
Figure:		2			
Formula/Theory:		2			
Observation:		7			
Calculation and Res	ult:	3			
Precautions:		1			
(ii) Viva voce		10			
(iii) Records		10			
Total		50 Marks			
	LIST OF EXPERIMENTS				

Section A

- Determination of wavelength using Biprism.
- Determination of wavelength of light using diffraction grating.
- Michelson's interferometer and determination of D1& D2.
- To find out the wavelength of a monocHRomatic source of light using Newton's rings and find the refractive index of liquid.
- Determine the specific rotation of sugar solution by Polarimeter.
- Find out the melting point of a given substance using platinum resistance thermometer.
- Determine thermal conductivity of a bad conductor by Lee's method.
- Determine the thermodynamic constant (γ) using Clements's and Desormas method.
- Study of variation of total thermal radiation with temperature.
- Determination of Resolving Power of Telescope.
- Determine wavelength of laser light using a diffraction grating.
- Determination of wavelength of Mercury source.

Section B

- Study thermo emf versus temperature of a thermo couple. Find the neutral temperature and use it to find temperature of an unknown substance.
- Study the magnetic field along the axis of a current carrying circular coil for different currents. Plot the observations on a graph and find out the radius of a circular coil.
- Determination of band gap using a junction diode.
- Measurement of inductance of a coil by Anderson's bridge.
- Measurement of Capacitance and dielectric constant of a liquid by Desauty bridge.
- Study of decay of currents in LR and RC circuits.
- Response curve for LCR circuit and resonance frequency and quality factor.
- Determination of power factor ($\cos \theta$) of a given coil using CRO.
- Determination of velocity of sound in air by standing wave method using speaker microphone and CRO.
- Determine the Ballistic constant of a ballistic galvanometer.
- Determination of a small thickness by interference method.

B.Sc. B.Ed. 1I Year				
COURSE CODE: BSCBED-254 b I COURSE TYPE : CORE				
COURSE TITLE :		Paper-I : Inorganic Chemistry		
MAX.MARKS:	50	MIN.PASS M	ARKS	20
THEORY	40	MIN.PASS M		16
EXAMINATION:	+0		initio	10
CONTINUOUS	10	MIN.PASS M	ARKS	4
COMPREHENSIVE	10			·
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPEC	CTIVE SESSIC	DN	
ELIGIBILITY				
EXAMINATION	TERM END EXAMINA	TION	MO	NTHLY TEST
DURATION	03 HR			01 HR
Objectives:				
	e of this course is to tea	ch the Basic	concept	s of chemistry of
· ·	ls," F"Block elements, org		-	•
-	Ietal complexes and Oxidati	•	-	-
	fter completion of the course			
-	•			be able to -
-	und, properties of organo me	-		
	concepts of coordination co	mpounds, chen	nistry o	f lanthanide and ac
tinids.				
• Apply the principal of the principal o	oles of thermodynamics st	ability of met	al com	plexes and factors
affecting the stabil	ity.			
Ũ	of Metal complexes and Ox	idation Reduction	on	
	Coordination Compounds			
		and its experim	ontal ve	rification affective
	Verner's coordination theory and its experimental verification, effective			
	comic number concept, chelates, nomenclature of coordination			
UNIT-J HING H (12) (12)	ompounds, isomerism in coordination compounds, valence bond theory			
	f transition metal complexes. Limitations of valance bond theory, an			
	lementary idea of crystal field theory, thermodynamic aspects of			
	ystal field theory, crystal field splitting in octahedral, tetrahedral and			
S	quare planner complexes, fac	tors affecting th	ne crysta	al-field parameters
0	Chemistry of "F" Block elem	<u>ments</u> : Lantha	nide an	d Actinides
	hemistry of Lantha			and separation,
\smile	lectronicstructure, oxidation			1 '
	ontraction, spectral and mag			
0.7	oplications.	gnetie propertie	<i>s</i> , com	pick formation and
CHUNIT-2-TINU OH NƏL OH NƏL		1		· · · · · · · · · · · · · · · · · · ·
	hemistry of Actinides: E		-	
H	ctinide contraction, comp		-	-
q P	roperties, applications. Che	mistry of separ	ration o	of Np, Pu and Am
fi fi	from Uranium, similarities between the later actinides and 1			ctinides and later
	nthanides and applications of actinides.			
	rganometallic compounds			
		and classifica	ation	of organometallic
	ompounds, Preparation, prop			C
	nd aryls of Li, Mg, Fe, Sn	cities, conding	, and up	Productions of unkyls
	ווס מו אוס דו, זאוצ, ד'ד, אום וויס מו אויס וויס וויס מו			

		and Nitrosyls carbonyls: Metallic carbonyls General	methods of	
UNIT-4 TEACHING HOURS(13)		general properties, structure and nature of I		
SS(bonding in	carbonyls, Effective atomic number (EAN)	rules as applied	
5	to metallic	carbonyls. 18-electron rules applied to me	tallic carbonyls.	
UNIT-4 ING HO		properties and structure of nickel tetracarb	-	
	-	cHRomium hexa carbonyls, dimanganese	• •	
	dicobalt octa		,	
C		trosyls: Some metallic notrosyls: Metal nit	rosyl carbonyls,	
EA		syl halides, sodium nitroprusside (Prepara		
F		nd uses) structure and nature of M-N bond		
	Effective atomic number (EAN) rules as applied to metallic nitrosyls.			
		Metal complexes and Oxidation Reductio	•	
UNIT-5 IEACHING HOURS(13)		namic and Kinetic Aspects of Metal Com		
RS	-	nermodynamics stability of metal complex	-	
00		ne stability, stability constants of compl		
H H	-	on, substitution reactions of square planar con		
UNIT-5 ING HO		Reduction: Use of redox potential data-ar	-	
		stability in water – Frost, Latimer and Pou	•	
AC	-	volved in the extraction of the elements.	C	
EL	1			
TEACHING AND	1. Lect	ure method		
LEARNING	2. Problem Solving method			
STRATEGIES	3. Graphical method			
		 Seminar/Symposia Review of literature 		
		 Review of literature Report writing 		
	6. Report writing7. Group Discussion			
	8. Videos/Animation			
	9. Self-Learning/e-Learning			
		kshops/Experiments.		
	* The TEA	CHING AND LEARNING STRATEGIES	5 may be	
		er requirement of the students and their o	_	
CONTINUOUS COMPREHENSIVE	Details of C follows:	Continuous and Comprehensive Assessmen	nt (CCA) are as	
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)	5.110.	CCA- components	Allocation	
()	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature Creativity/Innovation	10	
	<u> </u>	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	

	Total 160 marks equivalent reduced to CCA original marks 30.		
EXAMINATION	Term-end examinations are organized by the university in the prescribed		
PATTERN	format to enable the scholars to achieve success in contemporary		
	competitions and to achieve their goals.		
PERIODICAL	1. ANNUAL		
REVISE OF	2. However, the Universitymay revise the syllabus at any time		
SYLLABUS	during the running session after giving a notice for a period one		
	month.		
SELECTED	• Mehrotra.R. C. and A. Singh (1991). organometallic Chemistry A		
READINGS	Unified Approach. Wiley.		
	• Shar.A. G. (2012).Inorganic Chemistry. Pearson.		
	 Bell and Lott (1995).Modern approach to Inorganic chemistry. Van 		
	Nostrand.		
	• Emelns and Anderson (1991). Principles of Inorganic Chemistry –		
	• Miessler.G. L. and D. A. (1962).Tarr Inorganic Chemistry. Prentice		
	Hall.		
	• Cotton and Wilkinson (2001).Advanced Inorganic Chemistry. 6th		
	Edition. Wiley.		
	• Lee, J.D.(1964).Concise Inorganic Chemistry. ELBS.		
	• Douglas, B.E. and Mc Daniel, D.H.(1994).Concepts & Models of		
	Inorganic Chemistry.		
	• Day, M.C. and Selbin, J. (1963). Theoretical Inorganic		
	Chemistry.ACS Publications.		
	• SHRiver and Atkins (2003).Inorganic Chemistry. W. H. Freeman		
	and Company		
	 James Huheey, (1972). Inorganic chemistry: Principles of Structure 		
	and Reactivity, Pearson Education India.		
	Pearson Education.		

B.Sc. B.Ed. 1I Year				
COURSE CODE:	BSCBED-254II COURSE TYPE : CORE			
COURSE TITLE :	Paper II :Organic Chemistry			
MAX.MARKS:	50	MIN.PASS MARKS	20	
THEORY	40	MIN.PASS MARKS	16	
EXAMINATION:				
CONTINUOUS	10	MIN.PASS MARKS	4	
COMPREHENSIVE				
ASSESSMENT				
(CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE	SEESSION		
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION	MONTHLY	TEST	
DURATION Objective:	03 HR	01 HI	R	
 chemistry along with the study of reaction mechanisms in various types of substitution and elimination reactions To predict the relationships between organic chemical structures and their reactivity. To learn the basic concept of Stereochemistry of Organic Compounds and <i>Pericyclic reactions</i>. To learn the principles of IUPAC nomenclature of alkanes, Cycloalkenes, Dienes, and Alkynes. Learning outcomes - After completion of the course, student-teachers will be able to:- <i>Know the background, properties of Fundamentals of Organic Chemistry</i>. Get familiar with concepts of isomerism and Types of isomerism. Apply the principles of IUPAC nomenclature of branched and unbranched alkanes. 				
	balkenes, Dienes, and Alkynes. Itals of Organic Chemistry			
 Structure and Bonding: Hybridization, bond lengths and bond angles, bondenergy, localized and delocalized chemical bonding, van der Waals interactions, inclusion compounds, clatherates, charge transfer complexes, resonance, hyperconjugation, aromaticity, inductive and field effects, hydrogen bonding. Mechanism of Organic Reactions: Curved arrow notation, drawing electronmovements with arrows, half-headed and double-headed arrows, homolytic and heterolytic bond breaking, Types of reagents - electrophiles and nucleophiles, Types of organic reactions, Energy considerations. Reactive intermediates Carbocations, carbanions, free radicals, carbenes, arynes and nitrenes (with examples). Assigning formal charges on intermediates and other ionic species. Methods of determination of reaction mechanism (product analysis, intermediates, isotope effects, kinetic and stereochemical studies) 				

	Stereochemistry of Organic Compounds
UNIT-2 TEACHIGN HOURS (10)	Concept of isomerism , Types of isomerism; Optical isomerism, elements of symmetry, molecular chirality, enantiomers, stereogenic center, optical activity, properties of enantiomers, chiral and achiral molecules with two stereogenic centers, diastereomers, tHReo and erytHRo diastereomers, meso compounds, resolution of enantiomer, inversion, retention and racemisation. Relative and absolute configuration, sequence rules, D & L and R & S systems of nomenclature. Geometric isomerism: determination of configuration of geometric isomers, E & Z system of nomenclature, geometric isomerism in oximes and alicyclic compounds. Conformational isomerism: conformational analysis of ethane and n-butane; conformations of cyclohexane, axial and equatorial bonds, conformation of mono substituted cyclohexane derivatives, Newman projection and Sawhorse formulae, Fischer and flying wedge formulae, Difference between configuration and conformation.
UNIT-3 TEACHING HOURS(12)	Alkanes, Cycloalkanes and Alkenes Alkanes:- IUPAC nomenclature of branched and unbranched alkanes, the alkyl group, classification of carbon atom in alkanes, Isomerism in alkanes, sources, methods of formation (with special reference to Wurtz reaction, Kolbe reaction, Corey-House reaction and decarboxylation of carboxylic acids), physical properties and chemical reactions of alkanes, Mechanism of free radical halogenation of alkanes: orientation, reactivity and selectivity Cycloalkanes: Nomenclature, methods of formation, chemical reactionsBaeyer's strain theory and its limitations. Ring strain in small rings (cyclopropane and cyclobutane), theory of strainless rings. The case of cyclopropane ring, banana bonds. Alkenes:- Nomenclature of alkenes, methods of formation, mechanisms ofdehydration of alcohols and dehydrohalogenation of alkyl halides, regionselectivity in alcohol dehydration, The Saytzeff's rule, Hofmann elimination, physical properties and relative stabilities of alkenes. Chemical reactions of alkenes, mechanism involved in hydrogenation, electrophilic and free radical additions, Markownikoff's rule, hydroboration oxidation, oxymercuration-reduction. Epoxidation, ozonolysis, hydration, hydroxylation and oxidation with KMnO ₄ , Polymerization of alkenes, Substitution at the allylic and vinylic positions of alkenes, Industrial applications of ethylene and propene.

	Cycloalkenes, Dienes, A	Ilumos		
UNIT-4 TEACHING HOURS(13)	Cycloalkenes: Methods ofcycloalkenes, Nomencl cumulated dienes, Struct polymerization, chemical Alkynes: Nomenclature, Chemical reactions of all nucleophilic addition react oxidation and polymeriza Pericyclic reactions: cla elctrocyclic, cyclo- addit Arenes and Aromaticit Aromatic nucleus and s kekule structure, stability	of formation, conformation and lature and classification of dienes: i cture of allenes and butadiene, n reaction – 1, 2 and 1, 4 additions, D structure and bonding in alkynes, kynes, acidity of alkynes, Mechanis ctions, hydroboration- <i>oxidation</i> , meta	solated, conjugatedand nethods of formation, iels-Alder reaction. Methods offormation, m of electrophilic and al-ammonia reductions, ard Hoffmann rules - nt. vatives, the arylgroup, nolecular formula and ofbenzene, resonance	
	Concept of Electrophilic			
UNIT-5 TEACHING HOURS(13)	Aromatic electrophilic s and π - complexes, Mecha and Friedel-Crafts reaction derivatives, Birch reduct alkylbenzenes alkynylben Alkyl and Aryl Halides formation, and chemical reactions of alkyl halide profile diagrams, Polyhale Methods of formation of a elimination and the elim	substitution: general pattern of the anism of nitration, halogenation, sulp on. Energy profile diagrams. Activ and ortho-para ratio, Side chain etion, Methods of formation and nzenes and biphenyl, naphthalene and s: Nomenclature and classes of alk d reactions, Mechanisms of nucl as, SN^2 and SN^1E_1 and E_2 reactions ogen compounds: Chloroform, carbo aryl halides, nuclear and side chain r mination-addition mechanisms of elative reactivities of alkyl halides v	phonation, mercuration ating and deactivating reactions of benzene chemical reactions of d AntHRacene. cyl halides, methodsof leophilic substitution reactions with energy on tetrachloride. eactions, The addition- nucleophilic aromatic	
	•	es of DD1 and BHC		
TEACHI NG AND	 Lecture method Problem Solving r 	method		
LEARNI	3. Graphical method			
NG	4. Seminar/Symposia	a		
STRATE	5. Review of literatu	re		
GIES	6. Report writing7. Group Discussion			
	8. Videos/Animation			
	9. Self-Learning/e-Learning			
	10. Workshops/Exper			
	* The TEACHING AND LEARNING STRATEGIES may be change as per requirement of the students and their capabilities.			
CONTIN		nd Comprehensive Assessment (CC	CA) are as follows:	
UOUS	S. No.	CCA- Components	Max. Marks	
COMPR		1	Allocation	
EHENSI		Monthly test	20*3 Test=60	
VE	2.	Quizzes and Assignments	10	

ASSESS	3.	Viva-voce	10	
MENT	4.	Seminar/Symposia	10	
(CCA)	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		ent reduced to CCA original marks 30		
EXAMI		are organized by the university in th	1	
NATION		chieve success in contemporary comp	etitions and to achieve	
PATTE	their goals.			
RN				
PERIOD	1. ANNUAL			
ICAL	,	ersitymay revise the syllabus at a	•	
REVISE OF	running sessionatter	giving a notice for a period one mo	ntn.	
OF SYLLAB				
US				
SELECT	• Ali Hashmat (201)	6). Reaction Mechanism in Organic	Chemistry S Chand	
ED	• All, Hashinat, (2010 publishing.	5). Reaction Mechanism in Organic	Chemistry. S. Chand	
READIN		n (1994) Stereochemistry of Organi	c Compounds: Wiley.	
GS	• Eliel, E. L. & Wilen, (1994).Stereochemistry of Organic Compounds; Wiley: London,			
	<i>,</i>	Organic Chemistry (Volume 1), Dorl	ing Kindersley (India)	
	Pvt. Ltd. (Pearson Ed			
	 Finar, I. L. (2011). Organic Chemistry (Volume 2: Stereochemistry and the Chemistry of Natural Products), Dorling Kindersley (India) Pvt. Ltd. (Pearson Education). 			
	 Singh, Harkishan and V. K. Kapoor. (1996). Medicinal and Pharmaceutical Chemistry. Vallabh Prakashan Publishers, Delhi. 			
	• John Leonard, Barr Chemistry. Third Ed	y Lygo (1995).Garry Procter Advan ition.	ced Practical Organic	
	March's Advanced Structure, Sixth Edit	(2001).Organic Chemistry: Reactio ion.	ns, Mechanisms, and	
	• Morrison, R. N. & (India) Pvt. Ltd. (Pea	Boyd, R. N. (2010) Organic Chemistration Education).	ry. Dorling Kindersley	

B.Sc. B.Ed. 1I Year						
COURSE CODE:	BSCBED-254 b III			COURSE TYPE : CORE		
COURSE TITLE :	Paper-III: Physical Chemistry					
MAX.MARKS:	50	MIN	PASS	MARKS	20	
THEORY	40	MIN	PASS	MARKS	16	
EXAMINATION:						
CONTINUOUS	10	MIN	PASS	MARKS	4	
COMPREHENSIVE						
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RESPECTIVE SEESSION					
ELIGIBILITY						
EXAMINATION	TERM END EXAMINATION		-	MONTHLY TEST		
DURATION	03 HR			01 HR		
Objective:						
• This course will equip students with the necessary chemical knowledge concerning the						
fundamentals in the basic areas of physical chemistry viz. thermodynamics, electrochemistry						
and chemical kinetics, with regard to various theories developed and their applicability for						

and chemical kinetics, with regard to various theories developed and their applicability for various systems under consideration.To develop the problem solving skills of students are expected to be enhanced tHRough due

• To develop the problem solving skills of students are expected to be enhanced tHRough due weightage given to numerical problems in each unit.

Learning outcomes - After completion of the course, student-teachers will be able to:-

- Know the background, properties of Fundamentals of thermodynamics.
- Get familiar with concepts of Chemical Kinetics and Catalysis.
- Apply the principles laws of thermodynamics and thermo chemistry.
- Learn the Electrical transport-conduction in metals and in electrolyte solutions, conductance.

conductance.					
	Thermodynamics-I				
	Thermodynamics: Definition of thermodynamic terms: system,				
	surroundingsetc. Types of systems, intensive and extensive properties,				
6	State and path functions and their differentials. Thermodynamic process,				
	Concept of heat and work .First Law of Thermodynamics: Statement, definition of internal energy and enthalpy. Heat capacity, heat capacities at constant volume and pressure and their relationship .Joule's law- Joule-Thomson coefficient and inversion temperature. Calculation of w,				
00-1-1					
H					
	q, dU, &dH for the expansion of ideal gases under isothermal and				
	adiabatic conditions for reversible process. Thermo chemistry : standard state, standard enthalpy of formation–				
UNIT-1 TEACHING HOURS (12)					
E	Hess'sLaw of heat summation and its applications. Heat of reaction at				
	constant volume. Enthalpy of neutralization. Bond dissociation energy				
	and its calculation from thermo-chemical data, temperature dependence				
	of enthalpy, Kirchhoff's equation				
	or entituipy, Knemion's equation				

UNIT-2 TEACHIGN HOURS (12)	 Thermodynamics-II Thermodynamics: Second law of thermodynamics: need for the law, different statements of the law. Carnot cycle and its efficiency, Carnot theorem. Thermodynamic scale of temperature. Concept of entropy: entropy as a state function, entropy as a function of V & T, entropy as a function of P & T, entropy change in physical change, clausius inequality, entropy as a criterion of spontaneity and equilibrium. Entropy change in ideal gases and mixing of gases. Third law of thermodynamics:Jule-thompson's experiment,Nernst heat theorem, statement and concept of residual entropy, evaluation of absolute entropy from heat capacity data. Gibbs and Helmholtz functions, Gibbs function (G) and Helmholtz function (A) as thermodynamic quantities, A & G as criteria for thermodynamic equilibrium and spontaneity, their advantage over entropy change. Variation of G with A with P, V and T.
UNIT-3 TEACHING HOURS (10)	 Electrochemistry-I Electrical transport-conduction in metals and in electrolyte solutions, specific conductance and equivalent conductance, measurement of equivalent conductance, variation of equivalent and specific conductance with dilution. Migration of ions and Kohlrausch law, Arrhenius theory of electrolyte dissociation and its limitations, weak and strong electrolytes, Ostwald's dilution law its uses and limitations. Debye-Huckel- Onsager's equation for strong electrolytes (elementary treatment only). Transport number, definition and determination by Hittorf method and moving boundary method. Applications of conductivity measurements:- determination of solubility product of sparingly soluble salt, conductometric titrations
UNIT-4 TEACHING HOURS (13)	 Electrochemistry –II Types of reversible electrodes-gas-metal ion, metal-insoluble salt anion and redox electrodes. Electrode reactions, Nernst equation, derivation of cell E.M.F. and single electrode potential, standard hydrogen electrode-reference Electrodes-standard electrode potential, sign conventions, electrochemical series and its significance. Electrolytic and Galvanic cells-reversible and irreversible cells, conventional representation of electrochemical cells.EMF of a cell and its measurements. Computation of cell EMF. Calculation of thermodynamic quantities of cell reactions (ΔG, ΔH, and ΔK), polarization, over potential, application of concentration cells, valency of ions, solubility product and activity coefficient, potentiometric titrations. Definition of pH and pK_a determination of pH using hydrogen, quinhydrone and glass electrodes, by potentiometric methods. Buffers-mechanism of buffer action, Handerson-hazel equation. Hydrolysis of salts. Corrosion-types, theories and methods of combating it.

	Chemical Kinet	tics and Catalysis		
		etics: Chemical kinetics and its scope	. rate of a reaction	
	factorsinfluencing the rate of a reaction-concentration, temperature,			
	pressure, solvent, light catalyst, concentration dependence of rates,			
	mathematical characteristics of simple chemical reactions- zero order,			
[3]		econd order, pseudo order, half-li		
2	*	i of the order of reaction differential		
B		ethod of half-life period and isolation	•	
0 0 0	•	decay: as a first order phenome		
UNIT-5 TEACHING HOURS (13)		•	*	
Z Ç	methods of chemical kinetics: conductometric, potentiometric, optical methods, polarimetry and spectrophotometer. Theories of chemical			
	kinetics: effect of temperature on rate of reaction, Arrhenius equation,			
CE		tivation energy. Simple collision the		
EA		l, transition state theory (equili	-	
E	-	r the rate constant based on equilib		
	thermodynami	-	mum constant and	
	-	naracteristics of catalysed reactions	classification of	
		ogeneous and heterogeneous catalysis		
	miscellaneous		s, enzyme eatarysis,	
TEACHING AND	1. Lecture	*		
LEARNING		Solving method		
STRATEGIES	3. Graphica			
STRILOLD	4. Seminar			
	5. Review	• 1		
	6. Report w			
	-	7. Group Discussion		
	8. Videos/Animation			
	9. Self-Learning/e-Learning			
	10. Workshops/Experiments.			
	* The TEACHING AND LEARNING STRATEGIES may be			
	change as per requirement of the students and their capabilities.			
CONTINUOUS	Details of Continuous and Comprehensive Assessment (CCA) are as			
COMPREHENSIVE	follows:	-		
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)			Allocation	
	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	б.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		s equivalent reduced to CCA original		
EXAMINATION		inations are organized by the universi		
PATTERN		ble the scholars to achieve success	s in contemporary	
	competitions an	d to achieve their goals.		

PERIODICAL REVISE OF	 ANNUAL However, the Universitymay revise the syllabus at any time
SYLLABUS	during the running year after giving a notice for a period one months.
SELECTED READINGS	 Adamson, A. (2012). A textbook of physical chemistry. Elsevier. Atkins, P., & De Paula, J. (2009). Elements of physical chemistry. Macmillan. Espenson, J. H. (1995). Chemical kinetics and reaction mechanisms (Vol. 102). New York: McGraw-Hill. Glasstone, S. (1951). Textbook of physical chemistry. Macmillan. Jones, H. C. (1903). The elements of physical chemistry. Macmillan Company. Kapoor, K. L. (2001). A textbook of Physical Chemistry (Vol. 3). Macmillan. Laidler, K. J. (2008). Chemical Kinetics. (3rdedn). Pearson Education. Lewis, D., & Glasstone, S. (1960). Elements of physical chemistry. Macmillan. Negi, A. S., & Anand, S. C. (1985). A textbook of physical chemistry. New Age International. Renner, T. (2007). Quantities, units and symbols in physical chemistry. Royal Society of Chemistry. Roussel, M. R. (2012). A life scientist's guide to physical chemistry. Cambridge University Press. Steinfeld, J. I., Francisco, J. S., & Hase, W. L. (1999). Chemical kinetics and dynamics (pp. 325-389). Upper Saddle River, NJ: Prentice Hall

CHEMISTRY PRACTICALS-2				
Duration:4 HR MAX.MARKS: 50 Min. Marks: 20				
Inorganic Chemistry				
Ex.1 Volumetric analysis	15			
Organic Chemistry				
Ex.2 Identification of given organic compounds tHRough functional group analysis	15			
Physical Chemistry				
Ex.3 Perform one of the experiments mentioned in the syllabus.	10			
Ex. 4Vive-Voce	05			
Ex.5 Practical-Record	05			
Total 50 Marks				
LIST OF EXPERIMENTS				
Inorganic Chemistry				
Quantitative Analysis: Volumetric Analysis-				
• Determination of acetic acid in commercial vinegar using NaOH.				
• Determination of alkali content – antacid tablet using HCl.				
• Estimation of calcium content in chalk as calcium oxalate by permanganometry.				
• Estimation of ferrous and ferric by dicHRomate method.				
• Estimation of copper using thiosulphate.				
OrganicChemistry				
Qualitative Analysis				
• Detection of extra elements (N, S and halogens), solubility behaviour and fur	nctional			
groups (Alcoholic, phenolic, carboxylic, carbonyl, esters, carbohydrates, amines, amides,				
nitro and anilide) in simple organic compounds.				

• Identification of an organic compound tHRough the functional group analysis and preparation of suitable derivatives.

Physical Chemistry

 Determination of the transition temperature of the given substance by thermometric/ dialometric method (e.g.MnCl₂. 4H₂O/SrBr₂. 2H₂O).
 Phase Equilibrium

Phase Equilibrium

- Study of the effect of a solute (e.g. NaCl, succinic acid) on the critical solution temperature of two partially miscible liquids (e.g. phenol-water system) and to determine the concentration of that solute in the given phenol-water system.
- Construction of the phase diagram of two component (e.g. diphenylamine, benzophenone) system by cooling curve method. **Thermo chemistry**
- Determination of the solubility of benzoic acid at different temperatures and to determine ΔH of the dissolution process.
- Determination of the enthalpy of neutralization of a weak acid/weak base versus strong base/strong acid and determine the enthalpy of ionization of the weak acid/weak base.
- Determination of the enthalpy of solution of solid calcium chloride and calculate the lattice energy of calcium chloride from its enthalpy data using Born Haber cycle. **Distribution Law**
- Determination of the distribution coefficient for the distribution of iodine between water and CCl₄.
- Study the distribution of benzoic acid between benzene and water.

	B.SC.B.ED. I	[Year		
COURSE CODE:	BSCBED-254 c I	l I cai	COUF	RSE TYPE : CORE
COURSE TITLE :	Chordata		0001	
MAX.MARKS:			20	
THEORY	40	MIN.PASS MARKS		16
EXAMINATION:				10
CONTINUOUS	10	MIN.PASS M	ARKS	4
COMPREHENSIVE	10			•
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPE	CTIVE YEAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMINA	ATION	MC	ONTHLY TEST
DURATION	03 HR			01 HR
Objective :				
 To lear about understand in respect of vertebrates- their organizational hierarchies and complexities. To learn about the evolutionary trends in external morphology and comparative studies of internal structures; identification and classification with examples. To Learn them to understand various modes of adaptions in animals. Learning Out Come : After completion of the course, student-teachers will be able: 				
and complexities, the internal structures;	le students to understand in respect of vertebrates– their organizational hierarchies aplexities, the evolutionary trends in external morphology and comparative studies of structures; identification and classification with examples; to enable them to and various modes of adaptions in animals.			
UNIT-1 TEACHING HOURS (15)	An outline classification of chordates up to orders but up to sub- classes only in case of protochordate groups and mammals. Comparative anatomy of vertebrates from an evolutionary point of view of the following: Alimentary canal, Heart and aortic arches. Kidney and associated urinogenital ducts			
UNIT-2 TEACHIGN HOURS(15)	 Habit and habitat, structure, reproduction (excluding development) of following types:- Hemichordate: <i>Banlanoglossus</i> Urocohordata: <i>Herdmania</i>, ascidian tadpole larva and its metamorphosis. Cephalochordata: <i>Amphioxus</i> Agnatha: Petromyzon 			
UNIT-3 TEACHING HOURS (15)	 Habit and habitat, structure (morophology, digestive system, respiratory system, blood vascular system, nervous system (central and peripheral) and urinogenital system of the following types – Pisces: scoliodon Structure and development of placoid scales Amphibia: frog Parental care in Amphibia. 			
UNIT-4 TEACHIN G HOURS (15)	 Habit and habitat, st respiratory system, bloc and peripheral) and urine Reptile : uromastix, Po apparatus and biting med 	ructure (morop od vascular syst ogenital system visonous and no	em, nerv of the fo	vous system (central ollowing types – nous snakes, poison

S-LINO TEACHING AND LEARNING STRATEGIES	 Habit, habitat, structure, morophology, digestive system, respiratory system, blood vascular system, system nervous system (central and peripheral) and urinogenital system of the following types:- Aves: <i>Columba</i> ,Mammalia: <i>Rabbit</i>. Origin of birds, migration & flight adaptation of birds. Lecture method Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be change as per requirement of the students and their capabilities. 			
COMPREHENSIVE	follows:			
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)	1.	Monthly test	Allocation 20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.Report writing10			
	6.Workshop10			
	7.Review of literature10		10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		marks equivalent reduced to CCA original		
EXAMINATION PATTERN		examinations are organized by the universive enable the scholars to achieve success		
		ons and to achieve their goals.	s in contemporary	
PERIODICAL	1. ANNU			
REVISE OF		ver, the Universitymay revise the sylla	abus at any time	
SYLLABUS		g the running year after giving a notice	-	
	month	15.		
SELECTED		s, R. D. (1987). Invertebrate zoology	(No. Ed. 5). WB	
READINGS	Saund	lers Company.		
	• Brooks, W. K. (1882). Handbook of invertebrate zoology: for laboratories and seaside work. SE Cassino.			
		s, W. C., GutHRie, M. J., & Jeffers, K. R. (al zoology. Wiley.	(1938). Textbook of	

• Hegner, R. W., & Engemann, J. G. (1968). <i>Invertebrate zoology</i> (No. QL362 H4 1968). Macmillan.
• Hyman, L. H. (1940). <i>The invertebrates: mollusca I</i> (Vol. 6). McGraw-Hill.
• Jordan, E. L., & Verma, P. S. (1996). <i>Invertebrate Zoology</i> . Sixth revised and enlarged edition. <i>S. Chand and Company</i> , <i>Ltd.</i> 857pp.
• Jordan, E. L., & Verma, P. S. (2009). <i>Invertebrate zoology</i> . S. Chand & Company.
• Light, S. F. (1941). <i>Laboratory and field text in invertebrate zoology</i> . Associated Students Store, University of California.
• McEdward, L. (2020). <i>Ecology of marine invertebrate larvae</i> . CRC press.
• Parker, T. J., & Haswell, W. A. (1967). <i>Textbook of zoology</i> . Macmillan International Higher Education.
• Scott-Ram, N. R., & Scott-Ram, N. R. (1990). Transformed cladistics, taxonomy and evolution. Cambridge University Press.
• Sedgwick, A. (1905). A Student's Text-book of zoology (Vol. 2). Allen & Unwin.
• Verma, P. S. (2001). <i>Invertibrate Zoology</i> . S. Chand Publishing.

	B.Sc.B.Ed. I	I Year		
COURSE CODE:	BSCBED-254 c II			YPE : CORE
COURSE TITLE :	Animal Physiology an	Animal Physiology and Endocrinology		
MAX.MARKS:	50			20
THEORY	40			16
EXAMINATION:				
CONTINUOUS	10	MIN.PASS	MARKS	4
COMPREHENSIVI	E			
ASSESSMENT (CC	A)			
ATTENDANCE	80 PERCENT IN RESI	80 PERCENT IN RESPECTIVE YEAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMIN	NATION	MONTH	LY TEST
DURATION	03 HR		01 HR	
OBJECTIVE :				
organs and syster molecules and ph Learning outcomes • To enable studen organs and syster	 To learn about the comprehend the modern concepts of physiological aspects on various organs and systems of animals and human being to comprehend chemical nature, biological molecules and physiological roles. Learning outcomes - After completion of the course, student-teacher will be able:- To enable students to comprehend the modern concepts of physiological aspects on various organs and systems of animals and human being to comprehend chemical nature, biological molecules and physiological roles. 			
UNIT-1 TEACHING HOURS (15)	 Physiology of digestion: Chemical nature of food stuff (including micronutrients), various types of digestive enzymes and their digestive action in the alimentary canal, role of GI hormones in digestion, mechanism of absorption of digested food.Disorders of digestive system. Physiology of respiration: Mechanism and control of breathing, exchange of gases transport of respiratory gases (oxygen and carbon 			
UNIT-2 TEACHING HOURS (15)	 dioxide) .elementry idea about asthma and emphysema. Physiology of blood circulation: Composition and function of blood.Blood groups (ABO and Rh) Blood coagulation – factors, mechanism, theories and anticoagulants. Origin, conduction and regulation of heart beat in mammals. Cardiac cycle, ECG, Elementary ideas of cardiovascular disorders, hypertension, angina pectoris, myocardial infarction. 			
UNIT-3 TEACHING HOURS (15)	 Nerve physiology: structure and types of neuron, origin and conduction of nerve impulse, Synapse-structure, types, properties and signal transmission tHRough synapses. Muscle physiology: Ultra-structure and mechanism of contraction of skeletal muscle, summation and fatigue. Common disorders of muscles. Physiology of Excretion: Nitrogenous wastes, anatomy of mammalian kidney, structure of nepHRon, mechanism of urea and urine formation (including hormonal regulation),disorders of excretory system.Osmoregulation. 			

UNIT-4 TEACHING HOURS (15)	 Endocrinology: Types of endocrine glands and their disorders; pituitary, pineal body, Adreanl, Thyroid, Islets of langerhans, Testis and Ovary. Chemical nature of Hormone, hormone action(including role of cyclic AMP in hormone action). Preliminary idea of neurosecretion; Hypothalamic control of pituitary functions. Endocrine glands in Insects. Neurotransmitters. 		
UNIT-5 TEACHING HOURS (15)	 Endocrine glands;Ovary and Placenta,ovarian cycles and their neuroendocrine control.Endocrinology of Ovulation,Implantation,Parturition and Lactation. MTP and related social issues. Endocrine glands;Testis,testicular cycle and its hormonal control.hormonal control of reproduction using mammalian examples.reproductive disorders. 		
TEACHING AND LEARNING STRATEGIES	 Lecture method Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be change 		
CONTINUOUS COMPREHENSI	as per requirement of the students and their capabilities. Details of Continuous and Comprehensive Assessment (CCA) are as follows:		
VE ASSESSMENT	S. No.	CCA- Components	Max. Marks Allocation
(CCA)	1.	Monthly test	20*3 Test=60
(CCA)	<u> </u>	Monthly test Quizzes and Assignments	20*3 Test=60 10
(CCA)			
(CCA)	2.	Quizzes and Assignments	10
(CCA)	2. 3.	Quizzes and Assignments Viva-voce	10 10
(CCA)	2. 3. 4.	Quizzes and Assignments Viva-voce Seminar/Symposia	10 10 10
(CCA)	2. 3. 4. 5.	Quizzes and Assignments Viva-voce Seminar/Symposia Report writing	10 10 10 10 10
(CCA)	2. 3. 4. 5. 6.	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshop	10 10 10 10 10 10 10
(CCA)	2. 3. 4. 5. 6. 7.	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literature	10 10 10 10 10 10 10 10 10
(CCA)	2. 3. 4. 5. 6. 7. 8.	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activity	10 10 10 10 10 10 10 10 10 10 10 10 10 10 10 10 10 10 10
(CCA)	2. 3. 4. 5. 6. 7. 8. 9. 10. 11.	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activityAttendance	10 10
	2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 marks et	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activityAttendancequivalent reduced to CCA origin	10 11 10 10 10 10 11 10 10 10 11 11 11 11 11 11
EXAMINATION	2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 marks ex- Term-end examina	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activityAttendancequivalent reduced to CCA originations are organized by the univ	10 110 110 110 110 110 110 110 110 110 110 110 110 110 </th
	2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 marks end format to enable	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activityAttendancequivalent reduced to CCA originations are organized by the universitythe scholars to achieve succes	10 110 110 110 110 110 110 110 110 110 110 110 110 110 </th
EXAMINATION PATTERN	2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 marks en format to enable competitions and t	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activityAttendancequivalent reduced to CCA originations are organized by the univ	10 110 110 110 110 110 110 110 110 110 110 110 110 110 </th
EXAMINATION PATTERN PERIODICAL	2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 marks ex- Term-end examination format to enable competitions and the competition of the comp	Quizzes and Assignments Viva-voce Seminar/Symposia Report writing Workshop Review of literature Creativity/Innovation Experimental Skill Co-curricular activity Attendance quivalent reduced to CCA origin ations are organized by the unive the scholars to achieve suc	10 10
EXAMINATION PATTERN	2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 marks ex- Term-end examination format to enable competitions and to 1. ANNUAL 2. However the lagent	Quizzes and AssignmentsViva-voceSeminar/SymposiaReport writingWorkshopReview of literatureCreativity/InnovationExperimental SkillCo-curricular activityAttendancequivalent reduced to CCA originations are organized by the universitythe scholars to achieve succes	10 10

SELECTED READINGS	• Guyten and hall (2013), <i>A text book of Medical physiology</i> , Elsvier Pub (South Asia)
	• Schmidt – Nielson, (2013). Animal Physiology, 5thed, Cambridge Pub
	• D Voet & JG Voet (2011). <i>Biochemistry</i> , Wiley Arumugam Animal Physiology (Saras Publication, Nagercoil, Tamil Nadu).
	• K.V. Sastry (2008). <i>Animal Physiology and biochemistry</i> , Rastogi Publications.
	• Pandey, Kamleshwar and Shukla, J.P. (2008). <i>Regulatory mechanism in Vertebrates</i> –Jaipur: Rastogi Publications.
	• Goyal, K.A. and K.V. Sastry (2008). <i>Animal Physiology</i> , Rastogi Publication.
	• Sasyry, K.V. (2008). <i>Endocrinology and Reproductive Biology</i> , Jaipur: Rastogi Publication.
	• Arora, M.P. (1989). <i>Animal Physiology</i> –Shimla: Himalaya Pucations House.
	• Guyton, A.C. & Hall J.E (1996). <i>Textbook of Medical Physiology</i> . W.B. Saunders & Co.
	• Hoar, W.S. (1983). <i>General and Comparative Physiology</i> -Prentice Hall Publication).
	• Hurtkar, P.C. & Mathur, P.N. (1976). <i>A textbook of Animal Physiology</i> . S Chand & Co.
	• Turner, C.D. & Gangara, J.T. (1971). <i>General Endocrinology</i> - W. B. Saunders & Co.
	• Soni, KC (2001). <i>Animal Physiology and Immunology</i> , Hindi Edition, Jaipur: College book centre, Chaura Rasta.
	• Pawar, VS (2001). <i>Mammalian Endocrinology and Animal Behavior</i> , Hindi Edition, Jaipur: College book centre, Chaura Rasta.

	B.Sc.B.Ed.	II Year			
COURSE CODE:	BSCBED-254 c III			SE TYPE : CORE	
COURSE TITLE :		Evolution and Palentology			
MAX.MARKS:	50	MIN.PASS	MARKS	20	
THEORY	40		MIN.PASS MARKS		
EXAMINATION:			MIN.PASS MARKS 1		
CONTINUOUS	10	MIN.PASS	MIN.PASS MARKS		
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN R	ESPECTIVE Y	(EAR		
ELIGIBILITY					
EXAMINATION	TERM END EXA	MINATION MON		THLY TEST	
DURATION	03 HI	ł		01 HR	
Objective :					
 Embryology and Paleontology. To learn about apply the Theories of evolution in understanding of real life. To learn about the Mimicry and protective coloration Learning Out Come:- After completion of the course, student-teachers will be able: Know about Origin of life, Molecular basis of evolution and Variation. Get familiar with concepts of species/subspecies/sibling, Isolation, Embryology at Paleontology. Apply the Theories of evolution in understanding of real life. 				be able:	
UNIT-1 TEACHI NG HOURS (15)	• Origin of life (Abiogenesis and biogenesis): Evidence in favor of evolution: Evidences from morphology and comparative anatomy, Embryology and Paleontology. Molecular basis of evolution.				
UNIT-2 TEACHING HOURS (15)	• Theories of evolution: Lamarckism, inheritance of acquired characters and Neo-Lamarckism. Darwinism, theory of natural selection and Neo Darwinism, Mutation theory of Hugo de Vries, Weismann theory of germplasm, Recapitulation theory.				
G TEACHING 5) HOURS (15)	 Variation: Kinds, sources of variation, origin of new mutations. Isolation: Definition, mechanism and role of isolation in evolution. Adaptation: Introduction, kinds (structural, physiological and behavioural), Mimicry and protective coloration: Definition, kinds, condition necessary for mimicry, significance, divergent evolution, convergent evolution, Origin of species: Concept of species/subspecies/sibling, Factors 				
UNIT-4 TEACHING HOURS (15)	 causing genetic divergence in the population of species, genetic drifts, Bottle Neck effect founder's effect. Zoogeographical distribution of animals, geological time scale, eras origin and evolution of amphibian, reptiles, birds and mammals. 				

STEACHING TEACHING AND LEARNING	 Introduction, formation, kinds, determination of age of fossil and its significance. Dinosaurs, fossil evidence & reasons for extinction of dinosaurs. Evolution of man: Time of origin, compelling causes, ancestor of man, evolution from apes and evolutionary trends. 1. Lecture method 2. Problem Solving method 			
STRATEGIES	4. Seminar	 Graphical method Seminar/Symposia 		
	 Review of literature Report writing 			
	6. Report writing 7. Group Discussion			
	8. Videos/A	Animation		
		rning/e-Learning		
		ops/Experiments. ING AND LEARNING STRAT	EGIES may be	
	change as per r	requirement of the students and	their capabilities.	
CONTINUOUS COMPREHENSIVE	Details of Cont follows:	inuous and Comprehensive Ass	sessment (CCA) are as	
ASSESSMENT	S. No.	CCA- Components	Max. Marks	
(CCA)	·		Allocation	
	1.	Monthly test	20*3 Test=60	
	2.Quizzes and Assignments10			
	3. Viva-voce 10			
	4. Seminar/Symposia 10			
	5.Report writing10			
	6. Workshop 10			
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
		equivalent reduced to CCA orig		
EXAMINATION		inations are organized by the univ		
PATTERN		le the scholars to achieve such to achieve their goals.	cess in contemporary	
PERIODICAL	competitions and to achieve their goals. 1. ANNUAL			
REVISE OF		he Universitymay revise the	-	
SYLLABUS	during the months.	running year after giving a ne	buce for a period one	
SELECTED		B., Breedlove, S. M., Crews, D.	, & McCarthy, M. M.	
READINGS		2). Behavioral endocrinology. Mi		
		, R. E., & Blackwelder, R. E. (1 we book (No. OI 351 B54) New Y	· •	
	• Cope, E. D.	 <i>and reference book</i> (No. QL351 B54). New York: Wiley. Cope, E. D. (1904). <i>The primary factors of organic evolution</i>. Open 		
	Court.			

• Darlington, C. D. (1969). <i>The evolution of man and society</i> (p. 69ff). New York: Simon and Schuster.
• Gilbert, W. (1986). Origin of life: The RNA world. Nature, 319 (6055), 618-618.
• Haeckel, E. (1897). The evolution of man (Vol. 1). D. Appleton.
• Harvey, P. H., & Pagel, M. D. (1991). <i>The comparative method in evolutionary biology</i> (Vol. 239). Oxford: Oxford university press.
• Küppers, B. O. (1990). Information and the Origin of Life. Mit Press.
• Nelson, R. J. (2005). <i>An introduction to behavioral endocrinology</i> . Sinauer Associates.
• Provine, W. B. (1989). <i>Sewall Wright and evolutionary biology</i> . University of Chicago Press.
• Rose, M. R. (1994). <i>Evolutionary biology of aging</i> . Oxford University Press on Demand.
• Spencer, H. (1887). The factors of organic evolution. D. Appleton.
• Washburn, S. L. (1978). <i>The evolution of man. Scientific American</i> , 239(3), 194-211.

Practical

Maximum Marks - 50

Duration: 4 HR.

Learning outcome: -

• To develop in the student the skills of staining and mounting of materials (temporary and permanent) and of dissection, display and labeling as per UGC guidelines; Cadaver or otherwise collection, preservation, mounting, identification and labeling of specimens as per UGC guidelines: Field observation of animals. To enable students to develop the skills of hematology and endocrinology.

Course Content

- Study of Museum specimen with respect to levels and patterns of organization, biosystematics, biodiversity, adaptations, development stages, population dynamics, ecological implications ets.
- Hemichordata: Balanoglossus.
- Urochordata: Herdmania, Pyrosoma
- Cephalochordata: Petromyzon, Myxine
- Cyclostomata: Petromyzon, Myxine
- Pisces: Scoliodon, Sphyrna, Torpedo, Pristis, Trygon Lepidosteus, Clarias, Ophiocephalus, Anabas. Exocoetus, Hippocampus, Tetradon, Protopterus
- Amphibia: Icthyophis, Necturus, Proteus, Ambystoma, Axolotl larva, Triturus. Amphiuma, Alytes Bufo, Hyla, Rana.
- Reptilia: Testudo, Trionyx, Sphenodon, Hemidactylus, Draco, Calotes, Chamaeleon, Varanus, Heloderma, Typhlops, Eryx, Hydrophis, Viper, Bungarus, Naja Alligator,
- Aves: Pavo, Columba, Psitacula, Passer, Corvus, Archaeopteryx.
- Mammals: Ornithorhynchus, Echidna, Macropus, Loris, Manis, Rattus.

Study of Permanent Sildes

- Balanoglossus: T.S. of proboscis, collgar region and trunk
- *Amphioxus:* T.S. or oral hood, pharynx.
- *Mammals:* T.S., skin Stomach, Duodenum, IIeum, liver, Pancreas, spleen, lung, kindney Testis, Ovary.
- Study of skull bone of Frog, Varanus, Bird and Rabbits.
- Study of vertebral of Frog.Varanus, Bird and Rabbit.
- Study of girdles, forelimb and hind limb bones of Frog, Varanus, Bird and Rabbit.
- Dissections and/or its demonstration tHRough Charts / Models/Video/CD/digital alternatives etc and/or preparation of working
- Models of the different system of the following animals.
- *Scoliodon:* Afferent brachial systems, efferent branchial system, cranialnerves and internal ear.
- Frog: Anatomy, digestive, system, Urino-genital system
- Permanet /Temporary preparation of the follwing-:
- Scales: Placoid, Cycloid
- Blood film of any vertebrate

- Filoplumes of birds
- Thigh muscles of frog
- Estimation of Hemoglobin.
- Enumeration of RBC in blood samples.
- Enumeration of WBC in blood samples.
- Preparation of Haemin Crystals.
- Effect of different concentrations of NaCI on RBC.
- Measurement of blood pressure, Heart beat and Pulse rate.
- Study of bleeding time, Coagulation time of blood.
- Study of Histological slides of the following endocrine gland of mammal testis, ovary, thyroid, adrenal, pituitary, Islets of Langerhans

Note:

- Use of animals of materials for dissection or otherwise is subject to the condition that these are not banned under the Wildlife Protection Act.
- Report on study of animals from their natural habitat from their local surroundings or as assigned.

Guidelines/Instructions for Practical Examination Time allowed: 4 HRMax.Marks: 50

S. No.	Exercise	Marks
1.	Dissection* (Exposition, labeled diagram)	6
2.	Temporary mounting -one (Staining, identification, sketch	3
3.	Museum specimens – four or six (identification and classification	12
4.	Permanent slides – two or four (Identification with reasons)	4
5.	Osteology	3
6.	Physiological exercise	5
7.	Practical record and slides	5
8.	Viva	4
9.	Project report assignment	8
* as per UG	C guidelines	

B.SC.B.ED. II Year						
COURSE CODE:	BSCBED-254 d I COURSE TYPE : CORE			TYPE : CORE		
COURSE TITLE :	Diversity of Seed Plan breeding)	ts (Gymnospe	erm, Ang	iosperm and Plant		
MAX.MARKS:	50	MIN.PASS N	MARKS	20		
THEORY	40	MIN.PASS N	MARKS	16		
EXAMINATION:						
CONTINUOUS	10	MIN.PASS N	MARKS	4		
COMPREHENSIVE						
ASSESSMENT (CCA)			_			
ATTENDANCE	80 PERCENT IN RESPI	80 PERCENT IN RESPECTIVE YEAR				
ELIGIBILITY			M			
EXAMINATION DUPATION	TERM END EXAM	INATION	MIC	DNTHLY TEST		
DURATION Objective :	3 HR			1 HR		
 To learnaboutCharacteristics, Origin and evolution of seed plants. To learnaboutapply the principles of Asymptotes, multiple points, curve tracing in resituations. To learn about morphology and anatomy of vegetative and reproductive parts. Learning Out Come: After completion of the course, student-teachers will be able: Know aboutCharacteristics, Origin and evolution of seed plants. Get familiar with General features of gymnosperms. Apply the principles of Asymptotes, multiple points, curve tracing in real situations. Learn Morphology and anatomy of vegetative and reproductive parts. Characteristics of seed plants,(Gymnosperm & Angiosperm Homospory, Heterospory & orgion of Seed Plants,Evoulation Diversity of Gymnosperms,Fossil and Living seed Plants,Fos Gymnosperms.(williamsonia) Angiosperms: Origin and Evolution,Primitive Angiosperms. Diversiting Plant Forms-annuals, Biennials and Perennials,Convergence Evolution of Tree Habit inGymnosperms, Monocotyledons and Dicotyledons, 			parts. be able: al situations. & Angiosperm), Plants,Evoulation & seed Plants,Fossil giosperms. Diversity fals,Convergence of fonocotyledons and			
End of the second sec						
UNIT-4 TEACH ING HOURS (15)	Morphology and Anatomy of Vegetative and reproductive Parts and life cycle of PinusandEphedra, Economic importance of Gymnosperms					
Principles ofPlant breeding, Methods ofBreeding, Introducation Acctimatisation, Selection(mass, purelme, and clonal), Hydridizat pedi-gree analysis, Hybrid vigour, use of mutation and Polyploidy breeding.			onal), Hydridizaion,			

TEACHING AND	1. Lectur	re method			
LEARNING	2. Problem Solving method				
STRATEGIES		3. Graphical method			
	-	ar/Symposia			
	5. Review of literature				
	6. Report writing				
	7. Group Discussion				
		8. Videos/Animation			
		9. Self-Learning/e-Learning			
		10. Workshops/Experiments.			
		* The TEACHING AND LEARNING STRATEGIES may be			
		r requirement of the students and			
CONTINUOUS		ntinuous and Comprehensive Ass			
COMPREHENSIVE	follows:				
ASSESSMENT	S. No.	CCA- Components	Max. Marks		
(CCA)		r	Allocation		
	1.	Monthly test	20*3 Test=60		
	2.	Quizzes and Assignments	10		
	3.	Viva-voce	10		
	4.	Seminar/Symposia	10		
	5.	Report writing	10		
	6.Workshop10				
	7.	1			
	8.				
	9.Experimental Skill10				
	10.	Co-curricular activity	10		
	11.	Attendance	10		
	Total 160 mar	Total 160 marks equivalent reduced to CCA original marks 30.			
EXAMINATION		minations are organized by the univ			
PATTERN	format to enable the scholars to achieve success in contemporary				
		competitions and to achieve their goals.			
PERIODICAL	1. ANNUAL				
REVISE OF	2. However	the Unviersitymay revise the	syllabus at any time		
SYLLABUS	during th	e running year after giving a ne	otice for a period one		
	months.				
SELECTED	Ŭ	r, A.P. and Moitra, A. (1996). Gyn	nnosperms. New Delhi:		
READINGS	0	International Limited.			
		E.M. and Foster, A.S. (1988). Mor			
	of Vascular Plants. New York: W. H. Freeman & Company.				
	• Singh, D. (2009). Diversity and Systematics of Seed Plants. Jaipur:				
	Rastogi Publications.				
		K.R. (1965). The Morphology of C			
		on & Co. (Publishers) Ltd., Lor			
		Paleobotany and the Evolution	of Plants. Cambridge		
	-	y Press. Cambridge.			
		gooy, M. M. J. (1997). Malesian	n seed plants (Vol.1).		
	Rijksherb	arium/Hortus Botanicus.			

B.Sc.B.Ed. II Year				
COURSE CODE:				SE TYPE : CORE
COURSE TITLE :	Systematics of Angiospe	rms		
MAX.MARKS:	50 MIN.PASS MARKS 20		20	
THEORY	40	MIN.PASS	MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PASS	MARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPE	CTIVE YEA	R	
ELIGIBILITY				
EXAMINATION	TERM END EXAMIN	VATION	MC	NTHLY TEST
DURATION	3 HR			1 HR
Objective:				
 vector. TO Learnget famiclassification. TO Learnapply the To Learn about the To learn about developments as illustrated Learning Out Come:- Affiliar with conditional of the principles of the principles of the contained of the principles of the contained of the principles of the contained of the principles of the prin	 TO Learnget familiar with concepts of artificial, natural and phylogenetic system of classification. TO Learnapply the principles of Principles and rules of ICBN in real situations. To Learn about the Botanical Nomenclature. To learn about develop Herbarium and identified on the bases of diversity of flowering plants as illustrated by members of the families. Learning Out Come:- After completion of the course, student-teachers will be able: Know about the Continuity and differentiability of vector functions, unit tangent vector. Get familiar with concepts of artificial, natural and phylogenetic system of classification. Apply the principles of Principles and rules of ICBN in real situations. Learn the Botanical Nomenclature. 			logenetic system of situations. iversity of flowering vill be able: tangent vector. of classification. plants as illustrated
• Angiosperm taxonomy: Brief history, aims and fundament concepts of artificial, natural and phylogenetic system of classification, Bentham & Hooker, Engler & Prantle an Hutchinson, system of classification. • Herbarium; Tools and techniques, important herbaria an botanical gardens of India and their importance.			ogenetic system of ler & Prantle and rtant herbaria and	
UNIT-2 TEACH ING HOURS (15)	• Botanical Nomenclature: Principles and rules of ICBN, type of concept, taxonomical categories, principle of priority, identification keys, floras.			
UNIT-3 TEACHI NG HOURS (15)				Rutaceae, and
UNIT-4 TEACH ING HOURS (15)	• Diversity of Flowerin families–Ranunculace Asteraceae, Solanace	eae, Apocynac	eae,	by members of the Asclepiadaceae,

UNIT-5 TEACH ING HOURS (15)	• Diversity of flowering plants as illustration families–Amaranthaceae, Cucurbitat Liliaceae , and Poaceae .	•	
TEACHING AND LEARNING STRATEGIES	 Lecture method Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be 		
	change as per requirement of the studer capabilities.		
CONTINUOUS & COMPREHENSIVE ASSESSMENT (CCA)	Details of Continuous and Comprehen are as follows:		
ASSESSMENT (CCA)	S. CCA- Components No.	Max. Marks Allocation	
	1.Monthly test2.Quizzes and Assignments	20*3 Test=60 10	
	3.Viva-voce4.Seminar/Symposia	10	
	5. Report writing	10	
	6. Workshop7. Review of literature	10	
	8. Creativity/Innovation	10	
	9. Experimental Skill	10	
	10. Co-curricular activity	10	
	11. Attendance	10 A original marks 30	
EXAMINATION PATTERN	Total 160 marks equivalent reduced to CCA original marks 30. Term-end examinations are organized by the university in the prescribed format to enable the scholars to achieve success in contemporary competitions and to achieve their goals.		
PERIODICAL REVISE OF SYLLABUS	 ANNUAL However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one months. 		
SELECTED READINGS	 Davis, P.H. and Heywood, V.H. (1963). Principles of Angiosperm Taxonomy. Oliver and Boyd, London. Heywood, V.H. and Moore, D.M. (eds.) (1984). Current Concepts in Plant Taxonomy. Academic Press. London. Jaffrey, C, (1982). An Introduction to Plant Taxonomy, Cambridge University Press, Cambridge, London. Jones, S.B., Jr. and Luchsinger, A.E. (1986). Plant Systematics (2nd edition). Mc Graw-Hill Book Co., New York. 		

 Maheshwari, J.K. (1963). Flora of Delhi, CSIR, New Delhi. Radrord, A.E. (1986). Fundamentals of Plant Systematics. Harper and Pow New York
 Harper and Row, New York. Singh, F. (1999). Plant Systematics: theory and Practice. Oxford and IBH Pvt. Ltd., New Delhi.

B.Sc.B.Ed. II Year				
COURSE CODE:			RSE T	YPE : CORE
COURSE TITLE :	Plant Cell Biology and			
MAX.MARKS:	50			
THEORY	40	MIN.PASS M	ARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PASS M	ARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAM	IINATION	N	IONTHLY TEST
DURATION	3 HR			1 HR
Objective:				
 To learn about get familiar with concepts of Chromosome organization and gen expression. To learn apply the principles of Mendal's law of inheritance, Linkage and crossing o allelic and non-allelic interactions in real situations. To learnabout the Nuclear and extra nuclear genetic material. Learning Out Come: After completion of the course, student-teachers will be able: To know about the Ultrastructure of cell and cell organelles. To apply the principles of Mendal's law of inheritance, Linkage and crossing over; all and non-allelic interactions in real situations. To learn the Nuclear and extra nuclear genetic material. To learn the Nuclear and extra nuclear genetic material. Expression (C) Ultrastructure of cell and cell organelles: Cell wall, Plasmembrane, Golgi Complex, endoplasmic reticulum, Lysoson Ribosomes, peroxisomes, vacuoles, mitochondria, chloroph nucleus. 			e and crossing over; be able: etic expression. crossing over; allelic Cell wall, Plasma iculum, Lysosomes, ondria, chloroplast,	
UNIT-2 TEACHIN G HOURS (15)	• <u>Chromosome organization</u> :Structure/Morphology, centromere and telomere; cHRomosome aberrations – deletions, duplication, inversions and translocations; Variations in cHRomosome number – aneuploidy, euploidy, sex cHRomosomes. Cell division; Cell cycle and Mitosis and meiosis.			
	• <u>Nuclear and extra nuclear genetic material</u> : DN structure;Replication; DNA protein interaction; nucleosome mod Genetic code; satellite and repetitive DNA; mitochondrial and plas DNA; plasmid; gene mapping.		; nucleosome model;	
UNIT-4 TEACH ING HOURS (15)	• <u>Gene expression</u> : Structure proteins. Structure and regulation of gene expression	classification; t	ranscri	iption; translation;

UNIT-5 TEACHING TEACHING VIS	 <u>Genetic inheritance</u>: Mendelism, Mendal's law of inheritance, Linkage and crossing over; allelic and non-allelic interactions. Genetic variations: Mutations - spontaneous and induced; transposable genetic elements; DNA damage and repair. Mean,Mode,Median, Standard deviation,Tabular and graphical representation of data table, histogram, Pie diagram, bar diagram, line graph Lecture method 				
LEARNING		blem Solving method			
STRATEGIES		aphical method			
		ninar/Symposia			
		view of literature			
	-	port writing			
		bup Discussion			
		leos/Animation			
		f-Learning/e-Learning orkshops/Experiments.			
		ACHING AND LEARNING STRAT	EGIES may be		
		per requirement of the students and	•		
CONTINUOUS		Continuous and Comprehensive Ass			
COMPREHENSIVE	follows:				
ASSESSMENT	S. No.	CCA- Components	Max. Marks		
(CCA)			Allocation		
	1.	Monthly test	20*3 Test=60		
	2. Quizzes and Assignments 10				
	3. Viva-voce 10				
	4.Seminar/Symposia105.Report writing10				
	5.Report writing106.Workshop10				
	7.Review of literature10				
	7.Review of interature108.Creativity/Innovation10				
	9.	Experimental Skill	10		
	10.	Co-curricular activity	10		
	11.	Attendance	10		
		marks equivalent reduced to CCA origi			
EXAMINATION		examinations are organized by the univ	• •		
PATTERN		enable the scholars to achieve suc	cess in contemporary		
PERIODICAL	1.ANNUA	ns and to achieve their goals.			
REVISE OF			vllabus at any time		
SYLLABUS	2.However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one				
	months	ë.			
SELECTED READINGS	 Alberts, B., Bray, D., J., Raff, M., Roberts, K and Watson, I. D. 2001. Molecular Biology of Cell Garland, publishing Co. Inc., New York, USA. Atherely, G., Girton, J. R. and McDonald, J. F. 1999. The Science of Genetics Saunders College Publishing, Fort Worth, USA. 				
	 Gupta, Rastog 				

Molecular Biology (3rd edition). Harper Collins College Publishers,
New York, USA.
• Lodish, H., Berk, A., Aipursky, S.L., Matsudaira, P., Baltimore, D.
and Darnell, J.2000. Molecular Cell Biology. W.H. Freeman & Co.
New York, USA. Russel, P. J. 1998. enetics. The
Benjamin/Cummings Publishing Co. Inc., USA.
• Snustad, D.P. and Simmons, M. J. 2000. Principles of Genetics. John
Wiley & Sons, Inc., USA.
• Stent, G.S. 1986. Molecular Genetics. CBS Publications, Delhi.

BOTANY-PRACTICALS:-II				
Duration: 4 hours	MAX.MARKS: 50	Min. Pass Marks : 20		
Practical List				

Gymnosperms:

1. Cycas

Habit, armour of leaf bases on the stem, very young leaf (circinate vernation) and old foliage leaves, scale leaf, bulbils, male cone, microsporophyll, megasporophyll, and mature seed. Study tHRough temporary micro preparations or permanent slides – normal root, coralloid root, stem, rachis, leaflet, microsporophyll, pollen grains, and ovule.

2. Pinus

Habit, long and dwarf shoot showing cataphylls and scale leaves, T.S. wood showing growth rings, male cones, female cones, and winged seeds.

Study tHRough temporary micro preparations and/or permanent slides – stem (young and old), needle, pollen grains, root, female cone, ovule, embryo (W.M.) showing polycotyledonous condition.

3. Ephedra

Habit and structure of whole male and female cones.

Temporary micro preparations and/or permanent slides of node, internode, macerated stem to see vessel structure; epidermal peel mount of vegetative parts to study stomata, male cone, female cone, and pollen grains.

4. Angiosperms

Familiarity with the terms used in technical description of plants. Study of representative plants of families included in the syllabus.

5. Cyto-genetics

- 1. Study of cell structure from onion leaf peels; demonstration of staining and mounting methods.
- 2. Comparative study of cell structure in onion cells, Hydrilla and Spirogyra. Study of cyclosis in Tradescantia, staminal hairs.
- 3. Study of plastids to examine pigment distribution in plants (e.g., Cassia, Lycopersicon and Capsicum).
- 4. Examination of electron micrographs of eukaryotic cells with special reference to organelles.
- 5. Study of electron micrographs of viruses, bacteria, cyanobacteria and eukaryotic cells for comparative cellular organization.
- 6. Examination of various stages of mitosis and meiosis using appropriate plant material

(e.g. onion root tips, onion flower buds),.

- 7. Preparation of karyo types from dividing root tip cells and pollen grains.
- 8. Cytological examination of special types of cHRomosomes: bar body, lamp brush and polytene cHRomosomes.
- 9. Working out the laws of inheritance using seed mixtures.
- 10. Working out the mode of inheritance of linked genes from test cross and or F2 data.
- 11. Mean, Mode, Median, Standard deviation, Tabular and graphical representation of data table, histogram, Pie diagram, bar diagram, line graph.

B.Sc.B.Ed. II Year				
COURSE CODE:	BSCBED-254e-I		COUR	SE TYPE :
				CORE
COURSE TITLE :	Abstract Algebra			
MAX.MARKS:	67	MIN.PASS		27
		MARKS		
THEORY EXAMINATION:	60	MIN.PASS		24
		MARKS		
CONTINUOUS	7	MIN.PASS		3
COMPREHENSIVE	MARKS			
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION MONTHLY TEST			ΓEST
DURATION	3 HR 1 HR			
Objectives				

• This course is designed to give students a foundation for all future mathematics courses. The fundamentals of algebraic problem-solving are explained. Students will explore: foundations of Algebraic structures, Groups, Rings, Ideals, Fields, Homomorphism etc. The course also fulfills the objective to make students aware of the applicability of abstract mathematics in real world problems.

Learning Outcomes: The course will enable the students to:

- Recognize the mathematical objects called groups.
- Link the fundamental concepts of groups and symmetries of geometrical objects.
- Explain the significance of the notions of cosets, normal subgroups, and factor groups.
- Analyze consequences of Lagrange's theorem.
- Learn about structure preserving maps between groups and their consequences.
- Know the fundamental concepts in ring theory such as the concepts of ideals, quotient rings, integral domains, and fields.
- Learn in detail about polynomial rings, fundamental properties of finite field extensions, and classification of finite fields.

UNIT-1 TEACHING HOURS (12)	• Symmetries of a square, Definition and examples of groups including dihedral, permutation and quaternion groups, Elementary properties of groups. Subgroups and examples of subgroups, Cyclic groups, Properties of cyclic groups, Lagrange's theorem, Euler phi function, Euler's theorem, Fermat's little theorem.
UNIT-2 TEACHIGN HOURS (12)	• Properties of cosets, Normal subgroups, Simple groups, Factor groups, Cauchy's theorem for finite abelian groups; Centralizer, Normalizer, Center of a group, Product of two subgroups; Classification of subgroups of cyclic groups. Cycle notation for permutations, Properties of permutations, Even and odd permutations, alternating groups, Cayley's theorem and its applications.
UNIT-3 TEACHING HOURS (12)	• Group homomorphisms, Properties of homomorphisms, Group isomorphisms, Properties of isomorphisms; First, second and third isomorphism theorems for groups.

UNIT-4 TEACHING HOURS (12)	• Definition, examples and elementary properties of rings, Commutative rings, Integral domain, Division rings and fields, Characteristic of a ring, Ring homomorphisms and isomorphisms, Ideals and quotient rings. Prime, principal and maximal ideals, Relation between integral domain and field, Euclidean rings and their properties, Wilson and Fermat's theorems.			
UNIT-5 TEACHING HOURS (12)	The d lemma domai Algeb	• Polynomial rings over commutative ring and their basic properties, The division algorithm; Polynomial rings over rational field, Gauss lemma and Eisenstein's criterion, Euclidean domain, principal ideal domain, and unique factorization domain. Extension of a field, Algebraic element of a field, Algebraic and transcendental numbers, Perfect field, Classification of finite fields.		
TEACHING AND LEARNING STRATEGIES	2. F 3. C 4. S 5. F 6. F 7. C 8. V 9. S 10. V * The T	 Lecture method Problem Solving method 		
CONTINUOUS	Details of Continuous and Comprehensive Assessment (CCA) are			
&COMPREHENSIVE		as follows: S. No. CCA- Components Max. Marks		
ASSESSMENT (CCA)	S. No.	S. No. CCA- Components Max. Marks Allocation		
	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	<u>11.</u>	Attendance	10	
		0 marks equivalent reduced to CCA original i		
EXAMINATION PATTERN	Term-end examinations are organized by the university in the			
	prescribed format to enable the scholars to achieve success in			
PATIERN	-			
	contemp	orary competitions and to achieve their goals		
PERIODICAL	contemp	orary competitions and to achieve their goals		
	contemp 1. A 2. H	orary competitions and to achieve their goals	the syllabus at	

SELECTED	• Herstein, (1975). Topics In Algebra: Wiley Eastern, New Delhi,
READINGS	• Khanna V.K. And S.K. Bhambri, (1998). A Course In Abstract
	Algebra: Vikas Pub. House, New Delhi, 2nd Rev. Ed. 1998.
	 Vashistha, A.R. (1971). Modern Algebra: Krishna Prakashan
	Mandir, Meerut, 2ndrev.Ed.
	• Artin, M. (1991) Algebra: Prentice Hall.
	• Bhattacharya, P. B., Jain, S. K. & Nagpaul, S.R. (2003). Basic
	Abstract Algebra (2 nd edition). Cambridge University Press.
	• David S. Dummit & Richard M. Foote (2008). Abstract Algebra
	(2nd edition). Wiley.
	• John B. Fraleigh (2007). A First Course in Abstract Algebra (7th
	edition). Pearson.
	 Joseph A. Gallian (2017). Contemporary Abstract Algebra (9th)
	edition). Cengage.
	• Gopalakrishnan, N. S. (1986). University Algebra, New Age
	International Publishers.
	• Thomas W. Hungerford (2004). Algebra (8th edition). Springer.
	• Nathan Jacobson (2009). Basic Algebra I & II (2nd edition). Dover
	Publications.
	• Serge Lang (2002). Algebra (3rd edition). Springer-Verlag.
	• Luthar, I. S. & Passi, I. B. S. (2013). Algebra: Volume 1: Groups.
	Narosa.
	• Luthar, I. S. & Passi, I. B. S. (2012). Algebra: Volume 2: Rings.
	Narosa.
	1101080.

	B.Sc. B.Ed. II	Year			
COURSE CODE:	BSCBED-254e-II COURSE TYPE : CO			E TYPE : CORE	
COURSE TITLE :	Real Analysis				
MAX.MARKS:	67	MIN.PASS MARKS 27			
THEORY	60	MIN.PASS M	ARKS	24	
EXAMINATION:					
CONTINUOUS	7	MIN.PASS M	ARKS	3	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80PERCENT IN RESPEC	CTIVE YEAR			
ELIGIBILITY					
EXAMINATION	TERM END EXAMI	NATION	MONT	HLY TEST	
DURATION	3 HR		-	I HR	
Objective	·				
	ned to provide a deeper as	nd rigorous un	derstanding	of fundamental	
0	ous functions, sequences and	•	•		
the Riemann integral	etc. The main focus of this	course will be	on theoretic	al foundation of	
the above said concep	ts and it will cultivate the ri	gorous mathem	atical logics	and skills in the	
students.					
Learning Outcomes: Th	is course will enable the stu	dents to:			
Understand many	properties of the real line R	and learn to de	fine sequenc	e in terms of	
functions from R	to a subset of R.				
Recognize bound	ed, convergent, divergent, C	auchy and mono	otonic seque	nces and to	
calculate their lim	it superior, limit inferior, an	d the limit of a	bounded seq	uence.	
• Apply the ratio, re	oot, and alternating series an	d limit compari	son tests for	convergence	
and absolute conv	vergence of an infinite series of real numbers.				
• Learn some of the	properties of Riemann integrable functions, and the applications of the				
fundamental theo	ems of integration.				
• Learn the Uniform	orm convergence of series of function and fundamental theorem of integral				
calculus, Mean va	ue theorems of integral calculus.				
	<u>Real Number System</u>	eal Number System			
	Algebraic and order propert	ies of R, Absol	lute value of	f a real number;	
1 [12]	Bounded above and bounde	d below sets, S	upremum an	nd infimum of a	
UNIT-1 TEACHINO HOURS (12	nonempty subset of \mathbb{R} , The	completeness p	property of l	R, Archimedean	
	property, Density of ration	al numbers in	R, Definition	on and types of	
	ntervals, Nested intervals	property; Neig	hborhood of	f a point in \mathbb{R} ,	
	Open, closed and perfect set	ts in R, Connec	ted subsets	of \mathbb{R} , Cantor set	
	and Cantor function.				
	Sequences of Real Number				
[1] [0] [0] [0] [0] [0] [0] [0] [0] [0] [0	Convergent sequence, Limi	-		-	
	heorems, Monotone sequ			-	
UNIT-2 EACHIG OURS (1	Subsequences, Bolzano-Weierstrass theorem for sequences, Lim superior and limit inferior of a sequence of real numbers, Cauch				
1	sequence, Cauchy's converg	ence criterion.			
	Infinite Series				
175 NG	Convergence and divergence		-		
	Necessary condition for con	-	•	-	
		ce of positive term series; Basic comparison to		_	
	Limit comparison test, D'A		•		
	Integral test; Alternating ser				
(convergence, Rearrangemen	t of series and R	tiemann's th	eorem.	

4 H N	Riemann Integ	ration	
NIT ACH NG DUR 12)	Riemann integral, Integrability of continuous and monotonic functions,		
UNIT-4 TEACH ING HOURS (12)		neorem of integral calculus, First	
DF H	Bonnet and We	ierstrass forms of second mean va	lue theorems.
ひ බ		ergence and Improper integral	
UNIT-5 TEACHING HOURS (12)		iniform convergence of sequence	and series of functions,
LI HORE	Weierstrass's N		TT : C
		and Abel's test for uniform	
HCH		d continuity, Uniform convergent	
TEACHING AND	Improper integrals, Dirichlet test and Abel's test for improper integrals. 1. Lecture method		
LEARNING		a Solving method	
STRATEGIES		al method	
STICTIONS	4. Seminar		
		of literature	
	6. Report v	writing	
	7. Group I		
	8. Videos/		
		arning/e-Learning	
		ops/Experiments.	
		ING AND LEARNING STRAT	
CONTINUOUS		requirement of the students and tinuous and Comprehensive Ass	-
COMPREHENSIVE	follows:	tinuous and comprehensive Ass	essment (CCA) are as
ASSESSMENT	S. No.	CCA- Components	Max. Marks
(CCA)			Allocation
	1.	Monthly test	20*3 Test=60
	2.	Quizzes and Assignments	10
	3.	Viva-voce	10
	4.	Seminar/Symposia	10
	5.	Report writing	10
	6.	Workshop	10
	7.	Review of literature	10
	8.	Creativity/Innovation	10
	<u>9.</u> 10.	Experimental Skill Co-curricular activity	10 10
	10.	Attendance	10
		s equivalent reduced to CCA origi	
EXAMINATION		minations are organized by t	
PATTERN		rmat to enable the scholars t	
	contemporary competitions and to achieve their goals.		
PERIODICAL	1. ANNUA	AL	
REVISE OF		VER, THE UNIVERSITYmay	•
SYLLABUS	-	e during the running year after	giving a notice for a
	•	one months.	The Transit I
SELECTED READINGS		(2004) Mathematical Analysis: N	New Age International,
NEADINGS	New Delhi.		
	-	1. (1985) Real Analysis: Narosa I	Publishing House, New
	Delhi.		
	• Royden, H.L	. (1993).Real Analysis: Macmillan	n, 4th Edition.

• Rudin, W. (1976). Principles Of Mathematical Analysis: Mcgraw Hill,
3rd Edition 1976.
• Robert G. Bartle & Donald R. Sherbert (2015). Introduction To Real
Analysis (4 th Edition). Wiley India.
• Gerald G. Bilodeau, Paul R. Thie & G. E. Keough (2015). An
Introduction To Analysis (2nd Edition), Jones And Bartlett India Pvt.
Ltd.
• Ross, K. A. (2013). Elementary Analysis: The Theory Of Calculus
(2nd Edition). Springer.

B.Sc. B.Ed. I YearI				
COURSE CODE:	BSCBED-254e-III COURSE TYPE : CORE			
COURSE TITLE :	Differential Equations			
MAX.MARKS:	66	MIN.PA	SS MARKS	27
THEORY	60	MIN.PA	SS MARKS	24
EXAMINATION:				
CONTINUOUS	6	MIN.PA	SS MARKS	3
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAMINA	ATION	MON	THLY TEST
DURATION	3 HR 1 HR			1 HR
Objective				
• The Objective of this course is to introduce ordinary and partial differential equations and				
fundamental theorems for existence and uniqueness. This course further explains the analytic				
techniques in computing the solutions of various differential equations appearing in various				
fields of science and technology.				
Learning Outcomes: The course will enable the students to:				
• Understand the genesis of ordinary differential equations.				

- Learn various techniques of getting exact solutions of solvable first order differential equations and linear differential equations of higher order.
- Know Picard's method of obtaining successive approximations of solutions of first order differential equations, passing tHRough a given point in the plane and Power series method for higher order linear equations, especially in cases when there is no method available to solve such equations.
- Grasp the concept of a general solution of a linear differential equation of an arbitrary order and also learn a few methods to obtain the general solution of such equations.
- Apply a range of techniques to solve first & second order partial differential equations.
- Model physical phenomena using partial differential equations such as the heat and wave equations.

	First Order Differential Equations
UNIT-1 ING HOURS (10)	Basic concepts and genesis of ordinary differential equations, Order and degree of
SS	a differential equation, Differential equations of first order and first degree,
	Equations in which variables are separable, Homogeneous equations, Linear
HC HC	differential equations and equations reducible to linear form, Exact differential
	equations, Integrating factor, First order higher degree equations solvable for x, y
	and p. Clairaut's form and singular solutions. Picard's method of successive
TEACH	approximations and the statement of Picard's theorem for the existence and
EA	uniqueness of the solutions of the first order differential equations. Orthogonal
E	trajectories.

	Second Order Linear Differential Equations
	Statement of existence and uniqueness theorem for linear differential equations,
	General theory of linear differential equations of second order with variable
	coefficients, Solutions of homogeneous linear ordinary differential equations of
(16	second order with constant coefficients, Transformations of the equation by
SS	changing the dependent/independent variable, Method of variation of parameters
35	and method of undetermined coefficients, Reduction of order, Coupled linear
UNIT-2 TEACHIGN HOURS (16)	differential equations with constant coefficients.
	Higher Order Linear Differential Equations
1)IH	Principle of superposition for a homogeneous linear differential equation, Linearly
VCI	
LE∢	dependent and linearly independent solutions on an interval, Wronskian and its
	properties, Concept of a general solution of a linear differential equation, Linear homogeneous and non-homogeneous equations of higher order with constant
	coefficients, Euler-Cauchy equation, Method of variation of parameters and
	method of undetermined coefficients, Inverse operator method.
	Series Solutions of Differential Equations
HI -3	Power series method, Legendre's equation, Legendre polynomials, Rodrigue's
UNIT-3 TEACHI NG HOURS	formula, Orthogonality of Legendre polynomials, Frobenius method, Bessel's
	equation, Bessel functions and their properties, Recurrence relations.
	First Order Partial Differential Equations
S	Family of surfaces in tHRee dimensions and formulation of partial differential
10	equations; Order and degree of Partial differential equations (PDE), Concept of
HC HC	linear and non-linear partial differential equations, Solution of quasi-linear partial
NIT- NG (10)	differential equations of the first order, Cauchy's method of characteristics;
	Partial differential equations of the first order, Lagrange's
UNIT-4 TEACHING HOURS (10)	method, Some special type of equation which can be solved easily by methods
LE∢	other than the
F	general method, Charpit's general method.
×	Second Order Partial Differential Equations
	Classification of linear partial differential equations of second order,
IOI	Homogeneous and non-homogeneous equations with constant coefficients. Second
UNIT-5 HING H (14)	Order Partial Differential Equations with Variable Coefficients, Partial differential
INU ING (14)	equations reducible to equations with constant coefficient, Second order PDE with
CH CH	variable coefficients, Classification of second order PDE, Reduction to canonical
UNIT-5 TEACHING HOURS (14)	or normal form; Monge's method; Solution of heat and wave equations in one and
	two dimensions by method of separation of variables.

TEACHIN	1. Lecture me	ethod		
GAND	2. Problem Solving method			
LEARNIN	3. Graphical method			
G	4. Seminar/Symposia			
STRATEGI	5. Review of	× 1		
ES	6. Report wri			
LO	7. Group Disc	•		
	8. Videos/An			
		ing/e-Learning		
	10. Workshop:			
		G AND LEARNING STRATEGIES n	now ha ahanga ag nan	
			hay be change as per	
CONTINU		ne students and their capabilities.	(CCA) and as follows:	
OUS		uous and Comprehensive Assessment (Max. Marks	
COMPRE	S. No.	CCA- Components		
	1		Allocation	
HENSIVE	1.	Monthly test	20*3 Test=60	
ASSESSM	2.	Quizzes and Assignments	10	
ENT (CCA)	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
	Total 160 marks e	quivalent reduced to CCA original marks	s 30.	
EXAMINAT		nations are organized by the univer		
ION		ble the scholars to achieve succe		
PATTERN		to achieve their goals.	····· ··· ··· ··· ··· ··· ··· ··· ···	
PERIODIC	1. ANNUAL			
AL REVISE		the Universitymay revise the syllabus	at any time during the	
OF		ear after giving a notice for a period of	•	
SYLLABUS	i unning y	cui unter grong a nouce for a period of	it monting.	
SELECTED	• Boyce, W. E	., Diprima, R. C., & Meade, D. B. (2017)). Elementary	
READINGS	Differential I	Equations. John Wiley & Sons.		
	• Mao, X. (199	94). Exponential Stability Of Stochastic I	Differential Equations.	
	Marcel Dekker.			
	• Hale, J. K. (1971). Functional Differential Equations. In Analytic Theory Of			
	Differential Equations (Pp. 9- 22). Springer, Berlin, Heidelberg.			
	 Sewell, G. (2005). The Numerical Solution Of Ordinary And Partial 			
	Differential Equations (Vol. 75). John Wiley & Sons.			
		d Partial Differential Equations (Vol. 103		
	•	4). Numerical Solution Of Ordinary And		
	· · · · ·	ased On A Summer School Held In Oxfo		
	1961. Elsevi		, rugast september	
		E. A. (2012). An Introduction To Ordina	ry Differential	
	•	ourier Corporation.	u y Differential	
1	Equations. C			

THIRD YEAR

Course code	ode Title of the course		EVALUATION		
		External	Internal	Practical	Total
BSCBED-350	ICT in Education*	70	30		100
BSCBED-351	Gender. School & Society	35	15		50
BSCBED-352	Creating an Inclusive School	35	15		50
BSCBED-353	Educational aspects of Geeta	35	15		50
BSCBED-354	Pedagogy of School subject-I				
BSCBED-354-I	Mathematics	35	15		50
BSCBED-354-II	General Science	35	15		50
BSCBED-354-III	Chemistry	35	15		50
BSCBED-354-IV	Biology	35	15		50
BSCBED-354-V	Physics	35	15		50
BSCBED-355	Content: (BCZ& PCM)				
BSCBED-355 a I	1. Physics I	40	10	-	1 = 0
BSCBED-355 a II	2. Physics II	40	10	- 50	150
BSCBED-355 b I	3. Chemistry I	40	10	50	150
BSCBED-355b II	4. Chemistry II	40	10	- 50	150
BSCBED-355 c I	5. Zoology I	40	10	50	150
BSCBED-355 c II	6. Zoology II	40	10		150
BSCBED-355 d I	7. Botany I	40	10	50	150
BSCBED-355 d II	8. Botany II	40	10		150
BSCBED-355 e I	9. Mathematics I	60	15		150
BSCBED-355e II	10. Mathematics II	60	15		150
	CCA				25
Prayer, Yoga, Meditation & Festival etc					25
Internship (4 Weeks)					50
Total					750

	B.Sc.B.Ed. III Year			
COURSE CODE:	BSCBED-330 COURSE TYPE: CORE		: CORE	
COURSE TITLE:	ICT in Education (Compulsory			
MAX.MARKS:	100	MIN. PASS	MARKS:	40
THEORY	70	MIN. PASS	MARKS:	28
EXAMINATION				
CONTINUOUS	30	MIN. PASS	MARKS:	12
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIV	E YEAR		
ELIGIBILITY			_	
EXAMINATION	TERM END EXAMINA	TION	MONTH	ILY TEST
DURATION	03 HR		01	HR
Objectives :				
• To enable to underst	and the Role of ICT in education.			
• To enable to underst	and Instructional Technology.			
• To understand about	the Multimedia techonolgy.			
• To enable to underst	and Database Management System			
• To know about Inter	net and it's uses.			
• Make to know about	E-Commerce and E-Governance.			
 developed softwa To train the teach Graduation/B.Ed. To acquaint the state To develop the skew 	cher educators to computer science as a subject at Secondary level and l. (Computing) level. students with Author ware package. kills of writing programs to analyse and process the statistical data. rstand and appreciate ICT as an effective learning tool for learners and as			
	• Computer Fundamentals:Intro	duction to 1	Information'	Technology,
UNIT-1 TEACHING HOURS	Generation of Computers, Types of computers: Micro, Mini, Mainframe, Super, Architecture of Computer System: CPU, ALU Primary Memory: RAM, ROM, Cache memory, Secondary Memories, Input/Output device, Pointing device. Number System (binary, octal, decimal and hexadecimal) and their conversions, Logic gates, Languages: machine, assembly and high-level languages including 3GL, 4GL.			
UNIT-2 TEACHING HOURS	 Word Processing packages: wrap, text formatting, para merge. Spreadsheet Packages: Ty calculations, formula and st charts, Sorting, searching, f 	graph formatt ype of entri atistical funct	ing, effect t es, Simple tions, Differ	to text, mail arithmetic

	\mathbf{D}_{1}
	• Power point: - Slide creation, slide show, adding graphics,
	formatting, customizing and printing.
S	• Multimedia techonolgy Introducing framework for multimedia
CI II	devices, imae compression standrsa, JPEG, MPEG, MIDI
Ō	formats.
UNIT-3 TEACHING HOURS	• Database Management System: Data, fileds and records,
	information database, creation of a database file, inserting,
D H	deletion and updating of records, modifying structure, editing
	and browsing of records, searching, sorting and indexing of
L	records.
75	• Concept of Operating System, need and types of operating
4 X X	systems: batch, single user, multiprocessing, and time sharing,
UNIT-4 EACHIN HOURS	introduction to Unix/Linux, Windows and its simple commands.
UN YOH	• Type of networks, LAN, MAN and WAN, concept of topology,
	bridges, routers, gateways, modems, ISDN leased lines,
	teleconferencing and videoconferencing.
S	• Internet: Concept, email services, www, web browsers, search
	engines, simple programs in HTML, type of HTML document,
Ō	documents structures: element, type and character formatting,
6-H 1-D	tables, frames and forms, E-mail.
UNIT-5 IFACHING HOURS	• E-Commerce: Concept of e-commerce, benefits and growth of e-
CH HO	commerce, e-commerce categories, e-Governance, EDI,
EA.	electronic funds transfer on EDI networks Electronic payment
H	system.
TEACHING AND	1. Lectures
LEARNING	2. E-learning
STRATEGIES	3. Videos
	4. Extension Lectures
	5. Content Review
	6. Self-Learning
	7. Group Discussions
	8. Field Visit
	9. Survey 10. Documentaries
	10. Documentaries 11. Short Films
	12. Team Teaching
	* The teaching strategies are subject to change as per requirement
	of the students and their capabilities.

CONTINUOUS&C	Details of Continuous and Comprehensive Assessment (CCA) are as				
OMPREHENSIVEA	follows:				
SSESSMENT (CCA)	SR. NO.	CCA: COMPONENT		MAXIMU MARKS	
		Manthle Tast		10X6 Test =	
	1	Monthly Test			
	2	Presentation		10	
	3	Group Discussion	10		
	4	Debate	10		
	5	Participation and Presen	10		
	6 7	Report Writing	10		
	-	Viva Voce	10		
	8Attendance*9Co-curricular Activity			10	
	-	Co-curricular Activity		10	
	10	Team Teaching			
		EXPLANATION (METHOD TO ASCERTAIN MARKS FOR CCA)			
	 CCA will be reduced to 30 marks or 15 marks (as per course weightage). Formula: Marks obtained/Total marksX30 For example: 60÷160X30 =11.25 PROVISO-I: Provided that a candidate shall be granted a relaxation in th form of exemption from CCA components, however, not more than 3 in a respective course. PROVISO-II: Provided further that this will be mandatory for a candidat appear in the monthly test conducted in the respective course. *Attendance in Lectures and Practical 				
	Percentage Marks Allotted				
		75% to 80%	02		
	81% to 85% 04				
		86% to 90%	06		
		91% to 95%	08		
		Above 96%	10		
EXAMINATION	Term-end examinations are organized by the university in the prescribe				
PATTERN	format to enable the scholars to achieve success in contemporary competitions and to achieve their goals.				
PERIODICAL	1. ANNUAL				
REVISION OF	2. HOWEVER, THE UNIVERSITY may revise the syllabus at				
SYLLABUS	any time during the running year after giving a notice for a				
		period one month.			
SELECTED	• Bott, E., Siechert, C., & Stinson, C. (2009). Windows 7 inside out.				
READINGS	 Pearson Education. Comer, D. E. (2018). The Internet book: everything you need to know about computer networking and how the Internet works. CRC Press. Emberton, D. J., & Hamlin, J. S. (2000). Flash 4 magic. New Riders 				
	Publishing.				
	C				
	• Geoghan, D. (2011). Visualizing Technology, Introductory. Delhi:				
	Pearson Higher Ed.				
	• Melton, B., Dodge, M., Swinford, E., & Schorr, B. (2013). <i>Microsoft</i>				
	Office Home and Student 2013 Step by Step. Pearson Education.				
	• Mohanty, L., & VoHRa, N. (2006). ICT strategies for schools:				
	• Monanty, L., & Vorna, N. (2000). ICI strategies for schools. F.				

guide for school administrators. SAGE Publishing India.
• Rathbone, A. (2012). <i>Windows 8 for dummies</i> . John Wiley & Sons.
• Saxena, J. (2008). Role of Ict& Total Quality Management in
Professional Education. New Delhi: APH Publishing Corporatio.
• Shaikh, I. R. (2013). Introduction to Educational Technology & ICT.
Tata McGraw-Hill Education.
• Solomon, G., & ScHRum, L. (2007). Web 2.0. New tools, new
schools. Eugene, Oregon, Washington, DC: ISTE.
• Solomon, G., & ScHRum, L. (2007). Web 2.0: New tools, new
schools. ISTE (Interntl Soc Tech Educ.

B.Sc.B.Ed. III Year				
COURSE CODE:	BSCBED-351	COURSE T	YPE: CORE	
COURSE TITLE:	Gender, School and Society	7		
MAX.MARKS:	50	MIN. PASS	MARKS:	20
THEORY	35	MIN. PASS	MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN. PASS	MARKS:	6
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPEC	TIVE YEAR		
ELIGIBILITY				
EXAMINATION	TERM END EXAMIN	ATION	MONTHI	LY TEST
DURATION	03 HR		01 H	HR
Objectives.				

Objectives:

- To describe the concept of gender and sex.
- To describe the social construction of gender.
- To understand different types of gender roles.
- To understand Analysis the gender relationship matrix.
- To identify the division of gender and valuation of work based on gender.

Learning outcome: After completion of the course, student-teachers will be able -

- Develop basic understanding and familiarity with key concepts–gender, gender bias, gender stereotype, empowerment, gender parity, equity and equality, patriarchy and feminism.
- Understand the gradual paradigm shift from women's studies to gender studies and some important landmarks in connection with gender and education in the historical and contemporary period.
- Learn about gender issues in school, curriculum and textual materials across disciplines, pedagogical processes and its intersection with class, caste, religion and region.
- Understand how gender, power and sexuality are related to education (in terms of access, curriculum and pedagogy).
- Develop an understanding of the paradigm shift from women studies to gender studies, based on the historical backdrop.
- Student to construct critically the impact of policies programmes and scheme for promotion of gender equality and empowerment.
- Apply the conceptual tools learnt regarding gender and sexuality to understand issues related to Sexual Harassment at the workplace and Child Sexual Abuse.
- Develop an understanding of different theories on gender and education and relate it to power relations. The institutions involved in socialization processes would be analyzed to see how socialization practices impact power relations and identity formation.
- Understand how gender relates to education and schooling. The students will be able to understand on how school as an institution addresses gender concerns in curriculum, textual materials and pedagogy. It will enable the student to draw linkages between life skills and sexuality.

	Gender Issues an Introduction
[]	• Meaning of Sex and Gender.
	Key concept of Gender Studies.
	• Purpose of Gender Studies.
	• Gender socialization and Gender Roles.
	• Gender discrimination at Social, Cultural, Religious, Economic,
01-1	Political, and Educational stage.
	Assignment:
UNIT-1 TEACHINGHOURS (12)	• Organize cultural seminars/symposia with school-students and prepare
	a report on gender equality.
AC	• Collect material related to Women Role Models in various fields with
LE	Emphasis on Women in Unconventional Roles and prepare a brief
	report.
	• Collect thoughts of Eminent Men and Women of India on Girls
	Education and Women's Empowerment.
	Gender Identities and Socialization Practices in
	Gender Identity: Definitions and concept.
<u>.</u>	• Gender Identity and Socialization Practices in Family, School, and
(17)	other formal and informal organizations.
SS SS	Social construction of Gender.
UNIT-2 TEACHING HOURS (12)	 Schooling of Girls: Inequalities and resistances.
HC HC	• Gender Concerns related to access, enrolment, retention, participation
	and overall achievement.
	Assignment:
H	• Collection of folklores reflecting socialization processes and drafts a
JA C	report on entire program.
II	• Analyze of textual materials from the perspective of gender bias and
	stereotype.
	• Find out the concept of women empowerment in ancient Indian
	culture and analyze its relevance at present scenario. Creating Gender Inclusive Classroom
[3]	Gender Inclusive Classroom-tips/activities/toolkit.
2	 Developing Positive Self-Concept and Self-Esteem among Students.
K	 Teaching-Learning Materials to teach Gender Issues.
o lo 0	 Classroom Transaction in relation to Gender.
H H	 Teacher as an Agent of Change in the Context of Gender and Society.
UNIT-3 TEACHING HOURS	Assignment:
	• Write a survey based report on financial allocations/field
G	conditions/policies/ imperatives of schools.
EA	 Field visits to schools, to observe the schooling process from a gender
H	perspective.
400	Gender Issues in Curriculum
UNIT-4 TEACHING HOURS (11)	• Gender and Education (Indian context): Socialization-theory and
N H S	Structural-theory.
	• In the Culture, Gender and Institution, Girls as Learners, Curriculum,
HC	Gender Culture and Hidden Curriculum, Gender- Education content
	and Construction of Knowledge.
	• Curriculum frame-work and Pedagogy based on gender issues.

	Assignment:		
	• Debate on women role models in various fields with emphasis on		
	women in unconventional roles.		
	 Prepare tools to analyze reflection of gender in curriculum and draft a 		
	report after administration of scoring and prepare a report. Report will		
	be presented in seminar.		
	Gender, Sexual Harassment and Legislative Action		
	 Meaning, Definition, Concept, types and identification of term Gender/Sexual harassment. 		
11	Institutions redressing sexual harassment and abuse.		
S	Prenatal Diagnostic Technique Act, 1994.		
	• The draft sexual Law Reforms in India-2000.		
0	Domestic Violence Act, 2005.		
	• Reservation for Women.		
	• Constitutional provisions against sexual harassment.		
UNIT-5 TEACHING HOURS (11)	Assignment:		
	• Gathering Information on Laws by Compiling Violence against		
E A	Women in India.		
	• Case study on how students perceive role models in their own lives.		
	• Draft a report with the help of field interview while studying the issue		
	of reservation as an equalitarian policy.		
	Group Assignment on Examining Policies and Schemes on Girls.		
TEACHING AND	1. Lectures		
LEARNING	2. E-learning		
STRATEGIES	3. Videos		
	4. Extension Lectures		
	5. Content Review		
	6. Self-Learning		
	 Group Discussions Field Visit 		
	9. Survey 10. Documentaries		
	10. Documentaries 11. Short Films		
	11. Short Films 12. Team Teaching		
	* The teaching strategies are subject to change as per requirement of the		
	students and their capabilities.		
	suuents anu men capabilities.		

CONTINUOUS & COMPREHENSIVE	Details of follows:	Continuous and Comprehensive Assessmen	t (CCA) are as
ASSESSMENT (CCA)	SR. NO.	CCA: COMPONENT	MAXIMUM MARKS
	1	Monthly Test	10X6 Test = 60
	2	Presentation	10110 1252 = 00
	$\frac{2}{3}$	Group Discussion	10
	4	Debate	10
	5	Participation and Presentation in Seminar	10
	6	Report Writing	10
	7	Viva Voce	10
	8	Attendance*	10
	9	Co-curricular Activity	10
	10	Team Teaching	10
	EXPLANA	ATION (METHOD TO ASCERTAIN MARK	S FOR CCA):
	Formula: N	be reduced to 30 marks or 15 marks (as per cours Marks obtained/Total marksX30	se weightage).
	PROVISC form of exc	 le: 60÷160X30 =11.25 D-I: Provided that a candidate shall be granted a memption from CCA components, however, not memption from CCA components, however, not memption from CCA components, however, not memory and the statement of the statement	
		course.)-II: Provided further that this will be mandatory n the monthly test conducted in the respective co	
		ice in Lectures and Practical	
	1 ittenau	Percentage Marks Allotted	
		75% to 80% 02	
		81% to 85% 04	
		86% to 90% 06	
		91% to 95% 08	
		Above 96% 10	
EXAMINATION	Term-end e	xaminations are organized by the university	in the prescribed
PATTERN	format to	enable the scholars to achieve success s and to achieve their goals.	-
PERIODICAL REVISION OF SYLLABUS	1. ANNUA 2. However		
	-		
SELECTED READINGS	Gender, Prakasha	ndramoulesh G K, Mr. Manjunath D R & Mr School and Society. (ISBN13:978938184 na. Kanpur. Pp.no188.	46728): Sirivara
	Formalis	arah M. 2013. "They Always Call Me an Investm m and Latino / a College Pathways. Gender & So 3213508308.)	
	• Trived, Publicati	Vinoti Ojha (2016). Gender school and son: C.	society. Agrawal
	identity:	(2010). Textbook regimes. A feminist critiqu New Delhi.	
	• A. banon Delhi.	n, Robent (2010). Social Psychology. Pearson	education: New

• Mathur, Savitri (2008). Sociological Foundation of Education. Kavita prakashan: Jaipur.
• Sidhu, Ramindra (2009). Sociology of Education. SHRi Sai Printo Graphers: New Delhi.
• Mudgal, S.D. (2007). Social Work Education Today and Tomorrow. Book Enclave: Jaipur.
• Nath, pramanik rathindra (2006). Gender Lhequality and women's empowerment. abhijeet publication: Delhi
• Malik, C.D (2008). Social and Political Thought B.R. Ambedkar. Arise Publishers and Distributors: New Delhi.
• Naik, S.C. (2005) Society and Environment. Oxford & 1B Publishing Co. Pvt.ltd: New Delhi.
• Runela, satypal (2009). Society of the Indian Education. Rajasthan Hindi Granth Akadmi: Jaipur.

		B.Sc.B.Ed. III Year	•		
COUF	RSE CODE:	BSCBED-352		COURSE TYP	E: CORE
COUF	RSE TITLE:	Creating an Inclusive School			
MAX.	MARKS:	50	MIN.	PASS MARKS:	20
THEC	ORY	35	MIN.	PASS MARKS:	14
EXAN	MINATION				
CONT	ΓINUOUS	15	MIN.	PASS MARKS:	6
COM	PREHENSIVE				
	SSMENT (CCA)				
1	ENDANCE	80 PERCENT IN RESPECTI	VE YE	AR	
	IBILITY				
	AINATION	TERM END EXAMINATI	ON	MONTHLY	
	ATION	03 HR		01 HI	R
Objec [®]					
•		or inclusive schools.	1		
•		ptual understanding of inclusive lerstanding of difference betw			Internated
•	Education and Incl	usive Education	ween S	pecial Education	, integrated
•		inderstanding of the recommen	dations	of various comm	nissions and
	committees toward	ls teacher preparation for inclusiv	ve educa	ation.	
•		standing for curriculum and peda			
•		nducive teaching learning enviro			
•		nducive Assessment and Evaluat			1.
		completion of the course, studen			
•		ot, meaning and significance of in nderstanding of the culture, po			need to be
-		to create an inclusive school.		ind procees that	need to be
•	Appreciate the nee	ed for promoting inclusive practic	ce and t	he roles and respo	nsibilities of
	the teachers.				
•	Develop critical understanding of the recommendations of various commissions and committees towards teacher preparation for inclusive education; understand the nature of			nissions and	
	difficulties encoun		ve euuc	ation, understand	
•		or inclusive schools.			
•		education, integrated education,	mainst	ream and inclusiv	ve education
	practices.	·	• 1	. ,.	
•		e existing resources for promoting			the might to
•	education of all lea	e attitude and sense of commit	ment to	warus actualizing	the fight to
•		ve teaching learning environment	in vari	ed school settings.	
•	• Develop the ability to conduct and supervise action research activities.				
	Introduct	tion, Issues & perspectives of Ir	nclusive	Education	
	• De	efinitions, Concept and Important	ce of In	clusive Education,	
(12		fference between Special Ed			ication and
SS		clusive Education.		,	
			for Ed	ucation for all Ch	ildron in the
HO H		lvantages of Inclusive Education			nuren in tile
UNIT-1 TEACHINGHOURS (12)		ntext of Right to Education.			
	Assignme			·	
CI		ganize a group discussion on D		-	I Education,
EA	Int	tegrated Education and Inclusive	Educat	on.	
E	• To	investigate the opinion of teach	ers on t	he integration of s	tudents with
	dis	sability in normal schools.			

	Concept & Policy Perspective
	• Recommendations of the Indian Education Commission (1964-66),
\sim	National Curriculum Framework, 2005 NCERT.
12	• The Convention on the Rights of the Child (Specific articles related to
RS	inclusive education).
5	• The World Declaration on the Survival, Protection and Development of
F-2 HC	Children.
UNIT-2 TEACHING HOURS 12)	Promoting Inclusion Preventing Exclusion.
5 Ż	• UNESCO Conventions, Declaration and Recommendations related to
H	Rights of persons with Disabilities.
AC AC	Assignment:
E	• To study the conceptions of teachers about the need of inclusive
	education in primary schools, collect views of teachers of school.
	Analyze in the light of inclusive education and write a report.
	• Organize a seminar on constitutional provisions on inclusive education.
[2]	Diversity in the Classroom
	Diversity due to disability: Nature, Characteristic and Needs.
	• Diversity due to Socio-Cultural and Economic factors: Discrimination, Language Attitudes, Violence and Abuse.
or or	 Concept, Nature, and Characteristics of Multiple Disabilities, Classroom
Ξ.	Management for Inclusive Education.
UNIT-3 NG HO	 MDGs (Millennium Development Goals) UNESCO.
UNIT-3 TEACHING HOURS (12)	Assignment:
C	• Conduct an awareness program on millennium goal of UNESCO.
EA	• Conduct a survey on the type of supportive service needed for inclusion
F	of children with any disability and share the findings in the class.
	Curriculum & Pedagogy in Inclusive School
UNIT-4 NG HOURS (11)	 Inclusive Curriculum- Meaning and Characteristics.
S	• Teaching and Learning Environment with special reference to Inclusive
10	School.
HO HO	• Individual differences, children with disabilities: Hearing Impairment,
UNIT-4 NG HO	visual Impairment, voice Impairment and orthopedic.
5 ž	Assignment:
H	• Planning and conducting multi level teaching in the persons with
TEACHI	disabilities (two classes).
LE	• Write a report on entire activity and present it in classroom presentation.
	(among peer group)
UNIT-5 TEACHING HOURS (11)	Assessment in Inclusive School
	• Alternative means for Assessment and Evaluation in Inclusive Classroom.
	 Utilization of records/case profiles for identification, assessment and
OU	intervention for Inclusive Classrooms.
E H	• Evaluation and follow-up Programmes for improvement of teacher
	preparation programmes in Inclusive Education.
	Assignment:
	• Discussion, group work and presentation by students on any topic of
	course.
	• Study the assessment and evaluation practice being followed in a school.
	Critically reflect on the practice in the context of inclusive education.
TEACHING	1. Lectures
AND	2. E-learning

I FADNINC	3. Videos		
LEARNING STRATEGI	4. Extension Lectures		
ES	 Extension Lectures Content Review 		
E9	6. Self-Learning		
	7. Group Discussions		
	8. Field Visit		
	9. Survey		
	10. Documentaries		
	11. Short Films		
	12. Team Teaching		
	* The teaching strategies are subject to change as per	requirement of the	
	students and their capabilities.	1	
CONTINUO US &	Details of Continuous and Comprehensive Assess follows:	sment (CCA) are as	
COMPREH	SR. CCA: COMPONENT	MAXIMUM	
ENSIVE	NO.	MARKS	
ASSESSME	1 Monthly Test	10X6 Test =	
NT (CCA)		60	
	2 Presentation	10	
	3 Group Discussion	10	
	4 Debate	10	
	5 Participation and Presentation in Seminar	10	
	6 Report Writing	10	
	7 Viva Voce	10	
	8 Attendance*	10	
	9 Co-curricular Activity	10	
	10 Team Teaching	10	
	EXPLANATION (METHOD TO ASCERTAIN MA)	RKS FOR	
	CCA):		
	CCA will be reduced to 30 marks or 15 marks (as per course		
	weightage).		
	Formula: Marks obtained/Total marksX30		
	For example: 60÷160X30 =11.25	l a relavation	
	PROVISO-I: Provided that a candidate shall be granted a relaxation in the form of exemption from CCA components, however, not more		
	than 3 in a respective course.		
	PROVISO-II: Provided further that this will be mandat	cory for a	
	candidate to appear in the monthly test conducted in the	-	
	course.		
	*Attendance in Lectures and Practical		
	Percentage Marks Allotted		
	75% to 80% 02		
	81% to 85% 04		
	86% to 90% 06		
	91% to 95% 08		
	Above 96% 10	1 1 1 0	
EXAMINA	Term-end examinations are organized by the university		
TION	to enable the scholars to achieve success in contempora	ary competitions and to	
PATTERN	achieve their goals.		

PERIODIC AL REVISION OF SYLLABUS	 ANNUAL However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one month.
SELECTED READINGS	 Siddiqui, Hena (2016). Inclusive education. Agraval Publication: Agra. Mitara, Krishna and Saxena, vandana (2008). Inclusive Issues and Perspectives. Arihant Publication: Jaipur. Sharma, P.L (2003). Planning Inclusive Education in Small Schools. R .I E. Mysore . Delpit, L.D. (2012). Multiplication is for white people: raising expectations for other people's children. The new press: USA. GOI, (1966). Report of the education commission: Education and national development. Ministry of education: New Delhi. Govinda R. (2011). Who goes to school? Exploring exclusion in Indian education. Oxford University Press: United Kingdom. Parekh, B.C. (2000). Rethinking multiculturalism: Cultural diversity and political theory. Palgrave: 213-230. UNESCO (2006). United Nations convention on the rights of persons with disabilities. UNESCO. (2009). Policy guidelines on inclusion in education UNESCO.

	B.Sc.B.Ed. III Year			
COURSE CODE:	BSCBED-353	-	COURSE TYPE	: CORE
COURSE TITLE:	Educational Aspects of the G	eeta		
MAX.MARKS:	50	-	PASS MARKS:	20
THEORY	35	MIN.	PASS MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN.	PASS MARKS:	6
COMPREHENSIVE				
ASSESSMENT (CCA	·			
ATTENDANCE	80 PERCENT IN RESPECTI	IVE YE	CAR	
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATI	ION	MONTHLY	
DURATION	03 HR		01 HR	
Objectives:				
	he philosophy of the Geeta.			
	lifferent Aspects of the Geeta Educa			
Ŭ	nificance of the teachnigs of the Ge		-	
Learning outcome:Af	ter completion of the course, studen	t-teache	ers will be able -	
Develop unders	tanding about the Meaning & Natur	e of Th	e Geeta Philosophy.	
• Understand imp	pact of the Geeta on Education.			
Understand Cor	ncept related to the Geeta Philosoph	v.		
	al theories with special reference to	•	eta in education	
-	ntribution of the Geeta in the variou			
		5 menus		
	Introduction of the Geeta			
	General Assumption and I			
	Need, Importance and Cor			1.
5 <u>C</u>		Different Aspects of the Geeta Education.		
IT-1 NHIN RS (1	Educational Aims and Plac		e Geeta.	
VIT- CHI JRS (Life philosophy in the Gee Educational Elements – 		r Student Teachin	ng Method
OC EA U	Curriculum.	Teache		ig Methou,
ΞĦ				
	Assignment:			
	• Conduct a drama to demonstrate life philosophy in the Geeta.			
	• By playing method show teacher student relationship according			p according
	to the Geeta. Philosophical Aspects			
RS	God, Human being, Nature	e. Unive	erse.	
DO	 Human life and Duties. 			
HŬ	Soul, Knowledge and Science.			
VG 12)	• Religion, Morality and Nishkam Karma (Service of Selflessness)			
UNIT-2 TEACHING HOURS 12)	Assignment:			
CH		tition	and human hair	notices and
EA	 Poster presentation competition on god, human being, nature a universe according to the Geeta darshan. 		, nature and	
F	 Essay competition on Nish 			Geeta)
	- Essay competition on Misi	ikalli Ka	anna (concept of the	(Occia)

UNIT-3 TEACHING HOURS (12)	 Social Aspects Man and its Social Nature. Social duty, Understanding and Coordination. Concept and Significance of Lok Sangrah (Public Collection). Importance of The Geeta in present Social Scenario. Social life skill and Management. Assignment: Conduct an awareness programme in society with the help of students on social duties/social life skill then assess the impact of program and present the report in class. Conduct one week orientation program in school on the Geeta Darshan and evaluate the effectiveness of the program tHRough examination.
UNIT-4 TEACHING HOURS (11)	 Psychological Aspects Nature of Man- Satvik, Rajashi & Tamasi. Nature, Types and forms of intelligence in the Geeta. Concept of mind in the Geeta. Guidance and motivation in the Geeta. Assignment: Conduct a nukkar natak, to demonstrate satvik, rajsi & tamsi nature of man.
UNIT-5 TEACHING HOURS (11)	 Multi-dimensional aspects and current significance of the Geeta Yoga and Spiritualism. Religion, Religious – Secularism, Peace and Ahinsa. Universal values and decision making system (Conviction). Cosmic order and symbolism in the Geeta, Expansion of all religions fellow feeling. Environmental Conservation. Assignment: Practice Surynamaskar with school students and explain them importance of the yoga. Organize a Drama/ play on universal values as described in The Geeta Darshan.
TEACHING AND LEARNING STRATEGIES	 Lectures E-learning Videos Extension Lectures Content Review Self-Learning Group Discussions Field Visit Survey Documentaries Short Films Team Teaching * The teaching strategies are subject to change as per requirement of the students and their capabilities.

CONTINUOUS &		of Continuous and Comprehensive Assessmen	t (CCA) are as	
COMPREHENSIVE	follows:			
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM	
(CCA)	NO.		MARKS	
	1	Monthly Test	10X6 Test =	
			60	
	2	Presentation	10	
	3	Group Discussion	10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	
		ANATION (METHOD TO ASCERTAIN MA		
	$\begin{bmatrix} \mathbf{E}\mathbf{A}\mathbf{F}\mathbf{L}\mathbf{F}\\ \mathbf{C}\mathbf{C}\mathbf{A} \end{bmatrix}$	MATION (METHOD TO ASCENTAIN MA		
	· · · · ·	ill be reduced to 30 marks or 15 marks (as per co	ourse	
	weighta	· 1	Juise	
	Ŭ Ŭ	a: Marks obtained/Total marksX30		
	For example: $60 \div 160X30 = 11.25$			
	PROVISO-I: Provided that a candidate shall be granted a relaxation			
	in the form of exemption from CCA components, however, not more			
	than 3 in a respective course.			
	PROVISO-II: Provided further that this will be mandatory for a			
	candidate to appear in the monthly test conducted in the respective			
	course.			
		dance in Lectures and Practical		
	1 itten	Percentage Marks Allotted		
		75% to 80% 02		
		81% to 85% 04		
		86% to 90% 06		
		91% to 95% 08		
		Above 96% 10		
EXAMINATION	Term-en	d examinations are organized by the university i	n the prescribed	
PATTERN		o enable the scholars to achieve success in	-	
		ions and to achieve their goals.	r contemporary	
PERIODICAL	1. ANN			
REVISION OF	 ANNUAL HOWEVER, THE UNIVERSITY may revise the syllabus at any time 			
SYLLABUS		• •	-	
SELECTED	 during the running year after giving a notice for a period one month. Radha Krishanan, S (2009). Bhagwat the Geeta. Hindi Pocket Books. 			
READINGS				
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Aayam. Sharda Pustak Bhawan: Allahabad

B.Sc.B.Ed. III Year				
COURSE CODE:	BSCBED-354-I COU		URSE TYPE: CORE	
COURSE TITLE:	Pedagogy of Mathen	natics		
MAX. MARKS:	50	MIN. PA	ASS MARKS:	20
THEORY EXAMINATION	35	MIN. PA	ASS MARKS:	14
CONTINUOUS	15 MIN. PASS MARKS:		6	
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END		MONTHI	AY TEST
	EXAMINATIO	DN		
DURATION	03 HR		01 H	łR
Objectives:				
The demonstrate of the second free demonstrates of the demonstrate in the second s				

- Understand the need for teaching-learning of Mathematics in secondary classes.
- Develop a critical understanding about the aims and objectives of Mathematics in a Democratic and Secular country.
- Understand the nature of Mathematics curriculum and its pedagogical issues.
- Critique and develop suitable evaluation mechanisms in Mathematics
- Develop the ability to organize co-curricular activities and community resources for promoting Mathematics learning.
- Understand the Approaches to teaching of mathematics
- Understand the Concept of Teaching Skills

Learning outcome: After completion of the course, student-teachers will be able -

- Develop insight into the meaning, nature, scope and objective of mathematics education.
- Learn important mathematics: mathematics is more than formulas and mechanical procedures.
- See mathematics as something to talk about, to communicate tHRough, to discuss among them-selves to work together on; Pose and solve meaningful problems.
- Appreciate the importance of mathematics laboratory in learning mathematics.
- Stimulate curiosity, creativity and inventiveness in mathematics.
- Develop competencies for teaching-learning mathematics tHRough various measures.

	Nature and Scope of Mathematics	
12	 Meaning, Nature and Scope of Mathematics teaching. 	
UNIT-1 ING HOURS(12)	• History of mathematics teaching and contribution of mathematician with reference to–Bhaskaracharya, Aryabhatta, Bamanuian Euclid Bithagaragata	
HC -T	Ramanujan, Euclid, Pythogores etc.	
N D	• Importance of Aims and Objectives of Mathematics Teaching.	
PZ	• Co-relation with other subjects of Mathematics.	
H	Assignment:	
TEAC	• Prepare a model of Pythagoras Theorem/Cube, Cuboid, and Cone etc.	
	• Organise seminar on contribution of mathematician.	

	Aims and Objectives of Teaching School Mathematics	
UNIT-2 TEACHING HOURS(12)	 Aims and general objectives of teaching mathematics, Bloom's Digital Taxonomy (Cognitive, Effective and Psychomotor) in terms of Instructional Behaviour, the objectives of school education; writing specific objectives of various content areas in mathematics like algebra, geometry, trigonometry, etc. Approaches to teaching of mathematics – Analytic-Synthetic, Inductive-Deductive, Heuristic, Problem Solving, Project and Laboratory Using various techniques of teaching mathematics viz-oral, written, drill, assignment, team teaching, supervised study and programmed learning. Assignment: Organise workshop on techniques of teaching mathematics. 	
	Organise a training program on Bloom's Digital Taxonomy.	
UNIT-3 TEACHING HOURS(12)	 Planning Concept, Meaning and Objectives of Mathematics teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these plans. Meaning and Concept of Teaching Skills. Micro Teaching-Meaning, Need and Importance of Micro Teaching Cycle and its Features. Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good Curriculum. Dale cone of experiences. Assignment: Prepare and present a lesson tHRough power point presentation on any topic of your choice. Organise a training program on Micro Teaching. 	
	Teaching-learning Resources in Mathematics	
UNIT-4 TEACHING HOURS(11)	 Teaching-learning Resources in Mathematics Meaning, Objectives, Scope, Characteristics, Types, Preparation, Presentation and Importance of Teaching Learning Material. Planning and Importance of Mathematics Laboratories and its uses. Qualities of Mathematics Text Books at Secondary Level. Audio-Visual Aids- Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the teaching of Mathematics. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of Mathematics. Make a report on Mathematics Teaching Planning and Importance of Mathematics Laboratories and its uses. 	
1) G	Assessment and Evaluation	
UNIT-5 TEACHING HOURS(11)	 Meaning, concept and construction of Achievement test, diagnostic test and remedial teaching. Types of Questions, Characteristics of a good test. Blue print: Meaning, concept, need and construction. Continuous and Comprehensive Evaluation: Meaning, concept, importance and limitations. Models of Teaching. 	

	Assign	ment:			
	• Construction, administration and interpretation				
	achievement test of any standard of school.				
	•				
		after discussion with concerning teacher an			
		measure.	6		
TEACHING AND	1.	Lectures			
LEARNING	2.	E-learning			
STRATEGIES	3.	Videos			
	4.	Extension Lectures			
	5.	Content Review			
	6.	Self-Learning			
	7.	Group Discussions			
	8.	Field Visit			
	9.	Survey			
	10.	Documentaries			
	11.	Short Films			
		Team Teaching			
		teaching strategies are subject to change as pe	er requirement		
		students and their capabilities.			
CONTINUOUS &		s of Continuous and Comprehensive Assessme	ent (CCA) are as		
COMPREHENSIV	follows	5:			
E ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM		
(CCA)	NO.		MARKS		
	1	Monthly Test	10X6 Test =		
			60		
	2	Presentation	10		
	3	Group Discussion	10		
	4	Debate	10		
	5	Participation and Presentation in Seminar	10		
	6	Report Writing	10		
	7	Viva Voce	10		
	8	Attendance*	10		
	9	Co-curricular Activity	10		
	10	Team Teaching	10		
	EXPL	ANATION (METHOD TO ASCERTAIN MA	RKS FOR		
	CCA):				
		vill be reduced to 30 marks or 15 marks (as per c	ourse		
	weight				
		la: Marks obtained/Total marksX30			
	For example: 60÷160X30 =11.25				
		ISO-I: Provided that a candidate shall be grante			
		m of exemption from CCA components, howeve	r, not more than		
		espective course. ISO-II: Provided further that this will be manda	tory for a		
		ate to appear in the monthly test conducted in the	-		
	course.		-		
	*Atten	dance in Lectures and Practical			
		Percentage Marks Allotted			
		75% to 80% 02			
		81% to 85% 04			

	86% to 90% 06
	91% to 95% 08
	Above 96% 10
EXAMINATION	Term-end examinations are organized by the university in the prescribed
PATTERN	format to enable the scholars to achieve success in contemporary
	competitions and to achieve their goals.
PERIODICAL	1. ANNUAL
REVISION OF	2. However, the Universitymay revise the syllabus at any time
SYLLABUS	during the running year after giving a notice for a period one
	month.
SELECTED	•धाकड़, परशुराम एवं त्रिवेदी, शिल्पा (२००९). गणित
READINGS	शिक्षण विधियाॅ. साहित्यागार चोड़ा रास्ताः जयपुर.
	•मंगल, एस.के. (२००५). गणित शिक्षण. आर्य बुक डिपो. नई दिल्ली.
	•शर्मा, एच.एस. (२००५). गणित शिक्षण. राधा प्रकाशन मन्दिरः आगरा.
	•नेगी, जे.एस. (२००७). गणित शिक्षण. विनोद पुस्तक मन्दिर. आगरा.
	 सिंह,योगेश कुमार (2010). गणित शिक्षण आधुनिक पद्धतियाँ. ए.पी.एच.पब्लिशिंग कॉरपोरेशनः नई दिल्ली-02. क्वरशेष्ट अप्रया करणप (2012) रुणिर शिक्षण अपर
	•कुलश्रेष्ठ, अरूण कुमार (२०१३). गणित शिक्षण. आर. लाल.बुक डिपोः मेरठ

B.Sc.B.Ed. III Year				
COURSE CODE:	BSCBED-354-II COUR CORE		RSE TYPE: E	
COURSE TITLE:	Pedagogy of General Scien	ice		
MAX. MARKS:	50	MIN. PA	ASS MARKS:	20
THEORY EXAMINATION	35	MIN. PA	ASS MARKS:	14
CONTINUOUS COMPREHENSIVE ASSESSMENT (CCA)	15	MIN. PA	ASS MARKS:	6
ATTENDANCE ELIGIBILITY	80 PERCENT IN RESPECTIVE YEAR			
EXAMINATION	TERM END EXAMINATION MONT		THLY TEST	
DURATION	03 HR			01 HR

Objectives:

- To develop the ability to Students insight on the meaning and nature of General science for determining aims and strategies of teaching- learning.
- To develop the ability to Students appreciate the fact that every child possesses curiosity about his/her natural surroundings.
- Students will be able to identify and relate everyday experiences with learning of science.
- Students will be able to integrate the science knowledge with other school subjects.
- Students will be able to analyze the contents of science with respecttopots, branches, process skills, knowledge organization and other critical issues.
- Students will be able to identify the concepts of science.

Learning outcomes: After completion of the course, student-teachers will be able to:-

- Develop insight on the meaning and nature of General science for determining aims and strategies of teaching-learning.
- Appreciate that science is a dynamic and expanding body of knowledge.
- Appreciate the fact that every child possesses curiosity about his/her natural surroundings.
- Identify and relate everyday experiences with learning of science.
- Appreciate various approaches of teaching- learning of science.
- Explore the process skill in science and role of laboratory in teaching-learning.
- Use effectively different activities /experiments /demonstrations /laboratory experiences for teaching-learning of science.
- Integrate the science knowledge with other school subjects.
- Analyze the contents of science with respecttopots, branches, process skills, knowledge organization and other critical issues.
- Develop process-oriented objectives based on the content themes/units.
- Identify the concepts of science that are alternatively conceptualized by teachers and students in general.

[2] 4 2 7	Nature and Scope of General Science Teaching			
UNIT-1 TEACHING HOURS(12)	 Meaning, Nature and Scope of General Science teaching. Contribution of Scientist:-Har Govind Khurana, J. C.Boss, C.V. Raman. Chander Shekhar, A.P.J.Kalam. Importance, Aims and Objectives of General Science Teaching. Co-relation with other Subjects Journal and Referenced Book and daily routine. Assignment: Group discussion on importance of General Science. Organise seminar on Contribution of Scientist in General Science and preapre a report. 			
UNIT-2 TEACHING HOURS(12)	 <u>Aims and Objectives</u> Bloom's Taxonomy (Cognitive, effective and psycho. motor) In terms of Instructional Behavior. Quality and responsibilities of Science teacher. Methods of General Science teaching Subject- lecture method, Demonstration, Lab Method, Problem Solving, Heuristics Project Method, Inductive and deductive Method. Techniques of General Science Teaching. Assignment: Organise workshop on techniques of teaching General Science. Organise a training program on Bloom's Digital Taxonomy. 			
UNIT-3 TEACHING HOURS (12)	 Planning Concept Meaning and Objectives Teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these Plans. Meaning and Concept of Teaching Skills- micro Teaching - Meaning. Need and Importance. Micro-Teaching Cycle and its features. Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good curriculum. Evaluation of General Science Syllabus at Secondary level Assignment: Prepare and present a lesson tHRough power point presentation on any topic of your choice. Organise a training program on Micro Teaching. 			
UNIT-4 TEACHING HOURS(11)	 <u>Teaching-learning Resources inGeneral Science</u> Meaning, Objectives, Scope, Characteristics, Types, Preparation, Presentation and Importance of Teaching Learning Material. Planning and Importance of General Science Laboratories and its uses. Qualities of General Science Text Books at Secondary Level. Audio-Visual Aids- Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the teaching of General Science. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of General Science. 			

	• Make a report on General Science Teaching Planning and Importance
	of General Science Laboratories and its uses.
UNIT-5 TEACHINGHOURS(11)	 <u>Assessment and Evaluation</u> Meaning, concept and construction of Achievement test, diagnostic test and remedial teaching. Types of Questions, Characteristics of a good test. Blue print: Meaning, concept, need and construction. Continuous and Comprehensive Evaluation: Meaning, concept, importance and limitations. Models of Teaching. Assignment: Construction, administration and interpretation of an achievement test of any standard of school. Make a diagnostic test of your subject and apply it in school, after discussion with concerning teacher and give remedial measure.
TEACHING AND LEARNING STRATEGIES	 Lectures E-learning Videos Extension Lectures Content Review Self-Learning Group Discussions Field Visit Survey Documentaries Short Films Team Teaching * The teaching strategies are subject to change as per requirement of the students and their capabilities.

CONTINUOUS &	Details follows	of Continuous and Comprehensive Assess	ment (CCA) are as		
COMPREHEN	SR.	CCA: COMPONENT	MAXIMUM		
SIVE	NO.		MARKS		
ASSESSMENT	1	Monthly Test	10X6		
(CCA)			Test = 60		
	2	Presentation	10		
	3	Group Discussion	10		
	4	Debate	10		
	5	Participation and Presentation in Seminar	10		
	6	Report Writing	10		
	7	Viva Voce	10		
	8	Attendance*	10		
	9	Co-curricular Activity	10		
	10	Team Teaching	10		
		ANATION (METHOD TO ASCERTAIN MA			
	CCA)				
	· · · · ·	• will be reduced to 30 marks or 15 marks (as per o	Culted		
	weight		Jourse		
	U U	ila: Marks obtained/Total marksX30			
	For example: 60÷160X30 =11.25 PROVISO-I: Provided that a candidate shall be granted a relaxation in the form of exemption from CCA components, however, not more than 2 in a respective course				
	3 in a respective course. PROVISO-II: Provided further that this will be mandatory for a candidate to appear in the monthly test conducted in the respective course.				
	*Attendance in Lectures and Practical				
		PercentageMarks Allotted75% to 80%02			
		81% to 85% 04			
		86% to 90% 06 91% to 95% 08			
		Above 96% 10	• • • • • •		
XAMINATIO		nd examinations are organized by the univers	•		
PATTERN		to enable the scholars to achieve succe	ss in contemporary		
		itions and to achieve their goals.			
RIODICAL	1. ANN				
VISION OF		ever, the Universitymay revise the syllabus at	•		
LLABUS	running year after giving a notice for a period one month.				
LECTED	• Solomon, J., & Aikenhead, G. (1994). STS Education: International				
EADINGS	Pers	pectives on Reform. Ways of Knowing Science	Series. NY: Teachers		
		lege Press.			
	• Lawson, A. E. (1995). Science teaching and the development of thinking.				
			elopment of thinking.		
		nont, CA: Wadsworth.	elopment of thinking.		
	Belr				
	Belr • Ellis	mont, CA: Wadsworth.	A Materials Science		
	Belr • Ellis Con	mont, CA: Wadsworth. s, A. B. (1993). Teaching General Chemistry: npanion. American Chemical Society, Distribution	A Materials Science on Office Department		
	Belr • Ellis Con 225,	mont, CA: Wadsworth. s, A. B. (1993). Teaching General Chemistry:	A Materials Science on Office Department		

Ltd.
• Hodson, D. (2009). Teaching and learning about science: Language,
theories, methods, history, traditions and values. Brill Sense.
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science laboratory (Vol. 16). Springer Science & Business Media.
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Publishers.
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PublishersEdger, Marlow & Rao, D.B. (2003). Teaching Scioence
Successfully.New Delhi: Discovery Publishing House.

	B.Sc.B.Ed. III Yea	ır			
COURSE CODE:	BSCBED-354-III	-	COURSE TYPE	C: CORE	
COURSE TITLE:	Pedagogy of Chemistry				
MAX. MARKS:	50	MIN.	PASS MARKS:	20	
THEORY	35	MIN.	PASS MARKS:	14	
EXAMINATION					
CONTINUOUS	15	MIN. PASS MARKS: 6			
COMPREHENSIVE					
ASSESSMENT (CCA) ATTENDANCE	80 PERCENT IN RESPECT	IVF V	FAD		
ELIGIBILITY	OUTERCENT IN RESPECT				
EXAMINATION	TERM END EXAMINATIO)N	MONTHLY T	TEST	
DURATION	03 HR		01 HR		
Objectives:					
• To understand abil	ity to gain insight on the meaning	ng and	nature of chemistry.		
• To develop ability	to determining aims and strateg	gies of t	eaching-learning.		
• To develop abilit	y to use effectively differen	t activi	ties/ demonstration/	laboratory	
-	ching-learning of chemistry.			•	
	ity to integrate in chemistry kno	owledge	e with other school sul	ojects.	
	er completion of the course, stud	U U		0	
-	e meaning and nature of chemis				
of teaching-learnin	•	<u> </u>	6	0	
Ŭ	ence is a dynamic and expandir	ıg body	of knowledge.		
	fact that every child posses		-	er natural	
surroundings.	all that every ennu posses		moonly about ms, m	or matural	
Ŭ	everyday experiences with lear	ning ch	emistry		
•	approaches of teaching-learning	-	•		
**	process of science and role of labo	•	•	tuations	
_	ifferent activities/demonstration	-			
learning of chemis		11/10012	uory experiences for	teaching-	
Ũ	•	ol aubio	ota		
-	stry knowledge with other school	51 subje	cis.		
	<u>f Chemistry Teaching</u> Meaning, Nature and Scope of C	Thomist	ry toophing		
Ξ	Nature of Science with special r		•	<u>C1</u>	
	 Nature of Science with special reference to chemistry. History and Contribution of Chemistry teaching, history of Chemistry with special reference to India. Importance Aims and Objectives of Chemistry teaching Objective of teaching Chemistry at secondary/senior secondary level. Co-relation with other Subjects Journal and Referenced Book. Assignment: Group discussion on importance of Chemistry. 				
	 with special reference to India. Importance Aims and Objectives of Chemistry teaching Objective of 				
I • I • I • I • I • I • I • I • • I	1 0		•	bjective of	
	eaching Chemistry at secondary		•		
	Co-relation with other Subjects	Journal	and Referenced Book		
Assignm		a ~-			
	Group discussion on importance		•	_	
	Organise seminar on Contribution of Scientist in Chemistry and preap			nd preapre	
8	a report.				

	Instructional objectives and methods
5	• Instructional objectives and methods Bloom's Taxonomy (Congnative, effective and psychomotor).
S (1	 In terms of instructional behavior Methods of Chemistry teaching
L K	Subject · lecture method. Demonstration Method, lab based method.
HO HO	• Inductive & deductive method. Problem Solving. Heuristics & Project
IN 5	Method Techniques of Chemistry Teaching Approaches of Chemistry
UNIT-2 TEACHING HOURS (12)	teaching- Inquiry approach, programmed instruction, Group discussion,
ACI	team teaching, CAL, SEMINARS & WORKSHOP.
LE	Assignment:
	Organise workshop on techniques of teaching Chemistry.
	Organise a training program on Bloom's Digital Taxonomy.
	Planning
	• Concept, Meaning and Objectives of Chemistry Teaching Plan (Lesson
[13]	Plan, Unit Plan, Yearly Plan) and Preparation of these Plans.Meaning and Concept of Teaching Skills. Micro Teaching - Meaning,
ßS	 Meaning and Concept of Teaching Skins. Micro Teaching - Meaning, Need and Importance, Micro-Teaching Cycle and its features.
001 01	 Concept, Meaning, Principles and Objectives of Curriculum,
	Characteristics of good curriculum and Evaluation of Chemistry
	Syllabus at Secondary Level.
	Assignment:
UNIT-3 TEACHING HOURS (12)	• Prepare and present a lesson tHRough power point presentation on any
E E	topic of your choice.
	• Organise a training program on Micro Teaching.
	Instructional Support System
	• Meaning, Objectives, Scope, Characteristics, Types, Preparation,
	Presentation and Importance of Teaching Learning Material.Dales' Cone of Experiences. Planning and Importance of Chemistry
[1]	• Dates Cone of Experiences. Planning and Importance of Chemistry Laboratories and Its uses. Qualities of good Chemistry Text Books at
IRS	Secondary Level.
10H	• Qualities and Characteristics Chemistry Teacher. Audio-Visual Aids -
UNIT-4 ING HO	Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the Teaching of Chemistry. Utilization of
	Community Recourses in the Teaching of Chemistry Teaching.
ACI	Assignment:
UNIT-4 TEACHING HOURS (11)	• Conduct a training program on use of Different types of Audio-Visual
	Aids in the teaching of Chemistry.Make a report on Chemistry Teaching Planning and Importance of
	Chemistry Laboratories and its uses.

	F 1				
	Evalua				
(11)		Meaning & Objective of Evolution.			
S		Types of Test Items and their Construction.			
5					
UNIT-5 TEACHING HOURS (11)		Characteristics of a good Test. Concept and Preparation of Diagnostic Test. Remedial Teaching and Envision and Preparation.			
		Test, Remedial Teaching and Enrichment Programme.			
5 <u>ž</u>		Use of ICT: Video clips, Power points presentations, films etc.			
H		signment:			
AC AC		Construction, administration and interpretation of	of an achievement test		
E		of any standard of school.	. 1		
		Make a diagnostic test of your subject and ap			
	1	discussion with concerning teacher and give remo	ediai measure.		
TEACHING AND		Lectures			
LEARNING		E-learning Videos			
STRATEGIE		Extension Lectures			
S		Content Review			
5					
		6. Self-Learning7. Group Discussions			
		8. Field Visit			
		. Survey			
		10. Documentaries			
	11. Short Films				
	12.7	Team Teaching			
	* The te	eaching strategies are subject to change as per	requirement of the		
		s and their capabilities.	-		
CONTINUOU	Details	of Continuous and Comprehensive Assess	ment (CCA) are as		
S &	follows:				
COMPREHE	SR.	CCA: COMPONENT	MAXIMUM		
NSIVE	NO.		MARKS		
ASSESSMEN	1	Monthly Test	10X6		
T (CCA)			Test = 60		
	2	Presentation	10		
	3	Group Discussion	10		
	4	Debate	10		
	5	Participation and Presentation in Seminar	10		
	6	Report Writing	10		
	7	Viva Voce	10		
	8	Attendance*	10		
	9	Co-curricular Activity	10		
	10	Team Teaching	10		

	EXPLANATION (METHOD TO ASCERTAIN MARKS FOR		
	CCA will be reduced to 30 marks or 15 marks (as per course		
	weightage).		
	Formula: Marks obtained/Total marksX30		
	For example: $60 \div 160X30 = 11.25$		
	PROVISO-I: Provided that a candidate shall be granted a relaxation in		
	the form of exemption from CCA components, however, not more than		
	3 in a respective course.		
	PROVISO-II: Provided further that this will be mandatory for a		
	candidate to appear in the monthly test conducted in the respective		
	course.		
	*Attendance in Lectures and Practical		
	Percentage Marks Allotted		
	75% to 80% 02		
	81% to 85% 04		
	86% to 90% 06		
	91% to 95% 08		
	Above 96% 10		
EXAMINATI	Term-end examinations are organized by the university in the prescribed		
ON	format to enable the scholars to achieve success in contemporary competitions		
PATTERN	and to achieve their goals.		
PERIODICA	1. ANNUAL		
IERIODICA	I. ANNUAL		
I DEVISION	2 However, the University may revise the syllabus of any time during the		
L REVISION	2. However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one month		
OF	2. However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one month.		
OF SYLLABUS	running year after giving a notice for a period one month.		
OF SYLLABUS SELECTED	 running year after giving a notice for a period one month. Anderson, R. G. W. (1978). The Playfair Collection and the teaching of 		
OF SYLLABUS	 running year after giving a notice for a period one month. Anderson, R. G. W. (1978). The Playfair Collection and the teaching of chemistry at the University of Edinburgh, 1713-1858. Brill. 		
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B.Sc.B.Ed. III Year				
COURSE CODE:	BSCBED-354 -IV COURSE TYPE: CORE			
COURSE TITLE:	Pedagogy of Biology			
MAX. MARKS:	50	MIN	. PASS MARKS:	20
THEORY	35	MIN	. PASS MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN. PASS MARKS: 6		6
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION MONTHLY TEST			TEST
DURATION	03 HR 01 HR			
Objectives:				
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• To understand the ability to develop insight on the meaning and nature of biological science.

• To understand the ability to integrate the biological science knowledge with other school subjects.

• Develop the ability be to identify and relate everyday experiences with learning of biological science.

• To understand the ability to appreciate various approaches and methods of teachinglearning of biological science.

• Develop the ability to explore the process skill in science and role of laboratory in teaching-learning.

• Develop the ability to identify the concepts of biological science that are alternatively conceptualized by teachers and students in general.

Learning outcome: After completion of the course, student-teachers will be able -

- Develop insight on the meaning and nature of biological science for determining aims and strategies of teaching- learning.
- Integrate the biological science knowledge with other school subjects. Identify and relate everyday experiences with learning of biological science.
- Appreciate various approaches and methods of teaching- learning of biological science.
- Explore the process skill in science and role of laboratory in teaching- learning.
- To understand meaning, concept and various types of assessment.
- Identify the concepts of biological science that are alternatively conceptualized by teachers and students in general.

	Basics of Biology Teaching
UNIT-1 NG HOURS (12)	 Meaning, Nature and Scope of Biology teaching. Main discoveries and development in Biology Place and Values of Teaching Biology in School level. Correlation of Biology and other Subjects. Objectives of teaching Biology at School Level. Assignment:
C CI	• Group discussion on importance of Biology teaching.
TEAC	• Organise seminar on Contribution of main discoveries and development in Biology and preapre a report.

	Instructional objectives and methods
UNIT-2 TEACHING HOURS (12)	 Bloom's Taxonomy (Cognitive, effective and psychomotor), In terms of Instructional behavior. Methods of Biology teaching Subject- lecture method, Demonstration Method, Inductive & deductive method, Problem Solving, Heuristics & Project Method. Inquiry approach programmed Instruction, Group discussion Self Study team teaching, Seminar and workshops. Assignment: Organise workshop on techniques of Biology teaching. Organise a training program on Bloom's Digital Taxonomy.
	Planning
UNIT-3 TEACHING HOURS (12)	 Concept, Meaning and Objective of Biology Teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these Plans. Meaning and Concept of Teaching Skills. Micro Teaching - Meaning, Need and Importance, Micro-Teaching Cycle and its features. Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good curriculum and Evaluation of Biology Syllabus at Secondary Level. Assignment: Prepare and present a lesson tHRough power point presentation on any topic of your choice. Organise a training program on Micro Teaching.
	Instructional Support System
UNIT-4 TEACHING HOURS (11)	 Meaning, Objectives, Scope, Characteristics, Types, Preparation, Presentation and Importance of Teaching Learning Material. Dales' Cone of Experiences. Planning and Importance of Biology Laboratories and Its uses. Qualities of good Biology Text Books at Secondary Level. Qualities and Characteristics Biology Teacher. Audio-Visual Aids - Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the Teaching of Biology. Utilization of Community Recourses in the Teaching Biology Teaching. Use of ICT: Video clips, Power points presentations, films etc. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of Biology. Make a report on Biology Teaching Planning and Importance of <i>Chemistry</i> Laboratories and its uses.
UNIT-5 TEACHING HOURS(11)	 Evaluation Meaning and Objectives of Evaluation. Types of Test Items and their Construction. Preparation of Blue-Print and Achievement Test. Characteristics of a good Test. Concept and Preparation of Diagnostic Test, Remedial Teaching and Enrichment Programme. Assignment: Construction, administration and interpretation of an achievement test of any standard of school. Make a diagnostic test of your subject and apply it in school, after discussion with concerning teacher and give remedial measure.

TEACHING AND	1. I	Lectures	
LEARNING		E-learning	
STRATEGIES		Videos	
		Extension Lectures	
		Content Review	
		Self-Learning	
		Group Discussions	
		Field Visit	
		Survey	
		Documentaries	
		Short Films	
		Feam Teaching	
		aching strategies are subject to change as per	requirement
		udents and their capabilities.	requirement
CONTINUOUS &		of Continuous and Comprehensive Assessmen	t (CCA) are as
COMPREHENSIVE	follows:	-	are as
ASSESSMENT			MANIMINI
(CCA)	SR. NO.	CCA: COMPONENT	MAXIMUM MARKS
	1	Monthly Test	$\frac{10X6 \text{ Test}}{10}$
	1	Wontiny Test	$\frac{10X0}{60}$
	2	Presentation	10
	$\frac{2}{3}$	Group Discussion	10
	4	Debate	10
	5		10
	<u>5</u> 6	Participation and Presentation in Seminar	10
	0 7	Report Writing Viva Voce	
			10
	8	Attendance*	10
	9	Co-curricular Activity	10
	10 EVDI	Team Teaching	10 DVS FOD
	CCA):	ANATION (METHOD TO ASCERTAIN MA	KKSFUK
		······································	
		vill be reduced to 30 marks or 15 marks (as per co	ourse
	weighta	age).	ourse
	weighta Formul	age). a: Marks obtained/Total marksX30	ourse
	weighta Formul For exa	age). a: Marks obtained/Total marksX30 mple: 60÷160X30 =11.25	
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EXAMINATION PATTERN	weight: Formul For exa PROV in the f than 3 i PROV candida course. * Atten	age). a: Marks obtained/Total marksX30 ample: $60 \div 160X30 = 11.25$ ISO-I: Provided that a candidate shall be granted orm of exemption from CCA components, howe in a respective course. ISO-II: Provided further that this will be mandate that the to appear in the monthly test conducted in the dance in Lectures and Practical Percentage Marks Allotted 75% to 80% 02 81% to 85% 04 86% to 90% 06 91% to 95% 08	a relaxation ver, not more cory for a respective n the prescribed

	competitions and to achieve their goals.		
PERIODICAL	1. ANNUAL		
REVISION OF	2. However, the Universitymay revise the syllabus at any time		
SYLLABUS	during the running year after giving a notice for a period one		
	month.		
SELECTED READINGS	 Agarwal, D.D. (2004). Modern methods of Teaching Biology. Saruk &Sons: New Delhi. Miller, David F. & blaydes, Gllenn W. (1938). Methods and materials for teaching biological sciences. Mc GRAW Hill book company Inc: New York and London. Choudhary, S. (2010). Teaching of Biology. APH Publishing Corporation: New Delhi. Sood, J.K. (1987). Teaching of Life Science. Kohli publishers: Chandigarh. Yadav, M.S. (2000). Modersn Methods of Teaching Science. Anmol Publishers: New Delhi. Bhar, Suraj prakash (2006). Teacher Training Lotus Press: New Delhi. Singh, Veena (2007). Teaching of Biology. Adhyanyan Publishers & Distributors: New Delhi. मंगल, एस.क. (2010). जैविक विज्ञान शिक्षण. लायल बुक डिपो: मेरठ. भूषण, शैलेन्द्र (2008). जीव विज्ञान शिक्षण. विनोद पुस्तक मन्दिर: आगरा. 		
	 कुलश्रेष्ठ, एस.पी. (2005). जीव विज्ञान शिक्षण. लायल बुक डिपोः मेरठ. गानेपत्री ती के (2002) जीव विज्ञान शिक्षण. गर्ण प्रतिप्रकेषनः फेरन 		
	• माहेश्वरी, बी.के. (2003). जीव विज्ञान शिक्षण. सूर्या पब्लिकेशनः मेरठ.		
	 रावत एवं अग्रवाल (2014). नवीन विज्ञान शिक्षण. श्री विनोद पुस्तक मन्दिरः आगरा. 		
	अगरा. • श्रीमाली, भूषण एवं रिहानी (2013). विज्ञान शिक्षण. राजस्थान हिन्दी ग्रन्थ अकादमीः जयपुर		

	B.Sc.B.Ed.III Y	ear			
COURSE CODE:	BSCBED-354 -V	cui	COURSE TY	PE: CORE	
COURSE TITLE:	Pedagogy of Physics		COURSETT		
MAX. MARKS:	50	MIN	PASS MARKS:	20	
THEORY	35		PASS MARKS:	14	
EXAMINATION	55	10111 1.		14	
CONTINUOUS	15	MIN	PASS MARKS:	6	
COMPREHENSIVE	15	10111 4.	I ASS WARD.	0	
ASSESSMENT					
(CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR				
ELIGIBILITY	OUT ERCENT IN RESI EC				
EXAMINATION	TERM END EXAMINAT	TION	MONTHI	VTFST	
DURATION	03 HR		01 H		
Objectives:	03 IIK		011		
0	e need for teaching-learning of	Physics	in secondary class	Ses	
	ritical understanding about th				
Democratic and S		ie units	una objectives c	i i i i joico ili a	
	e nature of Physics curriculum	and its r	bedagogical issues		
	ritique and develop suitable eva				
	bility to organize co-curricula				
promoting Physic				5	
	e Approaches to teaching of Ph	iysics			
	e Concept of Teaching Skills	2			
	er completion of the course, s				
	the meaning and nature of Pl	hysics s	cience for determ	ining aims and	
strategies of teach					
	e everyday experiences with le				
	us approaches of teaching-learn				
	different activities/demonstrati	ons/labo	oratory experience	es for teaching-	
learning of Physi		1.1.			
	ics knowledge with other school	of subject	its;		
	 <u>Nature and Scope of Physics</u> Meaning, Concept, Nat 	ura Sac	no Dhysics Touchi	na	
<u>ප</u> බ	Weaning, Concept, NatContribution of Indian				
	S.N.Bose, H.J.Bhabha,			man, J.C.Dose,	
	 Correlation of Physics 				
UNIT-1 TEACHING HOURS (12)	Assignment:	with oth	er benoor buojeett	•	
HO HO	• Organise workshop on	Contrib	ution of Indian scie	entist.	
	• Write a report on Obj				
	level of School.		Jan	8	
5.5	Teaching-learning of Physics	;			
E	• Aims and general ob	-	of teaching ph	ysics, Bloom's	
N SS	Digital Taxonomy (Co	•		-	
15	terms of Instructional	-		-	
OH	education; writing spec				
UNIT-2 TEACHING HOURS (12)	Physics.	- J			
ž	• Approaches to teach	ing of	Physics - And	lytic-Synthetic	
E	Inductive-Deductive, H	-	· · · · · · · · · · · · · · · · · · ·	• •	
	Laboratory.	iouristic		is, i toject and	
E	•	ues of t	anching Physics	viz oral writton	
F					
	drill, assignment,	Team	teaching, supe	ervised study	

	and programmed locating
	andprogrammed learning. Assignment:
	• Prepare a lesson plan based on team teaching and execute it in
	school.
	• Make any two teaching aids with the low cost material.
	<u>Planning</u>
UNIT-3 TEACHING HOURS (12)	 Concept, Meaning and Objectives of Physics teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these plans Meaning and Concept of Teaching Skills, Maxims of Teaching Micro Teaching-Meaning, Need and Importance of Micro Teaching Cycle and its Features Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good Curriculum. Dale cone of experience.
	Organsie training program on Micro Teaching.
	Prepare a working model based on Dale cone of experience.
UNIT-4 TEACHING HOURS (11)	 <u>Teaching-learning Resources in Physics</u> Meaning, Objectives, Scope, Characteristics, Types, Preparation, presentation and Importance of Teaching - Learning Material. Planning and Importance of Physics Laboratories and its uses. Qualities of Physics Text Books at Secondary Level. Audio-Visual Aids- Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the teaching of Physics. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of Physics Make a report on Physics Teaching Planning and Importance of Physics Laboratories and its uses.
UNIT-5 TEACHING HOURS (11)	 <u>Assessment and Evaluation</u> Meaning, concept and construction of Achievement test, diagnostic test and remedial teaching. Types of Questions, Characteristics of a good test. Blue print: Meaning, concept, need and construction. Continuous and Comprehensive Evaluation: Meaning, concept, importance and limitations.Models of Teaching. Assignment: Construction, administration and interpretation of an achievement test of any standard of school. Make a diagnostic test of your subject and apply it in school, after discussion with concerning teacher and give remedial measure.

TEACHING AND	1 I.	ectures	
LEARNING			
	2. E 3. V	learning	
STRATEGIES			
		xtension Lectures	
		ontent Review	
		elf-Learning	
		roup Discussions	
	8. Fi	eld Visit	
	9. S		
	10. D	ocumentaries	
	11. S	nort Films	
	12. T	eam Teaching	
		ching strategies are subject to change as per	requirement
		idents and their capabilities.	-
		L	
CONTINUOUS &	Details o	f Continuous and Comprehensive Assessme	nt (CCA) are as
COMPREHENSIVE	follows:	•	
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM
(CCA)	NO.		MARKS
	1	Monthly Test	10X6
	1	Withing Test	Test = 60
	2	Presentation	$\frac{1000}{10}$
	2		
	3	Group Discussion	10
	4	Debate	10
	5	Participation and Presentation in Seminar	10
	6	Report Writing	10
	7	Viva Voce	10
	8	Attendance*	10
	9	Co-curricular Activity	10
	10	Team Teaching	10
	EXPLA	NATION (METHOD TO ASCERTAIN MA	RKS FOR
	CCA):		
	CCA wi	ll be reduced to 30 marks or 15 marks (as per c	ourse
	weighta		
	Formula	: Marks obtained/Total marksX30	
	For example	nple: $60 \div 160 \times 30 = 11.25$	
		SO-I: Provided that a candidate shall be grante	d a relaxation
		rm of exemption from CCA components, howe	
		a respective course.	,
		SO-II: Provided further that this will be manda	tory for a
		the to appear in the monthly test conducted in the	-
	course.	to appear in the monthly test conducted in the	
		ance in Lectures and Practical	
	Attill	Percentage Marks Allotted	
		75% to $80%$ 02	
		81% to 85% 04	
		86% to 90% 06	
		91% to 95% 08	
		Above 96% 10	

EXAMINATION PATTERN	Term-end examinations are organized by the university in the prescribed format to enable the scholars to achieve success in contemporary competitions and to achieve their goals.
PERIODICAL REVISION OF SYLLABUS	 ANNUAL However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one month.
SELECTED READINGS	 Mangal, S.K (2005). Teaching of Physics. Arya book depot: New Delhi. Joshi, S.R (2008). Teaching of science. A.P.H Publishing Corporation: New Delhi. Das, R.C, (2000). Science teaching in schools. Sterling Publishers private limited: New Delhi. Prasad, J. (1999). Practical aspects in teaching of science. Kanishka publisher: New Delhi. Nanda, V.K. (1997). Science education today. Anmol publications Pvt. Ltd.: New Delhi. Bhan, Suraj Prakash. (2006). Teacher training. Lotus press: New Delhi. etaöls, glGa. (2006). भौतिक विज्ञान शिक्षण. शिक्षा प्रकाशनः जयपुर. भटलागर, ए.बी. (2000). भौतिक विज्ञान शिक्षण. द्रार्थी पब्लिकेशन्सः मेरठ. नेगी, जे.एस. (2008). भौतिक विज्ञान शिक्षण. विनोद पुस्तक मन्दिरः आगरा. शर्मा, आर.सी. (2007). आधुनिक विज्ञान शिक्षण. दाशाप. सम्पत्तराय पब्लिशिंग कम्पनी (प्रा.) लि. सिंह, विजयपाल (2005-06). भौतिक विज्ञान शिक्षण. साहित्य प्रकाशनः आगरा. त्यागी, एस.के.(2000). भौतिक विज्ञान शिक्षण. साहित्य प्रकाशनः आगरा कुलश्रेष्ठ, एस.पी. शैक्षिक तकनीकी एवं उसके मूल आधार. विनोद पुस्तक मन्दिर. आगरा-02.

B.Sc. B.Ed. III Year					
COURSE CODE:	BSCBED-355 a I			COURSE TYPE : CORE	
COURSE TITLE :	Paper-I: Solid State Physics, Solid State Devices and Electronics				
MAX.MARKS:	50	MIN.PASS M	ARKS	20	
THEORY	40	MIN.PASS M	ARKS	16	
EXAMINATION:					
CONTINUOUS	10	MIN.PASS M	ARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR				
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATION		MONTHLY TEST		
DURATION	3 HR		1 HR		
OBJECTIVE:					

• This course aims to develop the fundamental, theoretical and experimental knowledge of solid state physics, solid state devices and electronics by learning various topics viz. crystal structure, Thermal properties of solids, band structure and motion of electron, circuit analysis, network theorems, P-N diode equation, semiconductor diode, rectifiers, filters, transistors and transistor amplifiers and their analysis. Students will also learn logic gates, amplifiers with feedback and oscillators. The course helps them to develop skills to design electronic circuits for various applications.

Learning Outcomes: After completion of the course, student-teachers will be able to-

• Understand thermal properties and band structure.

• Apply law of motion and semiconductors.

• To explain Semiconductor devices.

• Explain the overview of crystalline and glassy forms, liquid crystals, glass transition.

• Discuss on general principles of operation, classification, distortion of Small Signal Amplifiers.

_	Overview: Crystalline and glassy forms, liquid crystals, glass transition.
15)	Structure: Crystal structure, periodicity, lattices and bases, Miller indices,
SS(unit cell, Wigner-seitz cell, different crystal system, allowed rotations,
	lattice types, lattice planes, common crystal structures. Laue's theory of
UNIT-1 ING HOURS(15)	X-ray diffraction, Bragg's law, Laue patterns.
IN DE	Bonding: Potential between a pair of atoms, Lennard-Jones potential,
CHI	concept of cohesive energy, covalent, Van der walls, ionic, and metallic
	crystals.
TEA	Magnetism: Atomic magnetic moment, magnetic susceptibility, Dia-,
	Para- and Ferro-magnetism, Ferromagnetic domains, hysteresis.

UNIT-2 TEACHIGN HOURS(15)	 Thermal properties: Lattice vibrations, simple harmonic oscillator, second order expansion of Lennard-Jones potential about the minimum, vibrations of one-dimensional monatomic chain under harmonic and nearest neighbour interaction approximation, concept of phonons, density of modes (1-D), Dulong-Petit law, Einstein and Debye model; lattice specific heat, low temperature limit. Band structure: Electrons in periodic potential, nearly free electron model (qualitative), energy bands, energy gap, metals, insulators, semiconductors. Motion of electrons: Free electrons, conduction electrons, electron collisions, mean free path, conductivity and Ohm's law, Density of states, Fermi energy, Fermi velocity, and Fermi-Dirac distribution.
UNIT-3 TEACHING HOURS(10)	 Semiconductors:Intrinsic semiconductors, electrons and holes, Fermi level, Temperature dependence of electron and hole concentrations, Doping, impurity states, n and p type semiconductors, conductivity, mobility, Hall effect, Hall coefficient. Semiconductor devices: Metal-semiconductor junction, p-n junction, majority and minority carriers, diode, Zener and tunnel diodes, light emitting diode, transistor, and solar cell. Advanced Materials: Superconductors, fullerenes, carbon nanotubes, graphenes, and nanomaterials. Circuit analysis: Networks- some important definition, loop and nodal
UNIT-4 TEACHING HOURS(20)	equation, Kirchhofs laws, driving points and transfer impedance, four terminal netwok parameters. Networks theorems- Superposition, Thevenin, Norton, Maximum power transfer and Miller theorems Rectification and Power supply: Half wave and full wave rectifier, Ripple factor, efficiency and regulation, bridge rectifier, filters, Voltage regulation and voltage stabilization, Zener diode, voltage, voltage multiplier circuit, characteristics of a transistor in CB, CE and CC mode , graphical analysis of the CE configuration, low frequency equivalent circuits, h-parameters, bias stability, thermal runaway. Field effect transistors: Basic construction of JFET and MOSFET , Drain and transfer characterstics of JFET, operating regions and pinch off voltage.
UNIT-5 TEACHING HOURS(15)	 Small Signal Amplifiers: General principles of operation, classification, distortion, RC coupled amplifier, gain, frequency response, input and output impedance, multistage amplifiers, transformer coupled amplifiers, Equivalent circuits at low, medium and high frequencies, emitter follower, low frequency common-source and common-drain amplifier, Noise in electronic circuits. Operational amplifier: Characteristics of OP-AMP, differential amplifier, CMRR, inverting and non-inverting configuration, Application of OP-AMP: Unity gain buffer, Adder, Subtractor, Integrator and Differentiator. Digital circuit: Number systems, logic fundamentals, AND, OR, NOT, NOR, NAND, XOR gates, Boolean theorems, circuit realization using DTL and TTL logic.

TEACHING	1 Lootu	ra mathad				
AND	 Lecture method Problem Solving method 					
		0				
LEARNING	-	ical method				
STRATEGIES		ar/Symposia				
		w of literature				
	6. Repor	-				
	7. Group Discussion					
		s/Animation				
		earning/e-Learning				
		shops/Experiments.				
		ing and Learning Strategies may be ch	ange as per			
		of the students and their capabilities.				
CONTINUOUS		ontinuous and Comprehensive Assess	sment (CCA) are as			
COMPREHENSI	follows:					
VE	S. No.	CCA- Components	Max. Marks			
ASSESSMENT			Allocation			
(CCA)	1.	Monthly test	20*3 Test=60			
	2.	Quizzes and Assignments	10			
	3.	Viva-voce	10			
	4.	Seminar/Symposia	10			
	5.	Report writing	10			
	6.	Workshop	10			
	7.	Review of literature	10			
	8.	Creativity/Innovation	10			
	9.	Experimental Skill	10			
	10.	Co-curricular activity	10			
	10.Co-curricular activity1011.Attendance10					
	Total 160 marks equivalent reduced to CCA original marks 30.					
Exam Pattern	Term-end examinations are organized by the university in the					
	prescribed format to enable the scholars to achieve success in					
		y competitions and to achieve their go				
	contemporary competitions and to achieve their goals, semester.					
PERIODICAL	1. ANNU	IAL.				
REVISE OF	2. HOWEVER, THE UNIVERSITY may revise the syllabus at any					
SYLLABUS	time during the running year after giving a notice for a period one					
51LLilbeb	month		shee for a period one			
SELECTED			to solid state physics			
READINGS	• Kittel, C., & Mc Euen, P. (1976). Introduction to solid state physics (Vol. 8). New York: Wiley.					
	 Millman, J., & Grabel, A. (1987). Microelectronics. McGraw-Hill, Inc. Pyder, J. D. (1964). Electronic fundamentals and applications. (Book) 					
	• Ryder, J. D. (1964). Electronic fundamentals and applications. (Book					
	on solid state, vacuum and gaseous forms of electronic devices). Englewood Cliffs, N. J., Prentice-Hall.					
	Ŭ		Organia Salid Stata			
		B., Singh, R. J., & Singh, N. P. (1994)	. Organic sond-state			
	-	Tetrahedron, 50(22), 6441-6493.	physica New Della			
		ra, J. P. (2014). Elements of solid-state	physics. new Deini:			
		ling Pvt. Ltd.	• • • •			
	-	W. D. (1989). Electronic Devices: Circu	its and Applications.			
	Prentice H					
		• Streetman, B. G., & Banerjee, S. (1995). Solid state electronic devices (Vol. 4). Englewood Cliffs, NJ: Prentice hall.				

	B.Sc. B.Ed. III	[YearI					
COURSE CODE:	BSCBED-355 a II		COURS	E TYPE : CORE			
COURSE TITLE :	Paper-II: Quantum Me	chanics and	l Statistical	Physics			
MAX.MARKS:	50		S MARKS	20			
THEORY	40	MIN.PAS	S MARKS	16			
EXAMINATION:							
CONTINUOUS	10	MIN.PAS	S MARKS	4			
COMPREHENSIVE							
ASSESSMENT (CCA)							
ATTENDANCE	80 PERCENT IN RESPE	CTIVE YE	AR				
ELIGIBILITY							
EXAMINATION	TERM END EXAMINA	TION	MOI	NTHLY TEST			
DURATION	3 HR			1 HR			
OBJECTIVE:							
• This course will	enable the students to o	develop the	e basic kno	wledge of quantum			
	application to various pro	-		• •			
	like ScHRödinger equation			-			
	quantum nature of e. m. ra			-			
	particles, the notion of quan		-				
• •	re to students about statis						
	cluding Astrophysics, Sem	niconductors	s, Plasma P	hysics, Bio-Physics,			
Chemistry and in a	many other directions.						
Learning Outcomes: Aft	ter completion of the course	e, student-te	achers will b	be able to:-			
• Understand the Or	rigin of the quantum theory						
	tistical basis of thermodyna		ution of spec	us in an ideal gas.			
	particle duality and uncertai		0				
	•	• • •					
Apply the principle of quantum mechanics in further study. Origin of Quantum theory: Failure of classical Physics to explain the							
	8 6 6			· 1			
-	henomenon such as blac	• •					
ອີ 🖸	hotoelectric effect and E	-		-			
	Broglie'' hypothesis, evic						
H H S H H H S H H H S H H H S H H H S H H H S H H H S H H H H H H H H H H	articles. Uncertainly prin						
ND SAC	ingle slit, particle in a bo						
H H e	lectron in nucleus, (ii) Grou						
	energy of narmonic oscillator. Energy-time uncertainty, Borrk's						
	quantization of angular momentum and its application to hydrogen atom, limitations of BoHR's theory.						
	Schrodinger Equation: Ti		ant and Tim	e independent form			
	Physical significance of	-		-			
	Probability current density,			-			
	Hermitian operators. Expect	-	-				
	position, momentum, energy		-				
	nechanics, eigen function		-	-			
	of eigen functions, commut	-	-				
	of group and phase velocities, wave packet.						

	Application	s of quantum mechanics:Particle	in a ana dimansianal		
5 C	and tHRee-dimensional box, <i>particle in finite well</i> , harmonic oscillator, Reflection and transmission by a step potential and by a				
			ep potential and by a		
LI H S	-	r potential barrier.			
UNIT-3 TEACHING HOURS(12)	• •	atom: Time independent ScHR	0 1		
	hydrogen at	om in spherical co-ordinates, Natural	occurrence of n, 1 and		
	m quantum	numbers, the related physical quan	tities, comparison with		
	BoHR's the	ory, Wave functions, Probabilistic int	erpretation.		
		stical basis of thermodynami			
		mic probability, principle of equal	•		
	•		1 1 1		
2)	probability distribution and its narrowing with increase in number of				
3(1)	particles, <i>Macro-state and Microstate</i> , the expressions for average properties, Constraints, accessible and inaccessible states, distribution of				
R.	· · ·		-		
400	-	th a given total energy into a discre	•••		
HC HC		tomic ideal gas and barometric relation			
		ersal laws: The mu space represent			
UNIT-4 TEACHING HOURS(15)	space into	energy sheets and into phase co	ells of arbitrary size,		
H	application	to one-dimensional harmonic oscill	ator and free particles,		
AC	Equilibrium	between two systems in thermal	contact, bridge with		
E	macroscopio	e physics, Probability and entrop	y, Boltzmann entropy		
E	-	atistical interpretation of second la			
		canonical distribution law and its app	-		
		ition of energy. Partition function an	-		
		distribution of speeds in an idea			
	speeds and of velocities, experimental verification, distinction between				
2) 2)	mean, rms and most probable speed values, Doppler broadening of				
UNIT-5 TEACHING HOURS(15)	spectral lines.				
	Transition to quantum statistics: h'as a natural constant and its implications, cases of particle in a one-dimensional box and one-				
	-	▲			
		harmonic oscillator, indistinguishab	• •		
-	-	es, Bose-Einstein and Fermi-Dirac			
		elium, free electrons in a metal and	photons in blackbody		
		ermi level and Fermi energy.			
TEACHING AND		ure method			
LEARNING		lem Solving method			
STRATEGIES	3. Grap	bhical method			
	4. Sem	inar/Symposia			
	5. Revi	ew of literature			
	6. Repo	ort writing			
		p Discussion			
		cos/Animation			
		Learning/e-Learning			
	10. Workshops/Experiments.				
1	* The Teaching And Learning Strategies may be change as per				
		hing And Learning Strategies may	he change as ner		
	* The Teac				
CONTINUOUS	* The Teach requirement	t of the students and their capabili	ties.		
CONTINUOUS	* The Teach requirement Details of C		ties.		
COMPREHENSIVE	* The Teach requirement Details of C follows:	nt of the students and their capabili Continuous and Comprehensive Ass	ties. cessment (CCA) are as		
COMPREHENSIVE ASSESSMENT	* The Teach requirement Details of C	t of the students and their capabili	ties. essment (CCA) are as Max. Marks		
COMPREHENSIVE	* The Teach requirement Details of C follows: S. No.	at of the students and their capabili Continuous and Comprehensive Ass CCA- Components	ties. eessment (CCA) are as Max. Marks Allocation		
COMPREHENSIVE ASSESSMENT	* The Teach requirement Details of C follows:	nt of the students and their capabili Continuous and Comprehensive Ass	ties. essment (CCA) are as Max. Marks		

	3. Viv	va-voce	10		
		ninar/Symposia	10		
		port writing	10		
		rkshop	10		
	9.Experimental Skill10				
		curricular activity	10		
I		endance	10		
		equivalent reduced to CCA origi			
Exam Pattern		nations are organized by the univ			
		e the scholars to achieve suc			
		l to achieve their goals.	1		
PERIODICAL	1. ANNUAL				
REVISE OF	2. However, the Universitymay revise the syllabus at any time				
SYLLABUS		during the running year after giving a notice for a period one			
	months.				
SELECTED		0) Eurodomantala of statistical	and themes 1 mbrusies		
READINGS	• Kell, F. (200 Waveland Pre	09). Fundamentals of statistical	and thermal physics.		
KLADINGS					
	• Huang, K. (2009). Introduction to statistical physics. CRC press.				
	• Mandel, L., & Wolf, E. (1965). Coherence properties of optical				
		fields. Reviews of modern physics, 37(2), 231.			
	• ·	F. K., Kennard, E. H., & C	1 /		
		to modern physics (Vol. 747).	New York: McGraw-		
	Hill.		0		
		K., & Lokanathan, S. (2004).	Quantum mechanics:		
	theory and applications. Macmillan.				
	• •	▲			
	• Lifshitz, E. I	M., & LD and Sykes Landau (Non-relativistic Theory, Pergamo			

PHYSIC	S PRACTICALS-III
Duration: 4 HR MAX.MARKS: 50	Min. Marks: 20
Any twelve of the following experiments a	are to be performed. Few more experiments may be set
at the institutional level. In examination tw	wo experiments are to be performed taking at least one
from each section.	
The distribution of marks in the practical e	examination will be as follows:
(i) Two experiments	30 Marks
For each experiment, distribution of marks	s will be as follows:
Figure:	2
Formula/Theory:	2
Observation:	7
Calculation and Result:	3
Precautions:	1
(ii) Viva voce	10
(iii) Records	10
Total	50 Marks

LIST OF EXPERIMENTS

- Study the characteristic of a given transistor PNP/NPN (CE, CB and CC configuration) and determine alpha and beta values.
- Study the characteristic of junction diode & Zener diode.
- Designs a Zener regulated power supply and study the regulation with various loads.
- Study of Half wave & Full wave rectification and application of L and P section filters.
- Study the characteristics of a field effect transistor (FET) and design and study of amplifier of finite gain.
- Study the characteristics of a unijunction transistor (UJT).
- Determination of Stefan's constant.
- Study the frequency responses of a transistor amplifier (bipolar/FET) obtain the input and output impedance of the amplifier.
- To study hysteresis by B-H curve using CRO.
- Using discrete component, study OR, AND, NOT logic gates, compare with TTL integrated circuits IC's.
- Application of operational amplifier as: (i) Inverter (ii) Non-Inverter (iii) differentiator (iv) Integrator.
- Design and study of an R-C phase shift oscillator.
- Study of a RC transmission line at 50 Hz.
- Study of LC transmission line at fixed and variable frequency.
- Study of RC circuits as integrating and differentiating systems with square input.
- To study the Networks theorems.
- To study the characteristics of a Thermister.

	B.Sc. B.Ed-III Year				
COURSE CODE:	CODE: BSCBED-355 b	I	COUH	RSE TYPE : CORE	
COURSE TITLE :	Paper I: Organic Chem	nistry			
MAX.MARKS:	50	MIN.PAS	S MARKS	20	
THEORY	40	MIN.PAS	S MARKS	16	
EXAMINATION:					
CONTINUOUS	10	MIN.PAS	S MARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPENTIVE SESSION				
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATION MONTHLY TEST				
DURATION	03 HR 1 HR				
Objective:					
• To learn aboutChemistry	y of hydroxy compounds,	Carbonyl co	ompounds, C	Carboxylic Acids and	
its derivatives.					
• To be familiar with chemistry of Nitrogen containing compounds, Biomolecules, polymer					
and Drugs.					
• To learn various techniques of spectrometric identification (U.V & I.R.) of					
organiccompounds.					
Learning outcomes: - After completion of the course, student-teachers will be able to-					

• Develop an understanding of chemistry of hydrocarbons and their halogenated derivatives.

- Get familiar with chemistry of Nitrogen containing compounds, *Biomolecules, polymer and Drugs*
- Apply the principles of Alcohols classification and nomenclature in real life situation.
- Apply the knowledge of Ultraviolet (UV) and Infrared (IR) absorption spectroscopy to explain natural physical phenomena.
- Define the characteristics of carbonyl compounds, Carboxylic Acids and its derivatives.

	Chemistry of hydroxy compounds
UNIT-1 TEACHING HOURS(15)	 Alcohols classification and nomeniclature .Monohydric alcohols-nomenclature, methods of formation by reduction of aldehydes, ketones, carboxylic acids and esters. Hydrogen bonding. Acidic nature. Reactions of alcohols. Dihydric alcohols-nomenclature, methods of formation, chemical reactions of vicinal glycols, oxidative cleavage [Pb(OAc)₄ and HlO₄] and pinacol-pinacolone rearrangement. Trihydric alcohols- nomenclature and methods of formation, chemical reactions of glycerol. Phenols: Nomenclature, structure and bonding. Preparation of phenols, physical properties and acidic character. Acidic strengths of alcohols and phenols, resonance stabilization of phenoxide ion. Reactions of phenols-electrophilic aromatic substitution, acylation and carboxylation. Mechanism of Fries rearrangement, Claisen rearrangement, Gatterman synthesis, Hauben-Hooesch reaction, Lederer- Manasse reaction and Reimer-Tiemann reaction. Ethers and Epoxides: Nomenclature of ethers and methods of their formation, physical properties, Chemical reactions, cleavage and auto oxidation, Ziesel's method.Synthesis of epoxides. Acid and base-catalyzed ring opening of epoxides, orientation of epoxide ring opening, reactions of Grignard and Organolithium reagents with epoxides.
UNIT-2 TEACHIGN HOURS (15)	 <u>Carbonyl compounds</u> <u>Aldehydes and Ketones:</u> Nomenclature and structure of carbonyl group.Synthesis of aldehydes and ketones with particular reference to the synthesis of aldehydes from acid chlorides, synthesis of aldehydes and ketones using 1, 3-dithianes, synthesis of ketones from nitriles and from carboxylic acid, Physical properties. <u>Mechanism of nucleophilic additions</u> to carbonyl group with particular emphasis on Benzoin, Aldol, Perkin and Knoevenagel condensations, Condensation with ammonia and its derivatives, Witting reaction, Mannich reaction. <u>Use of acetals as protecting group</u>, Oxidation of aldehydes, Baeyer-villiger oxidation of ketones, Cannizzaro reaction, MPV, Clemmensen, Wolff-kishner, LiAlH4 and NaBH₄ reductions, Halogenation of enolizable ketones.

	Carboxylic Acids and its derivatives					
	Carboxylic Acids: Nomenclature, structure and bonding, physical properties,					
	acidity of carboxylic acids, effects of substituents on acid strength. Preparation of					
4	 carboxylic acids. Reactions of carboxylic acids. Hell-Volhard-Zelinsky reaction. Synthesis of acid chlorides, esters and amides. Reduction of carboxylic acids. Mechanism of decarboxylation. Methods of formation and chemical reactions of unsaturated monocarboxylic acids. 					
9(1)						
IRS						
01-3						
UNIT-3 TEACHING HOURS (14)						
15 Z	Dicarboxylic acids: Methods of formation and effect of heat and dehydrating					
H	agents.					
EAC	Carboxylic Acid Derivatives: Structure and nomenclature of acid chlorides,					
H	esters, amides and acid anhydrides. Relative stability of acyl derivatives. Physical					
	properties, interconversion of acid derivatives by nucleophilic acyl substitution.					
	Preparation of carboxylic acid derivatives, chemical reactions, mechanisms of					
	esterification and hydrolysis (acidic and basic).					
	Nitrogen containing compounds					
	Nitroalkanes and Nitroarenes: Preparation of nitroalkanes and					
	nitroarenes.Chemical reactions of nitroalkanes.					
	Mechanisms of nucleophilic substitution in nitroarenes and their reductions in					
	acidic, neutral and alkaline media. Picric acid. Halonitroarenes: Reactivity.					
	Amines: Structure and nomenclature of amines, physical					
G	properties.Stereochemistry of amines, Separation of a mixture of primary,					
UNIT-4 CHING HOURS(15)	secondary and tertiary amines, Structural features effecting basicity of amines,					
IRS	Amines salts as phase-transfer catalysts, Preparation of alkyl and aryl amines					
40	(reduction of nitro compounds, nitriles), reductive amination of aldehydic and					
UNIT-4 ING HO	ketonic compounds, Gabriel-phthalimide reaction, Hofmann bromamide reaction.					
N N	Reactions of amines: Electrophilic aromatic substitution in aryl amines, reaction					
H	of amines with nitrous acid. Synthetic transformation of aryl diazonium salts, azo					
CAC	coupling.					
TE	Biomolecules , polymer and Drugs :					
	Biomolecules:- carbohydrates ,proteins, vitamins, nuclic acids .					
	Polymers : -Types of polymerization ,Natural and synthetic polymers and their					
	uses.					
	Drugs :-antacids, antihistamines , analgesics , antipyretics, antibiotics, and					
	antifertility					
	Chemistry in everyday life: - in field of medicines, in food, in cleanising and					
	sanitization.					

		Electromagnet	c Spectrum: Absorption Spectra			
Spectra Lambertlaw), molar absorptivity, types of electronic transitions, effect of conjugation. Concept of elRomophore and auxoetRome. Bathoc/Romit hypsocHRomic, hypercHRomic and hypocHRomic shifts. UV spectra of conjugated enes and enones. Woodward Fieser rules for calculation of absorption maxima in dienes and unsaturated carbonyl compounds. Infrared (IR) absorption spectroscopy: molecular vibrations, Hooke's law selection rules, intensity and position of IR bands, fingerprint regior characteristic absorption of various functional groups and interpretation of IR spectra of simple organic compounds. TEACHIN 1. Lecture method G AND 2. Problem Solving method LEARNIN 6. Report writing 7. Group Discussion 8. Videos/Animation 9. Self-Learning/e-Learning 10. Workshops/Experiments. * The Teaching and Learning Strategies may be change as per requirement of the students and their capabilities. CONTINU Details of Continuous and Comprehensive Assessment (CCA) are as follows: OUS S. No. CCA- Components COMPRE 1. Monthly test 20*3 Test=60 ASESSM 2. Quizzes and Assignments 10 5. Report writing 10 10 6. Report writing 10 10 7. Review of literature 10 10 6. Workshop 10 10 <th></th> <th colspan="4"></th>						
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10.Co-curricular activity1011.Attendance10Total 160 marks equivalent reduced to CCA original marks 30.Exam PatternTerm-end examinations are organized by the university in the prescriber format to enable the scholars to achieve success in contemporar competitions and to achieve their goals.PERIODIC AL REVISE OF SYLLABU1. ANNUAL2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.			*			
11.Attendance10Total 160 marks equivalent reduced to CCA original marks 30.Exam PatternTerm-end examinations are organized by the university in the prescriber format to enable the scholars to achieve success in contemporary competitions and to achieve their goals.PERIODIC AL REVISE OF SYLLABU1. ANNUAL 2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.			·			
Total 160 marks equivalent reduced to CCA original marks 30.Exam PatternTerm-end examinations are organized by the university in the prescriber format to enable the scholars to achieve success in contemporar competitions and to achieve their goals.PERIODIC AL REVISE OF SYLLABU1. ANNUAL 2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.			·			
Exam PatternTerm-end examinations are organized by the university in the prescribed format to enable the scholars to achieve success in contemporar competitions and to achieve their goals.PERIODIC AL REVISE OF SYLLABU1. ANNUAL 2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.						
Patternformat to enable the scholars to achieve success in contemporar competitions and to achieve their goals.PERIODIC1. ANNUALAL REVISE OF SYLLABU2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.	Fyom					
competitions and to achieve their goals. PERIODIC AL AL REVISE OF SYLLABU competitions and to achieve their goals. 1. ANNUAL 2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.						
PERIODIC 1. ANNUAL AL 2. However, the Universitymay revise the syllabus at any time during th REVISE running session after giving a notice for a period one months. OF SYLLABU	1 aut 11			s in contemporary		
AL REVISE OF SYLLABU2. However, the Universitymay revise the syllabus at any time during th running session after giving a notice for a period one months.	DEDIODIC	_	Souther then South			
REVISE If the vertex, the conversity may revise the synaptic at any time during the running session after giving a notice for a period one months. OF SYLLABU						
OF SYLLABU						
SYLLABU		running session	after giving a notice for a period one	months.		
8						
	S					

SELECTE	• Atkins, R. C., & Carey, F. A. (1997). Organic chemistry: a brief course.						
D	McGraw-Hill Science, Engineering & Mathematics.						
READING	• Atkins, R. C., & Carey, F. A. (1997). Organic chemistry: a brief course.						
S	McGraw-Hill Science, Engineering & Mathematics.						
	• Bahl, A., & Bahl, B. S. (1968). <i>A textbook of organic chemistry</i> . S. Chand & Company.						
	• Bruckner, R. (2001). Advanced organic chemistry: reaction mechanisms. Elsevier.						
	• Kalsi, P. S. (2000). <i>Organic reactions and their mechanisms</i> . New Delhi: New Age International.						
	• Kalsi, P. S. (2007). <i>Organic Reactions: Stereochemistry and Mechanism</i> . New Delhi: New Age International.						
	• Kalsi, P. S. (2008). <i>Stereochemistry conformation and mechanism</i> . New Delhi: New Age International.						
	• Lowry, T. H., & Richardson, K. S. (1987). <i>Mechanism and theory in organic chemistry</i> (pp. 60-71). New York: Harper & Row.						

	B.Sc. B.Ed-III Year					
COURSE CODE:						
COURSE TITLE :						
MAX.MARKS:	50	MIN.PASS	MARKS	20		
THEORY	40	MIN.PASS	MARKS	16		
EXAMINATION:						
CONTINUOUS	10	MIN.PASS	MARKS	4		
COMPREHENSIVE						
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RES	SPECTIVE S	ESSION			
ELIGIBILITY						
EXAMINATION	TERM END EXAMI	INATION	MC	ONTHLY TEST		
DURATION	03 HR			1 HR		
Objective:						
	s in chemistry of Eleme			ics, Molecular Orbital		
•	mistry and physical prop					
	entals of various technic					
	pectrum, Raman Spe	ectrum, Ele	ctronic Spe	ctrum) of organic		
compounds.						
č	f this course is to p	-		0		
-	t synthesis and character	erization of	inorganic c	omplexes and Organic		
1	ation of metal ions.					
Learning outcomes: - Af	-					
	ementary Quantum Mec		nemical Bo	nding.		
	Molecular Orbital Theor	•				
	nal and Vibrational Spe		, , .			
	entals of Spectroscopy		tuation.			
	<u>Elementary Quantum N</u>					
UNIT-1 TEACHING HOURS(15)	Black-body radiation,					
rss(f hydrogen atom (no		
5		-		De Broglie hypothesis, soidal wave equation,		
HO H				e equation and its		
	1		•	e function, postulates of		
15 X	quantum mechanics, pa	-		-		
H H H				separation into tHRee		
	equations (without					
L						
importance, hydrogen like wave functions, radial wave functions, angular wave functions.						
Л	Iolecular Orbital Theor					
			from A.O, c	onstruction of M.O's		
UNIT-2 TEACHIGN HOURS (14)	Basic ideas, criteria for forming M.O from A.O, construction of M.O's $byLCAO-H_2^+$, ion, calculation of energy levels from wave functions,					
UNIT-2 IEACHIGN 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	hysical picture of bondi					
	rbitals and their charac					
	alculation of coefficient.					
	Introduction to valence bond model of H_2 , comparison of M.O. and V.B.					
	10dels.					

	Fundamentals of Spectroscopy-I			
	• Spectroscopy: Introduction: electromagnetic radiation, regions of			
	thespectrum, basic features of different spectrometers, statement of			
12	the Born-Oppenheimer approximation, degrees of freedom.			
S	• Rotational and Vibrational Spectrum: Diatomic molecules, Energy			
X	levels of a rigid rotor(semi-classical principles), selection rules,			
ο Ω Ω	spectral intensity, distribution using population distribution			
	(Maxwell-Boltzmann distribution) determination of bond length,			
UNIT-3 NG HO	qualitative description of non-rigid rotor, isotope effect.			
UNIT-3 TEACHING HOURS (15)	• Vibrational Spectrum: Infrared spectrum: energy levels of simple			
5	harmonicoscillator, selection rules, pure vibrational spectrum,			
EA	intensity, determination of force constant and qualitative relation of			
E	force constant and bond energies, effect of anharmonic motion and			
	isotope on the spectrum, idea of vibrational frequencies of different			
	functional groups.			
	Fundamentals of Spectroscopy-II			
	• Raman Spectrum : Concept of polarisability, pure rotational and			
12) NG	purevibrationl Raman spectra of diatomic molecules, selection			
HI-T	rules, energy levels and the respective transitions.			
UNIT-4 IEACHIN HOURS(1)	• Electronic Spectrum: Concept of potential energy curves for			
D E O	bonding and antibonding molecular orbitals, qualitative description			
	of selection rules and Frank-Condon principles. Qualitative			
	description and their energy levels and the respective transitions.			
	Photochemistry and physical properties of matter			
	• Photochemistry: Interaction of radiation with matter, difference			
	betweenthermal and photochemical processes. Law of			
UNIT-5 HING HOURS (16)	photochemistry: Grothus-drapper law, Stark-Einstein law, Jablonski			
SZ	diagram depicting variousprocesses occurring in the excited state,			
5	qualitative description of fluorescence, phosphorescence, non-			
I-5 HO	radiative processes (internal conversion, intersystem crossing),			
UNIT-5 NG HO	quantum yield, photosensitized reactions-energy transfer processes			
5 Ż	(simple examples).			
	• Physical properties of matter: Optical activity, polarization			
TEAC	(Clausius–Mossotti equation), orientation of dipoles in an electric			
E	field, dipole, moment, induced dipole moment, measurement of			
-	dipole moment temperature method and refractivity method, dipole			
	moment and structure of molecules, magnetic properties-			
	paramagnetism, diamagnetism and ferromagnetics.			
TEACHING AND	1. Lecture method			
LEARNING	2. Problem Solving method			
STRATEGIES	3. Graphical method			
	 Seminar/Symposia Review of literature 			
	6. Report writing 7. Group Discussion			
	7. Group Discussion8. Videos/Animation			
	9. Self-Learning/e-Learning			
	10. Workshops/Experiments.			
	* The Teaching and Learning Strategies may be change as per			
	requirement of the students and their capabilities.			
	requirement of the students and then capabilities.			

Details of Continuous and Comprehensive Assessment (CCA) are as				
follows:				
S. No. CCA- Components Max. Ma				
	-	Allocation		
1.	Monthly test	20*3 Test=60		
2.	Quizzes and Assignments	10		
3.	Viva-voce	10		
4.	4. Seminar/Symposia 10			
5.	Report writing	10		
6.	Workshop	10		
7.	Review of literature	10		
8.	Creativity/Innovation	10		
9.	Experimental Skill	10		
10.	Co-curricular activity	10		
11.	Attendance	10		
Total 160 m	arks equivalent reduced to CCA origi	nal marks 30.		
Term-end	examinations are organized by t	the university in the		
prescribed	prescribed format to enable the scholars to achieve success in			
contemporary competitions and to achieve their goals.				
1. ANNUAL				
2. However, the Universitymay revise the syllabus at any time				
during the running session after giving a notice for a period one				
months.				
• Banwell, C. N., & McCash, E. M. (1994). Fundamentals of				
molecular spectroscopy (Vol. 851). New York: McGraw-Hill.				
• Chandra, A. K. (1994). Introductory quantum chemistry. Tata				
McGrav	v-Hill Education.			
• Levine,	I. N., Busch, D. H., & Shull, H. (200	9). Quantum chemistry		
(Vol. 6)	. Upper Saddle River, NJ: Pearson Pre	entice Hall.		
• Lewis, l	D., & Glasstone, S. (1960). Elements	s of physical chemistry.		
Macmillan.				
• Linderb	erg, J., & ÖHRn, Y. (2004). Pro	pagators in quantum		
	chemistry. John Wiley & Sons.			
• Lowe, J	. P., & Peterson, K. (2011). Quantum	chemistry. Elsevier.		
		-		
		·		
	follows: S. No. 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. Total 160 m Term-end prescribed contempora 1. ANNUA 2. Howeve during t months. Banwell molecul Chandra McGrav Levine, (Vol. 6) Lewis, I Macmill Linderb chemistri Lowe, J Szabo, introduct	follows: S. No. CCA- Components 1. Monthly test 2. Quizzes and Assignments 3. Viva-voce 4. Seminar/Symposia 5. Report writing 6. Workshop 7. Review of literature 8. Creativity/Innovation 9. Experimental Skill 10. Co-curricular activity 11. Attendance Total 160 marks equivalent reduced to CCA origit Term-end examinations are organized by the prescribed format to enable the scholars the contemporary competitions and to achieve the the scholars the contemporary competitions and to achieve the during the running session after giving a months. • Banwell, C. N., & McCash, E. M. (19 molecular spectroscopy (Vol. 851). New Yor • Chandra, A. K. (1994). Introductory quaded McGraw-Hill Education. • Levine, I. N., Busch, D. H., & Shull, H. (2000) (Vol. 6). Upper Saddle River, NJ: Pearson Pro- • Lewis, D., & Glasstone, S. (1960). Elements Macmillan. • Linderberg, J., & ÖHRn, Y. (2004). Pro-		

CHEMISTRY PRACTICALS-3				
Duration:4 HR MAX.MARKS: 50 Min. Marks: 20				
Note: The students should be given exposure of any research lab	s and instrumentation			
center/reputed university-lab/industry/ government labs of north	nern region.			
Inorganic Chemistry				
Ex.1. Synthesis of an inorganic compound and gravimatric analysis	15			
Organic Chemistry				
Ex.2 Synthesis of Organic compounds 15				
Physical Chemistry				
Ex.4 Perform one of the experiments mentioned in the syllabus.	10			
Ex.5 Vive-Voce 05				
Ex.6 Practical-Record	05			
Total 50 Marks				
LIST OF EXPERIMENTS				

B.Sc.B.Ed. Year III				
COURSE CODE:	BSCBED-355 c I		COURS	SE TYPE : CORE
COURSE TITLE :	Development Biology			
MAX.MARKS:	50			
THEORY	40	MIN.PA	ASS MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PA	ASS MARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPEC	CTIVE Y	EAR	
ELIGIBILITY				
EXAMINATION	TERM END EXAMINAT	ION		ILY TEST
DURATION	3 HR		1	HR
<u>Objective</u> :				
• To learn about th	e comprehend the modern co	ncepts of	developmental	biology.
• To learn about th	e developmental sequences in	n vertebra	ates.	
• To learn about th	e Compare the development	of organs	and systems.	
	er completion of the course,			ble:
8	modern concepts of develop			
*	evelopmental sequences in ve		0.	
	· ·		•	
• Compare the dev	 elopment of organs and syste Historical perpective , 		1 0 1	1 . 1 11
UNIT-1 TEACHING HOURS (15)	definition and phase of development, theories of development, epigensist and preformation, mosaic, regulative, gradient, spemanns, theory of organizers, Gemetogensis, origin of primordial germ cell(PGC), spermatogenesis, morphology of mature sperm, types of sperm, Oogenesis, General feature of mature ovum, polarity of egg. Symmetry of egg. Types of egg.			
UNIT-2 TEACHI GN HOURS (14)	• Reproductive cycles (estrous and menstrual cycle). Events of fertilization, Mechanism of sperm transfer, Polyspermy preventing mechanism. Errors of fertilization and significance of fertilization, parthenogenesis, evolution of viviparity.			
INIT-3 ING HOURS (16)	 Cleavage- Definiton, types, pattern and planes of cleavage, morulation, blastulation, types of blastula, gastrulation, Types of gastrulation mechanism, fate maps (with suitable examples), morphogenetic cell movement and their significance in gastrulation. Embryonic Induction, organizers, Competence. Differentiation and organogenesis- differentiation, growth and organogenesis, development defects (teratology), senescence and ageing. Brief idea of animal tissue culture, Invitro fertilization, stem cells and culture media. 			
UNIT-4 TEACH ING HOURS (15)	 Development of frog upto formation of advanced tadpole. Metamorphosis of tadpole. Hormonal control of metamorphosis. 			

	1					
	Embryoge	enesis of chick upto Neurulation				
	• Extra En	nbryonic membrances in chick a	nd salient feature of			
12 X C		ent of chick development up to 72 H				
-H S						
UNIT-5 TEACHING HOURS (15)		norphology and histology. Functions	-			
	Regeneration	tion :- Regenration mechanism in	animal, steps of limb			
	regenerati	ion in Amphibians.				
TEACHING AND	1. Lectur	re method				
LEARNING	2. Proble	em Solving method				
STRATEGIES		ical method				
	-	ar/Symposia				
		w of literature				
	6. Repor	t writing				
	-	Discussion				
	8. Video	s/Animation				
	9. Self-L	earning/e-Learning				
	10. Works	shops/Experiments.				
	* The TEAC	HING AND LEARNING STRAT	EGIES may be			
		r requirement of the students and				
CONTINUOUS	Details of Co	ontinuous and Comprehensive Ass	essment (CCA) are as			
COMPREHENSIVE	follows:	_				
ASSESSMENT	S. No.	CCA- Components	Max. Marks			
(CCA)		-	Allocation			
	1.	1.Monthly test20*3 Test=60				
	2.	2.Quizzes and Assignments10				
	3. Viva-voce 10					
	4.	7 1				
	5.	5. Report writing 10				
	6.	Workshop	10			
	7.	Review of literature	10			
	8.	Creativity/Innovation	10			
	9.	Experimental Skill	10			
	10.	Co-curricular activity	10			
	11.	Attendance	10			
	Total 160 ma	rks equivalent reduced to CCA origi	nal marks 30.			
EXAMINATION		xaminations are organized by t	Ū.			
PATTERN	-	format to enable the scholars t				
		y competitions and to achieve the	r goals.			
PERIODICAL	1. ANNUAI					
REVISE OF	· · · · · · · · · · · · · · · · · · ·	, the Universitymay revise the s	•			
SYLLABUS	during the running year after giving a notice for a period one					
	months.					
SELECTED		ohan (1985). Chordate Embryology-				
READINGS	• Balinsky B.I. (1976). Introduction to Embryology-(W.B. Saunders,					
	Philadelphia,					
	• Hopper, A.F. and N.H. Hart (1981). Foundations of Animal					
	Developm	Development. New Yark: Oxford University Press.				
	MacBride	e, E. W., & Sir, J. G. K. (1919). Tex	t-book of Embryology.			
	(Vol. 2). l	Macmillan and Company.				

•	Marshall, A. M. (1893). Vertebrate embryology: a text-book for
	students and practitioners. GP Putnam's sons.
•	McEwen R.S. (1999). Vertebrate Embryology. New Delhi: Oxford
	& IBM Publishing CO.
•	McEwen, R. S. (1923). Vertebrate embryology. H. Holt.
•	Rugh R. (1998). Laboratory manual of Vertebrate Embryology-
	Allied Pacific Pvt. Ltd.
•	Sastry K.V. & Shukla Vinita (2008). Developmental Biology.
	Rastogi Publications,
•	Shumway, W. (2002). Introduction to vertebrate embryology. Daya
	Books.

B.Sc.B.Ed. Year III				
COURSE CODE:	BSCBED-355 c II COURSE TYPE : CORE			
COURSE TITLE :	Environmental Studies, E	Ethology ar		
MAX.MARKS:	50		S MARKS	20
THEORY	40	MIN.PAS	S MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PAS	S MARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPEC	TIVE YEA	AR	
ELIGIBILITY				
EXAMINATION	TERM END EXAMINA	ATION	MON	NTHLY TEST
DURATION	3 HR			1 HR
Objectives :				
• Tolearn about to	understand the energy source	es, flow of	energy and c	onservation;
• To learn about th	e understand the recycling o	f minerals a	and nutrients	in ecosystem;
	the understand the dynamic	cs of popu	lation; to u	nderstand causes of
pollution;				
	comprehend origin of life, and	nimal behav	vior and ecor	nomic importance of
animals with wild	*	a student	too oh ana will	ha ahlar
-	After completion of the cours			
• To enable students to understand the energy sources, flow of energy and conservation;				
	To understand the recycling of minerals and nutrients in ecosystem; To understand the dynamics of population; to understand causes of pollution;			
-			-	
	of life, animal behavior and	l economic	importance	of animals with wild
life protection.		1.1	1 11	1 1 1 1
	• Environment: - Atmosph	ere, litnosp	nere and nyc	rosphere as habitats
HOURS	and ecological factors.	4 and Tam		analasiaal fastam
10	• Abiotic factors: - Light and Temperature as ecological factors, limiting factors,			
	 Liebig's law of minimum 	and Chalf	and I are of t	1
NUC 12	0			
5H	• Ecosystem: Types, struc Ecosystem, Ecological P			· · ·
AC	 Biochemical cycle- wate 	•		• • •
UNIT TEACHING (15)	organic nutrients.	a, muogen	and surpriur	cycles legaluling of
	 Elementary-statistics: cer 	ntral tender	cy test of si	onificance
	 Air pollution: sources, a 			*
ZŦ	and control			a shios, prevention
[-2] [[[6]]] [1]	• Water pollution: sources,	preventior	and control	, eutrophication.
UNIT-2 TEACHIGN HOURS (14)	• Noise pollution: sources,	•		
U HOI	 Soil pollution: sources, p 	•		
		Provention and control		
	• Thermal pollution.			

		C 1		
એ C	-	of ozone layer.		
1-3 11N 12 12	Natural D			
UNIT-3 TEACHING HOURS (15	• Natural Resources and conservation – Non-Renewable and			
	Renewab	e		
	Biomagfi	cations		
URS	• Behavior: - Innate (tropism, Texas, reference instincts) ar Acquired (learning and reasoning)			
UNIT-4 HING HO (16)	• Motion: 0 & taxes	Classification of directional movement	s: - kinesis, tropism	
UNIT-4 IEACHING HOURS (16)		ication: - Definition, types of sig, , and visual), meta-comunication) phen		
TEA		characteristics and advantage with s e, and monkey	special reference to	
15)	Economic Sericultur	e Importance of Invertebrates (Apicure).	ulture, Aquaculture,	
URS (• Insects a Caterpille	s pests and their management (Lo r)	cust, Termite, and	
UNIT-5 ING HO	• Economic culture.)	e Importance of vertebrates (Fish c	culture and Poultry	
UNIT-5 TEACHING HOURS (15)	• Wild life of India, causes of depletion of wild life, modes of wild life conservation, Red data book. Environmental legislations (Wildlife Protection Act, Environment act. Bio divergent act). Wild life scenario in and around central foot hills of the Arawali and the thar.			
TEACHING AND	1. Lectur	re method		
LEARNING		em Solving method		
STRATEGIES	-	3. Graphical method 4. Seminar/Symposia		
	 Seminar/Symposia Review of literature 			
	6. Repor			
		Discussion		
		s/Animation		
		earning/e-Learning		
		shops/Experiments. HING AND LEARNING STRATEG	IFS may be	
		r requirement of the students and the	-	
CONTINUOUS &		ontinuous and Comprehensive Assess	_	
COMPREHENSIVE	follows:			
ASSESSMENT (CCA)	S. No.	CCA- Components	Max. Marks Allocation	
	1.	Monthly test	20*3 Test=60	
	, , , , , , , , , , , , , , , , , , ,		10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	

	б.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
	Total 160 mar	ks equivalent reduced to CCA original	marks 30.	
EXAMINATION PATTERN	prescribed f contemporar	xaminations are organized by the ormat to enable the scholars to y competitions and to achieve their g	achieve success in	
PERIODICAL REVISE OF SYLLABUS	any ti	JAL EVER, THE UNIVERSITYmay rev me during the running year after giv l one months.	•	
SELECTED		S. K. (2008). Fundamentals of ecology	. APH Publishing.	
READINGS		. J. (2020). <i>Statistics for environn</i> v. Routledge.	nental biology and	
	Chapman Cambridg	E (1988). <i>Ecology: Principle</i> eUniversity Prss	and Applications-	
	• Coleman, D. C., Callaham, M. A., & Crossley Jr, D. A. (2017). <i>Fundamentals of soil ecology</i> . Academic press.			
	• Dash, M. C. (2001). <i>Fundamentals of ecology</i> . Tata McGraw-Hill Education.			
	• Kormondy, E. J., & Brown, D. E. (1998). <i>Fundamentals of human ecology</i> . Pearson College Division.			
	• Kumar HD (1986). <i>Modern concept of ecology</i> Vikas Publication House.			
		S. (2003). Environmental biology of e University Press.	f agaves and cacti.	
		P., & Barrett, G. W. (1971). <i>Funde</i> 5). Philadelphia: Saunders.	amentals of ecology	
		S, (1998). <i>Mammalian Endrocrin</i> Hindi Edition, Jaipur: College Book C		
	• Sharma I Publicatio	PD (1991).Ecology and Environments.	nt. Jaipur: Rastogi	
	• Soni, K.C. (1999). Animal Ecology and Biostatistics. Hindi Edit Jaipur: College Book Centre.			
	 Woodward, F. I., & Sheehy, J. E. (2017). Principles a measurements in environmental biology. Elsevier. Verma, P. S., & Agarwal, V. K. (2003). Environmental Biology Principles of Ecology. Chand. 			
	Environm	D. A., Strom, P. F., & Allem ental biology for engineers and scient c: Wiley-Interscience.		

PRACTICAL

Duration:4 hours

Objectives: -

Max.Marks: 50

• To develop the skills of staining and mounting of embryos of chick/frog/insect as per UGC guidelines: to understand the development patterns of chick and frog. To enable students to analyses the physico-chemical and biological factors of water and soil sample; to identify and estimate quantitatively the aquatic organism and their adaptation; to observe the population growth patterns.

Practical work based of Paper I and II Course content-

- Study of different types of eggs (Insect, Frog, Hen)
- Study of eggs, cleavage, blastula, gastrula, neurula, tail bud, hatching, mature, tadpole larval metamorphic stages of toadle/forglet.
- Study of embryological slides of various stages of frog.
- Study of embryological slides of various stages of chick.
- Study of development of chick with the help of charts /CD/s /Video/ MM etc.
- Whole mounts: 18 HR, 24 HR, 33 HR, 48 HR, 56HR, 72HR, and 96 HR, of incubation period embryos.
- Study of primitive streak stage in living embryo after removal of the blastoderm from the egg or though multimedia film etsc
- Study of the embryo at various stages of incubation in vivo by making a window in the egg shell.
- Frog embryology Study of spawn, identification of different stages tHRough model/charts/multimedia etc
- Simulation of an ecosystem in the laboratory.
- Determination of oxygen cotent of water sample by Winkler's method.
- Determination of chloride content of water sample
- Determination of sulphates content of water sample
- Determination of dissolved CO2 content of water
- Determination of total solid content of water
- Determination of pH of soil sample
- Determination of water content in a given simple of soil
- Detection of salts i.e. phosphates, sulphates, nitrates and chlorides in a given sample of water.
- Exercise on mean, median, mode and test of significance
- Study and maintenance of Aquarium.
- Study of insect pests with respect to marks of identification, nature of damage and economic importance.
- Study of pest control appliances.
- Life cycle of honey bee, mouth parts, thoracic appendages (legs and wings) & sting apparatus of honey bee.

- Study of bee products, bee pests, bee enemies.
- Communication in honey bee.
- Study of life cycle of Bombyx mori.
- Study of any five equipments in sericulture.
- Submission of field visit report along with at least five photographs/sketch of insect pests.
- Visit of poultry culture/sericulture/apiculture.

Guidelines/Instructions for Practical Examination MAX.MARKS: 50

	Time allowed: 4 HR.
Exercise	Marks
Temporary mounting * one (Staining,	3
identification, sketch)	
Simulated ecosystem	3
Permanent slides -four (Identification	with 8
reasons)	
Applied Zoology/Ethology	8
Ecological experiment	6
Statistical exercise	5
Practical record and slides	5
Viva	4
Project report and assignment	8
	Temporary mounting * one (Staining, identification, sketch) Simulated ecosystem Permanent slides –four (Identification v reasons) Applied Zoology/Ethology Ecological experiment Statistical exercise Practical record and slides Viva

B.Sc.B.Ed. III Year				
COURSE CODE:	BSCBED-355 d I COURSE TYPE : CORE			
COURSE TITLE :	Structure, Development and Reproduction in Flowering Plants			
MAX.MARKS:	50	MIN.PASS		20
THEORY	40	MIN.PASS	MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PASS	MARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RES	PECTIVE YE	EAR	
ELIGIBILITY				
EXAMINATION	TERM END EXAM	INATION	MON	NTHLY TEST
DURATION	3 HR			1 HR
Objective:				
 flowering plants. To acquaint studies angiosperm plant To acquaint the sing out Come: - A To acquaint studies offlowering plant 	students with the structure and processes of embryology associated with lants. The students with the basic anatomical plan of floral plants. e:- After completion of the course, student-teachers will be able: students with the morphology, anatomy, reproduction and classification lants. students with the structure, development and processes associated with			
UNIT-1 TEACHING HOURS (16)	 <i>dicots.</i> The shoot system: The organization, vascular, and dicotyledons, for monopodial sympodi formation of secondar, in relation to conduct 	The shoot system: The shoot apical meristems and its histological organization, vascularization of primary shoot in monocotyledons and dicotyledons, formation of internodes, branching pattern, monopodial sympodial growth, cambium and its functions, formation of secondary xylem, a general account of wood structure in relation to conduction of water and minerals, characteristics of growth rings, sapwood and heart wood , secondary phloem-		
N TEACHIGN S HOURS (16)	 Leaf: Origin, development, arrangement and diversity in size and shape; internal structure in relation to photosynthesis and water loss, adaptations to water stress, stomatal types and trichomes, senescence and abscission. The Root System: The root apical meristem and its organization, differentiation of primary and secondary tissues and their roles, structural modifications for storage, respiration, reproduction and for interaction with microbes. Flower: A modified shoot, development, structure and function of 			
UNIT-3 TEACHIN G HOUR (15)	anther and pistil, Dev Types of <i>Embryo</i> , and megasporogenesis, Typ pollinators. <i>Types of Inp</i>	nd ovules, pes of pollina	Embryos tion, attracti	sac, micro and

LEARNING 2. Problem Solving method STRATEGIES 3. Graphical method 4. Seminar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion 8. Videos/Animation 9. Self-Learning/e-Learning 10. Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be change as per requirement of the students and their capabilities. Details of Continuous and Comprehensive Assessment (CCA) are as follows: ASSESSMENT (CCA) I. Monthly test 2. Quizzes and Assignments 1. Monthly test 2. Quizzes and Assignments 10 3. Viva-voce 10 3. Viva-voce 10 3. Viva-voce 10 5. Report writing 10 6. Workshop 10 7. Review of literature 10 8. Creativity/Innovation 9. Experimental Skill 10 11. Attendance 10 11. Attendance 10 11. ANNUAL <th>UNIT-4 TEACH ING HOURS (14)</th> <th>overcome</th> <th>stil interaction: Sexual inconsistil inconsistent states of rejection incompatibility. Fertilization: s, Parthenocarpy</th> <th></th>	UNIT-4 TEACH ING HOURS (14)	overcome	stil interaction: Sexual inconsistil inconsistent states of rejection incompatibility. Fertilization: s, Parthenocarpy			
LEARNING 2. Problem Solving method STRATEGIES 3. Graphical method 4. Seminar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion 8. Videos/Animation 9. Self-Learning/e-Learning 10. Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be change as per requirement of the students and their capabilities. CONTINUOUS Details of Continuous and Comprehensive Assessment (CCA) are as follows: SSESSMENT (CCA) S. No. CCA- Components Max. Marks Allocation 1. Monthly test 20*3 Test=60 2. Quizzes and Assignments 10 3. Viva-voce 10 4. Seminar/Symposia 10 3. Viva-voce 10 6. Workshop 10 5. Report writing 10 10 6. Workshop 10 7. Review of literature 10 10 11. Attendance 10 10 7. Review of literature 10 10 10 10 10 10 7. Review of literature 10 10 11. Attendance 10 10 8. Creat	UNIT-5 TEACHING HOURS (14)	and funct • Endospe Endospe	 Embryo: Embryo development in Dicots and monocots, structure and function of suspensor, Polyembryony. Endosperm: Types, Development, Structure and Functions of Endosperm, Haustorial and Ruminate Endosperm. 			
CONTINUOUS COMPREHENSIVE ASSESSMENT (CCA) Details of Continuous and Comprehensive Assessment (CCA) are as follows: S. No. CCA- Components Max. Marks Allocation 1. Monthly test 20*3 Test=60 2. Quizzes and Assignments 10 3. Viva-voce 10 4. Seminar/Symposia 10 5. Report writing 10 6. Workshop 10 7. Review of literature 10 8. Creativity/Innovation 10 9. Experimental Skill 10 10. Co-curricular activity 10 11. Attendance 10 12. Attendance 10 13. Attendance 10 14. Germental Skill 10 15. Report activity 10 16. Morthalce 10 7. Review of literature 10 10. Co-curricular activity 10 11. Attendance 10 <	TEACHING AND LEARNING STRATEGIES	 Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be 				
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1.Monthly test20*3 Test=602.Quizzes and Assignments103.Viva-voce104.Seminar/Symposia105.Report writing106.Workshop107.Review of literature108.Creativity/Innovation109.Experimental Skill1010.Co-curricular activity1011.Attendance1010.Total 160 marks equivalent reduced to CCA original marks 30.Term-end examinations are organized by the university in the prescribed format to enable the scholars to achieve success in contemporary competitions and to achieve their goals.PERIODICAL REVISE OF SYLLABUSI. ANNUAL 2. However, the Universitymay revise the syllabus at any time during the running year after giving a notice for a period one months.SELECTED READINGSPandey, A.K. (1997). Introduction to Embryology of Angiosperms.	ASSESSMENT		CCA- Components			
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3. Viva-voce 10 4. Seminar/Symposia 10 5. Report writing 10 6. Workshop 10 7. Review of literature 10 8. Creativity/Innovation 10 9. Experimental Skill 10 10. Co-curricular activity 10 11. Attendance 10 Total 160 marks equivalent reduced to CCA original marks 30. Term-end examinations are organized by the university in the prescribed format to enable the scholars to achieve success in contemporary competitions and to achieve their goals. PERIODICAL 1. ANNUAL REVISE OF SELECTED SELECTED • Johri, B.M. (1984). Embryology of Angiosperms.Berlin: Springer-Verlag. • Johri, B.M. (1984). Embryology of Angiosperms.Berlin: Springer-Verlag.						
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READINGS Verlag.• Pandey, A.K. (1997). Introduction to Embryology of Angiosperms.	SELECTED					
• Pandey, A.K. (1997). Introduction to Embryology of Angiosperms.	READINGS					
		• Pandey, A.K. (1997). Introduction to Embryology of Angiosperms.				
		• Shivanna, K.R. (2003). Pollen Biology and Biotechnology. New				

Delhi: Oxford & IBH Publishing Co.
• Singh, P.C. Pande, D.&Jain,K. (2005).A Textbook of Botany:
Structure, Development and Reproduction in Angiosperms. Meerut:
Rastogi Publications.
• Paula Rudall, (2007). Anatomy of flowering plants: an introduction to
structure and development. New York: Cambridge University Press
Cambridg,
• Bouman, (1978). Development of ovule and seed coat structure in
angiosperms.New Delhi: Today &Tomorrow's Printers & Publishers,

	B.Sc.B.Ed. Year III				
COURSE CODE:	BSCBED-355 d II COURSE TYPE : CORE				
COURSE TITLE :	Plant Physiology				
MAX.MARKS:	50	MIN.PASS MARKS 20			
THEORY	40		S MARKS	16	
EXAMINATION:					
CONTINUOUS	10	MIN.PAS	SS MARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPE	CTIVE YE	AR		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINAT	ΓΙΟΝ	MONTHLY	Y TEST	
DURATION	3 HR		1 HR		
Objective :					
• To inform the stu	dents about the sub-cell phys	siological p	henomena ir	plants.	
	idents about the relation of v				
	orking mechanism of plants	-		physiological point	
of view.	C I				
• To explain the	development, differentiatio	n and var	rious aspects	s of physiology in	
angiosperm plant	-		1	1 9 09	
U I I	After completion of the cou	rse, studen	t-teachers w	vill be able:	
0	nts with the sub-cellular phy				
-	water relations in plants;	51010 810 41		n pranos,	
	-	ne nhysiolo	gical point o	fviow	
	 To understand the functioning of plant from the physiological point of view; To understand various focate of growth differentiation and physiology of flowering in 				
	• To understand various facets of growth, differentiation and physiology of flowering in				
angiosperms.					
	• Plant water relations:	Importance	of water to	plant life: physical	
rh G	• Plant-water relations:				
1 NG (16)	properties of water; d	iffusion a	nd osmosis	; DPD and water	
(T-1 HING tS (16)	properties of water; d potential concept, abso	iffusion a orption and	nd osmosis d transport	; DPD and water of water, Ascent of	
NIT-1 ACHING URS (16)	properties of water; d potential concept, abso <i>sap</i> . Transpiration and	iffusion a orption and l mechani	nd osmosis d transport sm of open	; DPD and water of water, Ascent of	
UNIT-1 EACHING HOURS (16)	properties of water; de potential concept, abso sap. Transpiration and stomata. Factors affection	iffusion a orption an l mechani ng transpi	nd osmosis d transport sm of open ration.	; DPD and water of water, <i>Ascent of</i> ing and closing of	
UNIT-1 TEACHING HOURS (16)	 properties of water; di potential concept, abso sap. Transpiration and stomata. Factors affecti Translocation of orga 	iffusion a orption and l mechani ng transpi nic substa	nd osmosis d transport sm of open ration. ances: mecl	; DPD and water of water, Ascent of ing and closing of nanism of phloem	
UNIT-1 TEACHING HOURS (16)	 properties of water; di potential concept, abso sap. Transpiration and stomata. Factors affecti Translocation of orga transport; source – sink 	iffusion a prption and mechanic ng transpi nic substa relationshi	nd osmosis d transport sm of open ration. ances: mecl p; factors aff	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation.	
	 properties of water; di potential concept, abso sap. Transpiration and stomata. Factors affecti Translocation of orga transport; source – sink Photosynthesis: Histor 	iffusion a orption and mechanic ng transpi nic substa relationshi ical aspec	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. with the pigments;	
	 properties of water; dr potential concept, absorbed sap. Transpiration and stomata. Factors affective Translocation of orgation transport; source – sink Photosynthesis: Historbed action spectra and statement of the spectra and statement of t	iffusion a orption and mechaniang transpi mic substa relationshi ical aspecten enhanceme	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect;	; DPD and water of water, Ascent of ing and closing of manism of phloem fecting translocation. withetic pigments; concept of two	
	 properties of water; dr potential concept, absorber sap. Transpiration and stomata. Factors affecti Translocation of orgat transport; source – sink Photosynthesis: Historber action spectra and photosystems; photograms 	iffusion a orption and l mechanic ng transpi nic substa relationshi ical aspectenhanceme bhosphoryl	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect; ation; C3	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. withetic pigments; concept of two cycle; C4 cycle,	
	 properties of water; di potential concept, abso sap. Transpiration and stomata. Factors affecti Translocation of orga transport; source – sink Photosynthesis: Histor action spectra and photosystems; photog Chemosynthesis, CAM or set and set an	iffusion a orption and a mechanic ing transpi nic substa relationshi ical aspecten enhanceme bhosphoryl cycle, photo	nd osmosis d transport sm of open ration. ances: mecl p; factors aff cts; photosy nt effect; ation; C3 orespiration	; DPD and water of water, Ascent of ing and closing of nanism of phloem tecting translocation. withetic pigments; concept of two cycle; C4 cycle, ; factors influencing	
	 properties of water; dr. potential concept, absorsap. Transpiration and stomata. Factors affecti Translocation of orgatransport; source – sink Photosynthesis: Historaction spectra and photosystems; photog Chemosynthesis; C3& C4 	iffusion a orption and l mechania ing transpi nic substa relationshi ical aspect enhanceme bhosphoryl cycle, photo plants. Sig	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect; ation; C3 orespiration gnificance of	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. withetic pigments; concept of two cycle; C4 cycle, ; factors influencing photosynthesis.	
UNIT-2 TEACHIG TE N HOURS HC (16)	 properties of water; dr potential concept, abso sap. Transpiration and stomata. Factors affecti Translocation of orga transport; source – sink Photosynthesis: Histor action spectra and photosystems; photop <i>Chemosynthesis</i>, CAM or photosynthesis; C3& C4 Respiration: Aerobic 	iffusion an orption and and mechanic ing transpi- nic substarelationshi ical aspect enhanceme bhosphoryl cycle, photo plants. Sig- and anage	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect; ation; C3 orespiration gnificance of erobic resp	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. withetic pigments; concept of two cycle; C4 cycle, ; factors influencing photosynthesis. iration; respiratory	
UNIT-2 TEACHIG TE N HOURS HC (16)	 properties of water; dr. potential concept, absorvation and stomata. Factors affection and stomata. Factors affection Translocation of orgation transport; source – sink Photosynthesis: Historration spectra and photosystems; photographotosynthesis; C3& C4 Respiration: Aerobic substrates; Glycolytic participation 	iffusion a orption and l mechania ing transpi inic substa relationshi ical aspect enhanceme bhosphoryl cycle, photo plants. Sig and anage	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect; ation; C3- orespiration gnificance of erobic resp glucose degr	; DPD and water of water, Ascent of ing and closing of nanism of phloem ecting translocation. withetic pigments; concept of two cycle; C4 cycle, ; factors influencing photosynthesis. iration; respiratory cadation to pyruvic	
UNIT-2 TEACHIG TE N HOURS HC (16)	 properties of water; dr. potential concept, absorvation and stomata. Factors affections affection of orgation and stomata. Factors affections affection of orgation and stransport; source – sink Photosynthesis: Historration spectra and photosystems; photographotosynthesis; C3& C4 Respiration: Aerobic substrates; Glycolytic paracid; tricarboxylic acid 	iffusion a orption and l mechania ing transpi inic substa relationshi ical aspect enhanceme bhosphoryl cycle, photo plants. Sig and anace athway of d cycle; o	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect; ation; C3 orespiration gnificance of erobic resp glucose degree electron tra	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. withetic pigments; concept of two cycle; C4 cycle, ; factors influencing photosynthesis. iration; respiratory cadation to pyruvic nsport mechanism	
UNIT-2 TEACHIG TE N HOURS HC (16)	 properties of water; dr potential concept, abso sap. Transpiration and stomata. Factors affecti Translocation of orga transport; source – sink Photosynthesis: Histor action spectra and photosystems; photog <i>Chemosynthesis</i>; C3& C4 Respiration: Aerobic substrates; Glycolytic pa acid; tricarboxylic aci (chemi – osmotic 	iffusion a orption and l mechanic ing transpi inic substa relationshi ical aspect enhanceme bhosphoryl cycle, photo plants. Sig and anage athway of d cycle; of theory);	nd osmosis d transport sm of open ration. ances: mecl p; factors aff cts; photosy nt effect; ation; C3 orespiration gnificance of erobic resp glucose degre electron tra redox po	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. Anthetic pigments; concept of two cycle; C4 cycle, ; factors influencing photosynthesis. iration; respiratory cadation to pyruvic nsport mechanism otential; oxidative	
	 properties of water; dr. potential concept, absorvation and stomata. Factors affections affection of orgation and stomata. Factors affections affection of orgation and stransport; source – sink Photosynthesis: Historration spectra and photosystems; photographotosynthesis; C3& C4 Respiration: Aerobic substrates; Glycolytic paracid; tricarboxylic acid 	iffusion an orption and a mechanic ing transpi- nic substarelationshi ical aspec- enhanceme ohosphoryl cycle, photo plants. Sig- and anace athway of d cycle; of theory); ratory Que	nd osmosis d transport sm of open ration. ances: mecl p; factors aff ets; photosy nt effect; ation; C30 orespiration gnificance of erobic resp glucose degree electron tra redox po- potient (R.Q.)	; DPD and water of water, Ascent of ing and closing of nanism of phloem fecting translocation. Anthetic pigments; concept of two cycle; C4 cycle, ; factors influencing photosynthesis. iration; respiratory cadation to pyruvic nsport mechanism otential; oxidative	

UNIT-5UNIT-4TEACHING HOURSTEACHING HOURS(14)(14)	 macro and deficiency a nutrition. Nitrogen an importance assimilation; oxidation; mobilization Growth an development seed germin the concept 	 trition: criteria of essentialit micro elements and their and toxicity symptoms; water and lipid metabolism: Biology of nitrate reductase and its c) structure and function of lipid saturated and unsaturated f c) of fatty acids. d development: Definitions; ation and factors of their regulation and factors of their regulation ation and factors of their regulation ation and factors; biological 	r role; mineral uptake; culture technique; foliar y of nitrogen – fixation; regulation; ammonium s; fatty acid biosynthesis, fatty-acids; storage and phases of growth and growth; Seed dormancy, ation; plant movements; <i>lisation,</i> physiology of
INUT ING (14)		fruit ripening; plant hormon	
CH	•	abscisic acid and ethylene, h	
EA	•	and mechanism of action	· · · · · ·
Ē	phytocHRo	mes and cryptocHRome al role and mechanism of actio	
TEACHING AND	1. Lecture		-11.
LEARNING		n Solving method	
STRATEGIES	3. Graphic	-	
		r/Symposia	
		of literature	
	6. Report	-	
	7. Group Discussion		
	8. Videos/Animation 9. Solf Learning/a Learning		
	9. Self-Learning/e-Learning10. Workshops/Experiments.		
	* The Teaching and Learning Strategies may be change as per		
	requirement of the students and their capabilities.		
CONTINUOUS		tinuous and Comprehensive A	
COMPREHENSIVE	follows:	_	
ASSESSMENT	S. No.	CCA- Components	Max. Marks Allocation
(CCA)	1.	Monthly test	20*3 Test=60
	2.	Quizzes and Assignments	10
	3.	Viva-voce	10
	4.	Seminar/Symposia	10
	5.	Report writing	10
	6.	Workshop	10
	7.	Review of literature	10
	<u>8.</u> 9.	Creativity/Innovation Experimental Skill	10 10
	9.	Co-curricular activity	10
	10.	Attendance	10
		s equivalent reduced to CCA or	
PERIODICAL	1. ANNUAL		
REVISE OF		he Unviersity may revise th	e syllabus at any time
SYLLABUS		running year after giving a	
	months.		_
SELECTED	Hopkin	s W.G. 1995, Introduction to	Plant Physiology, John

READINGS		Willey & Sons, New York USA.
	•	Salisbury F.B. & Ross C.W. 1992 Plant physiology, Widsworth
		Publishing Co. California, U.S.A.
	•	Taiz L. & Zeiger E, 1998, Plant Physiology (Second Edition)
		Sinauer Associates Inc. Publishing U.S.A.

BOTANY-PRACTICALS:-III				
Duration	n: 4 hours MAX.MARKS: 50	Min. Pass Marks : 20		
S. No.	Questions	Marks		
1.	Experiment	4		
2.	Exercise -Morphology & Anatomy	4		
3.	Soil Water tests and Field Exercise	4		
4.	Experiments in tissue culture	5		
5.	Phyto chemical tests (two)	5		
6.	Spots (six). Two from each paper	12		
7.	Viva-voce	8		
8.	Records	8		
Total =	50			
Practical List				

The following experiments are to be conducted:

- 1. Anatomy of primary and secondary growth in monocots and dicots using hand sections (or prepared slides). Structure of secondary phloem and xylem. Growth rings in wood. Microscopic study of wood in T.S., T.L.S. and R.L.S. Anomalous Sec. growth in *Boerhaavia, Nyctanthus* and *Dracaena*
- 2. Anatomy of leaf and Peel mount for stomatal types/trichomes.
- 3. Anatomy of the root. Primary and secondary structure.
- 4. Examination of a wide range of flowers available in the locality and methods of their pollination.
- 5. Structure of anther, microsporogenesis (using slides) and pollen grains (using whole mounts). Pollen viability using in vitro pollen germination.
- 6. Structure of ovule and embryo sac development (using serial sections).
- 7. Simple experiments to show vegetative propagation: leaf cuttings in Bryophyllum, Sansevieria, Begonia; stem cuttings in rose, salix, money plant, sugarcane and Bougainvillea.
- 8. Germination of non-dormant and dormant seeds.
- 9. To demonstrate osmosis using egg membrane, onion/tomato peels, potato osmoscope.
- 10. To study the effect of temperature and alcohol on the permeability of membranes.
- 11. To demonstrate plasmolysis.
- 12. To compare the water holding capacity of soils (clay, peat and sand).
- 13. To demonstrate transpiration pull.
- 14. To compare the rates of transpiration in different environmental conditions.
- 15. To demonstrate the evolution of oxygen during photosynthesis.
- 16. To compare the rates of photosynthesis under different environmental conditions.
- 17. To demonstrate the necessity of light, CO2 and chlorophyll for photosynthesis.
- 18. Separation of photosynthetic pigments by paper cHRomatography.
- 19. Demonstration of aerobic respiration.
- 20. Demonstration of anaerobic respiration.
- 21. To demonstrate the liberation of CO2 during aerobic respiration.

B.Sc. B.Ed. III Year						
COURSE CODE:	BSCBED-355E-I COURSE TYPE :CORE					
COURSE TITLE :	Complex Analysis	Complex Analysis				
MAX.MARKS:	75 MIN.PASS MARKS 30			30		
THEORY	60	MIN.PA	SS MARKS	24		
EXAMINATION:						
CONTINUOUS	15	MIN.PA	SS MARKS	6		
COMPREHENSIVE						
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RESPEC	CTIVE YE	EAR			
ELIGIBILITY						
EXAMINATION	TERM END EXAMINATION MONTHLY TEST					
DURATION	3 HR 1 HR					
Objective:						
• The objective of this	course is to introduce an	nd develo	p a clear ur	nderstanding of the		
fundamental concepts of	of Complex Analysis such	h as anal	ytic function	s, Cauchy-Riemann		
relations and harmonic functions and to make students equipped with the understanding of the						
fundamental concepts of complex variable theory. In particular, to enable students to acquire						
skill of contour integration to evaluate complicated real integrals via residue calculus.						
Learning Outcomes: Thi	s course will enable the stud	dents to:				

- Visualize complex numbers as points of R..and stereographic projection of complex plane on the Riemann sphere.
- Understand the significance of differentiability and analyticity of complex functions leading to the Cauchy-Riemann equations.
- •Learn the role of Cauchy-Goursat theorem and Cauchy integral formula in evaluation of contour integrals.
- Apply Liouville's theorem in fundamental theorem of algebra.
- Understand the convergence, term by term integration and differentiation of a power series.
- Learn Taylor and Laurent series expansions of analytic functions, classify the nature of singularity, poles and residues and application of Cauchy Residue theorem.

	Complex Plane and functions
	Complex numbers and their representation, algebra of complex
	numbers; Complex plane, Open set, Domain and region in complex
	plane; Stereographic projection and Riemann sphere; Complex functions
UNIT-1 TEACHING HOURS (10)	and their limits including limit at infinity; Continuity, Linear fractional
	transformations and their geometrical properties.
S	Analytic Functions and Cauchy-Riemann Equations
	Differentiability of a complex valued function, Cauchy-Riemann
O	equations, Harmonic functions, necessary and sufficient conditions for
	differentiability, Analytic functions; Analyticity and zeros of
	exponential, trigonometric and logarithmic functions; Branch cut and
	branch of multi-valued functions. Construction of an analytic function,
VV	conformal mapping, bilinear transformation, and its properties,
UNIT-2 TEACHIGN HOURS (12)	elementary maps. F (z) = $\frac{1}{2}$ (z+1/z), z^2, 2z, sin z and log z

[
	-	eorems and Fundamental Theo			
(17 NC	Line integral, Path independence, Complex integration, Green's				
- H S	theorem, Anti-derivative theorem, Cauchy-Goursat theorem, Cauchy				
UNIT-3 ACHIN DURS (J	integral formula, Cauchy's inequality, Derivative of analytic function,				
UNIT-3 TEACHING HOURS (12	Liouville's t	theorem, Fundamental theorem	of algebra, Maximum		
	modulus theo	brem and its consequences.	-		
	Power Series	-			
UNIT-4 TEACHING HOURS (10)		-	avlor series and Laurent		
- H S	Sequences, series and their convergence, Taylor series and Laurent series of analytic functions, Power series, Radius of convergence,				
UNIT-4 ACHIN URS (J		•	•		
LE HO	Integration and differentiation of power series, Absolute and uniform convergence of power series.				
	-	*			
UNIT-5 TEACHING HOURS (16)	Singularities and Contour Integration				
UNIT-5 FEACHING HOURS (16)	*	e functions, Zeros and poles of	· ·		
	Nature of sin	ngularities, Picard's theorem, Re	esidues, Cauchy's residue		
	theorem, Ar	gument principle, Rouche's the	eorem, Jordan's lemma,		
E H	Evaluation of	proper and improper integrals.			
TEACHING AND	1. Lectu	re method			
LEARNING		em Solving method			
STRATEGIES	-	nical method			
		nar/Symposia			
		w of literature			
	-	6. Report writing			
	7. Group Discussion				
	 Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be 				
	change as per requirement of the students and their capabilities.				
CONTINUOUS		Details of Continuous and Comprehensive Assessment (CCA) are as			
COMPREHENSIVE	follows:	L.	· · · · ·		
ASSESSMENT	S. No.	CCA- Components	Max. Marks Allocation		
(CCA)	1.	Monthly test	20*3 Test=60		
	2.	Quizzes and Assignments	10		
	3.	Viva-voce	10		
	4.	Seminar/Symposia	10		
	5.	Report writing	10		
	6.	Workshop	10		
	7.	Review of literature	10		
	8.	Creativity/Innovation	10		
	9.	Experimental Skill	10		
	10.	Co-curricular activity	10		
	<u>11.</u>	Attendance	10		
		rks equivalent reduced to CCA or	0		
EXAMINATION		xaminations are organized by			
PATTERN	-	format to enable the scholars			
PERIODICAL	1.ANNUAL	contemporary competitions and to achieve their goals.			
REVISE OF		he Universitymay revise the syll	abus at any time during		
	2.110wever, l	ne Oniversitymay revise the syn	abus at any time uut mg		

SYLLABUS	the running year after giving a notice for a period one months.
SELECTED	Ahlfors, L. V. (1973). Complex Analysis.Mcgraw Hill
READINGS	• Conway, J. B. (2012). Functions Of One Complex Variable Ii (Vol.
	159). Springer Science & Business Media.
	• Stein, E. M., & Shakarchi, R. (2003). Princeton Lectures In Analysis.
	Princeton University Press.
	• Stein, E. M., & Shakarchi, R. (2005). Real Analysis, Princeton
	Lectures In Analysis Iii.
	• Stein, R. Shakarchi (2003) Complex Analysis: Princeton Lectures In
	Analysis. Princeton University Press
	• Purohit And Goyal (2005). Complex Analysis, Jph.
	• Vasishtha, A. R. (2010).Complex Analysis: Krishna Prakashan Media
	(P) Ltd., Meeruth, 11thed,
	• Rudin, W. (2006). Real And Complex Analysis. Tata Mcgraw-Hill
	Education.
	• Rudin, W. (1970). Real And Complex Analysis P. 2. Mcgraw-Hill.
	• Lars V. Ahlfors (2017). Complex Analysis (3rd edition). McGraw-
	Hill Education.
	• Joseph Bak & Donald J. Newman (2010). Complex Analysis (3rd
	edition). Springer.
	• James Ward Brown & Ruel V. Churchill (2009). Complex Variables and Applications (9th edition). McGraw-Hill Education.
	• John B. Conway (1973). Functions of One Complex Variable.
	Springer-Verlag.
	• Copson, E. T. (1970). Introduction to Theory of Functions of
	Complex Variable. Oxford University Press.
	• Theodore W. Gamelin (2001). Complex Analysis. Springer-Verlag.
	• George Polya & Gordon Latta (1974). Complex Variables. Wiley.
	• Priestley, H. A. (2003). Introduction to Complex Analysis. Oxford
	University Press.
	• Titchmarsh, E. C. (1976). Theory of Functions (2nd edition). Oxford
	University Press.

B.Sc. B.Ed. III Year						
COURSE CODE:	BSCBED-355e-II COURSE TYPE : CORE			SE TYPE : CORE		
COURSE TITLE :	Mechanics					
MAX.MARKS:	75	MIN.PA	30			
THEORY	60	MIN.PA	SS MARKS	24		
EXAMINATION:						
CONTINUOUS	15	MIN.PA	SS MARKS	6		
COMPREHENSIVE						
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RESPEC	CTIVE YE	EAR			
ELIGIBILITY						
EXAMINATION	TERM END EXAMINA	ATION	MON	THLY TEST		
DURATION	3 HR			1 HR		
Objective:						
	ended to provide a treatment	-				
will be able to ap	ply the techniques used in	deriving a	arrange of im	portant results. The		
objective is to pro	wide the student with know?	ledge of th	ne mechanics	and an appreciation		
of their application	n to real world problems.					
Learning Outcomes: Th	is course will enable the stud	lents to:				
• Familiarize with s	subject matter, which has be	een the sin	igle centre, to	which were drawn		
	• Familiarize with subject matter, which has been the single centre, to which were drawn mathematicians, physicists, astronomers, and engineers together.					
	• Understand necessary conditions for the equilibrium of particles acted upon by various					
forces and learn the principle of virtual work for a system of coplanar forces acting on a						
rigid body.						
• Determine the centre of gravity of some materialistic systems and discuss the equilibrium						
of a uniform cable hanging freely under its own weight.						
• Deal with the kinematics and kinetics of the rectilinear and planar motions of a particle						
including the cons	trained oscillatory motions of	of particles	5.			
• Learn that a partic	cle moving under a central f	force desci	ribes a plane	curve and know the		
-	Kepler's laws of the planetary motions, which were deduced by him long before the					
-	ry given by Newton.		j	6		
	Statics					
	Equilibrium of a particle,	Fauilibri	um of a s	vstem of narticles		
	Vecessary conditions of equi			I		
UNIT-1 TEACHING HOURS (12) (12) H M	•			-		
UNIT-I HING H (12)	Moment of a force about		-	_		
	Equipollent system of forces					
v C	rirtual work for a system of	-	-	-		
d E	lifferent points of a rigid boo	-	which can be	e omitted in forming		
the equations of virtual work.						

UNIT-2 TEACHIGN HOURS (12)	<u>Centres of Gravity and Common Catenary</u> Centres of gravity of plane area including a uniform thin straight rod, triangle, circular arc, semicircular area and quadrant of a circle, Centre of gravity of a plane area bounded by a curve, Centre of gravity of a volume of revolution; Flexible strings, Common catenary, Intrinsic and Cartesian equations of the common catenary, Approximations of the catenary.				
UNIT-3 TEACHI NG HOURS (10)	Simple harm SHM under e	Rectilinear Motion Simple harmonic motion (SHM) and its geometrical representation, SHM under elastic forces, Motion under inverse square law, Motion in resisting media, Concept of terminal velocity, Motion of varying mass.			
UNIT-4 TEACHI NG HOURS (12)	acceleration i	nd kinetics of the motion, Express in Cartesian, polar and intrinsic coo , projectiles in a vertical plane and c	ordinates; Motion in a		
UNIT-5 TEACHING HOURS (14)	<u>Central Orbits</u> Equation of motion under a central force, Differential equation of the orbit, (p, r) equation of the orbit, Apses and apsidal distances, Areal velocity, Characteristics of central orbits, Kepler's laws of planetary motion <i>Generalised coordinates; D'Alembert's principle and Lagrange's equations; Hamilton equations; moment of inertia; motion of rigid bodies in two dimensions.</i> <i>Equation of continuity; Euler's equation of motion for inviscid flow; stream-lines, path of a particle; potential flow; two-dimensional and axisymmetric motion; sources and sinks, vortex motion; Navier-stokes</i>				
TEACHING AND LEARNING STRATEGIES	 equation for a viscous fluid. 1. Lecture method 2. Problem Solving method 3. Graphical method 4. Seminar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion 8. Videos/Animation 9. Self-Learning/e-Learning 10. Workshops/Experiments. * The Teaching and Learning Strategies may be change as per requirement of the students and their capabilities. 				
CONTINUOUS COMPREHENSIVE ASSESSMENT (CCA)	Tequirement of the statents and their capabilities Details of Continuous and Comprehensive Assessment (CCA) are as follows: S. No. CCA- Components Max. Marks Allocation 1. Monthly test 20*3 Test=60 2. Quizzes and Assignments 10				
	3. 4.	Viva-voce Seminar/Symposia	10 10 10		

	5.	Report writing	10		
	6.	Workshop	10		
	7.	Review of literature	10		
	8.	Creativity/Innovation	10		
	9.	Experimental Skill	10		
	10.	Co-curricular activity	10		
	10.	Attendance	10		
		irks equivalent reduced to CCA origination	-		
EXAMINATION		examinations are organized by t			
PATTERN		format to enable the scholars to			
	-	ry competitions and to achieve their			
PERIODICAL	1. ANNUAL				
REVISE OF		the Universitymay revise the s	vllahus at any time		
SYLLABUS	· · · · · · · · · · · · · · · · · · ·	e running year after giving a no	• •		
STEERBOO	months.	e running year arter giving a no	ace for a period one		
SELECTED		&Hicher P V (1994) Elementa	ry Mechanics Of Soil		
READINGS		• Biarez, J., &Hicher, P. Y. (1994). Elementary Mechanics Of Soil Behaviour: Saturated			
		Remoulded Soils. Aa Balkema.			
	 De Gennes, P. G. (1985). Wetting: Statics And Dynamics. Reviews 				
	• De Gennes, P. G. (1985). Wetting: Statics And Dynamics. Reviews Of Modern Physics, 57(3), 827.				
		 Hansson, S. O. (2012). A Textbook Of Belief Dynamics: Theory 			
	Change And Database Updating. Springer.				
	-				
	• Loney, S. L. (1914). The Elements Of Statics And Dynamics. University Press.				
		 Nguyen, Q. S. (2000). Stability And Nonlinear Solid Mechanics. Poinsot, L. (1847). The Elements Of Statics. University Press. 			
		L. (1960). Classical Dynamics. In 1	-		
		cs And Field Theory/Prinzipien Der	*		
		theorie (Pp. 1-225). Springer, Berlin,			
		. L. (2006). An Elementary Treatise	-		
	•	nd of Rigid Bodies. Read Books.	on the Dynamics of a		
		, P. L. (1964). Elementary Dynamics	s Ram Narin I al Beni		
	Prasad Pu		. Rum Rum Lui, Dom		
	Allahaba				
		. L. & Griffith, B. A. (1949). Pri	nciples of Mechanics		
	McGraw-		nerpies of wicehames.		
		A. S. (2009). Statics. Cambridge Uni	versity Press		
	•	A. S. (2009). Dynamics. Cambridge Off	•		
	•		•		
l	• v arma, R	R. S. (1962). A Text Book of Statics.	roulisilala PVI. LIG.		

Phase-I: Internship (4 weeks)

	Assessment is based on the following activities –	
Planning	Content Analysis and mode of transaction (Assignment in teaching subject)	5
	Creating and maintaining teaching learning material for the school (which can become valuable resource for the regular teachers of the school). a) TLM in any teaching subject	5+5=10
	Make lesson plan using 10- different methods in which 5 must involve,student could develop their own method (fusion based) with the help on teacher educator.	5
Planning &	Identify a problem of action research and draft proposal on it.	5
Execution	*Innovative Micro Teaching (5 Skills) (Teaching Subjects at secondary to senior secondary level)	5
Execution	One-week, regular observation of regular teacher (at the beginning of practice teaching.)	5
	Delivery of Four lessons based on model of teaching. (After each lesson of practice teaching student teacher need to discuss with subject teacher on their pedagogy and new practices it must be seconded.)	5
Assessment & Evaluation	Draft a report based on: - continuously and comprehensively evaluating students' learning for feedback into curriculum and pedagogic practice.	5
Regularity and involvement in different school activities	Observation of day-to-day * school activities and report of an in- depth study of four activities.	5
	TOTAL MARKS	50

* Note: Any of the above activity may be replaced as per the need of the course

SUGGESTED SCHOOL ACTIVITIES: -

Select any one activity from each group given below: -

A Group: -

- Organization of cultural activities,
- Organization of literary activities
- Organization of games/sports.
- Framing of time table.
- Water resource management tHRough traditional methods.
- Prepare a report after interview of effective/good teachers.

B Group-

- Attending and organizing morning assembly
- Maintenance of classroom discipline
- Review of School Records
- Guidance and Counseling
- Gardening

CGroup -

- Organizing science fair, exhibition, science club, nature study
- Maintenance of School library
- Maintenance of School laboratories.
- Health and hygiene.
- Study on role of community for school improvement
- School mapping

D Group-

- Sensitization for environmental problems.
- Cleaning campaigning in school.
- School climate/Environment (any one aspect)
- Voluntary services.
- Mass awareness of social evils and taboos.
- Any other activity/s decided by the institute.

FOURTH YEAR

Course code	Title of the course	EVALUATION			
		External	Internal	Practical	Total
BSCBED-450	Environmental Studies*	70	30	-	100
BSCBED-451	Assessment for Learning	70	30	-	100
BSCBED-452	Agriculture	35	15	-	50
BSCBED-453	Pedagogy of school subject-II	-	1		
BSCBED-453 I	1. Mathematics	35	15	-	50
BSCBED-453 II	2. General Science	35	15	-	50
BSCBED-453 III	3. Chemistry	35	15	-	50
BSCBED-453 IV	4. Biology	35	15	-	50
BSCBED-453 V	5. Physics	35	15	-	50
BSCBED-454	Content: (PCB & PCM)				
BSCBED-454 a I	1. Physics I	40	10	50	100
BSCBED- 454 b I	2. Chemistry I	40	10	50	100
BSCBED- 454 c I	3. Zoology I	40	10	50	100
BSCBED- 454 d I	4. Botany I	40	10	50	100
BSCBED- 454 e I	5. Mathematics I	80	20	-	100
	CCA				
	Prayer, Yoga, Meditation &	Festival etc			25
BSCBED- 455	Internsh	up (16 weeks)		200
Total					

*Marks of Compulsory Subject shall not be added in the Total Marks. Internship (16Weeks) included in Total Marks.

B.Sc.B.Ed. IV Year					
COURSE CODE:	BSC BED-450		COURSE TYP	E: CORE	
COURSE TITLE:	Environmental Studies (Com	p.)			
MAX. MARKS:	50	MIN. PASS MARKS: 20			
THEORY	35	MIN. PASS MARKS: 14			
EXAMINATION					
CONTINUOUS	15	MIN	. PASS MARKS:	6	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECTI	IVE Y	EAR		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATION	ON	MONTHLY	TEST	
DURATION	03 HR		01 HR		
 To impart basic km To develop an attit To motivate put improvement. To acquire skills environmental pro Make enable to str 	eness about environmental probl lowledge about the environment tude of concern for the environm blic to participate in enviro s to help the concerned ind blems. iving to attain harmony with Nat r completion of the course, stud ckground of EVS as a composi	and its ent. nment ividua ure. lent-t	allied problems. protection and e is in identifying a eachers will be able -	nd solving	
science, social scieTo develop under	ence and environs mental educati standing about various ecosystem o develop the ability to plan co	on. ns and	biodiversity.		

- Understanding about the issues of conservation and environmental regeneration have been infused at appropriate places in all the textbooks.
- •
- To analyze and understand environment concerns tHRough the process of inquiry. To develop a sense of awareness about the environment hazards and its causes and • remedies.

Temedies.	
	Introduction to Environmental Studies
1 12)	• Environmental studies: Historical background, Concept, Nature and Scope of EVS.
	 Natural and Social Environment: Concept, its Components, and Relationship, Man & Environment, Man on Environment. Disciplinary and Multidisciplinary approach of EVS.
A CHI DURS	Assignment:
UTEA HOU	 Visit a natural or man made site (park/forest/zoo/KVK/forest department etc) then discuss with student about their perception on natural environmental and its relation with social environmental. Conduct awareness campaigning for plantation of Tulsi, Neem, and Khejri etc.

	Eco-systems
UNIT-2 TEACHING HOURS (12)	 Concept of an Ecosystem. Structure and function of an Ecosystem. Producers, Consumers and decomposers. Energy flow in Ecosystem. Ecological Succession. Food Chain, Food webs and Ecological pyramids. Introduction, Types, Characteristic Features, Structure and Function of the following Ecosystem- Forest Ecosystem. Grassland Ecosystem. Celebrate Important relevant days related to environmental conservation (such as earth day, world environmental days etc) in school or out of school with the help of students and make a systematic report on entire activities or work Make a bulletin board material on "Role of Individual in prevention of Pollution" and analyze its impact on students and
	teachers of schools.
UNIT-3 CHING HOURS (12)	 Biodiversity Its Conservation Introduction – Definition: Genetic, Species and Ecosystem Diversity. Value of Biodiversity & Biodiversity at Global, National & Local levels. Hot-Spots of Bio-diversity. THReats to Bio-diversity: Habitat loss, poaching of Wild life, Man Wild life Conflicts. Conservation of Bio-diversity: In-situ and Ex-situ conservation of Bio-diversity.
	Assignment:
UN TEACHING	 Conduct an activity in school, in which ask students to get opinion of their grandparents in changing life style and their merits and demerits/or collect their ideas on local products which can be helpful in healthy life style the pupil teacher will compile their experiences and draft a report then present it in class. Conduct a seminar in your village/ district /town in which invite local people of various socio-economic/socio cultural background to present their experiences on sustainable agriculture.
4 0 0	Environmental Issues
UNIT-4 TEACHING HOURS (11)	 Meaning, Concept, Process, Effects and Preventive. Action of Acid rain, Global warming and cooling. Natural & Manmade disasters. Meaning, Concept, Effects and Preventive action of Water pollution. Air Pollution. Noise Pollution. Land or Soil pollution.

	Assignme	ent:		
	 Conduct an orientation programme in rural / urban school on waste management. Organize a planned Visit to urban or rural area to study about waste 			
	produced by human, after visiting the site, present your report in local community to create awareness for resolution of the problem.			
	Environmental Conservation & Management			
(11)	 Meaning Manage 	g, Concept and Importance of Environmental (Conservation &	
RS		of women in Conservation: Chipko Mov	ement. Kheiri	
00 00	Movement.			
		erism and waste generation and its managemen		
UNIT-5 TEACHING HOURS (11)	•	tural/Urban waste: their impact and managemen	t.	
H H	Assignme	ent:		
		the direct or indirect message of Tradition		
		of your area for social or natural environment en t least one tree and cultivate it tHRoughout the		
TUTORIALS		rial class once a week (12)	years.	
TEACHING AND		ectures		
LEARNING		learning		
STRATEGIES	3. V			
		ktension Lectures		
		ontent Review elf-Learning		
		roup Discussions		
		eld Visit		
	9. Si	irvey		
		ocumentaries		
		nort Films		
		eam Teaching	• • • •	
		ching strategies are subject to change as per dents and their capabilities.	requirement	
CONTINUOUS &		f Continuous and Comprehensive Assessmen	t (CCA) are as	
COMPREHENSIVE	follows:	Continuous and Comprehensive Assessmen		
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM	
(CCA)	NO.		MARKS	
	1	Monthly Test	10X6 Test =	
			60	
	$\frac{2}{3}$	Presentation Group Discussion	<u>10</u> 10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	

	EXPLANATION (METHOD TO ASCERTAIN MARKS FOR			
	CCA):			
	CCA will be reduced to 30 marks or 15 marks (as per course			
	weightage).			
	Formula: Marks obtained/Total marksX30			
	For example: $60 \div 160X30 = 11.25$			
	PROVISO-I: Provided that a candidate shall be granted a relaxation			
	in the form of exemption from CCA components, however, not more			
	than 3 in a respective course.			
	PROVISO-II: Provided further that			
	candidate to appear in the monthly te	est conducted in the respective		
	course.			
	*Attendance in Lectures and Pract	tical		
	Percentage	Marks Allotted		
	75% to 80%	02		
	81% to 85%	04		
	86% to 90%	06		
	91% to 95%	08		
	Above 96%	10		
Examination	Term-end examinations are organized			
Pattern	format to enable the scholars to a			
	competitions and to achieve their goal	1 2		
PERIODICAL	1. ANNUAL	5.		
		anian the sullabora of our times		
REVISION OF	2. However, the Universitymay re			
SYLLABUS	J J J J J J J J J J J J J J J J J J J	during the running year after giving a notice for a period one		
	month.	Dharti li Dalan Dadhalarishaa		
SELECTED	• Bahuguna, Sundarlal (1996). I	Dharti ki Pukar. Radhakrishna		
SELECTED READINGS	• Bahuguna, Sundarlal (1996). I Publication: Delhi.			
	 Bahuguna, Sundarlal (1996). I Publication: Delhi. Kaushik, A. and Kaushik, C.P.(200 	4). Perspectives in Environmental		
	 Bahuguna, Sundarlal (1996). I Publication: Delhi. Kaushik, A. and Kaushik, C.P.(200 studies. New Age International(P) I 	4). Perspectives in Environmental Ltd. Publishers: New Delhi.		
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	B.Sc.B.Ed. IV Year	•]			
COURSE CODE:	BSCBED-451	[COURSE TYPE	· CODE			
COURSE CODE: COURSE TITLE:			COURSE I YPE	: COKE			
MAX. MARKS:	Assessment for Learning	MIN. PASS MARKS: 40			MIN PASS MARKS: 10		40
THEORY		MIN. PASS MARKS: 40 MIN. PASS MARKS: 28					
EXAMINATION	70	MIIN. PASS MARKS. 20					
CONTINUOUS	30	MIN. PASS MARKS: 12					
COMPREHENSIVE	50	MIIN. PASS MARKS: 12					
ASSESSMENT (CCA)							
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR						
ELIGIBILITY							
EXAMINATION	TERM END EXAMINATIO	DN	MONTHL	Y TEST			
DURATION	03 HR		01 H	IR			
Objectives:							
To Understand Con	ncept of Measurement Assessment	and H	Evaluation.				
• To Understand Typ	bes of Measurement and Evaluation	n.					
• Make to Know abo	ut Continuous and Comprehensive	e Eval	uation.				
• To Understand Eva	luation Tools.						
• Make to Know abo	ut Characteristics of Good Evalua	tion.					
• Make to Know abo	out analysis process of Students.						
Learning Outcomes: Af	ter completion of the course, stude	nts wi	ll be able to:				
• To understand a	ssessing children's progress, b	oth i	n terms of their	psychological			
development and	the criteria provided by the curricu	lum.					
 To provide broad 	outlook to go beyond the limited	conte	ext of syllabus-base	ed achievement			
testing, achieveme	ent scores in a subject linked with	the ch	ild's overall develo	pment.			
• Introduce student	teachers to the history of evaluation	n and	current practices.				
• Understand the d	ifferent dimensions of learning an	nd rel	ated Evaluation pr	ocedures, tools			
and techniques. A	nalyse, Manage and interpret asses	ssmen	t data.				
 Understanding the table 	ne policy perspectives on exa	amina	tion and evaluat	ion and their			
implementation pr	actices. Assessment for culturally	respo	nsive in diverse cla	issroom.			
• Develop critical u	nderstanding of issues in evaluation	on an	d explore realistic,	comprehensive			
and dynamic asses	ssment process which is culturally	respo	nsive for use in the	classroom.			
	processes which lead to better lean	ming	and more confident	and			
creative learners.							
	Concept of Evaluation						
$\widehat{}$	• Meaning & concept of As						
(5)	and their Interrelationship		-	-			
SS	6 6		oviding Feedbac	k, Promotion,			
5	Diagnosing). Principles of A						
HO HO	• Functions of Measurement		-				
UNIT-1 UNG HO	Process, Types of Measu			asurement and			
UNIT-1 TEACHING HOURS (24)	Evaluation. Techniques of H	zvalua	tion.				
E E	Assignment:	•		1			
(V)	• Observe the teaching learn	ing pi	cocess in class roor	n and prepare a			
H	report and feedback on it.		1.00 1	A			
	• Conduct a group discussi		n differnce betwee	en Assessment,			
	Measurement and Evaluati	on.					

UNIT-2 TEACHING HOURS (24)	 Type of Evaluation Classification of Assessment: Base on purpose (Prognostic, Formative, Diagnostic and Summative), Scope (Teacher made, Standardized) Attribute measured (Achievement, Aptitude, Attitude, etc.), Nature of information gathered (Qualitative, Quantitative) Mode of response (Oral and written), Nature of interpretation (norm referenced, criteria referenced). Item Analysis. Assignment: Presentation of papers on examination and evaluation policies. Prepare a diagnostic test of any subject and apply it on students and give suggestions for improvement.
UNIT-3 TEACHING HOURS (24)	 Continuous and Comprehensive Evaluation Meaning, concept need and process and characteristics of CCE. Assessment of Attitude and Values, Interest, Intelligence, Personality, self-concept items and procedures for their assessment. Grading: Concept, types and Application, Indicators for grading. Assignment: Prepare a plan for continuous and comprehensive evaluation of students. To assess self concept of the students in class room and provide them feedback for it.
UNIT-4 TEACHING HOURS (22)	 Evaluation Tools Use of projects, Assignments, Worksheet, Practical Work, and Performance based activities, seminars and reports as assessment devices. Self, Peer and Teacher Assessment. Commercialization of assessment. Construction an Achievement test Typology of questions. Assignment: Organise a group activity (like competition/story telling/reading/writing) and get it assessed by self, peer and teacher. Draft a feedback form about overall performance of students from parents and teacher, evaluate its effectiveness too.
UNIT-5 TEACHING HOURS (22)	 Evaluation Tools Construction and Selection of items, Guidelines for Construction of test items, assembling the test items, Guideline for administration. Characteristics of Good Evaluation System– Reliability, Validity, Objectivity, Comparability, Practicability. Analysis and interpretation of student's performance processing test, performance, calculation of percentage (Measures of Position), Central tendency measurement, Co-efficient of Correlation, Product Moment and Rank difference, Graphical Representations. Assignment: Determination of Reliability or Validity of any self made test. Construction, administration and interpretation of self made achievement text.

TEACHING AND	1	Lectures	
LEARNING		E-learning	
STRATEGIES		Videos	
SIKALEGIES		Extension Lectures	
		Content Review	
		Self-Learning	
		Group Discussions	
		Field Visit	
		Survey	
		Documentaries	
		Short Films	
		Team Teaching	
	* The t	eaching strategies are subject to change as pe	er requirement of
		dents and their capabilities.	
CONTINUOUS &		of Continuous and Comprehensive Assessme	ent (CCA) are as
COMPREHENSIVE	follows	:	
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM
(CCA)	NO.		MARKS
	1	Monthly Test	10X6 Test = 60
	2	Presentation	10
	3	Group Discussion	10
	4	Debate	10
	5	Participation and Presentation in Seminar	10
	6	Report Writing	10
	7	Viva Voce	10
	8	Attendance*	10
	<u>o</u> 9	Co-curricular Activity	10
	9 10	Team Teaching	10
	CCA)	ANATION (METHOD TO ASCERTAIN M	AKKSFUK
	· · · · ·	will be reduced to 30 marks or 15 marks (as per	2011#20
	weigh	· 1	course
	-	ila: Marks obtained/Total marksX30	
		cample: $60 \div 160 \times 30 = 11.25$	
		1	ad a malauration in
		VISO-I: Provided that a candidate shall be grant	
		rm of exemption from CCA components, howev respective course.	er, not more than
		1	atom for a
	PROV	ISO-II: Provided further that this will be mand	•
	PROV candid	ISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the	•
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the.	•
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the mance in Lectures and Practical	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the and ance in Lectures and Practical Percentage Marks Allotted	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the andance in Lectures and Practical Percentage Marks Allottee 75% to 80% 02	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the mance in Lectures and Practical Percentage Marks Allotted 75% to 80% 02 81% to 85% 04	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the mance in Lectures and Practical Percentage Marks Allotted 75% to 80% 02 81% to 85% 04 86% to 90% 06	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted in the mance in Lectures and Practical Percentage Marks Allotted 75% to 80% 02 81% to 85% 04 86% to 90% 06	ne respective
	PROV candic course	VISO-II: Provided further that this will be mand late to appear in the monthly test conducted	ne respective

EXAMINATION	Term-end examinations are organized by the university in the prescribed			
PATTERN	format to enable the scholars to achieve success in contemporary			
	competitions and to achieve their goals.			
PERIODICAL	1. ANNUAL			
REVISION OF	2. However, the Universitymay revise the syllabus at any time during			
SYLLABUS	the running year after giving a notice for a period one month.			
SELECTED	• अस्थाना, विपिन (२००९). मनोविज्ञान और शिक्षा में मापन			
READINGS	एवं मूल्यांकन. अग्रवाल प्रकाशनः आगरा.			
	• पाल, हंसराज एवं शर्मा, मंजूलता (२००९). मापन,			
	आकलन एवं मूल्यांकन. शिक्षा प्रकाशनः जयपुर.			
	• पाण्डेय, श्रीधर (२०१०). शिक्षा में मापन मूल्यांकन. भवदीय			
	प्रकाशनः फैजाबाद.			
	• शर्मा, ज्योति (२००९).शैक्षिक मापन एवं मूल्यांकन. अग्रवाल			
	प्रकाशनः आगरा.			
	• Ecclestone, Kathryn (2010). Transforming formative			
	assessment in lifelong learning. Mc Graw Hill. Eng.			
	• गुप्ता, एस. पी.(२०१०). आधुनिक मापन एवं मूल्यांकन.			
	प्रयागराज ः शारदा पुस्तक मन्दिर.			

B.Sc.B.Ed. IV Year						
COURSE CODE:	BSCBED-452		COU	RSE TYI	PE: CORE	
COURSE TITLE:	Agriculture (Specialization)	Agriculture (Specialization)				
MAX. MARKS:	50	MIN. PASS MARKS: 20			20	
THEORY	35	MIN. PASS MARKS: 14			14	
EXAMINATION						
CONTINUOUS	15	MIN. PASS MARKS: 6			6	
COMPREHENSIVE						
ASSESSMENT (CCA)						
ATTENDANCE	80 PERCENT IN RESPECTIVE	YEAI	k			
ELIGIBILITY						
EXAMINATION	TERM END EXAMINATIO	N	MO	NTHLY		
DURATION Objectives:	03 HR			01 HR		
 Understand al Acquire skills Understand pr Recognise dif 	 Learning outcomes: - On completion of the course, the student teacher will be able to Understand the meaning and scope of agriculture. Understand all about seeds, imported weed, manures etc. Acquire skills to practices of seed sowing, planting materials etc. Understand practices of different ornamental and horticulture crops. Recognise different field practices like earthing, hoeing, weeding watering etc. Inculcate healthy values related to work culture. 					
UNIT-1 TEACGIN G HOURS (10)	Agriculture: Meaning, definition, scope, history, branches and objectives					
Soil Science: Definition of pedology, soil management, soil erosion, soil conservation practices; structure of soil, soil profile; soil fertility and productivity, essential plant nutrients. Fertilizers and manures including bio-fertilizers. Identification of manures and fertilizers						
EXAMPLE 1 Irrigation: Definition, method of irrigation, systems of irrigation, drainage, irrigation pattern of India.						
UNIT-4 TEACGI NG HOURS (12)	Horticulture: Definition, branches of horticulture, layout of orchards, propagation by seeds and by vegetative means; Pot filling technique; Planning, planting and maintaining lawn; Practice related to landscaping.					
UNIT-5 TEACGING HOURS(12)	Agricultural practices: Preparation of land, selection of seeds, watering, thinning, noeing and weeding, harvesting of crop, identification of important agricultural ools, trees and crop plants. Minor project preparation on agriculture.					

	D	/T° - 1 - 1 XX/ 1- (4		
		n/Field Work(any two of the state of the sta	0.	Wheet Dains Mains
		ntification of an agronomy se etc.	y of following crops	. wheat, Бајга, Maize,
		ricultural Processes:Irrigation	on TrainingandD	uning,Hoeing and
		eding,SeedBed	on, manninganur	uning, noenig and
TEACHING AND	1. Le			
LEARNING		earning		
STRATEGIES	$\frac{2}{3}$. Vi	0		
SINALOILS		tension Lectures		
		ntent Review		
		f-Learning		
		oup Discussions		
		ld Visit		
	9. Su			
		cumentaries		
	11. Sh	ort Films		
	12. Te	am Teaching		
	* The tead	ching strategies are subjec	t to change as per re	equirement of the
	students a	nd their capabilities.		
CONTINUOUS &	Details of	Continuous and Compreh	ensive Assessment (CCA) are as follows:
COMPREHENSIVE				
ASSESSMENT	SR.NO.	CCA: COMP	ONENT	MAXIMUM
(CCA)				MARKS
	1	Monthly Test		10X6 Test = 60
	2	Presentation		10
	3	Group Discussion		10
	4	Debate		10
	5	Participation and Presen	tation in Seminar	10
	6	Report Writing		10
	7	Viva Voce		10
	8	Attendance*		10
	9	Co-curricular Activity		10
	10	Team Teaching		10
	EXPLANATION (METHOD TO ASCERTAIN MARKS FOR CCA):			
	CCA will be reduced to 30 marks or 15 marks (as per course weightage).			
	Formula: Marks obtained/Total marksX30			
	For example: 60÷160X30 =11.25			
	PROVISO-I: Provided that a candidate shall be granted a relaxation in the form			
	of exemption from CCA components, however, not more than 3 in a respective			
	course. PROVISO-II: Provided further that this will be mandatory for a candidate to			
				•
		appear in the monthly test conducted in the respective course. *Attendance in Lectures and Practical		
	1 ittenue	Percentage	Marks Allotted	
		75% to 80%	02	
		81% to 85%	04	
		86% to 90%	06	
		91% to 95%	08	
		Above 96%	10	

Examination	Term-end examinations are organized by the university in the prescribed format to
Pattern	enable the scholars to achieve success in contemporary competitions and to
	achieve their goals.
PERIODICAL	1. Annual
REVISION OF	2. However, the Universitymay revise the syllabus at any time during the
SYLLABUS	running year after giving a notice for a period one month.
SELECTED	• Bleasdale, J. K. A. (1973). Plant physiology in relation to horticulture.
READINGS	Macmillan International Higher Education.
	• Dubey, D. K. (2008). <i>Fruit Production in India</i> .Meerut: Rama Publishing House.
	• Edmond, J. B., Senn, T. L., Andrews, F. S., &Halfacre, R. G. (1975). <i>Fundamentals of horticulture</i> (No. 4th ed.). McGraw-Hill, Inc.
	• Panda, S. C. (2005). <i>Agronomy</i> . Agrobios. Varanasi:Kushal Publications and Distributors.
	• Sing, Jaiveer (2002). <i>Plant Propagation & Nursery Husbandry</i> . Meerut: Rama Publishing House.
	• Singh, J. (2014). <i>Basic Horticulture</i> . New Delhi" Kalyani publishers.

	B.Sc.B.Ed. IV Ye	ar		
COURSE CODE:	BSCBED-453-I		COURSE TYP	E: CORE
COURSE TITLE:	Pedagogy of Mathematics			
MAX. MARKS:	50	MIN. PASS MARKS: 20		20
THEORY	35	MIN. PASS MARKS: 14		14
EXAMINATION				
CONTINUOUS	15	MIN. PASS MARKS: 6		
COMPREHENSIVE				
ASSESSMENT				
(CCA)				
ATTENDANCE	80 PERCENT IN RESPECTI	VE YE	AR	
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATI	ON	MONTHI	
DURATION Objectives:	03 HR		01 H	łR
 Critique and deve Develop the abi promoting Mather Understand the A Understand the C Learning outcome: After Develop insight in Learn important procedures. See mathematics them-selves to wo Appreciate the im Stimulate curiosit 	nd Secular country. he nature of Mathematics curriculum and its pedagogical issues. develop suitable evaluation mechanisms in Mathematics ability to organize co-curricular activities and community resources for athematics learning. he Approaches to teaching of mathematics he Concept of Teaching Skills After completion of the course, student-teachers will be able - ght into the meaning, nature, scope and objective of mathematics education. tant mathematics: mathematics is more than formulas and mechanical attics as something to talk about, to communicate tHRough, to discuss among o work together on; Pose and solve meaningful problems. e importance of mathematics laboratory in learning mathematics. iosity, creativity and inventiveness in mathematics. petencies for teaching-learning mathematics tHRough various measures.			
	• Meaning Nature and Sco		lathematics teaching	ησ
UNIT-1 TEACHING HOURS (12)	 Meaning, Nature and Scope of Mathematics teaching. History of mathematics teaching and contribution of mathematician with reference to-Bhaskaracharya, Aryabhatta, Ramanujan, Euclid, Pythogores etc. Importance of Aims and Objectives of Mathematics Teaching. Co-relation with other subjects of Mathematics. Assignment: Prepare a model of Pythagoras Theorem/Cube, Cuboid, and Cone etc. Organise seminar on contribution of mathematician. 			

	Aims and Objectives of Teaching School Mathematics
UNIT-2 TEACHING HOURS (12)	 Aims and general objectives of teaching mathematics, Bloom's Digital Taxonomy (Cognitive, Effective and Psychomotor) in terms of Instructional Behaviour, the objectives of school education; writing specific objectives of various content areas in mathematics like algebra, geometry, trigonometry, etc. Approaches to teaching of mathematics – Analytic-Synthetic, Inductive-Deductive, Heuristic, Problem Solving, Project and Laboratory etc. Using various techniques of teaching mathematics viz-oral, written, drill, assignment, team teaching, supervised study and programmed learning. Assignment: Organise workshop on techniques of teaching mathematics. Organise a training program on Bloom's Digital Taxonomy.
	Planning
UNIT-3 TEACHING HOURS (12)	 Planning Concept, Meaning and Objectives of Mathematics teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these plans. Meaning and Concept of Teaching Skills. Micro Teaching-Meaning, Need and Importance of Micro Teaching Cycle and its Features. Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good Curriculum. Dale cone of experiences. Assignment: Prepare and present a lesson tHRough power point presentation on any topic of your choice. Organise a training program on Micro Teaching.
	Teaching-learning Resources in Mathematics
UNIT-4 TEACHING HOURS (11)	 Meaning, Objectives, Scope, Characteristics, Types, Preparation, Presentation and Importance of Teaching Learning Material. Planning and Importance of Mathematics Laboratories and its uses. Qualities of Mathematics Text Books at Secondary Level. Audio-Visual Aids- Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the teaching of Mathematics. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of Mathematics. Make a report on Mathematics Teaching Planning and Importance of Mathematics Laboratories and its uses.

	A		
1-5 12)		ment and Evaluation	A 1 •
	•	Meaning, concept and construction of	Achievement test,
		diagnostic test and remedial teaching.	
		Types of Questions, Characteristics of a good te	
UNIT-5 TEACHING HOURS (12)	•	Blue print: Meaning, concept, need and constru-	ction.
	•	Continuous and Comprehensive Evaluation: Me	eaning, concept,
		importance and limitations.	
		Models of Teaching.	
	Assign		
	-	Construction, administration and interpretation	of an achievement
		test of any standard of school.	of all define verificiti
	1	Make a diagnostic test of your subject and appl	ly it in school after
	•		
	1	discussion with concerning teacher and give rer	neurar measure.
TEACHING AND		Lectures	
LEARNING		E-learning	
STRATEGIES		Videos	
		Extension Lectures	
		Content Review	
		Self-Learning	
		Group Discussions	
		Field Visit	
		Survey	
		Documentaries	
		Short Films	
		Team Teaching	•
		teaching strategies are subject to change as pe	er requirement of
	the stu	dents and their capabilities.	
CONTINUOUC			
CONTINUOUS &	Details	of Continuous and Comprehensive Assessn	nent (CCA) are as
COMPREHENSIVE	Details follows	of Continuous and Comprehensive Assessn	·
COMPREHENSIVE ASSESSMENT	Details follows SR.	of Continuous and Comprehensive Assessn	MAXIMUM
COMPREHENSIVE	Details follows SR. NO.	of Continuous and Comprehensive Assessn :: CCA: COMPONENT	MAXIMUM MARKS
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1	of Continuous and Comprehensive Assessn :: CCA: COMPONENT Monthly Test	MAXIMUM MARKS 10X6 Test = 60
COMPREHENSIVE ASSESSMENT	Details follows SR. NO.	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation	MAXIMUM MARKS
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1	of Continuous and Comprehensive Assessn :: CCA: COMPONENT Monthly Test	MAXIMUM MARKS 10X6 Test = 60
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation	MAXIMUM MARKS 10X6 Test = 60 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion	MAXIMUM MARKS 10X6 Test = 60 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance*	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA)	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M :	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA	of Continuous and Comprehensive Assessing: CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M :: will be reduced to 30 marks or 15 marks (as per	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA weigh	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M : will be reduced to 30 marks or 15 marks (as per tage).	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA CCA weigh Formu	 of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M : will be reduced to 30 marks or 15 marks (as per tage). ala: Marks obtained/Total marksX30 	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10 10 10
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA CCA weigh Formu For ex	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M : will be reduced to 30 marks or 15 marks (as per tage). ala: Marks obtained/Total marksX30 cample: 60÷160X30 =11.25	MAXIMUM MARKS 10X6 Test = 60 10 10 10 10 10 10 10 10 10 10 10 20 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA CCA weigh Formu For ex PRO	 of Continuous and Comprehensive Assessing: CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M will be reduced to 30 marks or 15 marks (as per tage). ana Marks obtained/Total marksX30 cample: 60÷160X30 =11.25 VISO-I: Provided that a candidate shall be grant 	MAXIMUM MARKS 10X6 Test = 60 10 a relaxation in
COMPREHENSIVE ASSESSMENT	Details follows SR. NO. 1 2 3 4 5 6 7 8 9 10 EXPI CCA CCA weigh Formu For ex PROV the for	of Continuous and Comprehensive Assessn CCA: COMPONENT Monthly Test Presentation Group Discussion Debate Participation and Presentation in Seminar Report Writing Viva Voce Attendance* Co-curricular Activity Team Teaching ANATION (METHOD TO ASCERTAIN M : will be reduced to 30 marks or 15 marks (as per tage). ala: Marks obtained/Total marksX30 cample: 60÷160X30 =11.25	MAXIMUM MARKS 10X6 Test = 60 10 a relaxation in

	PROVISO-II: Provided further that this will be mandatory for a
	candidate to appear in the monthly test conducted in the respective
	course.
	*Attendance in Lectures and Practical
	Percentage Marks Allotted
	75% to 80% 02
	81% to 85% 04
	86% to 90% 06
	91% to 95% 08
	Above 96% 10
EXAMINATION	Term-end examinations are organized by the university in the prescribed
PATTERN	format to enable the scholars to achieve success in contemporary
	competitions and to achieve their goals.
PERIODICAL	1. ANNUAL
REVISION OF	2. However, the Unviersity may revise the syllabus at any time
SYLLABUS	during the running year after giving a notice for a period one
	month.
SELECTED	• धाकड़, परशुराम एवं त्रिवेदी, शिल्पा (२००୨). गणित
READINGS	शिक्षण विधियॉ. साहित्यागार चोड़ा रास्ताः जयपुर.
	 मंगल, एस.के. (2005). गणित शिक्षण. आर्य बुक डिपो. नई दिल्ली.
	• शर्मा, एच.एस. (२००५). गणित शिक्षण. राधा प्रकाशन
	मन्दिरः आगरा.
	 नेगी, जे.एस. (२००७). गणित शिक्षण. विनोद पुस्तक मन्दिर. आगरा.
	 सिंह,योगेश कुमार (2010). गणित शिक्षण आधुनिक
	पद्धतियाँ. ए.पी.एच.पब्लिशिंग कॉरपोरेशनः नई दिल्ली-02.
	• कुलश्रेष्ठ, अरूण कुमार (२०१३). गणित शिक्षण. आर. लाल.बुक डिपोः मेरठ

B.Sc.B.Ed. IV Year				
COURSE CODE:	BSCBED-453-II COURSE TY		YPE: CORE	
COURSE TITLE:	Pedagogy of General S	Pedagogy of General Science		
MAX. MARKS:	50	MIN. PASS	MARKS:	20
THEORY	35	MIN. PASS	MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN. PASS	MARKS:	6
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION MONTHLY TEST		NTHLY TEST	
DURATION	03 HR 01 HR			

Objectives:

- To develop the ability to Students insight on the meaning and nature of General science for determining aims and strategies of teaching- learning.
- To develop the ability to Students appreciate the fact that every child possesses curiosity about his/her natural surroundings.
- Students will be able to identify and relate everyday experiences with learning of science.
- Students will be able to integrate the science knowledge with other school subjects.
- Students will be able to analyze the contents of science with respecttopots, branches, process skills, knowledge organization and other critical issues.
- Students will be able to identify the concepts of science.

Learning outcomes: After completion of the course, student-teachers will be able to:-

- Develop insight on the meaning and nature of General science for determining aims and strategies of teaching-learning.
- Appreciate that science is a dynamic and expanding body of knowledge.
- Appreciate the fact that every child possesses curiosity about his/her natural surroundings.
- Identify and relate everyday experiences with learning of science.
- Appreciate various approaches of teaching- learning of science.
- Explore the process skill in science and role of laboratory in teaching-learning.
- Use effectively different activities /experiments /demonstrations /laboratory experiences for teaching-learning of science.
- Integrate the science knowledge with other school subjects.
- Analyze the contents of science with respecttopots, branches, process skills, knowledge organization and other critical issues.
- Develop process-oriented objectives based on the content themes/units.
- Identify the concepts of science that are alternatively conceptualized by teachers and students in general.

	Nature and Scope of General Science Teaching
UNIT-1 TEACHING HOURS (12)	 Meaning, Nature and Scope of General Science teaching. Contribution of Scientist:-Har Govind Khurana, J .C.Boss, C.V. Raman. Chander Shekhar, A.P.J.Kalam. Importance, Aims and Objectives of General Science Teaching. Co-relation with other Subjects Journal and Referenced Book and daily routine. Assignment: Group discussion on importance of General Science. Organise seminar on Contribution of Scientist in General Science and preapre a report.
	Aims and Objectives
UNIT-2 TEACHING HOURS (12)	 Bloom's Taxonomy (Cognitive, effective and psycho. motor) In terms of Instructional Behavior. Quality and responsibilities of Science teacher. Methods of General Science teaching Subject- lecture method, Demonstration, Lab Method, Problem Solving, Heuristics Project Method, Inductive and deductive Method. Techniques of General Science Teaching. Assignment: Organise workshop on techniques of teaching General Science. Organise a training program on Bloom's Digital Taxonomy.
	Planning Planning
UNIT-3 TEACHING HOURS (12)	 Concept Meaning and Objectives Teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these Plans. Meaning and Concept of Teaching Skills- micro Teaching - Meaning. Need and Importance. Micro-Teaching Cycle and its features. Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good curriculum. Evaluation of General Science Syllabus at Secondary level Assignment: Prepare and present a lesson tHRough power point presentation on any topic of your choice. Organise a training program on Micro Teaching.
	Teaching-learning Resources in General Science
UNIT-4 TEACHING HOURS (11)	 Meaning, Objectives, Scope, Characteristics, Types, Preparation, Presentation and Importance of Teaching Learning Material. Planning and Importance of General Science Laboratories and its uses. Qualities of General Science Text Books at Secondary Level. Audio-Visual Aids- Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the teaching of General Science. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of General Science. Make a report on General Science Teaching Planning and Importance of General Science Laboratories and its uses.

		t and Evaluation	ant toot discussed	
		aning, concept and construction of Achieveme	ent test, diagnostic	
		and remedial teaching.		
UNIT-5 TEACHING HOURS (11)	• Typ	bes of Questions, Characteristics of a good test.		
	• Blu	e print: Meaning, concept, need and constructio	n.	
0] -2		ntinuous and Comprehensive Evaluation: N		
		portance and limitations.	<i>d</i> , <i>i</i>	
UNIT-5 NG HO	-	dels of Teaching.		
	Assignmen	0		
	-		n achievement test	
EA		nstruction, administration and interpretation of a	in acmevement test	
		ny standard of school.		
		ke a diagnostic test of your subject and apply cussion with concerning teacher and give remediate		
TEACHING AND	1. Lec		lai measure.	
LEARNING	2. E-le			
STRATEGIES	2. L-R 3. Vid	0		
SIKAIEGIES		ension Lectures		
		ntent Review		
		f-Learning		
		bup Discussions		
	8. Fiel			
	9. Sur	•		
		cumentaries		
		11. Short Films		
	12. Team Teaching* The teaching strategies are subject to change as per requirement of the			
			equirement of the	
CONTINUOUS		nd their capabilities.		
CONTINUOUS &	Details of Continuous and Comprehensive Assessment (CCA) are as			
COMPREHENSIVE	follows:		1	
ASSESSMENT	SR. NO.	CCA: COMPONENT	MAXIMUM	
(CCA)			MARKS	
	1	Monthly Test	10X6 Test = 60	
	2	Presentation	10	
	3	Group Discussion	10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	
		ATION (METHOD TO ASCERTAIN MAR		
		be reduced to 30 marks or 15 marks (as per cou	rse weightage).	
		Marks obtained/Total marksX30		
		ple: $60 \div 160 \times 30 = 11.25$		
		O-I: Provided that a candidate shall be granted a		
	form of ex	components, however, not	more than 3 in a	
1	respective course.			
	PROVISO-III: Provided further that this will be mandatory for a candidate			
	PROVIS	O-II: Provided further that this will be mandaton	•	
	PROVIS		•	

	*Attendance in Lectures and Practical
	Percentage Marks Allotted
	75% to 80% 02
	81% to 85% 04
	86% to 90% 06
	91% to 95% 08
	Above 96% 10
EXAMINATION	Term-end examinations are organized by the university in the prescribed
PATTERN	format to enable the scholars to achieve success in contemporary
	competitions and to achieve their goals.
PERIODICAL	1. ANNUAL
REVISION OF	2. However, the Universitymay revise the syllabus at any time during the
SYLLABUS	running year after giving a notice for a period one month.
SELECTED READINGS	 Solomon, J., & Aikenhead, G. (1994). STS Education: International Perspectives on Reform. Ways of Knowing Science Series. NY: Teachers College Press. Lawson, A. E. (1995). Science teaching and the development of thinking. Belmont, CA: Wadsworth. Ellis, A. B. (1993). Teaching General Chemistry: A Materials Science Companion. American Chemical Society, Distribution Office Department 225, 1155 16th Street, NW, Washington, DC 20036. Das, R. C. (1990). Science teaching in schools. Sterling Publishers Pvt. Ltd. Hodson, D. (2009). Teaching and learning about science: Language, theories, methods, history, traditions and values. Brill Sense. National Research Council. (2007). Taking science to school: Learning and teaching science in grades K-8. National Academies Press. Staver, J. R. (2008). Teaching science (Vol. 17). APH Publishing. Psillos, D., & Niedderer, H. (Eds.). (2006). Teaching and learning in the science laboratory (Vol. 16). Springer Science & Business Media. Venkataih, S (2001.Science Education in 21st Century. Delhi:Anmol Publishers.
	• Yadav, M.S. (Ed.) (2000).Teaching Science at High Level.Delhi:Anmol PublishersEdger, Marlow & Rao, D.B. (2003). Teaching Scioence Successfully.New Delhi: Discovery Publishing House.

B.Sc.B.Ed. IV Year				
COURSE CODE:	BSCBED-453-III COURSE TY		(PE: CORE)	
COURSE TITLE:	Pedagogy of Chemistry			
MAX. MARKS:	50	MIN. P	ASS MARKS:	20
THEORY	35	MIN. P	ASS MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN. P	ASS MARKS:	6
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION MONTHLY TEST			
DURATION	03 HR 01 HR		IR	

Objectives:

- To understand ability to gain insight on the meaning and nature of chemistry.
- To develop ability to determining aims and strategies of teaching-learning.
- To develop ability to use effectively different activities/ demonstration/ laboratory experiences for teaching-learning of chemistry.
- To understand ability to integrate in chemistry knowledge with other school subjects.

Learning outcomes: After completion of the course, student-teachers will be able to:-

- Gain insight on the meaning and nature of chemistry for determining aims and strategies of teaching-learning.
- Appreciate that science is a dynamic and expanding body of knowledge.
- Appreciate the fact that every child possesses curiosity about his/her natural surroundings.
- Identify and relate everyday experiences with learning chemistry.
- Appreciate various approaches of teaching-learning of chemistry.
- Understand the process of science and role of laboratory in teaching-learning situations.
- Use effectively different activities/demonstration/laboratory experiences for teaching-learning of chemistry.
- Integrate in chemistry knowledge with other school subjects.

	Basic of Chemistry Teaching
	Meaning, Nature and Scope of Chemistry teaching.
	• Nature of Science with special reference to chemistry.
(12)	• History and Contribution of Chemistry teaching, history of
2	Chemistry with special reference to India.
UNIT-1 NG HOURS (• Importance Aims and Objectives of Chemistry teaching Objective of teaching Chemistry at secondary/senior secondary level.
	• Co-relation with other Subjects Journal and Referenced Book.
	Assignment:
C	Group discussion on importance of Chemistry.
U TEACHIN	• Organise seminar on Contribution of Scientist in Chemistry and preapre a report.
	I de la construcción de la const

Instructional objectives and methods			
	• Instructional objectives and methods Bloom's Taxonomy		
5	(Congnative, effective and psychomotor).		
2 3	• In terms of instructional behavior Methods of Chemistry		
B	teaching Subject lecture method. Demonstration Method, lab		
010	based method.		
UNIT-2 ING HO	Inductive & deductive method. Problem Solving. Heuristics &		
15 ž	Project Method Techniques of Chemistry Teaching Approaches of Chemistry teaching- Inquiry approach, programmed		
H	instruction, Group discussion, team teaching, CAL, SEMINARS		
UNIT-2 TEACHING HOURS (12)	& WORKSHOP.		
	Assignment:		
	Organise workshop on techniques of teaching Chemistry.		
	Organise a training program on Bloom's Digital Taxonomy.		
	Planning		
	• Concept, Meaning and Objectives of Chemistry Teaching Plan		
(12	(Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these Plans.		
UNIT-3 TEACHING HOURS (12)	• Meaning and Concept of Teaching Skills. Micro Teaching -		
300	Meaning, Need and Importance, Micro-Teaching Cycle and its		
UNIT-3 ING HO	features.		
	• Concept, Meaning, Principles and Objectives of Curriculum,		
	Characteristics of good curriculum and Evaluation · of Chemistry		
	Syllabus at Secondary Level.		
LE	Assignment:Prepare and present a lesson tHRough power point presentation		
	on any topic of your choice.		
	Organise a training program on Micro Teaching.		
	Instructional Support System		
	• Meaning, Objectives, Scope, Characteristics, Types, Preparation,		
	Presentation and Importance of Teaching Learning Material.		
	Dales' Cone of Experiences. Planning and Importance of		
	Chemistry Laboratories and Its uses. Qualities of good		
	Chemistry Text Books at Secondary Level.		
40	• Qualities and Characteristics Chemistry Teacher. Audio-Visual Aids - Meaning, Concept, Utility and Significance of Different		
H	types of Audio-Visual Aids in the Teaching of Chemistry.		
l S Ž	Utilization of Community Recourses in the Teaching of		
H H	Chemistry Teaching.		
UNIT-4 TEACHING HOURS (11)	Assignment:		
II	• Conduct a training program on use of Different types of Audio-		
	Visual Aids in the teaching of Chemistry.		
	Make a report on Chemistry Teaching Planning and Importance of Chemistry Laboratories and its uses.		
	or Chemisury Laboratories and its uses.		

ν, Ω Ξ	Evaluat	tion		
UNIT-5 TEACHING HOURS (11)		Meaning & Objective of Evolution.		
N H S		Types of Test Items and their Construction.		
		Preparation of Blue-Print and Achievement Test.		
HUCH		±	anotion of	
		Characteristics of a good Test. Concept and Prepa		
		Diagnostic Test, Remedial Teaching an	d Enrichment	
		Programme.		
		Use of ICT: Video clips, Power points presentation	ons, films etc.	
	Assignr		c c	
		Construction, administration and interpret	tation of an	
		achievement test of any standard of school.	1	
		Make a diagnostic test of your subject and app	-	
		after discussion with concerning teacher and	give remedial	
		measure.		
TEACHING AND		Lectures		
LEARNING		E-learning		
STRATEGIES		Videos Extension Lectures		
		Content Review		
		Self-Learning		
		Group Discussions Field Visit		
		Survey Documentaries		
		Short Films		
		Team Teaching		
		* The teaching strategies are subject to change as per requirement		
	of the students and their capabilities.			
CONTINUOUS &		of Continuous and Comprehensive Assessmer	nt (CCA) are as	
COMPREHENSIVE	follows			
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM	
(CCA)	NO.		MARKS	
	1	Monthly Test	10X6 Test = 60	
	2	Presentation	$\frac{10110}{10}$	
	3	Group Discussion	10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	
	1 * *		- v	

	EVELANATION (METHOD TO ASCEDIAN) MADIZ FOD CCA		
	EXPLANATION (METHOD TO ASCERTAIN MARKS FOR CCA)		
	CCA will be reduced to 30 marks or 15 marks (as per course weightage).		
	Formula: Marks obtained/Total marksX30		
	For example: $60 \div 160 \times 30 = 11.25$		
	PROVISO-I: Provided that a candidate shall be granted a relaxation in th		
	form of exemption from CCA components, however, not more than 3 in a		
	respective course.		
	PROVISO-II: Provided further that this will be mandatory for a candidat		
	appear in the monthly test conducted in the respective course.		
	*Attendance in Lectures and Practical		
	Percentage Marks Allotted		
	75% to 80% 02		
	81% to 85% 04		
	86% to 90% 06		
	91% to 95% 08		
	Above 96% 10		
EXAMINATION			
PATTERN	Term-end examinations are organized by the university in the prescribed		
PATIERN	format to enable the scholars to achieve success in contemporary		
DEDIODICAL	competitions and to achieve their goals.		
PERIODICAL	1. ANNUAL		
REVISION OF	2. However, the Universitymay revise the syllabus at any time		
SYLLABUS	during the running year after giving a notice for a period one		
	month.		
SELECTED	• Anderson, R. G. W. (1978). The Playfair Collection and the		
READINGS	teaching of chemistry at the University of Edinburgh, 1713-1858.		
	Brill.		
	Brill.Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of		
	• Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of		
	• Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry.		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). Nadendla, R. R. (2007). Principles of organic medicinal chemistry. 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). Nadendla, R. R. (2007). Principles of organic medicinal chemistry. New Age International. 		
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	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). Nadendla, R. R. (2007). Principles of organic medicinal chemistry. New Age International. Risch, B. (Ed.). (2010). Teaching chemistry around the world. Waxmann Verlag. 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). Nadendla, R. R. (2007). Principles of organic medicinal chemistry. New Age International. Risch, B. (Ed.). (2010). Teaching chemistry around the world. Waxmann Verlag. Smith, A., & Hall, E. H. (1902). The teaching of chemistry and 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). Nadendla, R. R. (2007). Principles of organic medicinal chemistry. New Age International. Risch, B. (Ed.). (2010). Teaching chemistry around the world. Waxmann Verlag. Smith, A., & Hall, E. H. (1902). The teaching of chemistry and physics in the secondary school. Longmans, Green, and Company. 		
	 Eilks, I., & Byers, B. (Eds.). (2015). Innovative methods of teaching and learning chemistry in higher education. Royal Society of Chemistry. Eilks, I., & Hofstein, A. (Eds.). (2015). Relevant chemistry education: From theory to practice. Springer. Ellison, M. D., & Schoolcraft, T. A. (2008). Advances in teaching physical chemistry. American Chemical Society. Herron, J. D. (1996). The Chemistry Classroom: Formulas for Successful Teaching. American Chemical Society, Product Services Office, 1155 16th Street NW, Washington, DC 20036 (cloth: ISBN-0-8412-3298-8; paperback: ISBN-0-8412-3299-7). Nadendla, R. R. (2007). Principles of organic medicinal chemistry. New Age International. Risch, B. (Ed.). (2010). Teaching chemistry around the world. Waxmann Verlag. Smith, A., & Hall, E. H. (1902). The teaching of chemistry and 		

B.Sc.B.Ed. IV Year				
COURSE CODE:	BSCBED-453 -IV COURSE TYP		PE: CORE	
COURSE TITLE:	Pedagogy of Biology			
MAX. MARKS:	50	MIN. P	ASS MARKS:	20
THEORY	35	MIN. P	ASS MARKS:	14
EXAMINATION				
CONTINUOUS	15	MIN. P	ASS MARKS:	6
COMPREHENSIVE				
ASSESSMENT (CCA)				
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR			
ELIGIBILITY				
EXAMINATION	TERM END EXAMINATION MONTHLY TEST		Y TEST	
DURATION	03 HR 01 HR		R	
Objectives:				

- To understand the ability to develop insight on the meaning and nature of biological science.
- To understand the ability to integrate the biological science knowledge with other school subjects.
- Develop the ability be to identify and relate everyday experiences with learning of biological science.
- To understand the ability to appreciate various approaches and methods of teachinglearning of biological science.
- Develop the ability to explore the process skill in science and role of laboratory in teaching-learning.
- Develop the ability to identify the concepts of biological science that are alternatively conceptualized by teachers and students in general.

Learning outcome: After completion of the course, student-teachers will be able -

- Develop insight on the meaning and nature of biological science for determining aims and strategies of teaching- learning.
- Integrate the biological science knowledge with other school subjects. Identify and relate everyday experiences with learning of biological science.
- Appreciate various approaches and methods of teaching- learning of biological science.
- Explore the process skill in science and role of laboratory in teaching- learning.
- To understand meaning, concept and various types of assessment.
- Identify the concepts of biological science that are alternatively conceptualized by teachers and students in general.

	Basics of Biology Teaching
.	• Meaning, Nature and Scope of Biology teaching.
(12)	Main discoveries and development in Biology
DURS	• Place and Values of Teaching Biology in School level.
-1	Correlation of Biology and other Subjects.
I-TINU L-TINU	Objectives of teaching Biology at School Level.
5 XII	Assignment:
CE	Group discussion on importance of Biology teaching.
TEA	• Organise seminar on Contribution of main discoveries and
F	development in Biology and preapre a report.

UNIT-2 TEACHING HOURS (12)	 Instructional objectives and methods Bloom's Taxonomy (Cognitive, effective and psychomotor), In terms of Instructional behavior. Methods of Biology teaching Subject- lecture method, Demonstration Method, Inductive & deductive method, Problem Solving, Heuristics & Project Method. Inquiry approach programmed Instruction, Group discussion Self Study team teaching, Seminar and workshops. Assignment: Organise workshop on techniques of Biology teaching. Organise a training program on Bloom's Digital Taxonomy.
UNIT-3 TEACHING HOURS (12)	 Planning Concept, Meaning and Objective of Biology Teaching Plan (Lesson Plan, Unit Plan, Yearly Plan) and Preparation of these Plans. Meaning and Concept of Teaching Skills. Micro Teaching - Meaning, Need and Importance, Micro-Teaching Cycle and its features. Concept, Meaning, Principles and Objectives of Curriculum, Characteristics of good curriculum and Evaluation of Biology Syllabus at Secondary Level. Assignment: Prepare and present a lesson tHRough power point presentation on any topic of your choice. Organise a training program on Micro Teaching.
UNIT-4 TEACHING HOURS (11)	 Instructional Support System Meaning, Objectives, Scope, Characteristics, Types, Preparation, Presentation and Importance of Teaching Learning Material. Dales' Cone of Experiences. Planning and Importance of Biology Laboratories and Its uses. Qualities of good Biology Text Books at Secondary Level. Qualities and Characteristics Biology Teacher. Audio-Visual Aids - Meaning, Concept, Utility and Significance of Different types of Audio-Visual Aids in the Teaching of Biology. Utilization of Community Recourses in the Teaching Biology Teaching. Use of ICT: Video clips, Power points presentations, films etc. Assignment: Conduct a training program on use of Different types of Audio-Visual Aids in the teaching of Biology. Make a report on Biology Teaching Planning and Importance of Chemistry Laboratories and its uses.

	Evaluat	tion		
11		Meaning and Objectives of Evaluation. Types of	Test Items and	
S. S.	t	their Construction.		
	•]	Preparation of Blue-Print and Achievement Test.		
<u> </u>	• (Characteristics of a good Test. Concept and	Preparation of	
ν H			d Enrichment	
Η̈́́́		Programme.		
UNIT-5 FEACHING HOURS (11)	Assignm	•		
	-	Construction, administration and interpret	ation of an	
JE J	6	achievement test of any standard of school.		
	• 1	Make a diagnostic test of your subject and app	oly it in school,	
		after discussion with concerning teacher and		
		measure.	C	
TEACHING AND	1. 1	Lectures		
LEARNING	2. 1	E-learning		
STRATEGIES	3. 1	Videos		
	4. 1	Extension Lectures		
	5. (Content Review		
		Self-Learning		
		Group Discussions		
	8. 1	Field Visit		
		Survey		
		Documentaries		
		Short Films		
		Team Teaching		
		* The teaching strategies are subject to change as per requirement		
	of the st	tudents and their capabilities.		
CONTINUOUS &		of Continuous and Comprehensive Assessmen	t (CCA) are as	
COMPREHENSIVE	follows			
ASSESSMENT	SR.	CCA: COMPONENT	MAXIMUM	
(CCA)	NO.		MARKS	
	1	Monthly Test	10X6 Test =	
			60	
	2	Presentation	10	
	3	Group Discussion	10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	

	EVELANATION (METHOD TO ACCEPTAIN MADIC FOD		
	EXPLANATION (METHOD TO ASCERTAIN MARKS FOR		
	CCA):		
	CCA will be reduced to 30 marks or 15 marks (as per course		
	weightage).		
	Formula: Marks obtained/Total marksX30		
	For example: 60÷160X30 =11.25		
	PROVISO-I: Provided that a candidate shall be granted a relaxation		
	in the form of exemption from CCA components, however, not more than 3 in a respective course.		
	PROVISO-II: Provided further that this will be mandatory for a		
	candidate to appear in the monthly test conducted in the respective		
	course.		
	*Attendance in Lectures and Practical		
	Percentage Marks Allotted		
	75% to 80% 02		
	81% to 85% 04		
	86% to 90% 06		
	91% to 95% 08		
	Above 96% 10		
EXAMINATION	Term-end examinations are organized by the university in the prescribed		
PATTERN	format to enable the scholars to achieve success in contemporary		
	competitions and to achieve their goals.		
PERIODICAL	1. ANNUAL		
REVISION OF	2. HOWEVER, THE UNIVERSITYmay revise the syllabus at any		
SYLLABUS	time during the running year after giving a notice for a period		
	one month.		
SELECTED READINGS	 Agarwal, D.D. (2004). Modern methods of Teaching Biology. Saruk &Sons: New Delhi. 		
	• Miller, David F. & blaydes, Gllenn W. (1938). Methods and		
	materials for teaching biological sciences. Mc GRAW Hill book		
	company Inc: New York and London.		
	• Choudhary, S. (2010). Teaching of Biology. APH Publishing		
	Corporation: New Delhi.		
	• Sood, J.K. (1987). Teaching of Life Science. Kohli publishers:		
	Chandigarh. • Vaday M.S. (2000) Modersn Methods of Teaching Science, Annual		
	• Yadav, M.S. (2000). Modersn Methods of Teaching Science. Anmol Publishers: New Delhi.		
	• Bhar, Suraj prakash (2006). Teacher Training Lotus Press: New Delhi.		
	• Singh, Veena (2007). Teaching of Biology. Adhyanyan Publishers & Distributors: New Delhi.		
	 मंगल, एस.के. (2010). जैविक विज्ञान शिक्षण. लायल बुक डिपोः मेरठ. 		
	• भूषण, शैलेन्द्र (2008). जीव विज्ञान शिक्षण. विनोद पुस्तक मन्दिरः आगरा.		
	 कुलश्रेष्ठ, एस.पी. (2005). जीव विज्ञान शिक्षण. लायल बुक डिपोः मेरठ. 		
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	• रावत एवं अग्रवाल (2014). नवीन विज्ञान शिक्षण. श्री विनोद पुस्तक मन्दिरः		
	आगरा.		
	 श्रीमाली, भूषण एवं रिहानी (2013). विज्ञान शिक्षण. राजस्थान हिन्दी ग्रन्थ अकादमीः जयपुर 		

B.Sc.B.Ed.IV Year			
COURSE CODE:	BSCBED-453 -V COURSE TYPE: CORE		
COURSE TITLE:	Pedagogy of Physics		
MAX. MARKS:	50	MIN. PASS MARKS:	20
THEORY	35	MIN. PASS MARKS:	14
EXAMINATION			
CONTINUOUS	15	MIN. PASS MARKS:	6
COMPREHENSIVE			
ASSESSMENT (CCA)			
ATTENDANCE	80 PERCENT IN RESPECT	IVE YEAR	
ELIGIBILITY			
EXAMINATION	TERM END EXAMINATION MONTHLY TEST		
DURATION	03 HR 01 HR		

Objectives:

- To understand the need for teaching-learning of Physics in secondary classes.
- To develop a critical understanding about the aims and objectives of Physics in a Democratic and Secular country.
- To understand the nature of Physics curriculum and its pedagogical issues.
- To understand Critique and develop suitable evaluation mechanisms in Physics.
- To develop the ability to organize co-curricular activities and community resources for promoting Physics learning.
- To understand the Approaches to teaching of Physics
- To understand the Concept of Teaching Skills

Learning outcome: After completion of the course, student-teachers will be able -

- Gain insight on the meaning and nature of Physics science for determining aims and strategies of teaching-learning.
- Identify and relate everyday experiences with learning Physics;
- Appreciate various approaches of teaching-learning of Physics;
- Use effectively different activities/demonstrations/laboratory experiences for teaching-learning of Physics;
- Integrate in Physics knowledge with other school subjects;

	Nature and Scope of Physics		
	Meaning, Concept, Nature, Scope Physics Teaching.		
.1 (12)	• Contribution of Indian scientist – Sir C.V.Raman, J.C.Bose,		
	S.N.Bose, H.J.Bhabha, M.N.Saha.		
	• Correlation of Physics with other School Subjects.		
	Assignment:		
UNIT-1 TEACHIN HOURS (1	Organise workshop on Contribution of Indian scientist.		
	• Write a report on Objectives of Physics Teaching at different level of		
	School.		
0 V 0	Teaching-learning of Physics		
UNIT-2 HOURS (12)	• Aims and general objectives of teaching physics, Bloom's Digital		
	Taxonomy (Cognitive, Effective and Psychomotor) in terms of		
	Instructional Behaviour, the objectives of school education; writing		
	specific objectives of various content areas in Physics.		
	• Approaches to teaching of Physics – Analytic-Synthetic, Inductive-		
G	Deductive, Heuristic, Problem Solving, Project and Laboratory.		
TEACHING	Deductive, ficultatic, ficoreni solving, ficject and Educitatory.		
E	• Using various techniques of teaching Physics viz-oral, written drill,		
	assignment, Team teaching, supervised study and		

	Programmed Learning.
	Assignment:
	• Prepare a lesson plan based on team teaching and execute it in
	school.
	• Make any two teaching aids with the low cost material.
	Planning
ି କ	• Concept, Meaning and Objectives of Physics teaching Plan (Lesson
	Plan, Unit Plan, Yearly Plan) and Preparation of these plans
RS S	Meaning and Concept of Teaching Skills, Maxims of Teaching
	• Micro Teaching-Meaning, Need and Importance of Micro Teaching
H	Cycle and its Features
UNIT-3 ING HO	Concept, Meaning, Principles and Objectives of Curriculum,
UNIT-3 TEACHING HOURS (12)	Characteristics of good Curriculum.
	• Dale cone of experience.
EA	Assignment:
	Organsie training program on Micro Teaching.
	Prepare a working model based on Dale cone of experience.
	Teaching-learning Resources in Physics
UNIT-4 TEACHING HOURS (11)	• Meaning, Objectives, Scope, Characteristics, Types, Preparation,
s	presentation and Importance of Teaching - Learning Material.
	• Planning and Importance of Physics Laboratories and its uses.
1 <u>4</u> 0	• Qualities of Physics Text Books at Secondary Level.
UNIT-4 ING HO	• Audio-Visual Aids- Meaning, Concept, Utility and Significance of Different types of Audio Visual Aids in the teaching of Physica
15ž	Different types of Audio-Visual Aids in the teaching of Physics. Assignment:
H	• Conduct a training program on use of Different types of Audio-Visual
	Aids in the teaching of Physics
II	• Make a report on Physics Teaching Planning and Importance of
	Physics Laboratories and its uses.
	Assessment and Evaluation
1	• Meaning, concept and construction of Achievement test, diagnostic
	test and remedial teaching.
R	• Types of Questions, Characteristics of a good test.
0 e	• Blue print: Meaning, concept, need and construction.
UNIT-5 ING HO	• Continuous and Comprehensive Evaluation: Meaning, concept,
	importance and limitations.Models of Teaching.
UNIT-5 TEACHING HOURS	Assignment:
AC	• Construction, administration and interpretation of an achievement test of any standard of school.
	 Make a diagnostic test of your subject and apply it in school, after
	discussion with concerning teacher and give remedial measure.
TEACHING AND	1. Lectures
LEARNING	2. E-learning
STRATEGIES	3. Videos
	4. Extension Lectures
	5. Content Review
	6. Self-Learning
	7. Group Discussions
	8. Field Visit
	9. Survey

	10. Documentaries 11. Short Films			
	12. Team Teaching			
	* The teaching strategies are subject to change as per requirement of the students and their canabilities			
	the students and their capabilities.			
CONTINUOUS & COMPREHENSIVE	Details of Continuous and Comprehensive Assessment (CCA) are as follows:			
ASSESSMENT				
(CCA)	SR. NO.	CCA: COMPONENT	MAXIMUM MARKS	
	1	Monthly Test	10X6 Test = 60	
	2	Presentation	10/10/10/10/10/10	
	3	Group Discussion	10	
	4	Debate	10	
	5	Participation and Presentation in Seminar	10	
	6	Report Writing	10	
	7	Viva Voce	10	
	8	Attendance*	10	
	9	Co-curricular Activity	10	
	10	Team Teaching	10	
	form o respect PROV appear * Atten	ISO-I: Provided that a candidate shall be grante f exemption from CCA components, however, r ive course. ISO-II: Provided further that this will be mandate in the monthly test conducted in the respective co dance in Lectures and Practical Percentage Marks Allotted 75% to 80% 02 81% to 85% 04 86% to 90% 06 91% to 95% 08 Above 96% 10	not more than 3 in a bry for a candidate to urse.	
EXAMINATION PATTERN	format	nd examinations are organized by the universi- to enable the scholars to achieve succes- tions and to achieve their goals.	•	
PERIODICAL REVISION OF SYLLABUS		NUAL vever,the Unviersity may revise the syllabus running year after giving a notice for a period	•	
SELECTED READINGS	 Joshi, New I Das, I limite Prasac publis 	al, S.K (2005). Teaching of Physics. Arya book S.R (2008). Teaching of science. A.P.H Pub Delhi. R.C, (2000). Science teaching in schools. Sterlin d: New Delhi. d, J. (1999). Practical aspects in teaching o her: New Delhi. a, V.K. (1997). Science education today. Ann	blishing Corporation: ng Publishers private f science. Kanishka	

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• भटनागर, ए.बी. (२०००). भौतिक विज्ञान शिक्षण. सूर्या पब्लिकेशन्सः मेरठ.
•नेगी, जे.एस. (२००८). भौतिक विज्ञान शिक्षण. विनोद पुस्तक मन्दिरः आगरा.
• शर्मा, आर.सी. (२००७). आधुनिक विज्ञान शिक्षण. धनपतराय पब्लिशिंग कम्पनी (प्रा.) लि.
• सिंह, विजयपाल (2005–06). भौतिक विज्ञान शिक्षण. राधा प्रकाशन मन्दिरः आगरा–02.
•त्यागी, एस.के.(२०००). भौतिक विज्ञान शिक्षण. साहित्य प्रकाशनः आगरा
• कुलश्रेष्ठ, एस.पी. शैक्षिक तकनीकी एवं उसके मूल आधार. विनोद पुस्तक मन्दिर. आगरा-०२.

B.Sc. B.Ed. Part-IV							
COURSE CODE:	BSCBED-454 a I			COURSE TYPE : CORE			
COURSE TITLE : Atomic, Molecular and Nuclear Physics							
MAX.MARKS:	50	MIN.PAS	SS MARK	KS 20			
THEORY	40	MIN.PASS MARKS		KS 16			
EXAMINATION:							
CONTINUOUS	10	MIN.PAS	SS MARK	KS 4			
COMPREHENSIVE							
ASSESSMENT (CCA)							
ATTENDANCE	80 PERCENT IN RESPECTIVE YEAR						
ELIGIBILITY							
EXAMINATION	TERM END EXAMIN	IATION MONTHLY TEST					
DURATION	3 HR		1 HR				
Objective:							
• This course aims to introduce various types of spectra for hydrogen, deuteron and alkali atoms. It also gives an introduction to X-ray spectra. Techniques of Molecular spectroscopy are also discussed in this course, which include IR and Raman spectra. In addition, the students enable to understand of the sub atomic particles and their properties. It will emphasize to gain knowledge about the different nuclear techniques and their applications in different branches of Physics and societal application. The acquire knowledge can be applied in the areas of nuclear, medical, archeology, geology and other interdisciplinary fields of Physics and Chemistry.							
Learning Outcomes: After completion of the course, student-teachers will be able to:-							
Understand atomic and molecular physics							

- Understand atomic and molecular physics.
- Apply Theory of Nuclear Fission and fusion in further study.
- Explain Structure of Nuclei and Elementary Particles.
- Discuss the Concept of Nuclear Properties and Theory of Nuclear Forces.
- Applies the Concept of Detector and Accelerator.

2	Atomic Physics:Spectra of hydrogen, Frank-Hertz experiment and		
S(1)	discrete energy states, Stern-Gerlach experiment, deuteron and alkali		
	atoms, spectral terms, doublet fine structure, screening constants for		
5 HOU	alkali spectra for s, p, d and f states, selection rules, L-S and J-J		
	couplings, Spectroscopic notation of atomic states, Atoms in a		
15 ž	magnetic field, Zeeman effect, Zeeman splitting.		
UNIT-1 TEACHING HOURS(15)	Weak spectra: continuous X-ray spectrum and its dependence on		
	voltage, Duane and Hunt's law, Characteristics X-rays, Moseley's law,		
E	doublet structure of X-ray spectra, X-ray absorption spectra.		
UNIT-2 TEACHIGN HOURS(10)	Molecular Physics: Discrete set of electronic energies of molecules,		
	quantisation of vibrational and rotational energies, determination of		
	internuclear distance, pure rotational and rotational-vibrational		
	spectra, Dissociation limit for the ground and other electronic states,		
	transition rules for pure vibrational and electronic vibrational		
	spectra. Raman effect, Stokes and anti-Stokes lines, complimentary		
	character of Raman and infrared spectra, experimental arrangements for		
	Raman spectroscopy.		

UNIT-3 TEACHING HOURS(15)	 Structure of Nuclei: Rutherford theory of alpha paticle scattering, properties of nucleus quadrupole moment and nuclear ellipticity. Quadrapole moment and nuclear spin parity, angular momentum and magnetic moment, nuclear mass and mass spectroscopy. Nuclear potential, Mass defect and Binding energy, Meson theory of Nuclear forces and properties of nuclear forces. Radioactive decay:Beta decay, range of alpha particles, Geiger-Nuttall law, Gamow's explanation of alpha decay, gamma decay, continuous and discrete spectra. Elementary Particles:Classification of Elementary Particles, Fundamental interactions, Unified Approach (basic ideas), The Conservation laws, Quarks (basic ideas), Charmed and Colour Quarks, Higgs Boson, Large Hadron Collider.
UNIT-4 TEACHING HOURS(10)	 Nuclear Model: Liquid drop model, Semi-emperical mass formula, condition of stability, Magic Numbers, Basic assumption of shell model-success and limitations, Fermi gas model. Nuclear Fission and Fusion: Theory of Nuclear Fission, Barrier Penetration-Theory of Spontaneous Fission, Nuclear Fission as a source of Energy, The Nuclear Chain reaction, Condition of controlled Chain Reaction, The Principle of Nuclear Reactors, Classification of Reactors, Power of Nuclear Reactors, Uncontrolled chain reaction, Nuclear fusion, Energy released in Nuclear Fusion, Fusion in stars. The Plasma: The fourth state of the matter.
UNIT-5 TEACHING HOURS(15)	 Accelerators: Ion sources, Cockcroft-Walton high voltage generators, Van de Graaff generators, Drift tube, Linear accelerators, Wave guide accelerators, Magnetic focusing in Cyclotron, SyncHRocyclotron, Betatron, The electromagnetic induction Accelerator, Electron syncHRotron, Proton SyncHRotron. Detectors: Interaction of charged particles and neutrons with matter, working of nuclear detectors, Geiger-Muller counter, proportional counter and scintillation counter, cloud chambers, spark chamber, emulsions.
TEACHING AND LEARNING STRATEGIES	 Lecture method Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. * The TEACHING AND LEARNING STRATEGIES may be change as per requirement of the students and their capabilities.

CONTINUOUS & COMPREHENSIVE	Details of Continuous and Comprehensive Assessment (CCA) are as follows:			
ASSESSMENT (CCA)	S. No.	CCA- Components	Max. Marks Allocation	
	1.	Monthly test	20*3 Test=60	
	2.	Quizzes and Assignments	10	
	3.	Viva-voce	10	
	4.	Seminar/Symposia	10	
	5.	Report writing	10	
	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
	Total 160 ma	arks equivalent reduced to CCA original	ginal marks 30.	
Exam Pattern	Term-end e	examinations are organized by	the university in the	
	-	format to enable the scholars		
		ry competitions and to achieve th	eir goals. semester.r.	
PERIODICAL REVISE OF SYLLABUS	 ANNUAL However, the Unviersity may revise the syllabus at any time during the running year after giving a notice for a period one months. 			
SELECTED READINGS				

PHYSICS PRACTICALS-IV				
Duration: 4 HR MAX.MARKS: 50	Min. Marks: 20			
Any twelve of the following experiments are	to be performed. Few more experiments may be set			
at the institutional level. In examination two	experiments are to be performed taking at least one			
from each section.				
The distribution of marks in the practical example.	nination will be as follows:			
(i) Two experiments	30 Marks			
For each experiment, distribution of marks wi	ll be as follows:			
Figure:	2			
Formula/Theory:	2			
Observation: 7				
Calculation and Result: 3				
Precautions:	1			
(ii) Viva voce	10			
(iii) Records	10			
Total 50 Marks				
LIST OF EXPERIMENTS				

- Determination e/m by Thomson method.
- Determination e/m by Magnet Method.
- Measurement of Hall coefficient of given material.
- Verification of inverse square law by photo cell.
- Determination of Planck's costant by photo cell.
- Determination of Planck's constant using solar cell.
- To study V-I characteristics of Light Emitting Diode (LED).
- To find the magnetic susceptibility of paramagnetic solution using Quinck's method.
- Study of Magnetic Hysteresis parameters using a CRO.
- Study of characteristics of a GM counter and verification of inverse square law for the same strength of radioactive source.
- Study of resistance characteristics of semi-conductor Material using four probe method.
- Polarization of light by reflection, verify Brewster's law & Law of Malus.
- To determine the value of a High resistance by Leakage method.
- To determine small thickness by using thin film interference.
- The study of frequency response and phase relationship in a series LCR circuit.
- Measurement of magnetic susceptibility.
- Study of statistical distribution: Gaussian and Poisson's spectral distribution using dices.

B.Sc. B.EdIV Year					
COURSE CODE:	BSCBED- 454 b I COURSE TYPE : CORE				
COURSE TITLE :	Chemistry : Advance Chemistry				
MAX.MARKS:				20	
THEORY	40	MIN.PASS N	MARKS	16	
EXAMINATION:					
CONTINUOUS	10	MIN.PASS N	MARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPI	ECTIVE SESS	SION		
ELIGIBILITY					
EXAMINATION	TERM END EXAMI	NATION	M	ONTHLY TEST	
DURATION	03 HR			1 HR	
OBJECTIVE:					
	istry the spectral and magne			*	
	chemistry of Heterocyclic c	-	-	-	
-	of Nuclear magnetic res	sonance (NMI	R)) spect	roscopy of organic	
compounds.		_			
_	fter completion of the course				
1	etral and magnetic propertie		meteal cor	nplexes.	
	eterocycles and Bioinorgani	•			
_	magnetic resonance (NMR)				
Apply the principle	s of Acid-base behaviorisoe	*			
5 3	Electronic spectra of Transition Metal Complexes				
	Types of electronictransi	tions, selection	on rules	for d-d transitions,	
UNIT-1 EACHIN DURS (J	spectroscopic ground states	s, spectro cher	mical serie	es, Orgel-energy level	
UNIT-1 TEACHING HOURS (12)	diagram for d^1 to d^9 states, discussion of the electronic spectrum of				
II	$[Ti(H_2O)_6]^{3+}$ complex ion.	$\operatorname{Ti}(\operatorname{H}_2\operatorname{O}_6]^{3+}$ complex ion.			
7 0	Magnetic Properties of Tr	ansition Meta	l Comple	xes	
IT-2 (HIGN RS (12)	Types of magneticbehav		_		
IT-2 JHIGI RS (12	susceptibility, spin-only for			0 0	
	values, orbital contribution				
L HO	moment data for 3d metal c		iomento, a	pplication of magnetic	
- , ,		ompiexes			
כי	Heterocyclic Chemistry	1., 1	1	. 1	
φĨS	Introduction: Molecular of	-			
UNIT-3 EACHIN HOURS (12)	pyrrole, furan, thiophene an			•	
	O reactions with particular emphasis on the mechanism of electrophilic substitution. Mechanism of nucleophilic substitution reactions in pyridine				
-	derivatives. Comparison of basicity of pyridine, piperidine and pyrrole.				
പ	Bioinorganic Chemistry				
JNIT-4 ACHIN HOURS (12)	with special reference to hemoglobin and myoglobin. Biological role of				
	alkali and alkaline earth metals ions with special reference to Ca^{2+} .				
H H	Nitrogen fixation.				
-	ivinogen jixunon.				

	N	(NIMD)			
RS		netic resonance (NMR) spectroscopy	1		
UNIT-5 TEACHING HOURS (15)	Proton magneticresonance (^I H-NMR) : spectroscopy, nuclear shielding				
		and deshielding, chemical shift and molecular structure, area of signals			
		and proton counting, splitting of signals, spin-spin coupling and coupling			
UNIT-5 HING F (15)		rpretation of NMR spectra of simple orga			
5 H	•	ide, ethanol, acetaldehyde, 1, 1, 2-tribror	noethane, ethyl		
A C		ne and acetophenone.			
E.	Problems pertaining to the structure elucidation of simple organic				
	compounds using UV, IR and NMR spectroscopic techniques.				
TEACHING AND	1. Lectur	re method			
LEARNING	2. Proble	em Solving method			
STRATEGIES	3. Graph	nical method			
	4. Semin	nar/Symposia			
	5. Revie	w of literature			
	6. Repor	t writing			
		Discussion			
	8. Video	os/Animation			
	9. Self-L	earning/e-Learning			
	10. Works	shops/Experiments.			
	* The TEAC	HING AND LEARNING STRATEGI	ES may be change		
	as per requir	ement of the students and their capab	ilities.		
CONTINUOUS	Details of Co	ontinuous and Comprehensive Assess	ment (CCA) are as		
COMPREHENSIVE	follows:	-			
ASSESSMENT	S. No.	CCA- Components	Max. Marks		
(CCA)		*	Allocation		
	1.Monthly test20*3 Test=60				
	2. Quizzes and Assignments 10				
	3. Viva-voce 10				
	4. Seminar/Symposia 10				
	5.Report writing10				
	6.	Workshop	10		
	7.	Review of literature	10		
	8.	Creativity/Innovation	10		
	9.	Experimental Skill	10		
	10.	Co-curricular activity	10		
	10.	Attendance	10		
		rks equivalent reduced to CCA original r	-		
Exam Pattern		xaminations are organized by the			
		format to enable the scholars to a			
	-	y competitions and to achieve their go			
	contemporar	y competitions and to demeve then go	ais. semester		
PERIODICAL	1. ANNUAI				
REVISE OF	2. However, the Unviersity may revise the syllabus at any time during				
SYLLABUS	the running year after giving a notice for a period one month.				
SELECTED	• Abraham, R. J., Fisher, J., & Loftus, P. (1998). <i>Introduction to NMR</i>				
READINGS	spectroscopy (Vol. 2). New York: Wiley.				
	• Anslyn, E. V., & Dougherty, D. A. (2006). <i>Modern physical organic</i>				
		<i>chemistry</i> . University science books.			
	• Chambers, R. D. (2004). <i>Fluorine in organic chemistry</i> . CRC press.				
	• Colthup, N. (2012). <i>Introduction to infrared and Raman spectroscopy</i> . Elsevier.				
	spectros	copy. Elsevier.			

•	Gordon, P. F., & Gregory, P. (2012). <i>Organic chemistry in colour</i> . Springer Science & Business Media.
•	Kalsi, P. S. (2007). Organic reactions stereochemistry and
	mechanism (THRough Solved Problems). New Age International.
•	Kumar, N., & Kumbhat, S. (2016). <i>Essentials in nanoscience and nanotechnology</i> . Weilly & Sons.
•	Levenson, M. (2012). <i>Introduction to Nonlinear Laser Spectroscopy</i> 2e. Elsevier.
•	Niemantsverdriet, J. W. (2007). Spectroscopy in catalysis: an introduction. John Wiley & Sons.
•	Ozin, G. A., & Arsenault, A. (2015). <i>Nanochemistry: a chemical approach to nanomaterials</i> . Royal Society of Chemistry.
•	Pavia, D. L., Lampman, G. M., Kriz, G. S., & Vyvyan, J. A. (2008). Introduction to spectroscopy. Cengage Learning.
•	Pradeep, T. (2007). NANO: The Essentials-Understanding Nanoscience and Nanotechnology. McGraw Hill.
•	Schwarzenbach, R. P., Gschwend, P. M., & Imboden, D. M. (2016). Environmental organic chemistry. John Wiley & Sons.
•	Sergeev, G. (2003). Nanochemistry. MGU Publishing.
	Sharma, Y. R. (2007). <i>Elementary organic spectroscopy</i> . S. Chand Publishing.
•	Sharma, Y. R. (2009). Elementary organic spectroscopy, principles and chemical application. <i>Chand and Company Ltd, New Delhi,</i>
	India, 23.
•	Smith, M. B. (2020). March's advanced organic chemistry: reactions, mechanisms, and structure. John Wiley & Sons.

CHEMISTRY PRACTICALS-3				
Duration:4 HR MAX.MARKS: 50 Min. Marks: 20				
Note: The students should be given exposure of any research center/reputed university lab/industry/government labs of northern reg				
Inorganic Chemistry				
Ex. 1Estimation of given substance by volumetrically ,iodometrically water. 15	and total hardness of			
Organic Chemistry				
Ex.2 Prepartion and determination of an Organic compounds				
15				
Physical Chemistry				
Ex.4 Perform one of the experiments mentioned in the syllabus.				
10				
Ex.5 Vive-Voce				
05				
Ex.6 Practical-Record	05			
Total	50 Marks			
LIST OF EXPERIMENTS				
A.Inorganic Chemistry				
• Quantitative estimation of one metal volumetrically from a give				
• To estimate magnesium volumetrically from a mixture conta	aining Ba ²⁺ and Mg ²⁺ Ions/			
Zn^{2+} and Mg^{2+} ions.				
• To estimate copper iodometrically from a given mixture conta	ining Pb^{2+} and Cu^{2+} ions.			

- Estimation of Glucose with the help of Fehling's solution.
- Determination of Total hardness of water.

B.Organic Chemistry

- Two stage preparation: p-nitroacetanilide from Aniline and p-Bromoacetanilide from Aniline.
- Determination of Iodine value of an oil/fat.
- Separation of two component mixture using water or NaHCO₃ solution & identification of the two components. Preparation of one derivative

C. Physical Chemistry

Colorimetry

- Determination of formula of complex by Job's method.
- Verification of Beer-Lambert law for KMnO₄/K₂Cr₂O₇and determine the concentration of the given solution of the substance.

Polarimetry

• Determination of the specific rotation of a given optically active compound and determination of the concentration of given solution of an optically active substance

Solvent Extraction

• Separation and estimation of Mg (II) and Zn (II)

Ion Exchange Method

• Separation and estimation of Mg (II) and Zn (II)

B.Sc.B.Ed. Year IV					
COURSE CODE:	BSCBED- 454 c I COURSE TYPE : CORE				
COURSE TITLE:	Molecular Genetics, Biotechnology and Biostatistics				
MAX.MARKS:	50	MIN.PASS MARKS 20			
THEORY	40			16	
EXAMINATION:					
CONTINUOUS	10	MIN	.PASS MARKS	4	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPECT	IVE Y	'EAR		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINATI	ON	MONTHL	Y TEST	
DURATION	3 HR		1 H	R	
genetics.To learn about get	comrehend the modern concept familiar with the molecular structer completion of the course, structer	ture o	f DNA and its cell	ular activities.	
• To Comrehend the	modern concepts and applied asparent molecular structure of DNA as	pects of	of molecular gener		
	 Introduction, concept and deve Genetic engineering. Nucletic acids: DNA (prokary) 	Introduction, concept and development of Molecular Genetics & Genetic engineering. Nucletic acids: DNA (prokaryotic and eukaryotic)- Structure, forms, chemical composition functions and units of DNA, Genetic			
UNIT-2 TEACHIGN HOURS (15)	 RNA: Genetic RNA, non-genetic RNAs (mRNA, tRNA, and rRNA) – Structure and functions. Gene mutation: nature of mutation, types of mutation and causes of mutation. DNA repair: mismatch repair, direct repair, base-excision, nucleotide-excision repair and other types of DNA repair. Genetic diseases and faulty DNA repair. 				
UNIT-3 TEACHING HOURS (15)	Gene expression: Transcription and translation of prokaryotes and eukaryotes. Regulation of gene expression in prokaryotes (Lac and tryptophan operon) Gene cloning,DNA Amplification by PCR, Tools and techniques of gene transfer, gene mapping, Human Genome Project, Clones,				
UNIT-4 TEACHING HOURS (15)	transgenic Animal, gene library. Introduction historical perspective, animal cell hybridoma, major areas and future prospects of biotechnology, Medicine and Biotechnology, Microbes in medicine. Antibiotics, vaccine, Antibodies, Antigens, Environmental biotechnology: use of micro- organisms in metal and petroleum recovery, pest control, waste treatment, processing of industrial waste.Degradation of xenobiotic compounds including pesticides and surfactants and oil pollutants. Food and drink biotechnology, ferment food, dairy product, food preservation, microbial spoilage, alcoholic beverages, Vinegar, monoclonal antibodies and their applications.				

STRATEGIES	 Introduction and understanding of concepts of descriptive and inferential statistics, frequency distribution, graphical presentation, mean, mode, median, standard deviation, standard error of mean. Productivity distribution, correlation and regression. Test of Significance, chi Square and t-test. Biostatistical analysis of gene distribution in population. Lecture method Problem Solving method Graphical method Seminar/Symposia Review of literature Report writing Group Discussion Videos/Animation Self-Learning/e-Learning Workshops/Experiments. 				
CONTINUOUS		ent of the students and their capa Continuous and Comprehensive			
&COMPREHENSIVE	as follows		Assessment (UCA) are		
ASSESSMENT (CCA)	S. No.	CCA- Components	Max. Marks Allocation		
	1.	Monthly test	20*3 Test=60		
	2.Quizzes and Assignments10				
	3.Viva-voce10				
	4.Seminar/Symposia10				
	5.Report writing10				
	6. Workshop 10				
	7. Review of literature 10				
	8. Creativity/Innovation 10				
	9.	Experimental Skill	10		
	10.	Co-curricular activity	10		
	11.	Attendance	10		
		marks equivalent reduced to CCA	-		
EXAMINATION PATTERN	prescribe contempo	l examinations are organized by d format to enable the scholar rary competitions and to achieve	s to achieve success in		
PERIODICAL DEVISE OF	1. ANNU		o gullohug of over the		
REVISE OF SYLLABUS	2. However, the Universitymay revise the syllabus at any time				
SILLADUS	during the running year after giving a notice for a period one				
SELECTED	Months.				
READINGS	 Alberts, B., Johnson, A., Lewis, J., Raff, M., Roberts, K., & Walter, P. (2015). <i>Molecular biology of the cell</i>. Garland Science. <i>New York</i>, 1227-1242. 				
	• Blomquist, G. J., & Bagnères, A. G. (Eds.). (2010). <i>Insect hydrocarbons: biology, biochemistry, and chemical ecology</i> . Cambridge University Press.				
	 Conklin, K. F., Doerfler, W., Grafstron, R. H., Groudine, M., Hamilton, D. L., Jaenisch, R.& Langner, K. D. (2012). DNA methylation: biochemistry and biological significance. 				

	Springer Science & Business Media.
•	Gupta, P. K. (2005). <i>Cell and molecular biology</i> . Rastogi Publications.
•	Karp, G. (2007). <i>Cell and Molecular Biology</i> . John Wiley & Sons Incorporated.
•	Karp, G. (2009). <i>Cell and molecular biology: concepts and experiments</i> . John Wiley & Sons.
•	Kotpal, R. L. (1967). Annelida. Jaipur: Rastogi Publications.
•	Lodish, H., Berk, A., Kaiser, C. A., Krieger, M., Scott, M. P.,
	Bretscher, A.& Matsudaira, P. (2008). Molecular cell biology.
	Macmillan.
•	Rosen, F. S., Steiner, L., & Unanue, E. (1989). Macmillan
	dictionary of immunology.
•	Tripathi, G. (2010). <i>Cellular and Biochemical Science</i> . IK International Pvt Ltd.
•	Wilson, E. B. K., & Walker, J. (2005). <i>Biochemistry and Molecular Biology. Kuudes painos.</i>
•	Animal biotechnology, M.M. Ranga, Agrobios India.
•	Industrial Biotechnology, G.Read, presscott and Dunns,
	Chapman and Hall.
•	Advances Biotechnology, Manjula k.saxena and B.B.S.
	Kapoor, madhupublications
•	Biostatistics, Veer bala Rastogi
•	Introduction to Biostatistics, Dr. Pranab Kumar Bangerjee. S.
	Chand.

ZOOLOGY-PRACTICALS

Duration: 4 HR.

Max. Marks - 50

Objectives: To develop the molecular and biotechnological techniques and to develop skills of preparing media, separation of nucleic acids and culture of animal cells.

Course Contents-

- Study of DNA by Fulgent reaction in the salivary gland cHRomosomes.
- Isolation of genomic DNA
- Molecular separations by cHRomatography, electrophoresis, precipitation etc.
- Isolation of milk protein from the milk sample.
- Separation of serum by using centrifuge
- Estimation of protein by Biuret Method.
- Separation of plasma by centrifugation.
- Separation of biomolecules by Course and gel cHRomatography.
- Preparation and use of culture media for microbes.
- Preparation and use of culture media for animal tissues.
- Media preparation media sterilization and inoculation.
- Cell culture techniques-Design and functioning of tissue culture laboratory, cell proliferation measurements, culture media preparation and cell harvesting methods.
- Isolation and staining of bacteria.
- Determination of pH value of different water samples, blood urine and saliva.
- Qualitative tests for carbohydrates.
- Qualitative test for proteins.
- Qualitative test for lipids.
- Effects of temperature on the activity of enzyme.
- Biostastistics Construction of frequency tables, Histogram, polygons, pie charts. Exercise on mean, median and mode. Test of significance :- t-test and chi square test.
- Chart, model Power point/multimedia presentation preparation related to evidence of evolution Human /Horse evolution, Geographical time scale etc.
- Students are expected to visit different laboratories (RRL, CSIR, ICMR, Science centers etc).

MAX.MARKS: 50 Time Allowed: 4 Marks. S. No.	Exercise*	Marks
1.	Molecular Biology Experiment	4
2.	Biotechnological Experiment	4
3.	Biochemical tests	4
4.	Biostastistics	4
5.	Bacteriological experiment	5
6.	Instrumentation-major	7
7.	Instrumentation - minor	3
8.	Practical record	5
9.	Viva	4
10.	Project report	10

Guidelines/Instructions for Practical Examination

B.Sc.B.Ed. Year IV				
COURSE CODE:	BSCBED- 454 d I COURSE TYPE : CORE			
COURSE TITLE:	Genetic Engineering, Molecular Biology, Ecology & Economic Botany			
MAX.MARKS:	50	MIN.PAS	SS MARKS	20
THEORY	40	MIN.PAS	SS MARKS	16
EXAMINATION:				
CONTINUOUS	10	MIN.PAS	SS MARKS	4
COMPREHENSIVE				
ASSESSMENT (CCA)	80 PERCENT IN RESPE		A D	
ATTENDANCE ELIGIBILITY				
EXAMINATION	TERM END EXAMIN	ATION	MON	NTHLY TEST
DURATION	3 HR			1 HR
Objective :				
	t understand plant tissue cult			eomics
	t get familiar with concept o			
	t explain the concept of gene	0	•	life
	t implement the application of fter completion of the cour		. .	
0	l plant tissue culture, genomi			III DE ADIE.
	ar with concept of cellular to		teonnes	
Ũ	e concept of genetic engineer	- ·		
-	t the application of biotechno	U	1 lifa	
				re (Flectrophoresis
2)	Tools and Techniques in Plant Tissue Culture (Electrophoresis, Centrifugation, CHRomatography, Colorimetery, Spectrophotometery, ELISA Leuringer air Gran showhar metadawa Ust air			
2 3				
	<i>ELISA, Laminar air flow chamber, autoclave, Hot air oven,</i> <i>Incubator</i>)Media Preparations , Solid media, Liquid media, sterilization			
- 1	_		-	
L-TINU NG HO	techniques, sterilization of glasswares and medium ,Aseptic			
	manipulation and Culture maintenance, Inoculation and Sub			
	culture.Concept of Cellular to	otipotency,	Methods of	application of micro
AC	propagation,			
TEAC	Haploid production, Zygot	ic embryo	o culture, H	Endosperm Culture,
	Somatic embryogenesis and	Synthetic S	eeds.	
	An Overview of Genetic Er	ngineering,	Tools & Te	chniques of genetic
UNIT-2 TEACHI GN HOURS (12)	engineering, recombinant DN	NA technol	ogy , Metho	ds and applications
I G G G E	in agriculture, horticulture			
DE H	Concept of genomics and pro	-		
	Atmosphere (gaseous co			
		-		
IS NO DO	factors, Global warming, Green house effect, Acid rains, Alnino effe ozone depletion, Population ecology with special reference to Grow			
DE E Curves acotypes acads and plant Indicators Ecological succession				
	Curves, ecotypes, ecads and plant Indicators.Ecological succession. Ecosystem,Structure and function, Abiotic & biotic components,			
				_
	food chain, food Web,	U	al pyrami	, .
$\begin{array}{c} \Sigma \\ \Sigma \\ \Sigma \\ \end{array} \end{array} \qquad \qquad$				sphorous.

UNIT-5 TEACHING HOURS (18)	Economic Botany: - Origin, Cultivation and value added products of following: Cereals: Rice, Wheat and Maize, Oil Yielding Plants: Mustard, Groundnut and Coconut, Fibre Yielding Plants: Cotton,						
UNIT-5 TACHIN JURS (1		Spices : Cardamom, Fennel,					
OCEAN		—					
ĒĒ	Medicinal plants : Opium, Cinchona, Sarpagandha. Beverages : Tea & Coffee.						
		1. Lecture method					
TEACHING AND LEARNING		em Solving method					
STRATEGIES		0					
SINAILGILS	 Graphical method Seminar/Symposia 						
		w of literature					
	6. Repor						
	-	Discussion					
	8. Video	s/Animation					
		earning/e-Learning					
		shops/Experiments.					
		ning and Learning Strategies ma	• • •				
COMPANY		of the students and their capabilit					
CONTINUOUS		ontinuous and Comprehensive Ass	essment (CCA) are as				
COMPREHENSIVE ASSESSMENT	follows: S. No.	CCA Components	Max. Marks				
(CCA)	5. INO.	CCA- Components	Allocation				
	1.Monthly test20*3 Test=60						
		1.Nonany test20 9 Pest-002.Quizzes and Assignments10					
		3. Viva-voce 10					
	4. Seminar/Symposia 10						
	5. Report writing 10						
	6.	Workshop	10				
	7.	Review of literature	10				
	8.	Creativity/Innovation	10				
	9.	Experimental Skill	10				
	10.	Co-curricular activity	10				
	11.	Attendance	10				
		rks equivalent reduced to CCA origin					
EXAMINATION		minations are organized by the univ					
PATTERN		able the scholars to achieve suc and to achieve their goals.	cess in contemporary				
PERIODICAL	1. ANNU						
REVISE OF			syllabus at any time				
SYLLABUS	2. However the Unviersitymay revise the syllabus at any time during the running year after giving a notice for a period one						
	months.						
SELECTED	 Vasil. I.K. and Thorpe, T.A. 1994, Plant Cell and Tissue Culture, 						
READINGS		Kluwer					
	Acade	mic Publishers, The Netherlands.					
		r, S.L. 1998. Economic Botany in	Tropics 2nd edition.				
		illan India Ltd. New Delhi.					
	• Simpson. B.B. and Conner-Ogorzaly, M. 1986. Economic						
	Botan	y – Plants in Our World, Mc. Graw I	Hill, New Delhi.				

	BOTANY-PRACTICALS:-III
	Duration: 4 hoursMAX.MARKS: 50Min. Pass Marks : 20
	Total = 50
	Practical List
1.	Basic requirements of a tissue culture laboratory.
	a) Common Glassware,
	b) test tubes, culture tubes and screw-capped tubes,
	c) Petridish
	d) Pipette
	a) Pipette
	b) Pasteur pipette
	c) Erlenmeyer
	d) flask
	e) Volumetric flask
	f) Cleaning glassware
	 g) Inoculation needle and inoculation loop b) Durger human (Spirit lawn)
	h) Bunsen burner (Spirit-lamp),i) water baths
	j) Autoclaves.k) laminar air flow
	1) Incubator
	m) Hot air oven
	n) Colony counter
	o) pH meter
	p) Electric balance
	q) Spectro photometer
	r) Centrifuse
	s) binocular Microscope.
1.	Method of using balance Preparation of temporary cotton plug Preparation of permanent
	cotton plugs. Preparation of culture media
	a) Preparation of liquid medium (broth)
	b) Preparation of Solid media (PDA medium and plates)
	c) Preparation of agar slants.
	d) Preparation of agar deep tubes.
2.	Methods of Sterilization.
3.	Demonstration of the techniques of micro-propagation by using different explants, e.g.
	axillary buds, shoot meristems etc.
4.	To determine the minimum size of quadrate by species area curve method.
5.	To determine the minimum number of quadrate to be laid down in field under study.
6.	To study the vegetation structure tHRough profile diagram
7.	To determine moisture content and water holding capacity of different types of soil
8.	To determine the dust holding capacity of different types to leaves.
9.	Fibres: Study of cotton flowers, sectioning of the cotton ovules/developing seeds to trace the
	origin and development of cotton fibres. Microscopic study of cotton and test for cellulose.
	Sectioning and staining of jute stem to show the location and development of fibres.
10.	Microscopic structure. Tests for lignocellulose.
11.	Spices: Examine Coriander, Fennel and Cumin (hand sections) and opened fruits of
	cardamom and describe them briefly.
12.	Preparation of an illustrated inventory of 5 medicinal plants used in indigenous systems
	of medicine or allopathy: Write their botanical and common names, parts and
	diseases/disorders for which they are prescribed

B.Sc. B.Ed. IV Year					
COURSE CODE:			RSE TYPE : CORE		
COURSE TITLE :					
MAX.MARKS:	100	MIN.PASS	-	40	
THEORY	80	MIN.PASS	MARKS	32	
EXAMINATION:					
CONTINUOUS	20	MIN.PASS	MARKS	8	
COMPREHENSIVE					
ASSESSMENT (CCA)					
ATTENDANCE	80 PERCENT IN RESPEC	CTIVE YEAI	R		
ELIGIBILITY					
EXAMINATION	TERM END EXAMINAT	TION	MO	MONTHLY TEST	
DURATION	3 HR			1 HR	
Objective:	•				
 This course is designed to introduce the basic concepts of Numerical Mathematics in order to solve the problems arising in various fields of application, for example in science, engineering and economics etc. that do not possess analytical solutions or difficult to deal with analytically. This course addresses development, analysis and application of different numerical methods to solve the problems, viz. system of linear & nonlinear equations, interpolation and numerical integration. Learning Outcomes: This course will enable the students to: Obtain numerical solutions of algebraic and transcendental equations. Find numerical solutions of system of linear equations and check the accuracy of the solutions. Learn about various interpolating and extrapolating methods. Solve initial and boundary value problems in differential equations using numerical methods. Apply various numerical methods in real life problems. To solve linear programming problems (LPP) by simplex method. To solve assignment and transportation problems. Understand and apply the programming concepts of C++ which is important for mathematical investigation and problem solving. 					
UNIT-1 UNIT-1 EACHING HOURS (12) M F T T T T T	Aumerical Methods for Aquations Acound-off error and compu- rrors, Algorithms and conv- nethod, Fixed point iteratio ecant method for solving eq Aumerical Methods for Sol ower and upper triangular pplications, Gauss elimin Gauss-Jacobi, Gauss-Seidel 1	ter arithmetic vergence; Bis n method, No uations. ving Linear : (LU) decorn nation meth	e, Local an ection me ewton-Raj Systems: nposition	nd global truncation thod, False position phson's method and of a matrix and its	

STOP Interpolation Lagrange and Newton interpolations, Piccewise linear interpolation, Cubic spline interpolation, Finite difference operators, Gregory-Newton forward and backward difference interpolation. Newton's formula for forward and backward difference interpolation formula, Lagrange's interpolation formula, error in interpolation for first derivative, Approximation for second derivative; Numerical integration: Trapezoidal rule, Simpson's rules and error analysis, Newton's cotes quadrature formula, gauss quadrature formula. Initial and Boundary Value Problems of Differential Equations: Euler's method, Runge-Kutta methods, Higher order one step method, Multi-step methods; Finite difference method, Shooting method, Real life examples. FUND S21 (1) Singer programming problems, basic solution, basic feasible solution and optimal solution; graphical method and simplex method of solutions; duality. Transportation and assignment problems. Strong S10 (1) Computer programming: • Binary system; arithmetic and logical operations on numbers; octal and hexakecinal system; conversion to and from decimal systems; algebra of binary numbers. * Elements of computer systems and concept of memory; basic logic gates and truth tables, Boolean algebra, normal forms. Representation of unsigned integers. Algorithms and flow charts for solving numerical analysis problems. TEACHING AND LEARNING 1. Lecture method StrateGles Self-Learning: • Group Discussion • Self-Learning StrateGles S. No. CCAC-Components Max. Marks Allocation • Self-Learning Self-Learning * Composite stand		Tratering alother	-		
Store Cubic spline interpolation, Finite difference operators, Gregory-Newton forward and backward difference interpolations. Newton's formula for forward and backward interpolation divided differences and simple differences. Newton's general interpolation formula, Lagrange's interpolation for approximation for second derivative; Numerical integration. Store Numerical Differentiation and Integration First order and higher order approximation for first derivative; Approximation for second derivative; Numerical integration: Trapezoidal rule, Simpson's rules and error analysis, Newton's cotes quadrature formula, gauss quadrature formula. Initial and Boundary Value Problems of Differential Equations: Euler's method, Runge-Kutta methods, Higher order one step method, Multi-step methods; Finite difference method, Shooting method, Real life examples. FUDD ON Linear programming problems, basic solution, basic feasible solution and optimal solution; graphical method and simplex method of solutions; duality. Transportation and assignment problems. Store Computer programming: • Binary system; arithmetic and logical operations on numbers; octal and hexadecimal systems; adjebra of binary numbers. • Elements of computer systems and concept of memory; basic logic gates and truth tables, Boolean algebra, normal forms. • Elements of computer systems and concept of memory; basic logic gates and truth tables, Boolean algebra, normal forms. • Elements of computer systems and concept of memory; basic logic gates and truth tables. Boolean algebra, normal forms.				a linear interpolation	
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STOPH SUPPER First order and higher order approximation for first derivative, Approximation for second derivative; Numerical integration: Trapezoidal rule, Simpson's rules and error analysis, Newton's cotes quadrature formula, gauss quadrature formula. Trapezoidal rule, Simpson's rules and error analysis, Newton's cotes quadrature formula, gauss quadrature formula. Initial and Boundary Value Problems of Differential Equations: Euler's method, Runge-Kutta methods, Higher order one step method, Multi-step methods; Finite difference method, Shooting method, Real life examples. FLOD Step of the examples. Linear programming problems, basic solution, basic feasible solution and optimal solution; graphical method and simplex method of solutions; duality. Transportation and assignment problems. STOP Differential Equations: Binary system; arithmetic and logical operations on numbers; octal and hexadecimal systems; conversion to and from decimal systems; algebra of binary numbers. Elements of computer systems and concept of memory; basic logic gates and truth tables, Boolean algebra, normal forms. Representation of unsigned integers, signed integers and reals, double precision reals and long integers. Algorithms and flow charts for solving numerical analysis problems. TEACHING AND LEARNING 1. Lecture method LEARNING 1. Lecture method STRATEGIES 1. Lecture method Seminar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion		interpolation	n formula, error in interpolation.		
Iife examples. FHOYAL Since programming problems, basic solution, basic feasible solution and optimal solution; graphical method and simplex method of solutions; duality. Transportation and assignment problems. Since your of the programming: • Binary system; arithmetic and logical operations on numbers; octal and hexadecimal systems; conversion to and from decimal systems; algebra of binary numbers. • Elements of computer systems and concept of memory; basic logic gates and truth tables, Boolean algebra, normal forms. Representation of unsigned integers, signed integers and reals, double precision reals and long integers. Algorithms and flow charts for solving numerical analysis problems. TEACHING AND LEARNING STRATEGIES 1. Lecture method StrateGies 1. Lecture method Beinar/Symposia 5. Review of literature 6. Report writing 7. Group Discussion 8. Videos/Animation 9. Self-Learning/e-Learning 10. Workshops/Experiments. * The Teaching and Learning Strategies may be change as per requirement of the students and their capabilities. CONTINUOUS COMPREHENSIVE ASSESSMENT (CCA) S. No. CCA-Components Max. Marks Allocation 3. Viva-voce 10 4. Seminar/Symposia 10		Numerical Differentiation and Integration			
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		5.	Report writing	10	

	6.	Workshop	10	
	7.	Review of literature	10	
	8.	Creativity/Innovation	10	
	9.	Experimental Skill	10	
	10.	Co-curricular activity	10	
	11.	Attendance	10	
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SYLLABUS	U	ne running year after giving a no	tice for a period one	
	month.			
SELECTED		S. (2012). Introductory Methods C	It Numerical Analysis.	
READINGS		ing Pvt. Ltd.		
		A., & Rabinowitz, P. (2001). A Firs	t Course In Numerical	
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		W. (1997). Numerical Analysis.	Springer Science &	
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	-	G. M., & Taylor, P. J. (Eds.).		
		ons Of Numerical Analysis. Elsevier		
		d, F. B. (1987). Introduction To	Numerical Analysis.	
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	• Gerald, C. F. & Wheatley, P.O. (2008). Applied Numerical Analysis			
	(7th edition), Pearson Education, India.			
	• Hildebrand, F. B. (2013). Introduction to Numerical Analysis: (2nd			
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	Methods for Scientific and Engineering Computation (6th edition).			
	New Age International Publishers.			
	• Robert J. Schilling & Sandra L. Harris (1999). Applied Numerical			
	Methods for Engineers Using MATLAB and C. Thomson-			
	Brooks/Cole.			
		& Chip Weems (2013).Programmin		
	with C++	(6 th edition). Jones & Bartlett Learnin	ng.	

Internship (16 weeks) TEACHING PRACTICE AND PRACTICAL WORK

Planning &	Preparation, administration & analysis of diagnostic test (s)	10
Execution	followed by remedial teaching any selected lesson.	
Execution	Execution of action research project	10
	Observation and preparation of report	10
	Working with community project of social welfare. (submission	10
	of report)	
	Exhibition of TLM in school prepared by student teacher	10
	Regular classroom teaching delivery of 70 lessons	5+5=10
Assessment &	Two Criticism Lesson in teaching subject	10+10=20
Evaluation	5 Lessons to be observe by teacher educator.	5+5=10
	Final Lesson (External evaluation)	50
Regularity &	Student teachers function in liaison with the regular teachers in	05
involvement in	the school in all day-to-day functioning along with teaching-	
different school	learning by mentor teachers	
activities	Participating in various 'out of classroom activities' in school	5
	Organizing events	
	Participation in any two co-curricular activity and preparation of	10
	report	
	Study (and preparation) of school calendar, time table, assessment	5
	schedule, library and laboratory.	
	Portfolio, including detailing of teaching-learning plans, resources	10
	used, assessment tools, student observations and records.	
	observations of work done by the students during the internship	05
	programme.(Seeking reactions of students,headmasters/	
	principals/ cooperating teachers and supervisors)	
	Preparation and maintenances of feedback diary	10
	A journal by student teacher in which he/she records one's	10
	experiences, observations, and reflections.	
	TOTAL MARKS	200